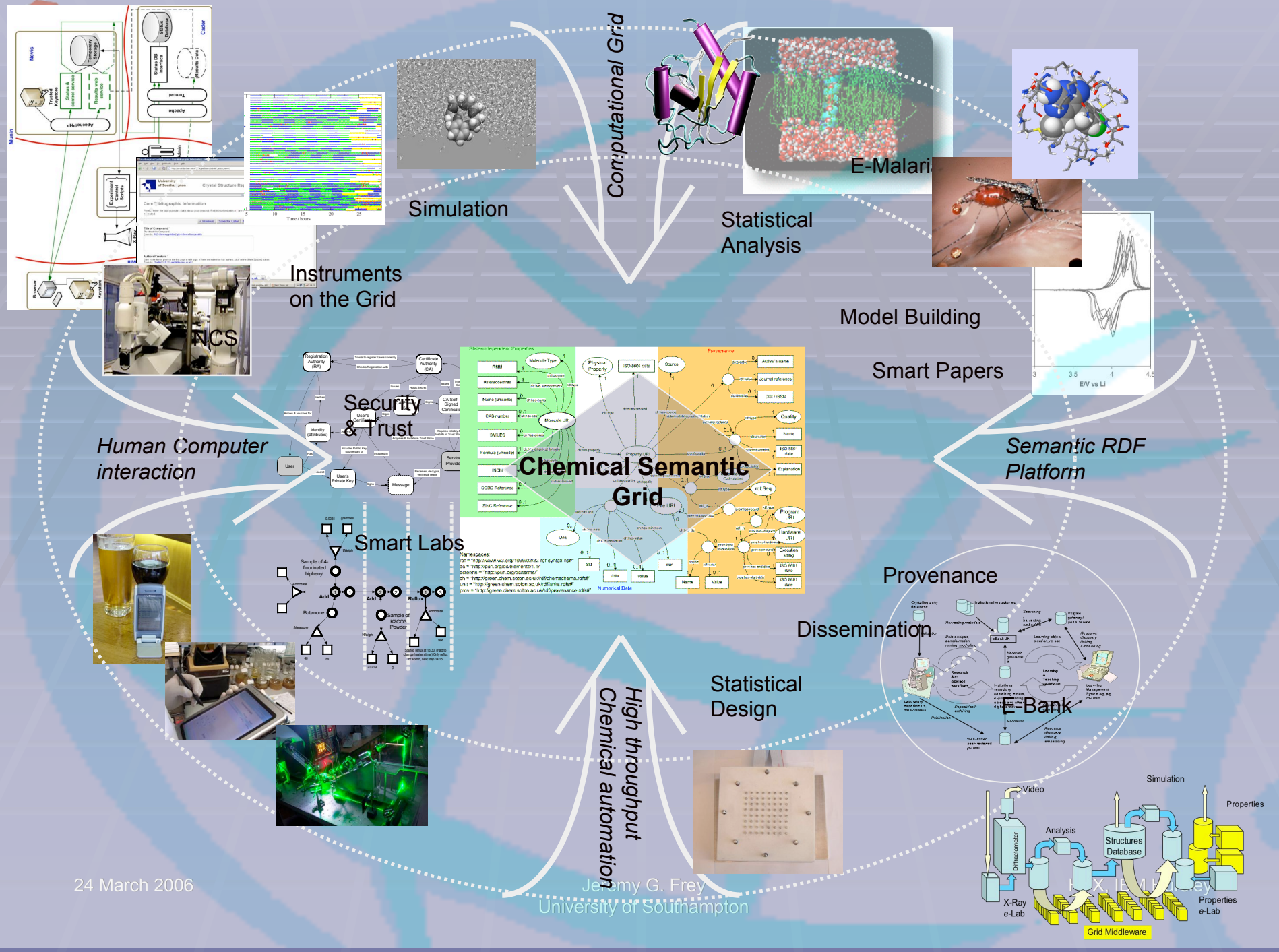


Future Lab - "Smart Not Dark"

Jeremy G. Frey
School of Chemistry, University of Southampton, UK







Molecular Beam van der Waals IR Diode Laser Spectroscopy

UV ns pump/probe mol beam photochemistry

Interfacial SHG ns studies

Confocal Raman in situ probe

E-Science - useful computer science!

1980

2000



OPO IR/VUV cluster spectra

Ab initio QM

Interfacial SHG ps / fs studies

Simulations

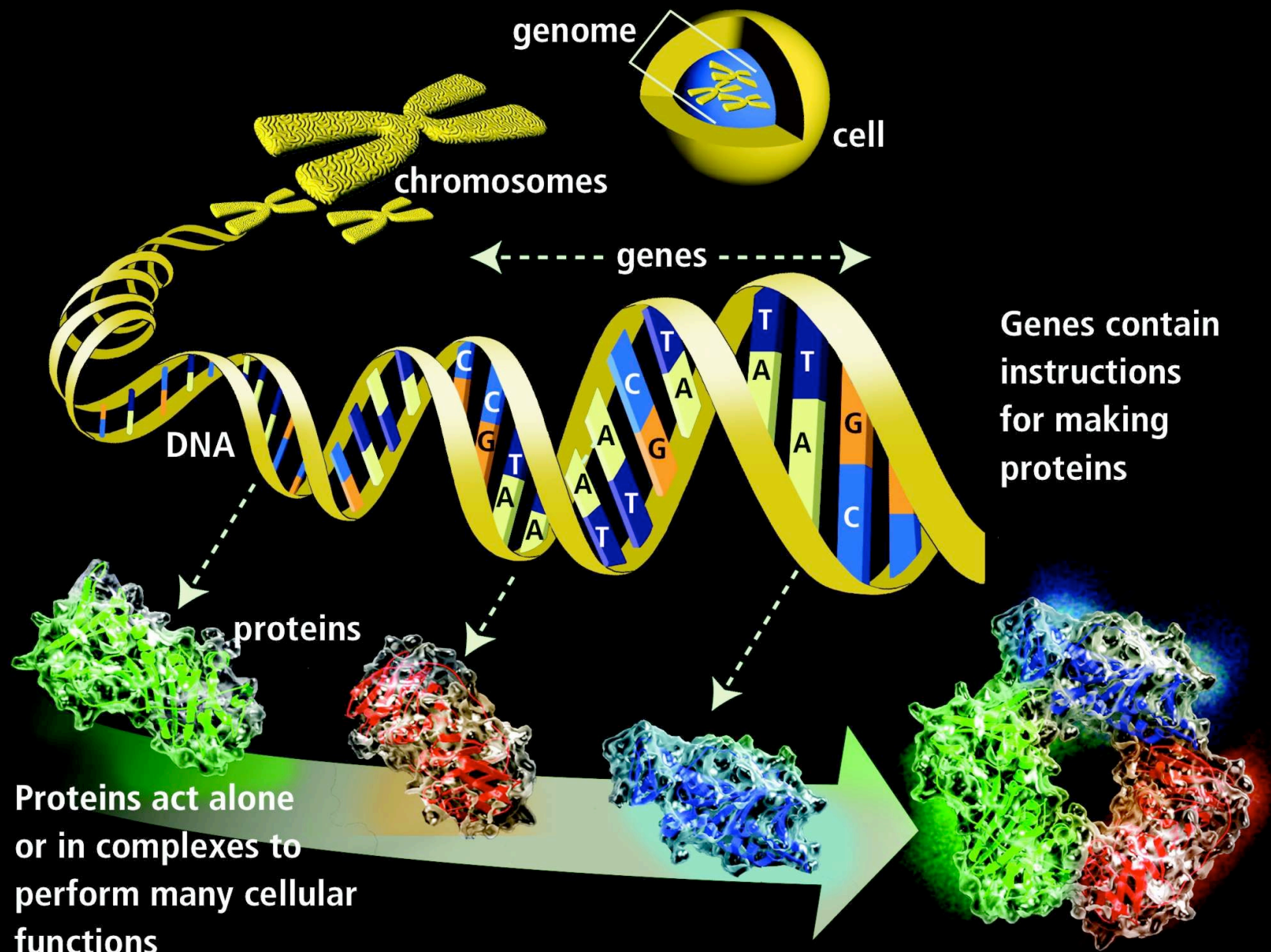
Single molecule IR spectra OPO/STM

Nano-scale x-ray studies

24 March 2006

Jeremy G. Frey
University of Southampton

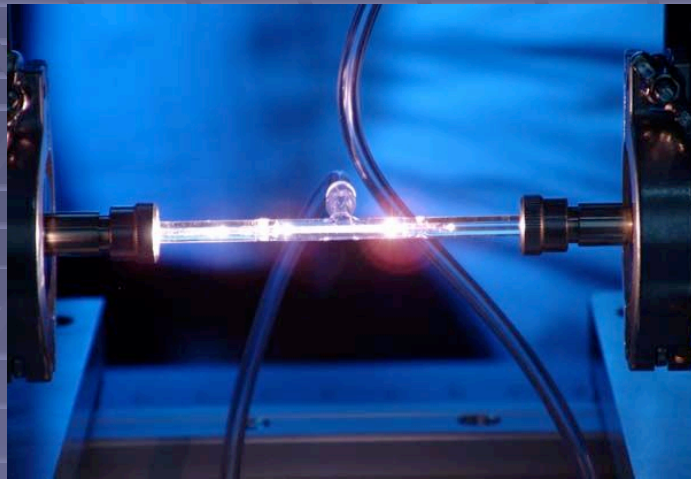
HTX, IBM Hursley



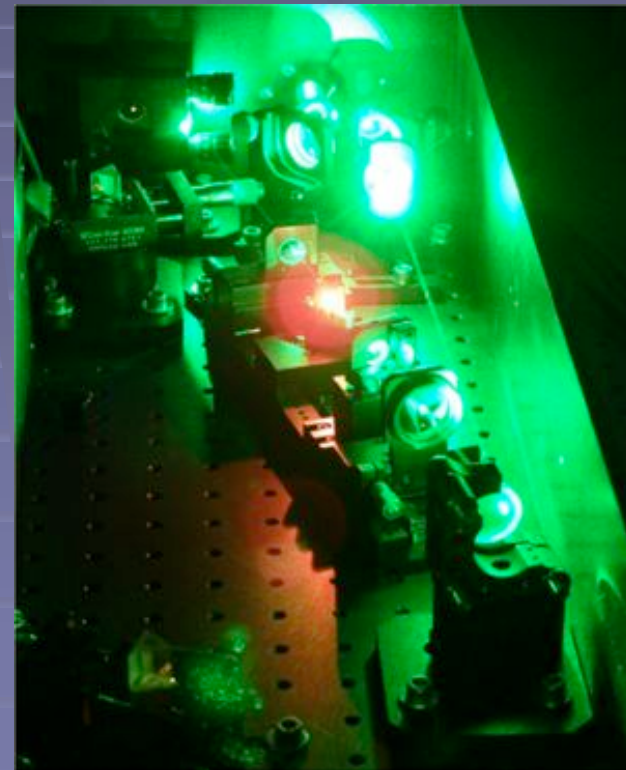
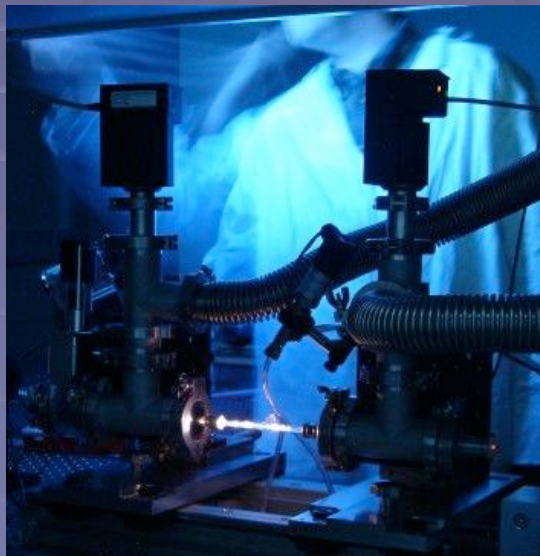
U.S. DEPARTMENT OF ENERGY



Laser X-ray generation



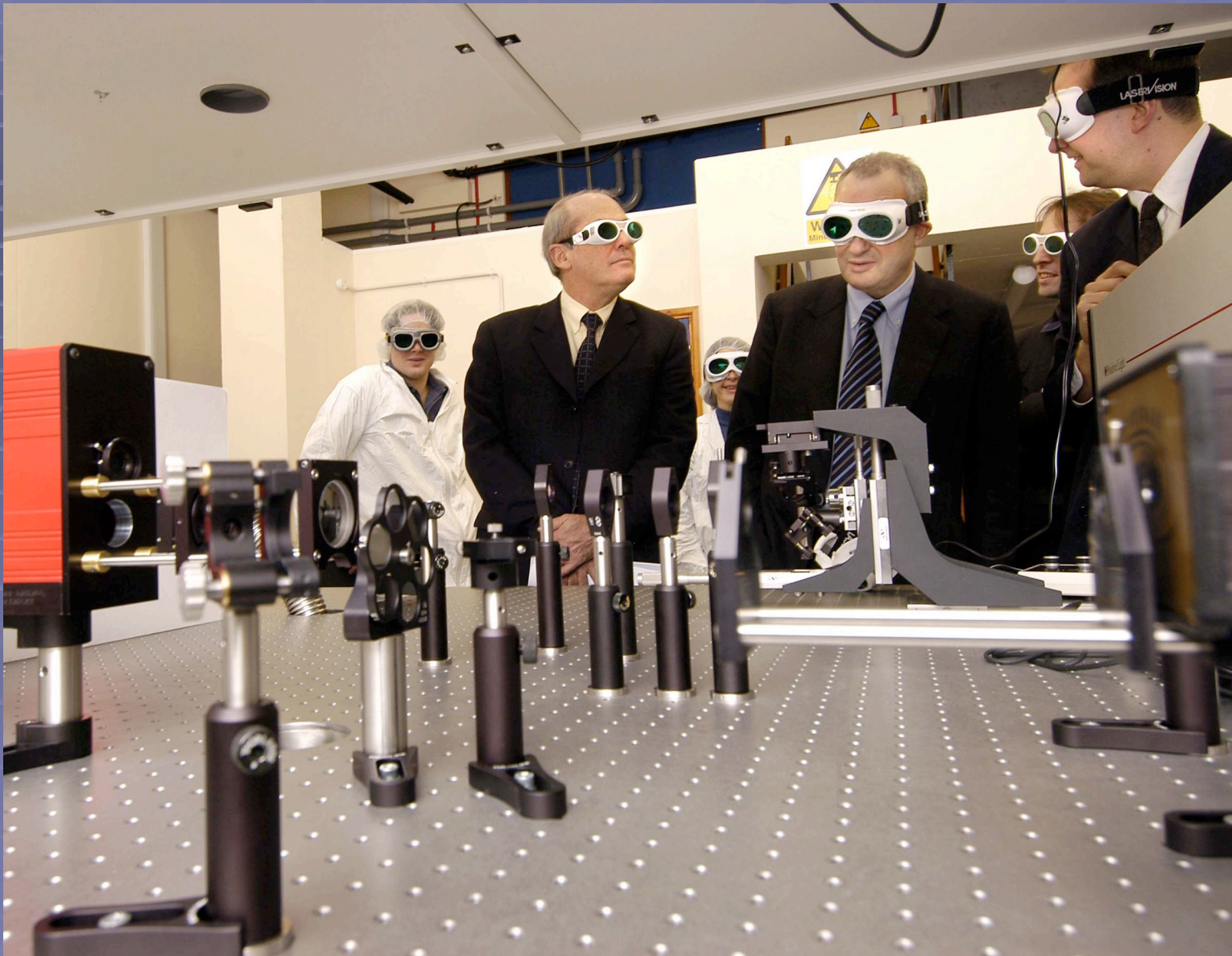
- Need **very high** laser powers:
 - Peak intensity $\sim 10^{15}$ W/cm²
 - Peak E-field ~ 100 GV/m



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University of Southampton

HTX, IBM Hursley



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University of Southampton

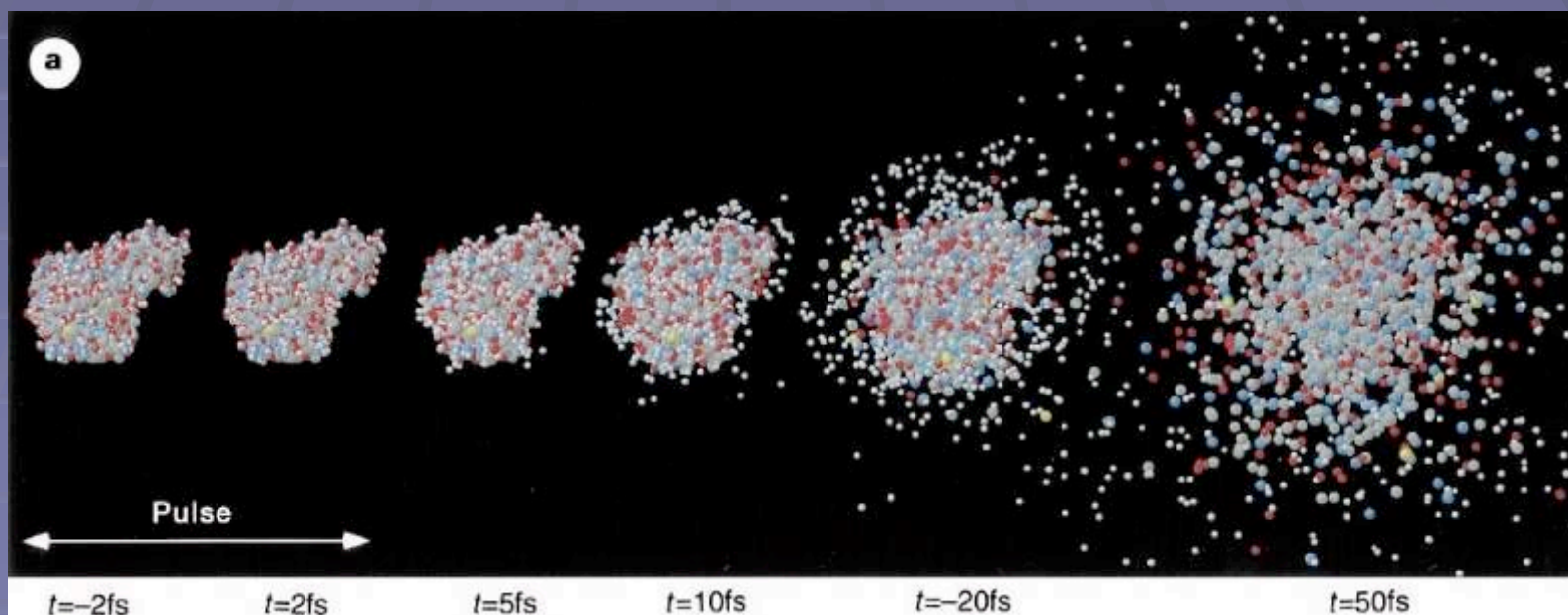
HTX, IBM Hursley



Damage limitation

- Conventional damage threshold: 200 ph/Å² Damage process
 - inelastic scattering of electrons out of molecule
 - Coulomb repulsion of remainder
- Timescale? fs

Neutze *et al.*, Nature (2000) **406**, 752



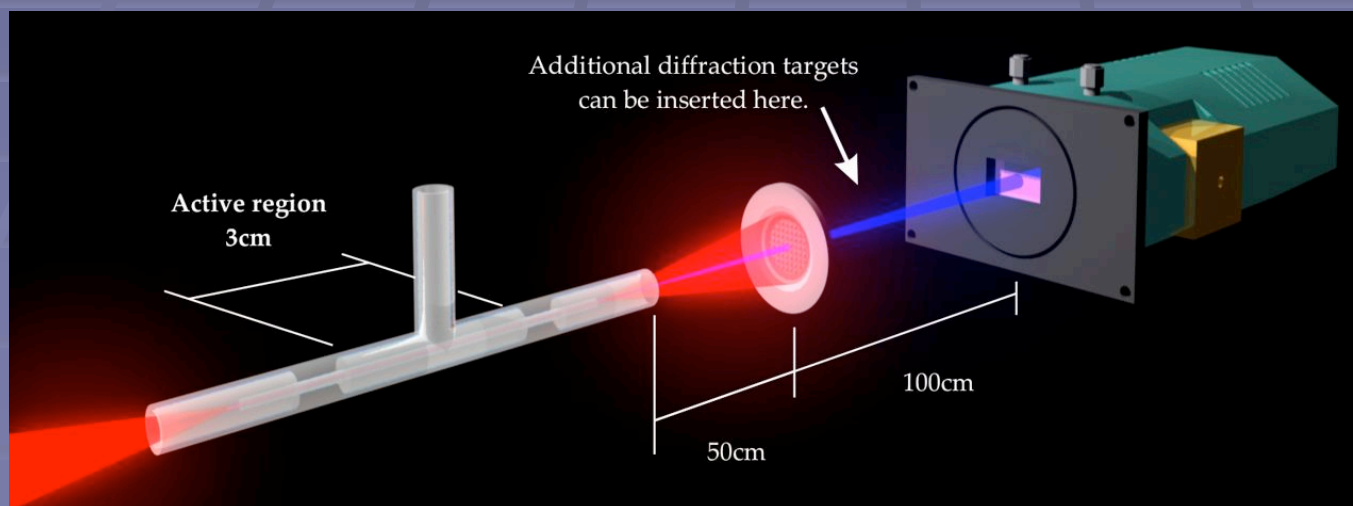
24 March

University of Southampton



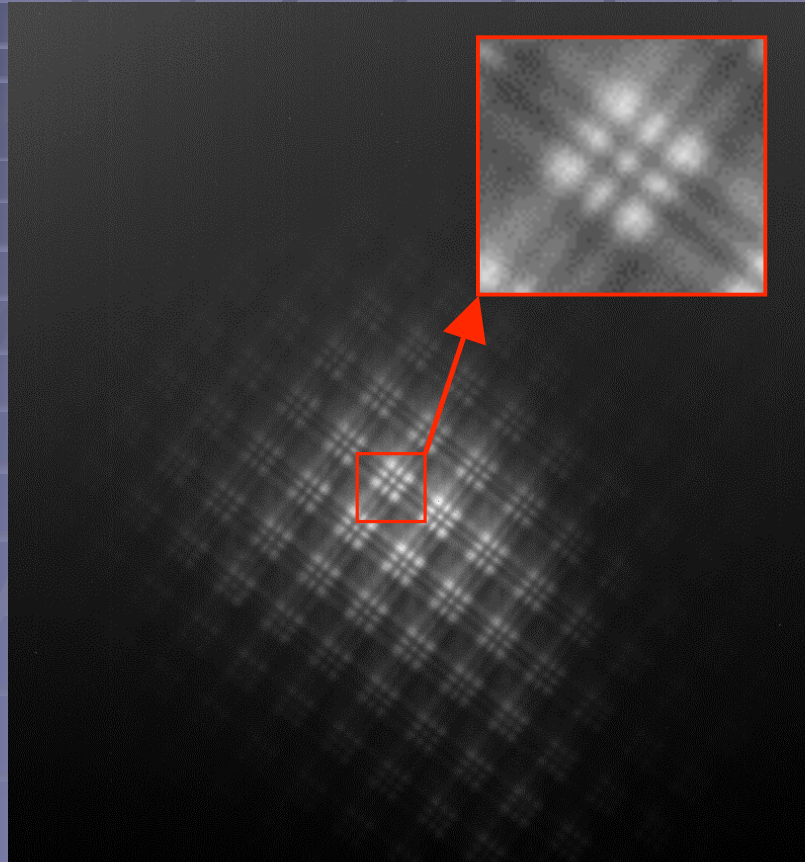
Square Grid Diffraction

- Record x-ray diffraction by filter support grid.
- 340 μm square apertures, 18 μm bars.
- 50cm source to grid, 100cm grid to camera.
- Observe spatial variation of diffraction patterns.
- We can extract the x-ray spectrum.





Fresnel diffraction of EUV beam



- Wire mesh: $18\mu\text{m}$ bars, $340\mu\text{m}$ spacing (Al filter support)
- Experiment and theory agree.
- Incoherent sum of all harmonics.



Chemical Grids

- Start in the laboratory – pervasive physical grid (Smart Tea)
- Computational chemistry very significant source – *in silico* grid (*Chem MyTea*)
- Used by chemists so must be simple to support & maintain - *simple, stable, secure, & autonomic*

What are the people up to?

Who is doing what?

Where are they doing it?

What is the environment like?



The laboratory notebook has been the way scientific research has been recorded for over 200 years

Can we do better now?

How about an electronic notebook?





Like cash machine
electronic notebooks
have taken many forms
we want one that is
simple to use in the lab
whilst doing actual
bench chemistry



ChemLab

The Chemistry 3/5 & 6
Laboratories

- ▶ General Information
- ▶ Instruments & Techniques
- ▶ Chemistry 3/5 Experiments
- ▶ Chemistry 6 Experiments

DARTMOUTH COLLEGE

Permanent,
primary
record
Observations

Write down what
you see

24 March 2006

Safety

[General Rules](#)
[Safety Equipment](#)
[Safety Hazards](#)
[Emergency Procedures](#)

Resources

[Applets](#)
[General FAQ](#)
[Uncertainty](#)

[ChemLab Home](#)

[Info](#) | [Techniques](#) | [Chem 3/5](#) | [Chem 6](#)

How to Keep a Notebook

One of the most useful skills you will acquire in the laboratory is the proper use of a laboratory notebook. Notebooks, or other formally kept records, are an essential tool in many careers, ranging from that of the research scientist to that of the practicing physician. The effort invested in developing good habits of notebook use will be amply repaid for students who pursue a future in the basic or applied sciences. Experience has indicated that skillful notebook use is developed by most students only through continued special effort--it does not come naturally. Some of the main principles of sound notebook use are outlined below.

The laboratory notebook is a permanent, documented, and primary record of laboratory observations. Therefore, your notebook will be a bound journal with pages that should be numbered in advance and never torn out. A notebook will be supplied to you before the first laboratory period. Write your name, the name of your TA, and your lab section on the cover of your notebook. All notebook entries must be in ink and clearly dated. No entry is ever erased or obliterated by pen or "white out". Changes are made by drawing a single line through an entry in such a way that it can still be read and placing the new entry nearby. If it is a primary datum that is changed, a brief explanation of the change should be entered (e.g. "balance drifted" or "reading error"). No explanation is necessary if a calculation or discussion is changed; the section to be deleted is simply removed by drawing a neat "x" through it.



necessary if a calculation or discussion is changed; the section to be deleted is simply removed by drawing a neat "x" through it.

In view of the fact that a notebook is a primary record, data are not copied into it from other sources (such as this manual or a lab partner's notebook, in a joint experiment) without clear acknowledgment of the source. Observations are never collected on note pads, filter paper, or other temporary paper for later transfer into a notebook. If you are caught using the "scrap of paper" technique, your improperly recorded data may be confiscated by your TA or instructor at any time. It is important to develop a standard approach to using a notebook routinely as the primary receptacle of observations.

Each week at the beginning of lab lecture, you will turn in your prelab problems from the manual for grading. Problems not turned in at the beginning of lab lecture will be

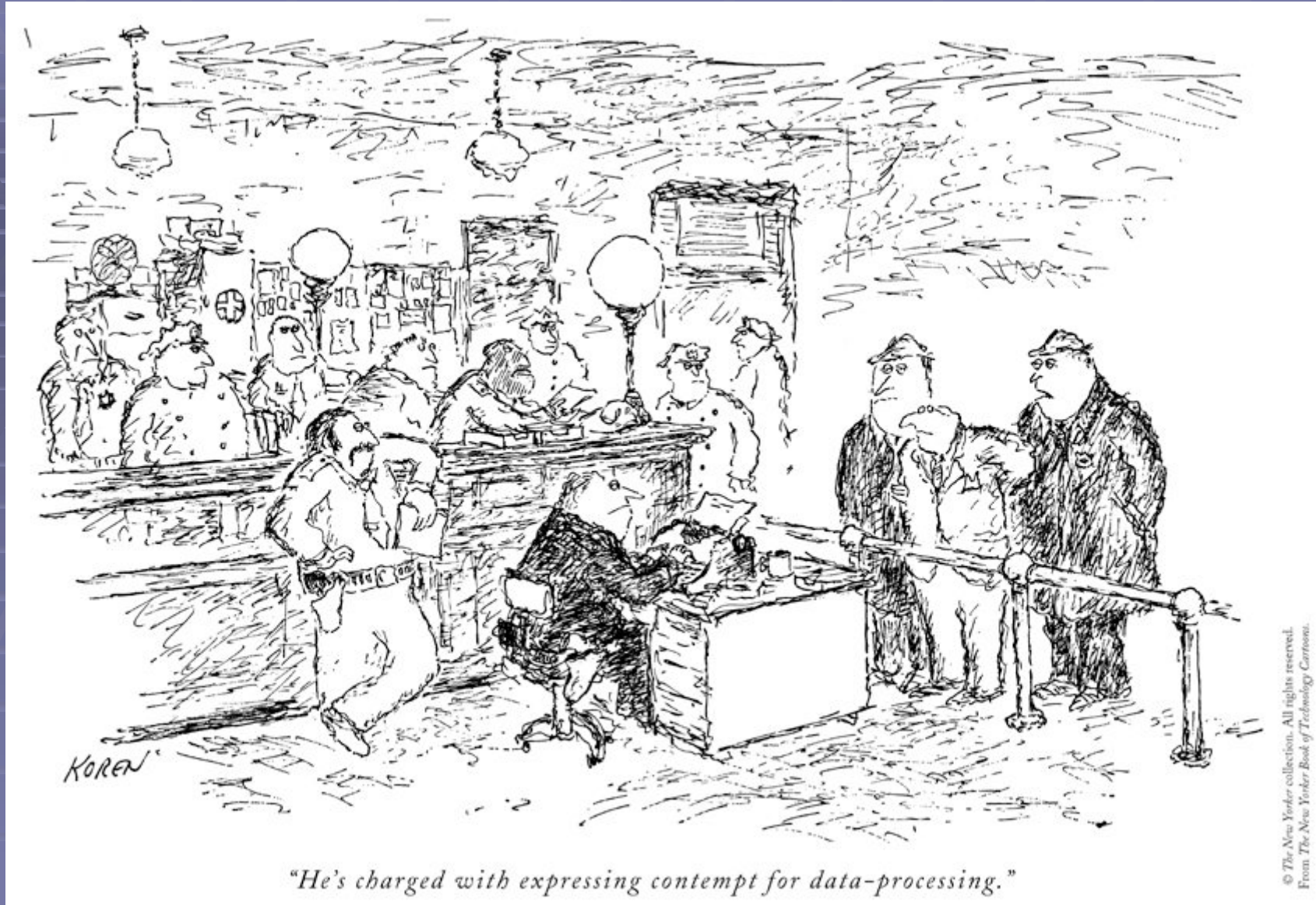
Observations are never collected on bits of paper to be written up later on!



If you are caught using the "scrap of paper" technique, your improperly recorded data may be confiscated by your teacher!

Jeremy G.
University of Southamp.

ursley



He is charged with expressing contempt for meta-data

24 March 2006

Jeremy G. Frey
University of Southampton

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Digital
record at
source
don't try
to add
metadata
after the
fact

*"We can pause, Stu—we can even try fast-forwarding—
but we can never rewind."*



Different labs

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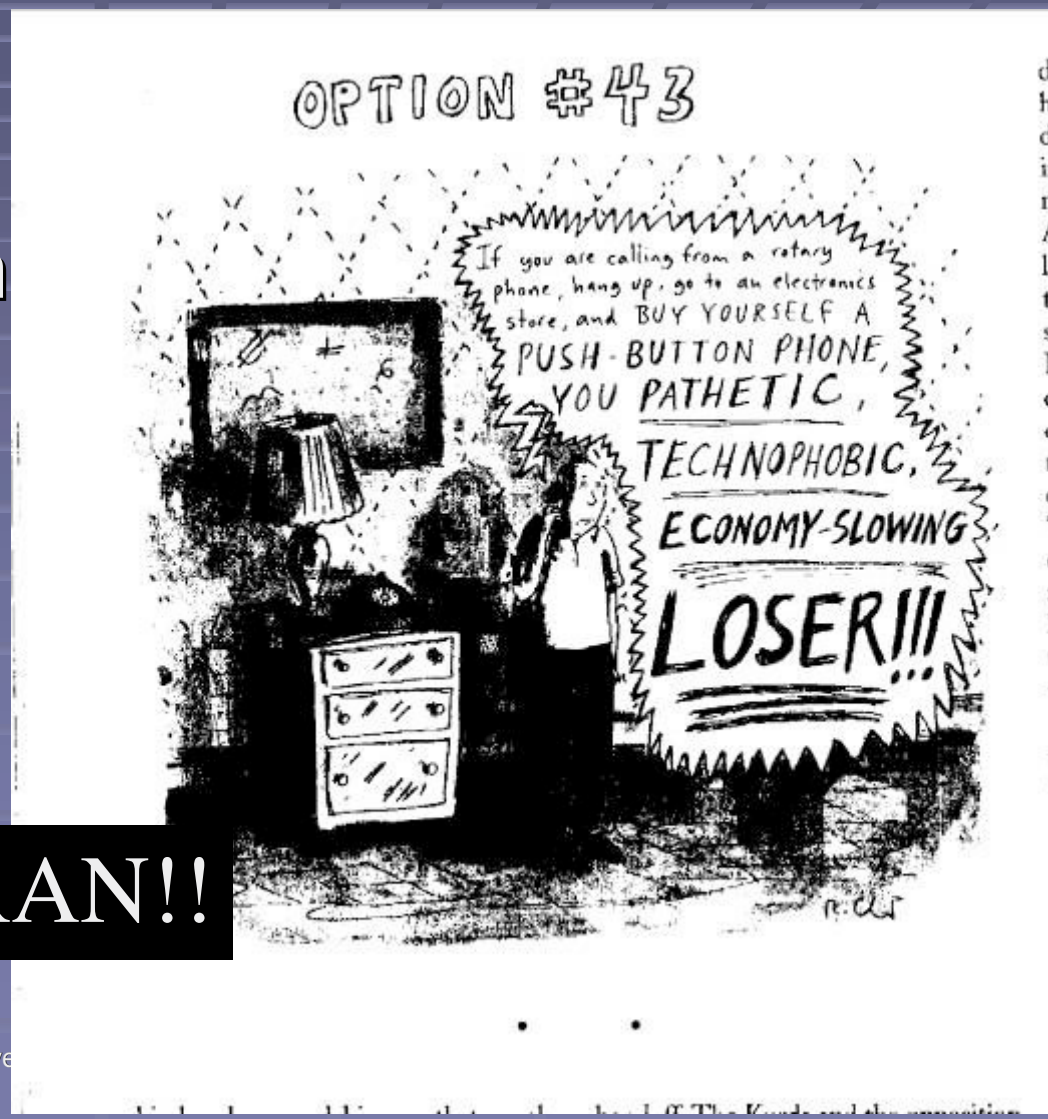
IBM Hursley



Chemists and programming

- Many Chemists think that they can program
- So leave the systems to the Chemists

You still use FORTRAN!!





e-Workflow

Some Chemists
can and leaving
it to the
computer
scientists can
give you a
perfect system
for a problem
you didn't
know you had!

24 March 2006



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What about that! His brain uses formal logic! No wonder it took so long to get a result

Jeremy G. Frey
University of Southampton

HTX, IBM Hursley



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Jeremy G. Frey
University of Southampton

HTX, IBM Hursley



COSHH

leverage off things we already have to do

COSHH ASSESSMENT FORM				Record No.
SUBSTANCE NAME	PHYSICAL FORM	QUANTITY	NATURE OF HAZARD	
Water	liquid	1000ml	None	
Dextrose	Soln	<20g	possible irritation to eyes and skin	
Caffeine	Solid (tea)	<1g	Harmful if swallowed, induce vomiting.	
Milk	liquid	<100ml	No particular hazards	
NATURE OF PROCESS liquid extraction of caffeine, followed by combination with dextrose to produce a sweet drink				
Is there a less hazardous substance? No If so, why not use it?				
CONTROL MEASURES REQUIRED (Local exhaust ventilation, personal protection, etc.) No specific measure required				



But How to get
chemists and
computer
scientists to
understand each
other

***By Making
Tea!***



24 March 2006

Jeremy G. Frey
University of Southampton

HTX, IBM Hursley



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Jeremy G. Frey
University of Southampton

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Result of extensive
collaborative HCI
research between
Computer
Scientists and
Chemists over Tea

Weigh-Station #1
11-Feb-2004 16:04:40

dj**djbj3403**

Experiment Details

Name	Planned	Actual
Fluorinated biphenyl	0.9000 g	0.9031 g
Br11OCB	1.5900 g	1.5918 g
Potassium Carbonate	2.0700 g	2.0719 g
Butanone	40.0 ml	40.0 ml

7

8

9

4

5

6

1

2

3

0

.

Enter

Del

All measurements completed.

djbj3

Merck2

Potassium Carbonate

Butanone

Escape


Quit

Weigh


Liquid-Measure

Bench

Store



Unregistered HyperCam 21#>
46:54 BST 2004 execQuery (experimentmetadata:http://smarttea.org/
000000nn)
00:15 BST 2004 execQuery (experimentmetadata:http://smarttea.org/
000000nn)
00:15 BST 2004 execQuery (ingredients:http://smarttea.org/#000000
n)
00:15 BST 2004 execQuery (observation:http://smarttea.org/#000000
u)
00:15 BST 2004 execQuery (observation:http://smarttea.org/#000000
6)
00:15 BST 2004 execQuery (observation:http://smarttea.org/#000000
x)
00:15 BST 2004 execQuery (observation:http://smarttea.org/#000000
o)
00:15 BST 2004 execQuery (steps:http://smarttea.org/#000000000000
0)
03:57 BST 2004 performRDQL:SELECT ?p, ?s WHERE (?p, <cec:experim
ng cec for <http://www.combechem.org/ontology/process/0.1#>
03:59 BST 2004 execQuery (experimentmetadata:http://smarttea.org/
000000nn)
04:01 BST 2004 execQuery (ingredients:http://smarttea.org/#000000
n)
04:01 BST 2004 execQuery (observation:http://smarttea.org/#000000
u)
04:01 BST 2004 execQuery (observation:http://smarttea.org/#000000
6)
04:01 BST 2004 execQuery (observation:http://smarttea.org/#000000
6)
Смотрите на экран! Студия Артемия Лебедева



The underlying digital world - The Matrix!

24 March 2006

Jeremy G. Frey
University of Southampton

HTX, IBM Hursley



Back to Safety issues

Incompatible Chemicals

http://ptcl.chem.ox.ac.uk/MSDS/incompatibles.html

Apple (147) Amazon eBay Yahoo! News (834)

Incompatible Chemicals

A wide variety of chemicals react dangerously when mixed with certain other materials. Some of the more widely-used incompatible chemicals are given below, but the absence of a chemical from this list should not be taken to indicate that it is safe to mix it with any other chemical!

I'm adding links to MSDS data for these chemicals, but if there is no link from this page to MSDS data for a chemical in which you are interested, try the main page on this web site, through which you can access MSDS data for a large number of chemicals. Click here to get to the [Safety web pages of the Physical and Theoretical Chemistry Laboratory, Oxford University](#).

- [acetic acid](#):
chromic acid, [ethylene glycol](#), [nitric acid](#), hydroxyl compounds, [perchloric acid](#), peroxides, permanganates
- [acetone](#):
concentrated sulphuric and nitric acid mixtures
- [acetylene](#):
[chlorine](#), [bromine](#), [copper](#), [fluorine](#), [silver](#), [mercury](#)
- alkali and alkaline earth metals:
[water](#), chlorinated hydrocarbons, [carbon dioxide](#), halogens, alcohols, aldehydes, ketones, acids
- [aluminium](#) (powdered):
chlorinated hydrocarbons, halogens, [carbon dioxide](#), organic acids.
- anhydrous ammonia:
[mercury](#), [chlorine](#), [calcium hypochlorite](#), [iodine](#), [bromine](#), [hydrofluoric acid](#)
- [ammonium nitrate](#):
acids, metal powders, flammable liquids, chlorates, nitrites, sulphur, finely divided organic combustible materials
- [aniline](#):
[nitric acid](#), [hydrogen peroxide](#)
- arsenic compounds:
reducing agents
- azides:
acids

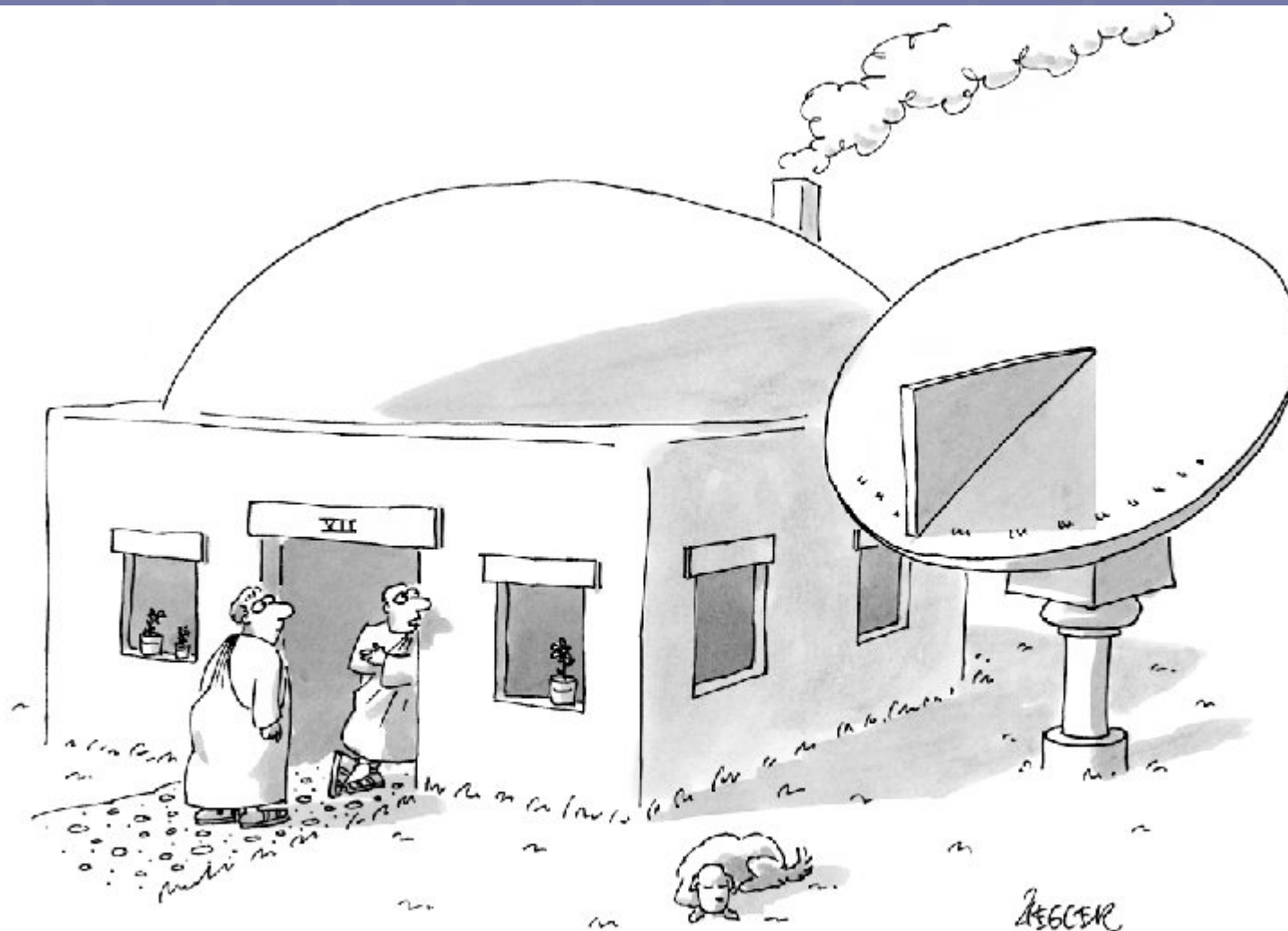


But what
about the
laboratory
envionement?



**"I just realized, Howard, that everything
in this apartment is more sophisticated
than we are"**

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"Sure, it's an eyesore, but we get better time than anyone else in the neighborhood."

Will the technology scale?

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University of Southampton

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Smart Places & Things



We need Smart Labs!

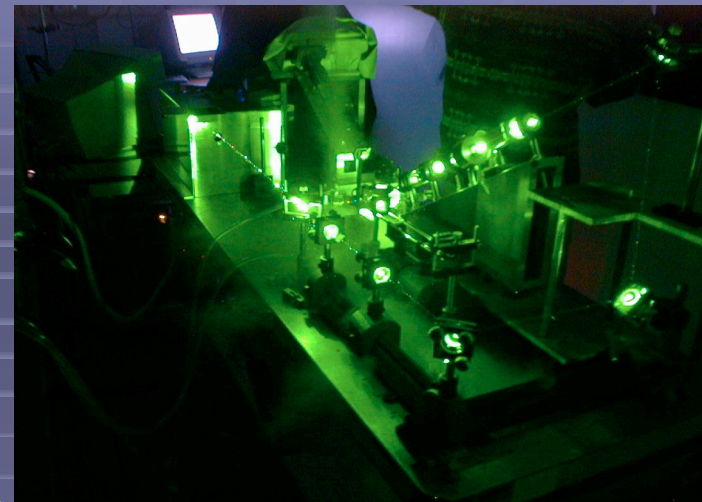
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Temperature
in the lab



Distribute



My PC

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University of Southampton

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Temperature
in the lab

My PC

Student's
PC

Archive



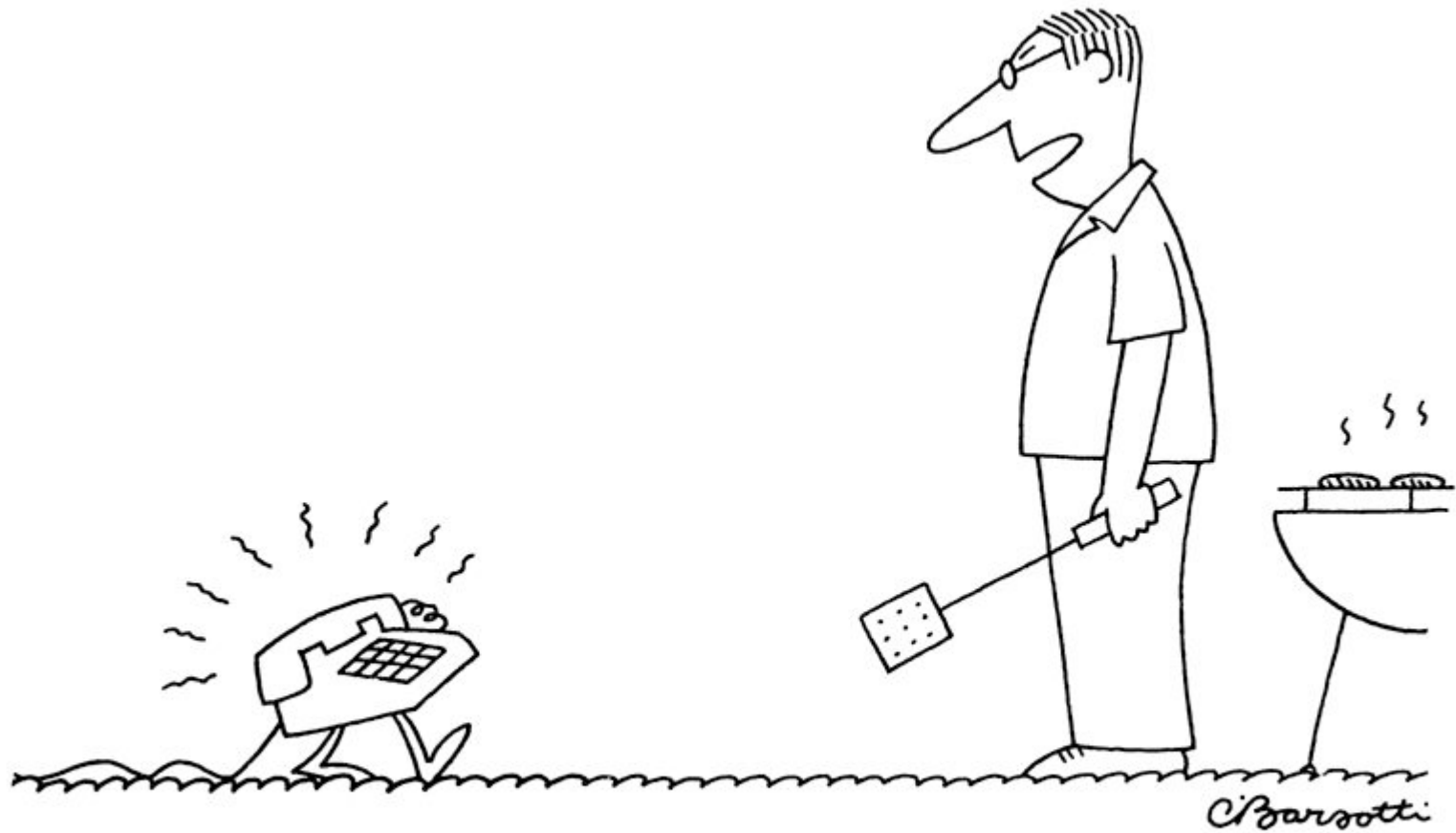
Temperature
in the lab

Door
opening

My PC

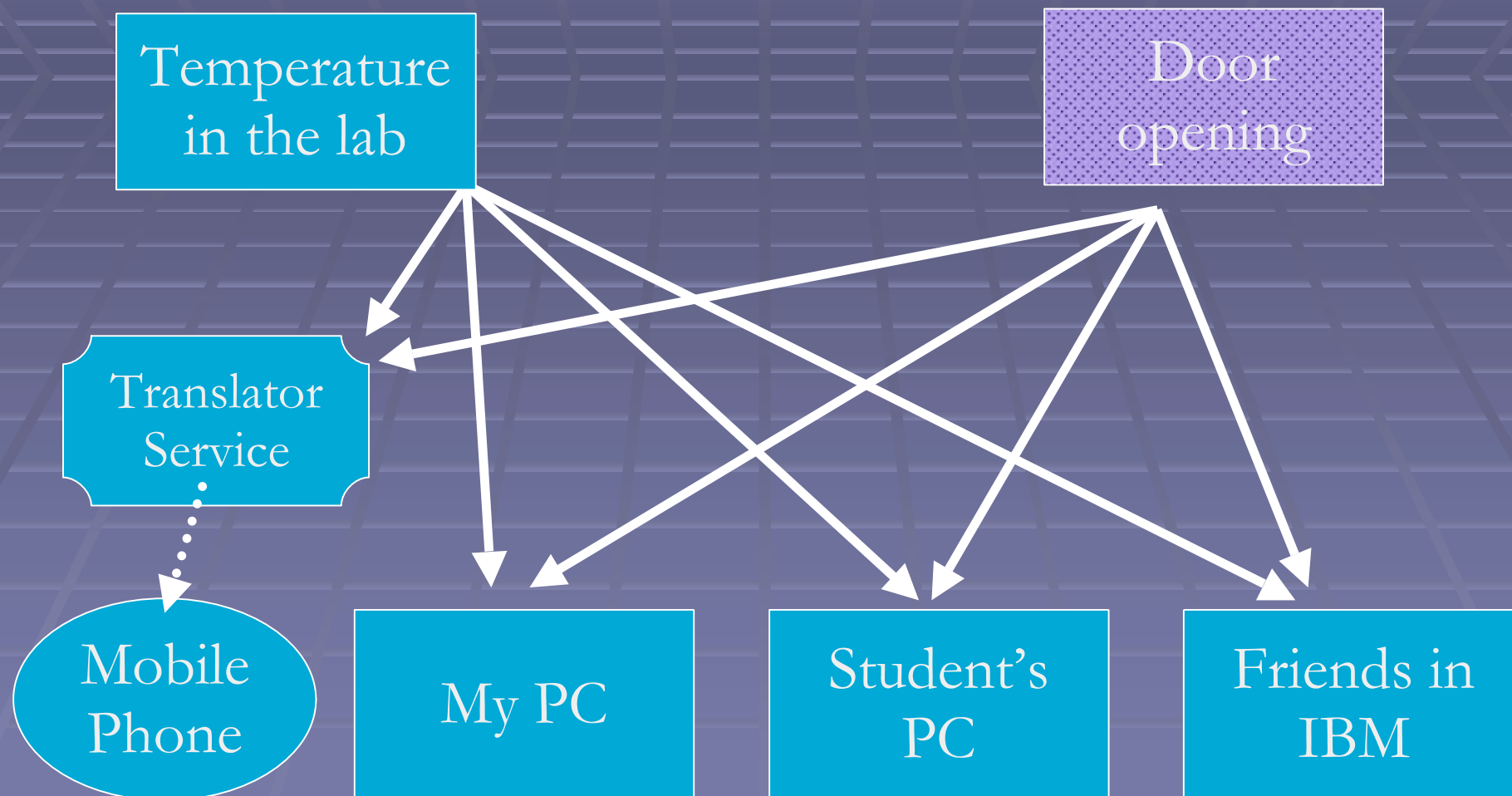
Student's
PC

Friends in
IBM



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“What the hell sort of convenient new feature is this?”

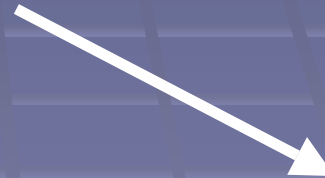




Temperature



Message
Agent



My PC



Temperature

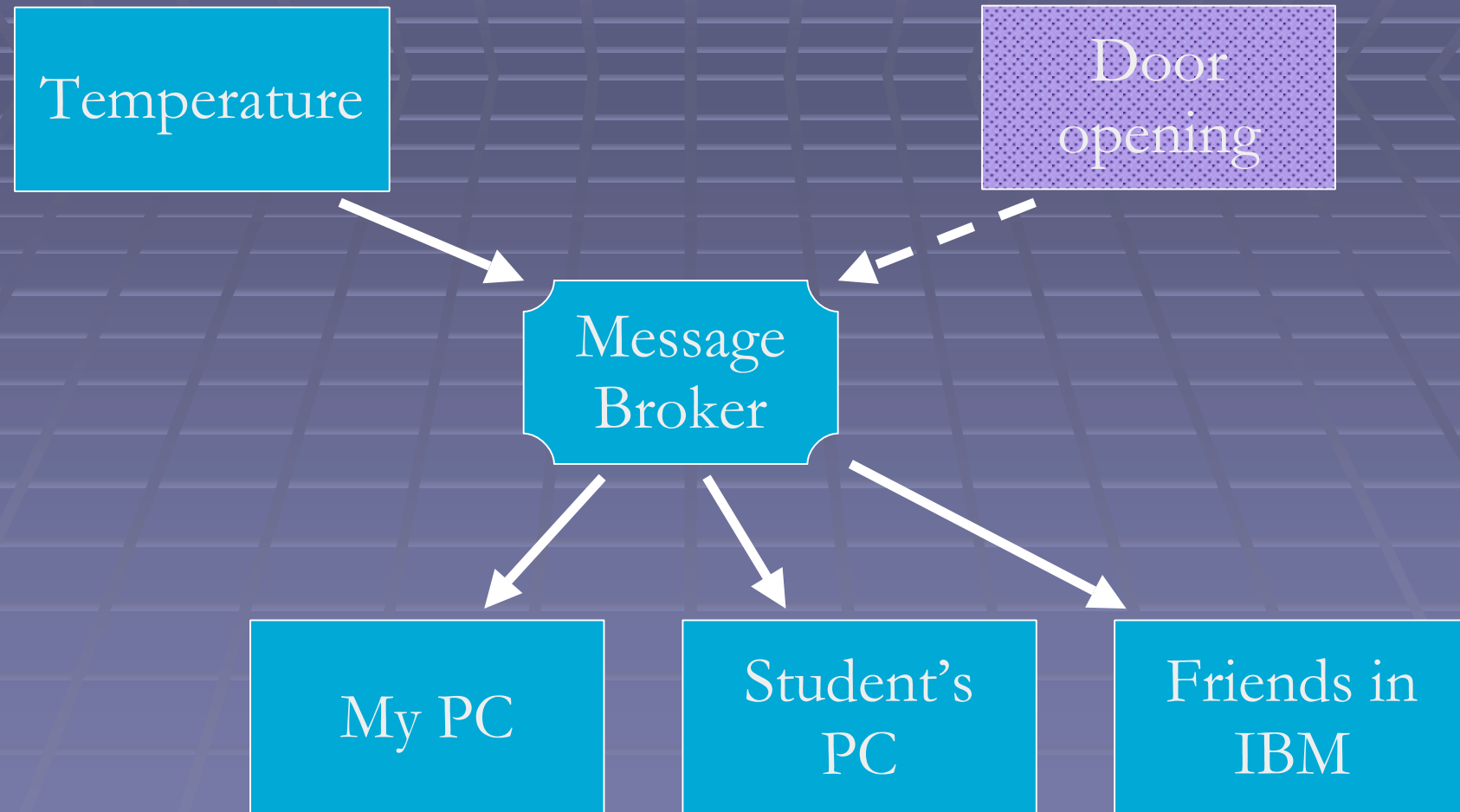
Door
opening

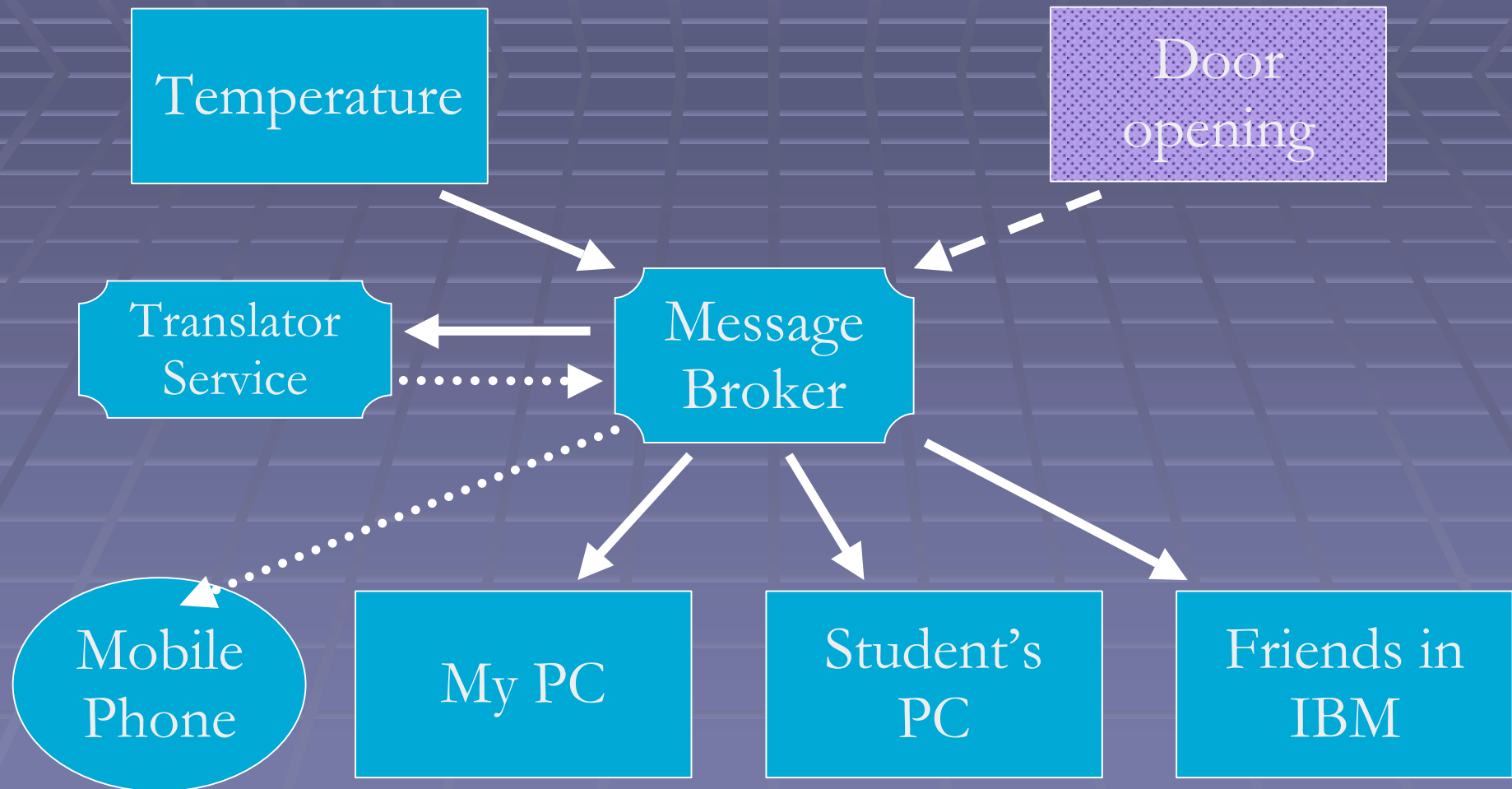
Message
Broker

My PC

Student's
PC

Friends in
IBM







Monitoring by Phone

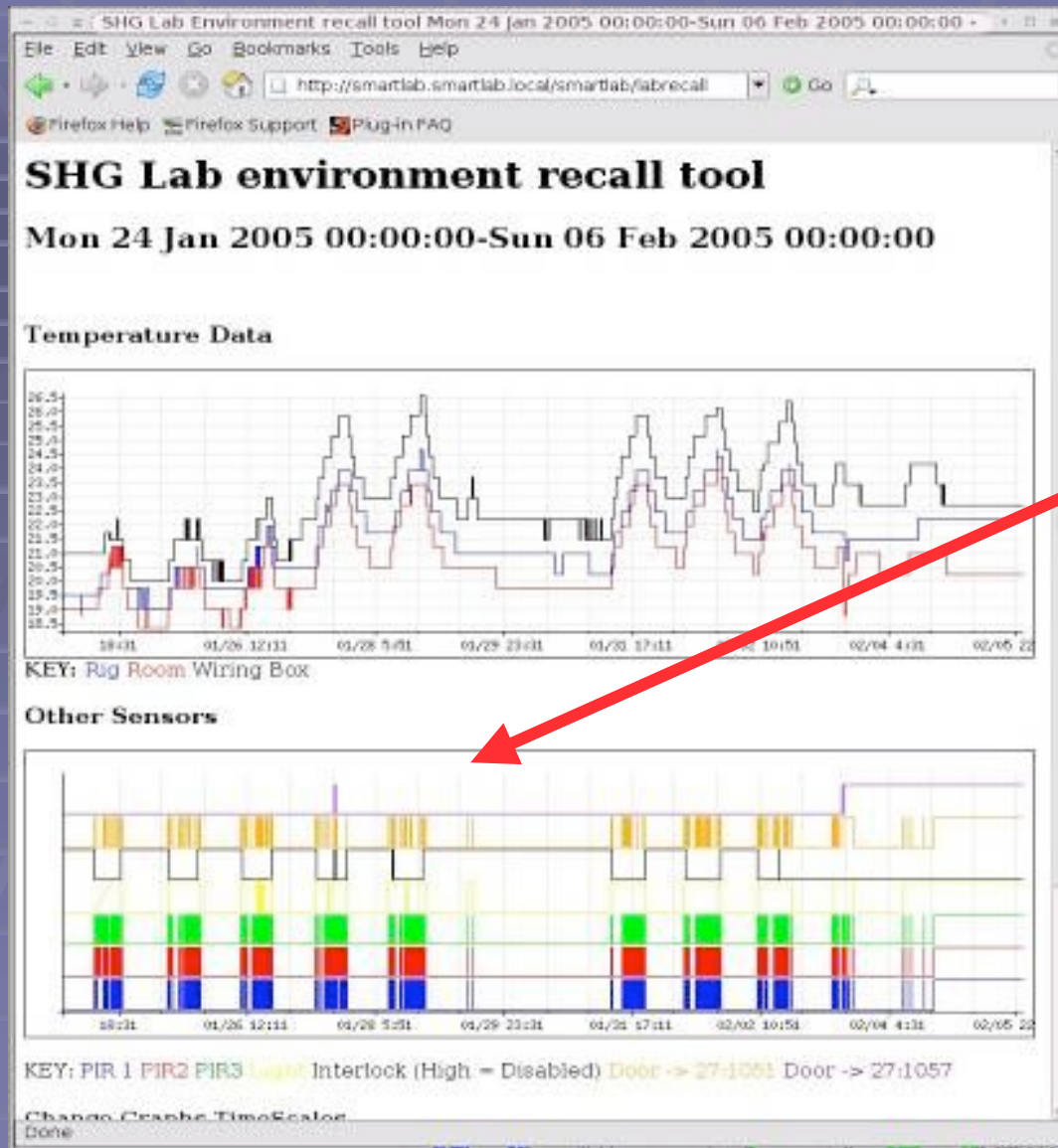




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"I love the convenience, but the roaming charges are killing me."

"I love the convenience, but the roaming charges are killing me"



Air Conditioning failed



What you have to do to get on to the BBC web site

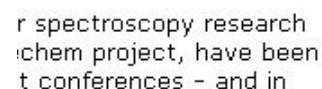


Pub/Sub for Laboratory data
using a broker and ultimately
delivered over GPRS



24 March 2006

Jeremy G. F.
University of Southampton



...so that at a push of a button, it is able to remotely change the temperature.

controlling home
and seamlessly.

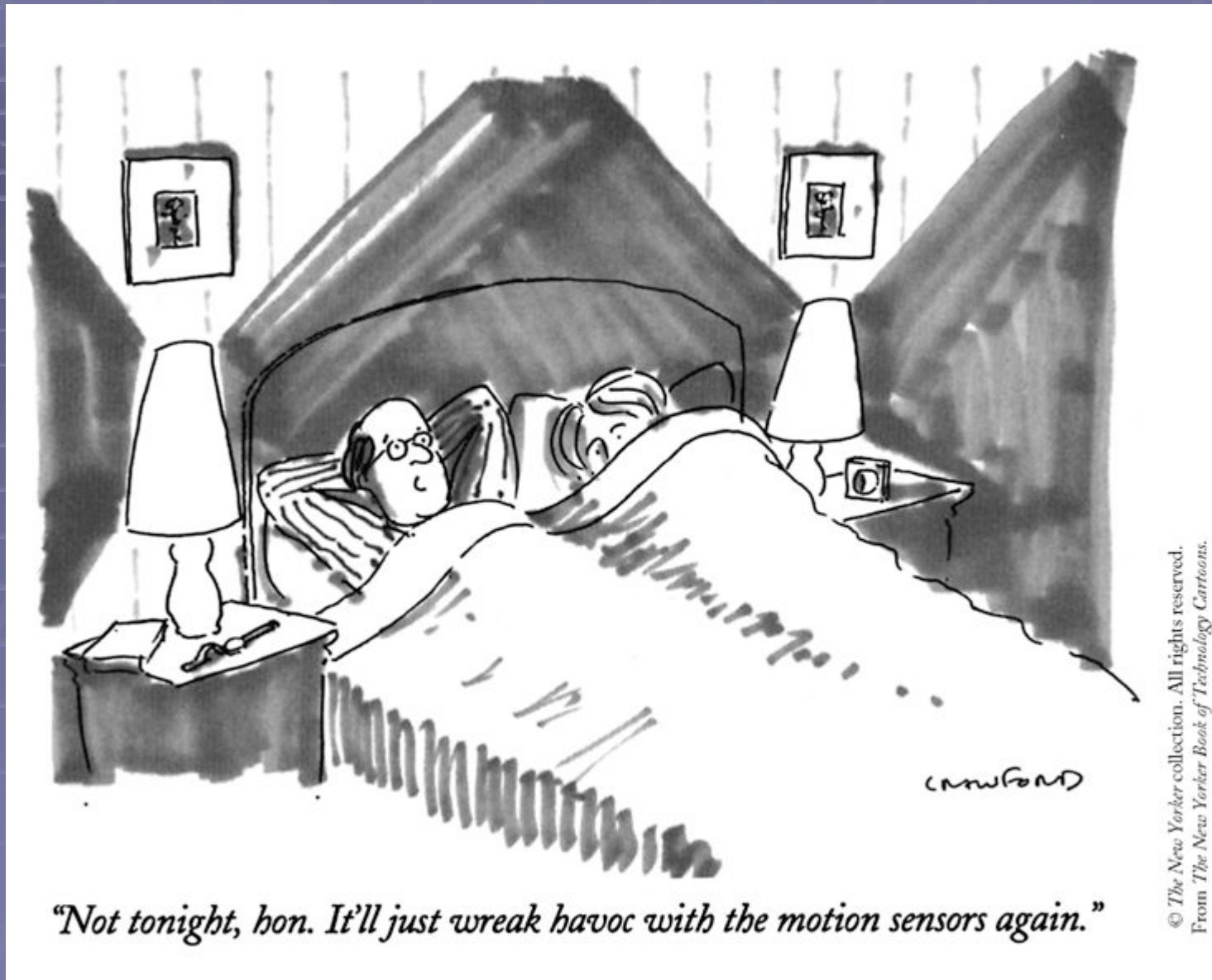


Chemists enjoy a drink at the bar while keeping an eye on the lab.

IBM won the Royal Academy of Engineering's MacRobert prize which rewards technological and engineering innovation for the program in June last year.

Used by top global banks, the WebSphere MQ family is a decade old.

University of Southampton





We don't
want to
take this
technology
too far.

There are
privacy
issues



"When we implant your pacemaker, we can, for a modest additional fee, also implant your beeper."

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"When we implant your pacemaker, we can, for a modest additional fee, also implant your beeper."



24 March 2006

Remote control?

Jeremy G. Frey
University of Southampton

HTX, IBM Hursley



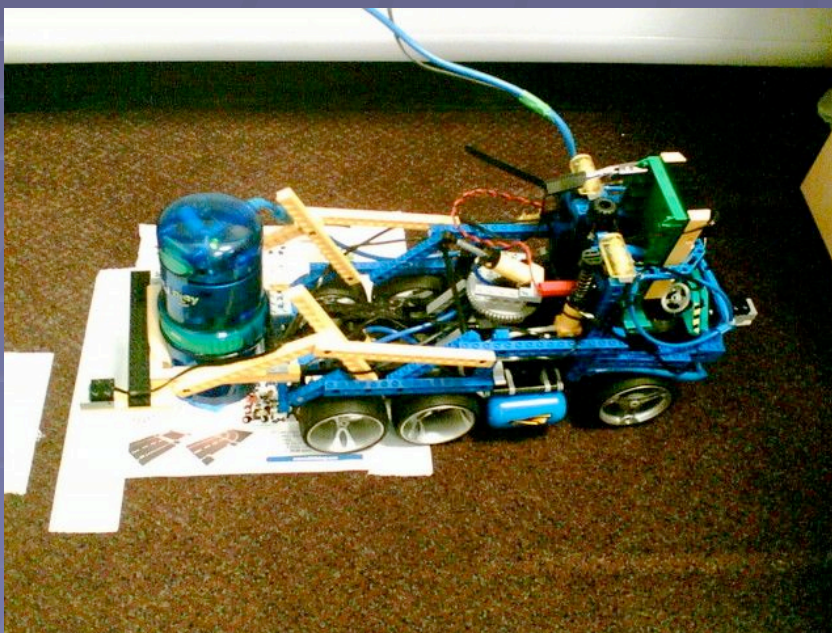
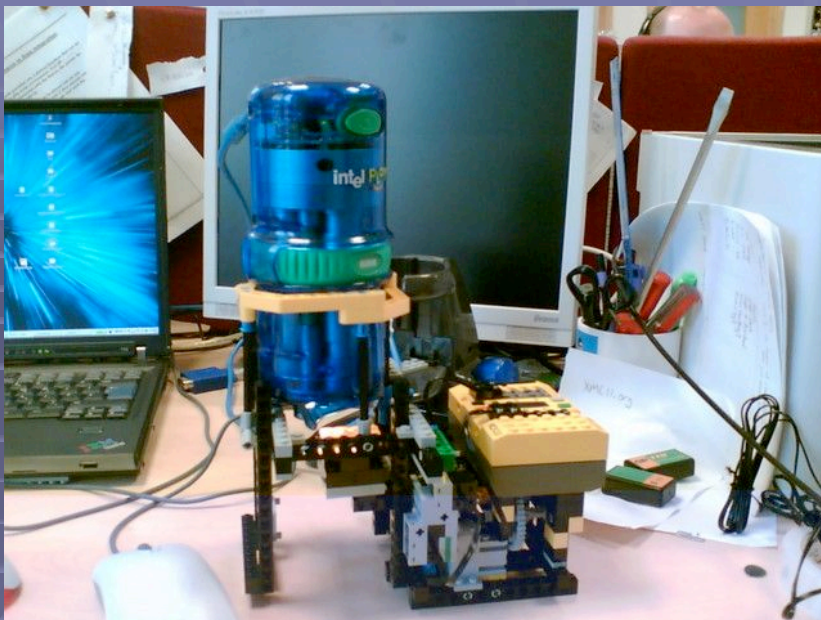
Security and trust for experiments and data



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University of Southampton

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Need to make
the data
available

Need to be
able to find it

But how to
expose it?



First, they do an online search



Fig. 2 (a) Phase behavior of 1,4-bis(4-polyphenylene-2,5-diyl)-2,2'-(p-phenylene)-4,4'-bis(4-phenylene-2,5-diyl) ether (PPE) in various solvents. (b) Phase behavior of 1,4-bis(4-polyphenylene-2,5-diyl)-2,2'-(p-phenylene)-4,4'-bis(4-phenylene-2,5-diyl) ether (PPE) in various solvents. (c) Phase behavior of 1,4-bis(4-polyphenylene-2,5-diyl)-2,2'-(p-phenylene)-4,4'-bis(4-phenylene-2,5-diyl) ether (PPE) in various solvents.

The screenshot shows a software window titled "NewImage2014-06-10 - NewImage". The left pane displays a list of coordinates (X, Y, Z) for 1000000 points. The right pane shows a grayscale image of a grid of bright spots, with a central dark region.

Free
uthampton

Intellect & Interpretation

Jeremy G. Frey
University of Southampton



```

C:\Users\user> g++ 1.cpp
C:\Users\user> .\1.exe
Sum of squares of first 100 natural numbers is: 338350

```

[illegible]

Including all the laboratory and environmental data

The screenshot shows a web browser window with a URL bar containing "http://sparta.earth.ac.uk/ncv/00000001-01/0001245-45". The main content area displays a table of environmental data with columns for time (e.g., 11:0, 11:15, 11:30) and values (e.g., 12.00, 12.00, 12.00). A red arrow points from the table to a "Data page report: Netscape" window. Another red arrow points from the "Data page report: Netscape" window to a "Summary page for Directory:" window. The "Summary page for Directory:" window shows the title "EPSRC National Data Centre" and a date "Report generated Sat 09/ 2007/ 10:13:41".

File Edit View Go Bookmarks Help

http://reports.eprdc.ac.uk/ncrhc/00000009/01/02SOT082.HTM

EPSRC National Crystallography Service

Data Collection Summary

Summary report for Directory: diska/02sot082

Report generated Jul 09, 2002, 10:13:51

Unit cell

15124 reflections with $2.91^\circ < \theta = 27.48^\circ$ (resolution between 7.00Å and 0.77Å) were used for unit cell refinement

Symmetry used	p222
a (Angstrom)	9.3113 ± 0.0003
b (Angstrom)	9.8424 ± 0.0003
c (Angstrom)	15.4441 ± 0.0004
alpha (°)	90.000
beta (°)	90.000
gamma (°)	90.000
Volume (Å ³)	1415.69 ± 0.07
Mosaicity (°)	0.743 ± 0.002

Start | Desktop | Microsoft Word | Microsoft Excel | Microsoft PowerPoint | Windows | Windows Explorer | Internet Explorer | Outlook Express

Jeremy G. Frey
University of Southampton

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Subversive
and furtive
sharing &
exploitation
of data



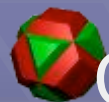
PERPETRATOR OF A DARING DAYLIGHT ILLEGAL ELECTRONIC
TRANSFER OF FUNDS FLEEING THE SCENE OF THE CRIME

24 March 2006

Data

Jeremy G. Frey
University of Southampton

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Chemistry Data

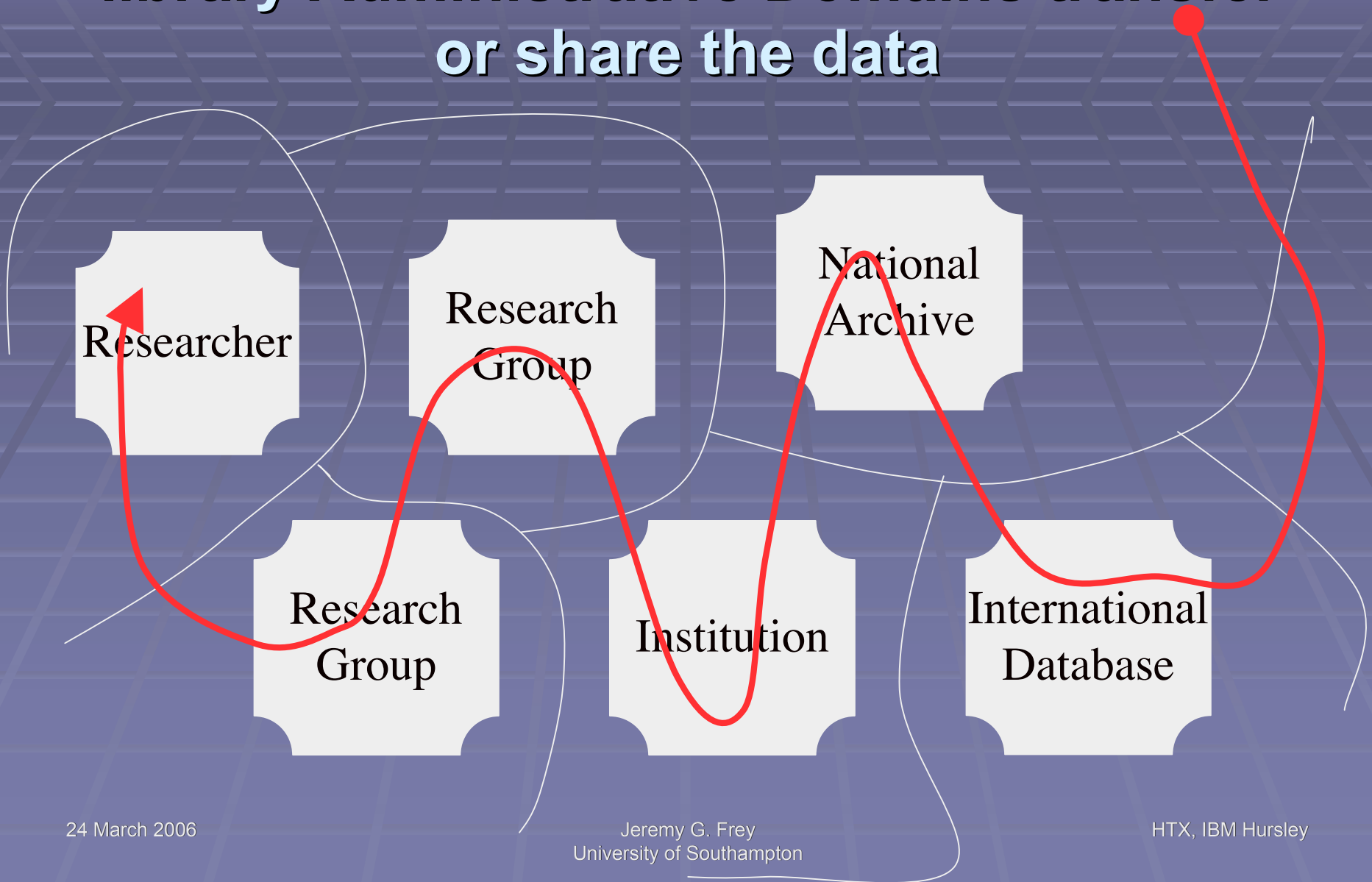
private or public,

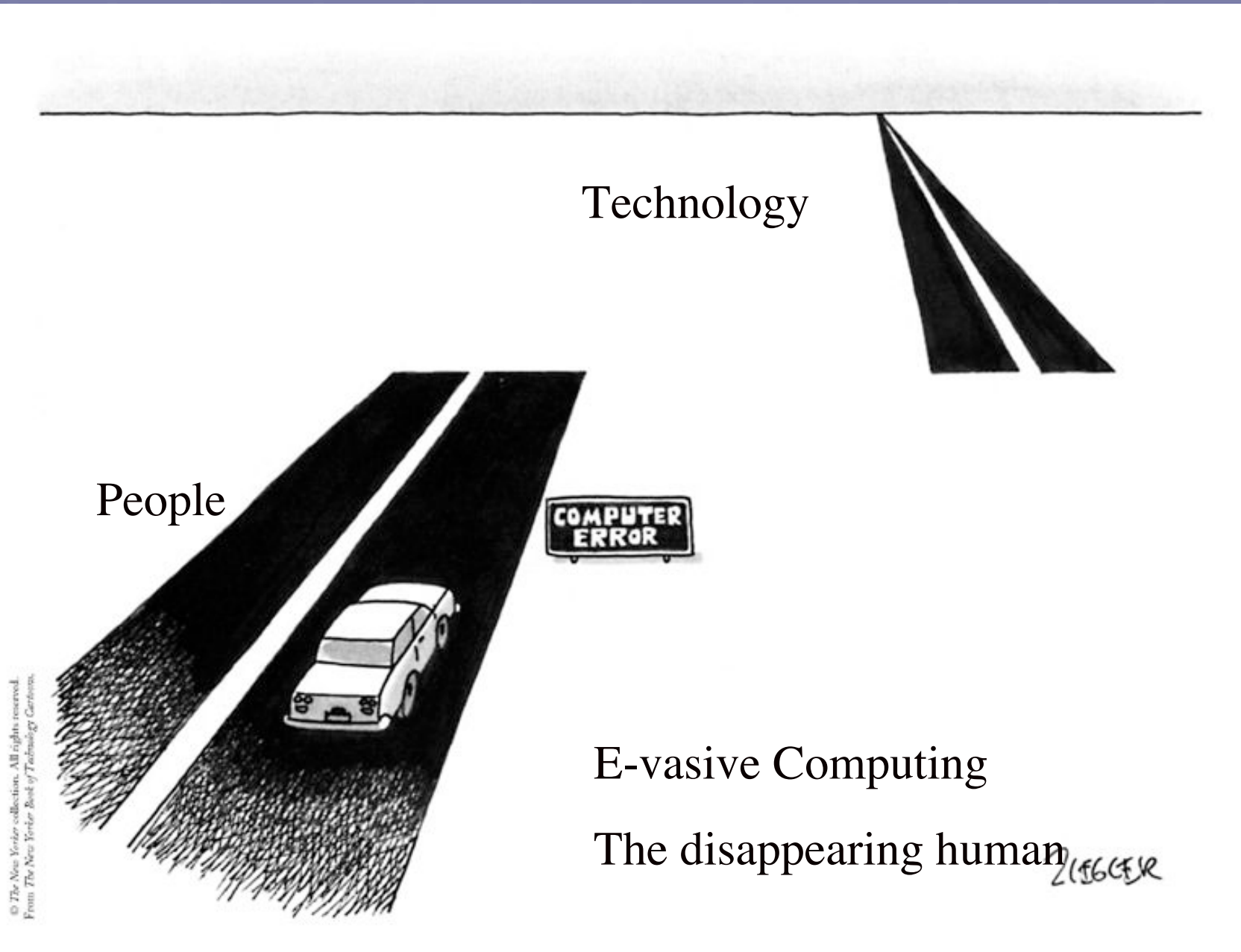
open or controlled access





Several groups making and analysing the library Administrative Domains transfer or share the data





24 March 2006

Jeremy G. Frey
University of Southampton

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University of Southampton

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