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Materials Research **Express**

Metal-free plasma-enhanced chemical vapor deposition of large area nanocrystalline graphene

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Received 6 March 2014, revised 16 April 2014 Accepted for publication 28 April 2014 Published 29 May 2014

Materials Research Express 1 (2014) 025031 doi:10.1088/2053-1591/1/2/025031

Abstract

This paper reports on large area, metal-free deposition of nanocrystalline graphene (NCG) directly onto wet thermally oxidized 150 mm silicon substrates using parallel-plate plasma-enhanced chemical vapor deposition. Thickness non-uniformities as low as 13% are achieved over the whole substrate. The cluster size L_a of the as-obtained films is determined from Raman spectra and lies between 1.74 and 2.67 nm. The film uniformity was further confirmed by Raman mapping. The sheet resistance R_{sq} of 3.73 k Ω and charge carrier mobility μ of 2.49 cm² V⁻¹ s⁻¹ are measured. We show that the NCG films can be readily patterned by reactive ion etching. NCG is also successfully deposited onto quartz and sapphire substrates and showed >85% optical transparency in the visible light spectrum.

Keywords: PECVD, nanocrystalline graphene, wafer-scale

1. Introduction

Graphene, a one-atom thin carbon layer with hexagonal honeycomb structure, has recently attracted considerable research interest. It is considered a possible successor of silicon in the post-Moore era due to its unique mechanical and electronic properties [1]. However, the application of these extraordinary properties at commercial scale is hindered by the lack of appropriate methods to obtain large area, single crystal graphene. The exfoliation from highly oriented pyrolytic graphite yields the highest quality graphene so far, however, it does not scale and mechanical exfoliation at wafer-scale remains doubtful. Epitaxial growth on silicon carbide

(SiC), on the other hand, allows good control over the number of layers, but the electrical properties of the films become dependent on the orientation due to the surface steps forming during growth, and the expensive SiC substrates have to be processed at temperatures exceeding $1200 \,^{\circ}$ C [2]. The third method to be mentioned here is graphene growth on transition metal catalysts, such as Ni [3], Cu [4] and Fe [5] among others. The catalyst is exposed to a carbon-rich atmosphere at elevated temperature, resulting in the carbon to solve inside the catalyst and crystallizing on the surface upon cooling. The carbon source can be either obtained purely through thermal decomposition (chemical vapor deposition (CVD)) or with the help of a plasma (plasma-enhanced chemical vapor deposition (PECVD)) from hydrocarbon gases, such as methane (CH₄). To make the obtained graphene film usable for electronic applications it is necessary to transfer the film from the catalyst onto a different substrate. This method is being constantly developed and a roll-to-roll process has been recently demonstrated [6]. Nevertheless, there is still scope for research in direct deposition of graphene and nanocrystalline graphene (NCG) layer onto insulator substrates [7]. There has been some research into metal-free deposition of graphene to achieve this goal, such as molecular beam epitaxy (MBE) [8, 9], CVD [7, 10, 11] and various plasma-enhanced methods based on partially modified tools [12–14]. Although these films have typically a higher defect rate compared to graphene obtained by exfoliation or CVD, they can be suitable for applications such as saturable absorber in mode-locked lasers [15].

Here, we demonstrate large area synthesis of NCG on wet thermally oxidized silicon substrates, quartz glass and sapphire glass by a capacitively coupled parallel-plate PECVD process at high pressure of 1.5 Torr. We investigate the quality of these films deposited at different temperatures and conditions, and fabricate two terminal devices to study the resistivity.

2. Deposition method and characterization

NCG films were deposited on wet thermally oxidized $(1000 \,^\circ\text{C}, 48 \,\text{min}, 244 \,\text{nm})$ oxide thickness) n-Si wafers with a diameter of 150 mm. We use an Oxford Instruments Nanofab 1000 Agile in this work, which can process substrates with up to 200 mm diameter. The deposition chamber is schematically illustrated in figure 1(a). The sample is placed on the electrically grounded, temperature controlled lower electrode (graphite, up to 1000 $^\circ$ C) which is used together with the upper electrode to generate the capacitively coupled plasma. The process gas is introduced uniformly through holes in the upper electrode.

The deposition process is illustrated in figure 1(b). First, the substrate is loaded into the process chamber at the loading temperature $T_{\text{load}} = 690^{\circ}\text{C}$ and heated to the deposition temperature T_{dep} . Next, the substrate is cleaned at T_{dep} by introduction of H₂ gas (200 sccm, 1000 mTorr) for typically $t_{\text{clean}} = 15$ min. In the deposition step, CH₄ and H₂ are introduced (1500 mT pressure) and the RF plasma is generated (100 W RF) for t_{dep} . Finally, the substrate is cooled down to T_{load} and unloaded. In case T_{dep} is below 690 °C the heating and cooling steps are omitted.

A total of eight wafers were deposited with varying process parameters, with the relevant conditions summarized in table 1. Between wafer 1 and wafer 3 only the methane flow rate $F[CH_4]$ was varied while keeping all other parameters fixed. Then, the effect of the deposition temperature was varied for wafers 4–7. Finally, wafer 8 was deposited to finalize the deposition

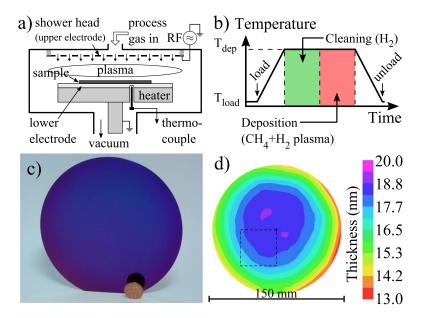


Figure 1. (a) Schematic representation of deposition chamber. (b) Illustration of deposition process. (c) Photograph of thermally oxidized 150 mm silicon wafer after PECVD deposition of NCG. (d) Thickness distribution of NCG film on different 150 mm wafer obtained by ellipsometry.

	T_{dep} (°C)	t _{dep} (min)	$F[H_2]$ (sccm)	<i>F</i> [CH ₄] (sccm)	$NU^{100} \ (\%)$	$d_{\rm mean}^{100}$ (nm)	rms (nm)	L _a (nm)
Wafer 1	900	15	75	60	7	35.5	2.04	1.79 ± 0.01
Wafer 2	900	15	75	30	6	31.5	2.25	1.81 ± 0.01
Wafer 3	900	15	75	10	7	18.0	1.17	2.06 ± 0.03
Wafer 4	900	5	75	60	4	15.7	1.27	1.74 ± 0.02
Wafer 5	850	5	75	60	6	12.4	0.91	1.83 ± 0.02
Wafer 6	800	5	75	60	11	8.9	0.56	2.09 ± 0.02
Wafer 7	750	5	75	60	25	5.2	0.41	2.39 ± 0.02
Wafer 8	900	1	75	60	7	3.0	0.23	2.67 ± 0.06

 Table 1. Deposition conditions and corresponding measurement results.

duration series (together with wafer 1 and 4). A photograph of a 150 mm wafer with deposited NCG is shown in figure 1(c).

The film thickness of the as-deposited substrates is then mapped using a Woollam M-2000 ellipsometer (5 mm offset from edge, 180 data points), with the result for Wafer 4 shown in figure 1(d). The accuracy of the ellipsometer measurement was confirmed to be $\pm 10\%$ through atomic force microscopy measurement of surface steps created by oxygen RIE on reference samples.

Raman spectra of carbon films are well understood and contain a large amount of information [16, 17]. From the ellipsometer measurement shown in figure 1(d) it is apparent that there is a slight non-uniformity across the 150 mm substrates, which is expected to be also prominent in the Raman spectra. We therefore performed Raman mapping using a Renishaw inVia Raman microscope. However, due to the limited travel range of the motorized sample

stage of the Raman spectrometer, $40 \times 40 \text{ mm}^2$ samples were first extracted through dicing (deposited film facing dicing tape) and used for the Raman mapping. The location of these quadratic samples is indicated in figure 1(d) by dashed lines. A total of 64 Raman spectra (5 mm spacing) were acquired on each sample using 532 nm excitation wavelength.

The surface topography was measured in an $1 \times 1 \mu m$ large area close to the center of each of the wafers by contact mode AFM in order to obtain the root mean square (rms) roughness values.

In order to measure the electrical properties of the films, simple two-terminal devices were fabricated. First, optical lithography was used to pattern 60 μ m wide lines with lengths of 80, 280, 500 and 680 μ m by photoresist, followed by an O₂ reactive ion etching step (20 W RF power, 20 mT pressure, 20 sccm O₂ gas flow).

3. Results and discussion

3.1. Ellipsometer

One of the key aspects of the reported deposition process is the uniformity across the whole substrate. Therefore, the non-uniformity NU of the thickness distribution obtained through ellipsometry was calculated by $NU = (\max(d) - \min(d))/(\max(d) + \min(d)) \times 100\%$. First, all 180 data points were considered, resulting in non-uniformity values between 13% (wafer 4) and 92% (wafer 1). The latter value was caused by several extreme data points at the edge of the wafer caused by local contamination. Additionally, the non-uniformity was calculated for the data points within the inner 100 mm diameter circle to compensate for the edge non-uniformity values for the 100 mm equivalent NU^{100} are significantly lower for all wafers with the largest drop for wafer 1 from 92% to 7% (the NU^{100} values are listed in table 1). A certain non-uniformity is common for plasma processes, as the plasma, temperature or gas distribution can suffer from edge effects. The cause in this case was not further investigated. Additionally, the mean thickness was calculated from the ellipsometer results in the 100 mm equivalent area and are shown as d_{mean}^{100} in table 1. There is a significant increase of non-uniformity with decreasing temperature which was not further investigated.

3.2. Raman

A representative Raman spectrum obtained on wafer 2 is shown in figure 2 (blue curve). Curve fitting was used to determine the position and intensity of the individual peaks (red curve). Here, we used Lorentzian curve shapes for all peaks with the exception of the D' peak (1627 cm⁻¹), which was fitted by a Fano line shape [18]. The individual components of the fit model are shown as black curves. The spectrum exhibits distinct D, G and 2D peaks, which are characteristic for defective graphene [17]. Other peaks observed in this spectrum, such as the D', D+D', D+D' and 2D' (not shown) had been reported for carbon films before, and are briefly explained in table 2 alongside the other three peaks.

Since the spectrum exhibits a D peak with an intensity larger than the G and 2D peaks, the films obtained in this work have a significant number of crystal defects in comparison to graphene or graphite. The intensity ratio of the D-peak to G-peak I(D)/I(G) is ≈ 2.7 for the

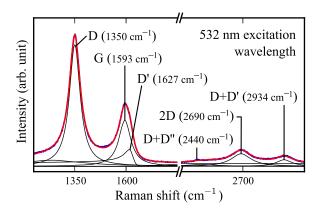


Figure 2. Raman spectrum obtained on 30 nm thick NCG film deposited at 900 °C. The peaks are explained in table 2.

Table 2. List of Raman peaks observed in NCG spectra (such as the one shown in figure 2), together with Raman shift and function used for fitting; LO for Lorentzian and FAN for Fano line-shape.

Peak	Shift	Description		
D	1350	[LO] Intervalley defect (A _{1g} -mode), present only in disordered		
		graphene or edge areas.		
G	1593	[LO] Radial breathing mode of sp2 carbon forms (E_{2g} -mode).		
\mathbf{D}'	1627	[FAN] Intravalley defect [16, 19]		
D+D″	2441	[LO] Intervalley double resonance process; sometimes referred to as G* [16, 20]		
2D	2690	[LO] Sum of double and triple resonances, sometimes referred to as G' [16, 19, 21]		
D+D'	2934	[LO] Defect related band, sometimes referred to as D+G. [19, 21]		
2D'	3248	[LO] (Not visible in figure 2) Overtone of D', sometimes referred to as G". [21, 22]		

shown spectrum. Furthermore, the G-peak location is not significantly affected by the excitation wavelength (confirmed for 532, 633 and 785 nm). These characteristics $(I(D)/I(G) \approx 2, \text{Pos}(G) \approx 1590 \text{ cm}^{-1}$ and Pos(G) unaffected by excitation wavelength) had been described previously and attributed to NCG [23]. Several recent publications reported on carbon films with very similar Raman spectra [13, 14] labeling them NCG or nanographene. Therefore, based on the Raman spectrum of our films and the current naming convention of thin carbon films in the scientific community, we label our films NCG.

The cluster diameter of the NCG can be calculated from the D and G peak intensities as $L_a = (I(G) \cdot C(\lambda))/I(D)$ with $C(\lambda)$ a variable scaling coefficient which depends on the excitation wavelength [24]. L_a for the spectrum shown in figure 2 is thus 1.81 nm, and the values of the remaining wafers (spectra obtained close to center of wafers) are listed in table 1.

As mentioned earlier, we also performed Raman mapping. The obtained spectra were analyzed as explained above, resulting in the maps shown in figures 3(a), (b) and (c) (wafer 3,

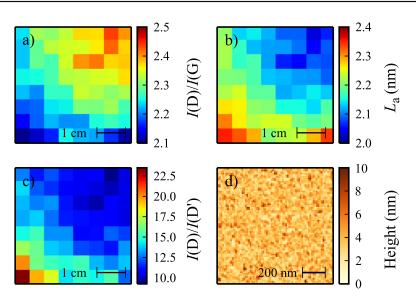


Figure 3. (a)+(c) Raman mapping results of area indicated in figure 1(d). (b) Cluster size derived from I(D)/I(G) map shown in (a). (d) Topography of typical NCG film acquired by AFM.

top right data point at center of wafer). The concentric pattern seen in the thickness distribution (compare figure 1(d)) is also clearly visible in all Raman results.

The I(D)/I(G) ratio map (figure 3(a)) and the derived L_a map (figure 3(b)) show that the cluster diameter is smallest at the center of the wafer, the place of the largest thickness. This trend was observed for all wafers deposited in this work. Furthermore, we did not find any evidence of a correlation between deposition temperature and cluster diameter. The I(D)/I(D') ratio map (figure 3(c)) shows a decrease towards the center of the wafer. A ratio of 10 indicates a mix of vacancy-like and sp3 defects, while a ratio of 13 and above shows a prevalence of sp3 defects [18].

3.3. Surface roughness

The Raman spectra clearly indicate nanocrystalline structure, however, it is also important to know the roughness of the films in relation to the cluster size. Therefore, atomic force microscopy in contact mode was used to measure the topography of $1 \times 1 \mu m$ areas close to the center of the wafers. The result for wafer 3 is shown in figure 3(d), indicating that the surface comprises round particles. The average distance between the particles was manually extracted from the AFM data and is between 26 and 37 nm (~27 nm for wafer 3), much larger compared to the respective cluster diameters. This observation suggests that the observable surface topography is not related to the cluster diameter. The extracted rms roughness values for all of the wafers are listed in table 1. These rms values are between 5 and 8% of the film thickness, independent of process conditions or film thickness.

3.4. Optical

NCG films were also deposited directly onto small quartz glass and sapphire glass samples $(T_{dep} = 900 \text{ °C}, t_{dep} = 15 \text{ min})$. A molybdenum disk was used as handle wafer for the deposition

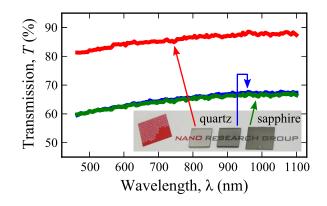


Figure 4. Measured optical transmission of the quartz and sapphire samples shown in the inset. Left: quartz with 5.5 nm NCG, middle: quartz with 15 nm NCG, right: sapphire with 15 nm NCG.

due to its higher thermal conductivity compared to Si. To obtain the film transmittance, substrate absorption and film reflection were measured separately and accounted for. Three samples are shown in figure 4 together with their film transmittance across the visible range. The left quartz glass sample with 5.5 nm thick film ($F[H_2] = 60 \text{ sccm}$, $F[CH_4] = 75 \text{ sccm}$) offers 85% transparency and 13 k Ω sheet resistance at 550 nm wavelength, while the middle quartz glass sample ($F[H_2] = 72 \text{ sccm}$, $F[CH_4] = 90 \text{ sccm}$) with 15 nm film thickness has only 65% transparency but much lower 2.5 k Ω sheet resistance (approximately linear decrease of transparency with NCG thickness). The sapphire sample, deposited with identical conditions as the middle quartz glass sample, has also a 15 nm thick NCG film and 65% transparency, however, a much higher sheet resistance values are considerably higher than the values reported for mono- and bi-layer graphene [2, 25], however the reported deposition technique has not been optimized for optical electrode applications and the deposition does not require a transfer step as is the case for CVD grown graphene. The slight decrease of transparency at lower wavelengths is also observed in graphene obtained by other methods [25, 26].

3.5. Electrical characteristics

In order to measure the electrical characteristics of the newly developed NCG films, test structures were fabricated on a reference sample ($T_{dep} = 900 \,^{\circ}\text{C}$, $t_{dep} = 15 \,^{\circ}\text{min}$, $F[\text{H}_2] = 90 \,^{\circ}\text{cm}$, $F[\text{CH}_4] = 72 \,^{\circ}\text{sccm}$) by using standard thin film fabrication techniques. 60 μ m wide lines were first patterned by optical lithography and oxygen based reactive ion etching (Oxford Instruments RIE80, 20 sccm O₂, 20 mTorr chamber pressure, 20 W RF power), followed by probe pad patterning using a titanium/gold lift-off process. An SEM micrograph of such a fabricated device is shown as inset of figure 5. Consecutively, the I_D/V_D characteristics were measured using an Agilent B1500A semiconductor device parameter analyzer. All devices showed metallic-like I_D/V_D characteristics between 273 and 79 K as shown in the upper left inset of figure 5 for a 80 μ m long device. The increasing resistivity with decreasing temperature is common for semiconducting materials. The device resistances measured at room temperature are plotted against the channel lengths in figure 5. The contact resistance of 0.69 k Ω is obtained by linear fitting, and the sheet resistance $R_{sq} = 9.3 \pm 1.9 \,^{\circ}{k\Omega} \,^{-1}$ is calculated. The charge

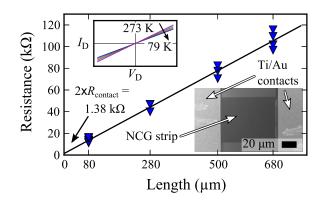


Figure 5. Measured resistance for NCG strips with various lengths. SEM micrograph of NCG strips contacted by Ti/Au lift-off shown as inset lower-right, and the I_D/V_D characteristics of an 80 μ m long device for temperatures between 273 and 79 k are shown as inset upper left.

carrier mobility μ of 2.49 cm² V⁻¹ s⁻¹ was measured using an *Accent HL5500 Hall System*, alongside with the sheet electron concentration of $n_s = 6.73 \times 10^{14} \text{ cm}^{-2}$. This corresponds to a bulk electron concentration of $n = 2.22 \times 10^{20} \text{ cm}^{-3}$.

The sheet resistance and charge carrier mobility are in a very similar range as values reported for NCG elsewhere. The films deposited by Kalita [13] using surface wave plasmaenhanced chemical vapor deposition (PECVD) have sheet resistances of 3–7 k Ω sq⁻¹. Charge carrier mobility and concentration were not reported. The ECR-CVD deposited films at 400° by Medina [27] have sheet resistances in the range of 5–20 k Ω sq⁻¹ without reporting the mobility or concentration of charge carrier. The charge carrier mobility of $\mu = 1 \text{ cm}^2 \text{ V}^{-1}\text{s}^{-1}$, as reported for MBE grown films on sapphire, is slightly lower than the values reported here, however no sheet resistance is provided [8].

4. Conclusion

Metal-free deposition of large area NCG using a parallel plate plasma-enhanced CVD tool was demonstrated. The deposition was carried out at temperatures between 750 and 900°. The high uniformity of the film thickness across the whole 150 mm substrates was demonstrated, and the domain size of these graphene based films were studied in detail by Raman spectroscopy. The cluster diameter L_a increases for thinner films, suggesting that the initially deposited films have a higher crystallinity. No correlation between deposition temperature and cluster diameter was observed.

The deposition of NCG on insulating, transparent substrates such as quartz glass or sapphire glass was demonstrated. The metallic electrical properties and the demonstrated patterning by dry oxygen etching make these films suitable for transparent electrode applications or wafer-scale electronic applications.

Acknowledgments

The corresponding author would like to acknowledge the supported by the University of Southampton, School of Electronics and Computer Science Scholarship, and the Southampton Nanofabrication Centre.

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