The Long-Lived Nuclear Singlet State of $^{15}$N-Nitrous Oxide in Solution

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Nuclear singlet states may have lifetimes $T_S$ that exceed the conventional magnetization relaxation time $T_1$ by an order of magnitude. 1-11 Applications of these states to the NMR measurements of slow molecular diffusion, chemical exchange, and the transport of hyperpolarized nuclear spin order have been demonstrated.9-11

So far, long-lived nuclear singlet states have only been observed for proton pairs. We now demonstrate an extraordinarily long lifetime ($T_S$) of $\sim 26$ min for the nuclear spin singlet of $^{15}$N$_2$-nitrous oxide (dinitrogen monoxide, N$_2$O) in solution. This result has high potential importance since nitrous oxide is soluble in many important fluids such as water, oil, and blood. It is used routinely as a food additive, a gas propellant, and an anesthetic.

Doubly labeled $^{15}$N$_2$O gas, purchased from CK-gas (UK), was dissolved in a degassed solution of DMSO-$d_6$ at a pressure of $\sim 3.5$ bar. The concentration of $^{15}$N$_2$O in solution was $\sim 0.3$ M as determined by comparison of the $^{15}$N signal strength with that from an external standard of $^{15}$N-benzamide. The $^{15}$N sites in $^{15}$N$_2$O are inequivalent and form a AX spin system with a difference in chemical shifts of $\Delta \delta = 82.3$ ppm and a $^{15}$N-$^{15}$N scalar coupling $J = 8.1$ Hz. The $^{15}$N NMR spectrum, obtained in a magnetic field $B_{high} = 7.0463$ T, is shown in Figure 1a.

The slow relaxation of singlet states is revealed by suppressing their interconversion with the triplet states, either by using a resonant radiofrequency field3-6 or by reducing the static magnetic field to a very low value.1,2 The radiofrequency method is not feasible for $^{15}$N$_2$O because of the large $^{15}$N chemical shift difference. The field-cycling procedure shown in Figure 2 was therefore used. This is a modified version of the method used in ref 1.

The spin system is allowed to reach thermal equilibrium, and two strong 90° pulses with a relative phase of 90° are applied at the mean chemical shift frequency of the two $^{15}$N sites. The delay between the pulses, $\tau_1 = 0.198$ ms, was chosen so that the transverse magnetization vectors of the two $^{15}$N sites precess through 180° relative to each other. The two pulses act as a selective 180° pulse on one of the $^{15}$N sites and lead to a spin density operator10 of the form

$$\hat{\rho}_2 = -\hat{I}_{j} + \hat{I}_{k} = \begin{pmatrix} \alpha \beta \rangle \langle \alpha \beta | - | \alpha \beta \rangle \langle \alpha \beta | \end{pmatrix}$$

where the two sites are denoted $j$ and $k$ and the selective inversion is assumed to act on site $j$. In this and the following equations, the subscript refers to a time point in Figure 2.

The sample is transported out of the magnetic field by activating a stepper motor to wind up a string attached to the sample holder. The transport process takes $\tau_{tr} = 40$ s and transports the sample into a region of low magnetic field, $B_{low} \approx 2$ mT, estimated by a Hall effect device. As shown in ref 2, slow adiabatic transport converts the population of each high-field state into that of the corresponding low-field state, leading to a density operator of the form

$$\hat{\rho}_3 = |S_0\rangle\langle S_0| - |T_{LF}\rangle\langle T_{LF}|$$

where the low-field eigenstates are $|S_0\rangle = 2^{-1/2}(|\alpha \beta \rangle - |\beta \alpha \rangle)$, $|T_{LF}+\rangle = |\alpha \alpha \rangle$, $|T_{LF}0\rangle = 2^{-1/2}(|\alpha \beta \rangle + |\beta \alpha \rangle)$, and $|T_{LF}-\rangle = |\beta \beta \rangle$. The sample is left in the low-field region for a variable time $\tau_{LF}$. During the first few minutes, the three triplet populations equilibrate with each other on a time scale set by the relaxation constant $T_1$. The density operator after several minutes in low magnetic field is therefore given approximately by

$$\hat{\rho}_4 = |S_0\rangle\langle S_0| - \frac{1}{3}|T_{LF}+\rangle\langle T_{LF}+| - \frac{1}{3}|T_{LF}0\rangle\langle T_{LF}0| - \frac{1}{3}|T_{LF}-\rangle\langle T_{LF}-|$$

The sample is transported back into the high-field region by running the stepper motor in the opposite direction. Adiabatic transport from the density operator in eq 3 leads to

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Figure 1. $^{15}$N spectra of $^{15}$N$_2$-N$_2$O dissolved in DMSO-$d_6$: (a) conventional NMR spectrum taken using a single $^{15}$N pulse in a magnetic field of 7.04 T; (b) spectrum taken using the experimental sequence in Figure 2, with $\tau_{LF} = 300$ s; (c) as in spectrum b but using $\tau_{LF} = 40$ min. All spectra result from the sum of eight transients.

Figure 2. Field-cycling pulse sequence for observing the long-lived singlet state of $^{15}$N$_2$O. The circled numbers refer to time points mentioned in the text.
This corresponds to an NMR signal from only one of the two forms. The advantage of this procedure is that the intensity of the DMSO-pulse with a relative phase of 45° to the spin density operator $\tau$.

In low magnetic field, the chemical shift anisotropy mechanism is the transport into the magnet. For comparison, the long waiting times is due to partial recovery of magnetization during the transport into the magnet. After equilibration in high field, the low-field waiting time $\tau$ values in a saturated solution of $^{15}$N$_2$O are 198 and 114 s for the terminal and central positions of spin-rotation, chemical shift anisotropy, intramolecular dipole–dipole coupling, and external random field mechanisms.

The relaxation of $^{15}$N nuclei in $^{15}$N$_2$O is caused by a superposition of spin-rotation, chemical shift anisotropy, intramolecular dipole–dipole coupling, and external random field mechanisms. In low magnetic field, the chemical shift anisotropy mechanism is absent, and the intramolecular dipole–dipole coupling does not cause singlet relaxation. In a forthcoming publication we show that for pure spin-rotation relaxation, the ratio of $T_S$ and $T_I$ is given by

$$\frac{T_S}{T_I} = \frac{1}{2}(C^2 + C^2)/(C^2 - C^2)^2$$

where $C_1$ and $C_2$ are the spin-rotation tensors for the two $^{15}$N sites. This equation takes into account the strong cross-correlation of the spin-rotation interactions in this linear molecule. The values of the tensors are known leading to an expected ratio $T_S/T_I = 11.4$. The observed ratio $T_S/T_I = 8.0$ is qualitatively consistent with this. The remaining discrepancies are attributed to other relaxation mechanisms. Even slower singlet relaxation is expected for suitable larger molecules, for which even the conventional $^{15}$N $T_1$ may be as long as 500 s.

We are now exploring the dependence of the $^{15}$N$_2$O singlet relaxation time on the solvent and the concentration of paramagnetic agents. For example, we have observed a $T_S$ of ~19 min for $^{15}$N$_2$O in undegassed olive oil.

Although the solubility of $^{15}$N$_2$O in blood at atmospheric pressure is about 15 times less than that observed under the current conditions, it should be possible to overcome this sensitivity loss by hyperpolarization techniques, enabling applications in medical imaging. The long singlet relaxation time of $^{15}$N-nitrous oxide should facilitate the transport of the agent to remote parts of the subject while minimizing the loss of nuclear spin order. Experiments are in progress for estimating the $T_S$ in blood at different oxygenation levels, which might provide a contrast mechanism. Applications are also anticipated for following the diffusion and flow of materials such as oil and food substances.

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