

1 Diagenetic Mg-calcite overgrowths on foraminiferal tests in the  
2 vicinity of methane seeps

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21

22 **ABSTRACT**

23 Methane is a potent greenhouse gas and some episodes of past global warming appear to coincide  
24 with its massive release from seafloor sediments as suggested by carbon isotope records of

25 foraminifera. Here, we present structural, geochemical, and stable carbon isotope data from single  
26 foraminiferal calcite tests and authigenic Mg-calcite overgrowths in a sediment core recovered from  
27 an area of active methane seepage in western Svalbard at ca. 340 m water depth. The foraminifera are  
28 from intervals in the core where conventional bulk foraminiferal  $\delta^{13}\text{C}$  values are as low as -11.3 ‰.  
29 Mg/Ca analyses of the foraminiferal tests reveal that even tests for which there is no morphological  
30 evidence for secondary authigenic carbonate can contain Mg-rich interlayers with Mg/Ca up to 220  
31 mmol/mol. Transmission electron microscopy (TEM) of the contact point between the biogenic  
32 calcite and authigenic Mg-calcite layers shows that the two phases are structurally indistinguishable  
33 and they have the same crystallographic orientation. Secondary ion mass spectrometry (SIMS)  
34 analyses reveal that the Mg-rich layers are strongly depleted in  $^{13}\text{C}$  ( $\delta^{13}\text{C}$  as low as -34.1 ‰). These  
35 very low  $\delta^{13}\text{C}$  values indicate that the authigenic Mg-calcite precipitated from pore waters containing  
36 methane-derived dissolved inorganic carbon at the depth of the sulfate-methane transition zone  
37 (SMTZ). As the depth of the SMTZ can be located tens of metres below the sediment-seawater  
38 interface, interpretation of low foraminiferal  $\delta^{13}\text{C}$  values in ancient sediments in terms of the history  
39 of methane seepage at the seafloor must be undertaken with care.

40

## 41 **1. Introduction**

42 Since the discovery of methane emissions at the seabed, investigators have explored the potential of  
43 using tests of modern and fossil calcitic foraminifera as archives of seepage history. Remarkable  
44 carbon isotope excursions (CIE), characterized by very low  $\delta^{13}\text{C}$  values in both benthic and  
45 planktonic foraminiferal calcite, have been interpreted to reflect widespread release of isotopically-  
46 light carbon from dissociating marine methane hydrates during the Paleocene-Eocene Thermal  
47 Maximum (PETM) at ~54.95 Ma (e.g., Dickens et al. 1995, Zachos et al. 2005). Massive methane  
48 releases have been also used to explain CIEs at other time intervals related to abrupt global warming  
49 (Hesselbo et al., 2000, Jiang et al. 2003). However, it remains unclear if the very low  $\delta^{13}\text{C}$  values of

50 foraminifera reflect  $\delta^{13}\text{C}$  values of dissolved inorganic carbon (DIC) in the primary biomineralization  
51 environment or whether they are associated with secondary overgrowths of authigenic carbonate.  
52 Very low  $\delta^{13}\text{C}$  values in foraminiferal calcite are widely considered to reflect seepage and  
53 incorporation of methane-derived,  $^{13}\text{C}$ -depleted DIC at the time when foraminifera calcify their tests  
54 (Wefer et al., 1994; Dickens et al., 1995; Kennett et al., 2000; Hill et al., 2004). Alternatively, low  
55 foraminiferal  $\delta^{13}\text{C}$  values may reflect precipitation of secondary overgrowths of methane-derived  
56 authigenic carbonates forming at the sulfate-methane transition zone (SMTZ) within subsurface  
57 sediments (Torres et al., 2010; Millo et al., 2005; Martin et al., 2010; Panieri et al. 2014, 2015;  
58 Consolaro et al., 2014). In this scenario, the original test surface merely acts as a template for  
59 authigenic carbonate precipitation. At the depth of the SMTZ, upward migrating methane and  
60 downward diffusing sulfate are consumed, by anaerobic oxidation of methane (AOM), according to  
61 the reaction (Boetius, 2000):



62  
63 The bicarbonate ( $\text{HCO}_3^-$ ) ions produced by AOM react with calcium and magnesium ions present in  
64 sediment pore waters to form Ca-Mg carbonates (Berner, 1970; Jørgensen et al., 2004; Ussler III and  
65 Paull, 2008) which are typically strongly  $^{13}\text{C}$  depleted.  
66 Methane-derived authigenic calcite occurring as sediment cements or as overgrowths on foraminifera  
67 tests are commonly rich in Mg with Mg/Ca often  $>20$  mmol/mol. Mg is effectively complexed by  
68 sulfate in normal seawater which inhibits Mg incorporation into calcite, but if sulfate concentrations  
69 are very low, e.g. in pore waters affected by AOM, Mg is not complexed and can be more readily  
70 incorporated in calcite (Walter, 1986; Wright and Wacey, 2004). Hence Mg enrichment observed in  
71 foraminiferal tests can be indicative of the presence of diagenetic phases.  
72 Here we present the results of analyses of individual foraminifera tests in a sediment core from an  
73 area of methane seepage west of Prins Karls Forland (ca. 340 m water depth), offshore western

74 Svalbard (Fig. 1). Scanning Electron Microscope Energy Dispersive X-ray Spectroscopy (SEM-EDS),  
75 Electron Probe Micro Analysis (EPMA), Transmission Electron Microscopy (TEM), and Secondary  
76 Ion Mass Spectrometry (SIMS) studies reveal that there is significant variation in the geochemical  
77 and carbon isotope compositions *within* individual tests, and highlight the complexity of  
78 distinguishing primary vs. secondary signals. We demonstrate that diagenetic overprints on  
79 foraminifera can be very extensive, but not always evident using traditional methods such as optical  
80 microscopy and SEM.

81

## 82 **2. Material selection**

83 Foraminiferal tests were picked from three intervals (core depths 102, 132 and 210 cm below  
84 seafloor (bsf); Table 1) within a 224 cm long sediment core PC06 (78:36.66 N; 9:25.53E; 374 m  
85 water depth) collected during RRS James Clark Ross cruise 253 in August 2011 (Fig. 1). The  
86 present-day depth of the SMTZ in this core is 184 cm bsf, and all sediments from between 100 and  
87 220 cm bsf contain foraminifera with distinctly negative bulk  $\delta^{13}\text{C}$  values (Panieri et al., 2016),  
88 outside of the range of normal marine values (ca. -1 to 1 ‰). From each of the three selected  
89 intervals calcitic tests of two benthic species, *Nonionella labradorica* (Dawson, 1860) and *Melonis*  
90 *barleeanus* (Williamson, 1858), were chosen for detailed study to assess whether the low  $\delta^{13}\text{C}$  values  
91 (Table 1) are associated with primary biogenic calcite, or secondary overgrowths.

92

## 93 **3. Strategy**

94 Sediment samples were washed through a 63  $\mu\text{m}$  sieve, and oven dried at 40 °C. Tests of calcitic  
95 benthic foraminifera, including both well preserved and altered (with obvious dissolution features  
96 and overgrowths) specimens, were picked from the larger size fraction, >100  $\mu\text{m}$ , under a light  
97 microscope. The selected specimens (F1-F5) are listed in Table 2, together with results of analysis of  
98 the carbon isotope composition of bulk foraminifera (Panieri et al., 2016).

### 100 **3.1 Scanning Electron Microscope equipped with Energy Dispersive X-ray Spectroscopy (SEM-** 101 **EDS)**

102 Polished cross-sections of a selection of tests were made by (i) placing the hand-picked tests on  
103 adhesive tape in a circular 25 mm diameter mold, (ii) mounting using Struers Epofix, and (iii)  
104 polishing to expose a cross-section through the interior of the tests. SEM-EDS examination of these  
105 polished sections was carried out at The Arctic University of Norway using a SEM Hitachi Tabletop  
106 Microscope TM-3000 equipped with Bruker Quantax 70 EDS Detector, and on a Zeiss Merlin  
107 Compact VP with the GEMINI I electron column, and a Zeiss Sigma.

108

### 109 **3.2 Electron Probe Micro Analysis (EPMA)**

110 Quantitative Mg and Ca spot analyses and semi-quantitative elemental mapping were carried out  
111 using a JEOL JXA-8500F electron microprobe (field-emission electron gun) equipped with a high  
112 resolution wavelength X-ray dispersive spectrometer (WDS) at the Department of Materials Science  
113 and Engineering in Trondheim. Diopside  $\text{MgCaSi}_2\text{O}_6$  (Mg = 11.23 wt%) and Calcite  $\text{CaCO}_3$  (Ca =  
114 39.98 wt%) were used as standards. The mapping was performed using a 0.1  $\mu\text{m}$  step size and beam  
115 scan with an image resolution of 512 x 512 pixels, 10 kV accelerating voltage and an electron beam  
116 current of 20nA with a minimum probe size of 50nm. Under such microprobe conditions the electron  
117 diffusion region of X-ray generation is in the range of 1 $\mu\text{m}$  (X-ray spatial resolution during mapping).  
118 Optimum quantitative counting strategies from Mg and Ca were obtained with a total counting time  
119 of 60 seconds in both peak and background position, from an excitation cross-section of approx. 2 $\mu\text{m}$ .  
120 The reproducibility of Mg and Ca determined from repeated analysis of standards was <1.5 %, and  
121 the detection limit was ~100 ppm. Raw data were reduced to weight percentages using the ZAF  
122 algorithm. Performing WDS on carbonate material can result in beam damage to the specimen and  
123 lower counts compared to materials such as silicates. However, the ratio of Mg to Ca is not severely

124 affected by these effects. Topographic variations and impurities within the foraminifera tests such as  
125 clay particles and organic matter can influence the quantification of Mg and Ca in carbonate by  
126 EPMA. To minimize these effects, the specimens were carefully examined by SEM- BSD/EDS, and  
127 areas that appeared to be impurity-free were selected for EPMA.

128

### 129 **3.3 Transmission Electron Microscopy (TEM)**

130 Focused ion beam (FIB) milling was used as a sample preparation method for measurements  
131 by transmission electron microscopy (TEM). FIB was conducted on a C-coated polished mount  
132 to extract two foils from a single foraminifera test using a FEI FIB2000-TEM equipped with a  
133 Ga-ion source at GeoForschungsZentrum Potsdam (GFZ; Germany). FIB milling occurred at 30  
134 keV. Final polishing of the surface with the ion beam almost parallel the surface occurred at a  
135 beam current 11 pA. This leads to minimal damage to the outer layer (10 - 15 nm) and, as the  
136 thickness of the foil is ~150 nm, the crystalline state of the carbonate that is documented in  
137 electron diffraction and lattice fringes, is unaffected. The experience of having cut more than  
138 5000 foils from different materials including many carbonates guarantees that a complete  
139 amorphisation of the carbonates by Ga ion sputtering is excluded. The FIB-foil dimension was  
140 15 x 10 x 0.15  $\mu\text{m}$  on average. Milled FIB foils were lifted out of the FIB sputtering site and  
141 subsequently placed on standard Cu TEM grids covered with perforated amorphous carbon  
142 film (Wirth, 2009). The TEM at GFZ used for this study was a FEI TecnaiG2 F20 X-TWIN  
143 equipped with a Gatan Imaging Filter GIF (Gatan Tridiem, Gatan, CA, USA) EDAX X-ray analyzer  
144 and a Fishione high-angle annular dark-field detector (HAADF). The electron source of the TEM  
145 was a field emission gun emitter, operated at an acceleration voltage of 200 kV. EDX analyses  
146 were performed in scanning transmission mode (STEM) by scanning the electron beam within  
147 a preselected area. This technique minimizes mass-loss caused by electron sputtering.

148

149 **3.4 Carbon isotope analyses by secondary ion mass spectrometry (SIMS)**

150 High spatial resolution carbon isotope analyses (c. 15  $\mu\text{m}$  analytical spots) were performed on two  
151 mounted foraminifera tests using a CAMECA IMS 1280 large geometry Secondary Ion Mass  
152 Spectrometry (SIMS) at the NordSIM facility in Stockholm, Sweden. A critically focused  $\text{Cs}^+$   
153 primary beam with 20 keV impact was used to sputter the sample, and a low-energy electron-  
154 flooding gun was used for charge compensation. The primary beam current was ca. 3 nA. Each  
155 analysis consisted of an initial pre-sputter over a rastered 20  $\mu\text{m}$  area to remove the gold coating,  
156 followed by centering of the secondary beam in the field aperture (field of view on the sample of 30  
157  $\mu\text{m}$  with 90x magnification transmission ion optics). Secondary ion signals were measured in  
158 multicollector mode using a Faraday detector for  $^{12}\text{C}^-$  (ca.  $2 \times 10^7$  cps), and an ion counting electron  
159 multiplier for  $^{13}\text{C}^-$  at a mass resolution of ca. 4000, sufficient to resolve  $^{13}\text{C}$  from  $^{12}\text{C}^1\text{H}$ . A within run  
160 correction was made for electron multiplier gain drift using the pulse height analysis curve. The  
161 secondary magnet field was locked at high stability using a nuclear magnetic resonance field sensor  
162 operating in regulation mode. All pre-sputter, beam centering and data acquisition steps were  
163 automated in the run definition. Typical internal precision obtained for individual run  $^{13}\text{C}/^{12}\text{C}$  ratios  
164 from twenty-four cycles of 4-second integrations was ca. 0.2 ‰ (SE). Sample chamber vacuum was  
165 maintained at  $<2 \times 10^{-8}$  mbar.

166 Fully automated sequences comprised two measurements of the reference carbonate, Brown Yule  
167 Marble calcite (BYM,  $\delta^{13}\text{C}$  VPDB = -2.28 ‰, kindly provided by J. Craven, University of  
168 Edinburgh), bracketing six measurements of unknown targets of the foraminiferal calcite and  
169 authigenic Mg-calcite. Calcite and Mg-calcite can be expected to display different instrumental mass  
170 fractionation (IMF) during SIMS analysis, hence using the same calcite reference material for  
171 foraminiferal calcite and authigenic Mg-calcite may not be fully appropriate. An IMF bias between  
172 calcite and various Mg-containing carbonates has been demonstrated for oxygen isotopes (Rollion-  
173 Bard & Marin-Carbonne, 2011) and measurements of carbon isotopes in dolomite and calcite

174 reference materials at NordSIMS have yielded a bias of -0.33‰ per wt. % of MgO (MJW  
175 unpublished data), similar to the reported bias for oxygen. This implies that Mg-calcite with Mg/Ca  
176 of 220 mmol/mol normalized against calcite will be approximately 2‰ too light. Assuming a similar  
177 linearity in the IMF bias for carbon as has been demonstrated for oxygen, the effect will be  
178 proportionally smaller at lower Mg abundance. The putative instrumental mass bias effect thus has a  
179 relatively minor influence on the acquired isotope ratios and, given the much greater magnitude of  
180 the observed differences, does not affect our interpretations. The regularly interspersed BYM  
181 measurements were used to correct measured isotope ratios for any drift during the analytical session  
182 (typically a cubic polynomial fit was made to the standard analyses) and for instrumental mass  
183 fractionation (IMF). External precision on  $\delta^{13}\text{C}$  based on multiple analyses of the BYM standard was  
184  $<0.4$  ‰ (SD). Carbon isotope compositions are reported in conventional  $\delta$  notation relative to Vienna  
185 PeeDee Belemnite (VPDB).

186

## 187 **4. Results and discussion**

### 188 **4.1 SEM evidence for the presence of [cryptic] authigenic carbonate**

189 SEM analyses reveal the presence of abundant diagenetic pyrite and authigenic carbonate on the  
190 foraminiferal test walls (Fig. 2). Co-precipitation of these phases can be expected at the depth of the  
191 SMTZ, because AOM results in the production of bicarbonate and hydrogen sulfide, which induces  
192 precipitation of authigenic carbonate minerals and pyrite (Peckmann et al., 2001). These authigenic  
193 carbonate overgrowths can be ‘cryptic’, i.e. they are not visible under optical microscope. For  
194 instance, relatively well preserved tests with typical granular microstructure and wall pores and no  
195 obvious evidence for authigenic carbonate overgrowths, contain clusters of framboidal pyrite inside  
196 the test chambers (Fig. 2a-b). The extent of authigenic carbonate overgrowths can be highly variable  
197 both between and within individual tests. In sediments from 132 cm bsf, tests without obvious  
198 overgrowths (Fig 2a) and tests with thick overgrowths (Fig. 2h) co-occur. In some tests there is a



199 gradient from thick overgrowths on the test walls of the outer chambers to thin overgrowths on the  
200 test walls of the inner chambers in the umbilical region (Fig. 2c). The pores appear to act as a nuclei  
201 for precipitation of authigenic carbonate, as suggested by linings of authigenic crystals around the  
202 pores in the inner chambers (Fig. 2d). The outer chambers are variably covered with the overgrowths;  
203 authigenic carbonate precipitation takes place in spaces between the pores (Fig. 2e) and eventually  
204 the whole inner side of the test can become covered with microcrystalline overgrowths completely  
205 masking the original surface features of calcite tests (Fig. 2f-h).

206

#### 207 **4.2 Mg/Ca in biogenic calcite and authigenic carbonate overgrowths**

208 Morphological evidence for authigenic carbonate overgrowths on the tests from core PC06 is  
209 supported by differences in the Mg/Ca ratio of the original test calcite and the overgrowths. EBS  
210 images and EPMA Mg/Ca mapping show the presence of low EBS and relatively Mg-rich  
211 overgrowths approximately 20  $\mu\text{m}$  thick, on both the inner and outer foraminiferal test walls (Fig. 3).  
212 The Mg/Ca ratio of the overgrowths is as high as 220 mmol/mol, while the Mg/Ca ratio of the  
213 original test calcite is  $< 10$  mmol/mol. The occurrence of Mg-rich authigenic phases is not however  
214 limited to the overgrowths on test walls; these authigenic phases can also fill the original wall  
215 porosity and pores within it.

216 Some specimens are compositionally laminated with distinct Mg-rich laminae (Fig. 3). The initial  
217 lamination was likely formed during ontogenesis as calcite laminae are added to the pre-existing test  
218 every time a new chamber is built. Calcite crystallites in such laminae have a radial appearance with  
219 C-axes lying perpendicular to the lamination, facilitating high porosity (Erez, 2003). Organic material  
220 deposited between the calcite laminae can also become porous upon mineralization of organic matter  
221 resulting in porous laminae. These pores are likely susceptible to filling by authigenic Mg-rich calcite  
222 during diagenesis. The result is a systematic banding with alternating high and low Mg/Ca.

223 Quantitative EPMA point analyses (ca. 2  $\mu\text{m}$  diameter spot) show that Mg/Ca in Mg-rich zones  
224 ranges from 10-60 mmol/mol in thin interlayers and in-filled pores while thicker overgrowths can  
225 have Mg/Ca up to 220 mmol/mol (Table 3, Fig. 4). Given that the thin Mg-rich layers and pores are  
226 commonly smaller than the analytical spot size of the EPMA, the measured Mg/Ca ratios of the  
227 authigenic carbonates are likely lower than the actual Mg/Ca ratios due to contributions from  
228 enclosing low-Mg phases. The Mg/Ca ratio of foraminiferal calcite is a widely accepted and applied  
229 proxy for ocean temperature (Bohaty et al., 2012; Elderfield and Ganssen, 2000; Garidel-Thoron et  
230 al., 2005; Lea et al., 2000; Nürnberg et al., 2000), and variable Mg/Ca with distinct banding or  
231 apparently random variations in calcite tests can be produced by temperature variations during  
232 foraminiferal growth (Sadekov et al., 2005; Branson et al., 2013). However, these variations are  
233 typically in the range 2-10 mmol/mol. Given that Mg/Ca ratios in the foraminifera from core PC06  
234 are typically far higher than would be expected due to temperature effects or other environmental  
235 factors (Toyofuku et al., 2000; Toler et al., 2001) it appears likely that the Mg-rich interlayers, in-  
236 filled pores and overgrowths within the foraminiferal tests from core PC06 represent authigenic  
237 precipitates. Mg abundance therefore serves as an effective tracer of the primary foraminiferal test  
238 calcite vs secondary authigenic Mg-calcite.

239

#### 240 **4.3 Contact relationship between foraminiferal calcite and authigenic carbonate overgrowths**

241 The TEM investigation of the contacts between the original foraminiferal calcite and the authigenic  
242 Mg-calcite overgrowths was undertaken using two FIB foils milled from F1 (Fig. 5). EDX analyses  
243 confirm the compositional contrast between the calcite test and the Mg-calcite overgrowth (Fig. 5b,  
244 e-f). In addition, a sharp contact between the test and the overgrowth can be identified on high-angle  
245 annular dark-field (HAADF) images due to changes in contrast caused by differences in chemical  
246 composition (Fig. 5b-d).

247 Selected area electron diffractograms (SAEDs) obtained from the contact zones covering both  
248 foraminiferal tests and overgrowths (Fig. 5g-h) show that the mineral phases on both sides of the  
249 contact are structurally identical, and the crystallographic orientation is maintained through the  
250 contact. High-resolution transmission electron microscope (HRTEM) images also show that the  
251 crystal lattice fringes are identical through the contact (Fig. 5i).

252 The TEM data thus indicate that foraminiferal calcite serves as an ideal nucleation template for the  
253 diagenetic Mg-calcite which, despite its elevated Mg abundance, has identical structural  
254 characteristics to the foraminiferal calcite. Thus the templating capabilities of foraminifera may  
255 therefore affect the preservation of foraminifera even in environments where only minor carbonate  
256 authigenesis occurs. Although there is a wide range of carbonates in marine sediments, and several  
257 crystallographic faces are exposed, the strength of bonding at the interface is clearly strongly  
258 dependent on the structure and chemistry of the substrate surface. Our investigation demonstrates that  
259 foraminiferal calcite and authigenic Mg-calcite can have identical lattice fringes, such that authigenic  
260 Mg-calcite precipitates preferentially on foraminifera tests.

261

#### 262 **4.4 Carbon isotopic composition of primary biogenic calcite vs secondary authigenic Mg-calcite**

263 Carbon isotope analyses on different parts of an individual test were performed on foraminifera F5  
264 (Fig. 2).  $\delta^{13}\text{C}$  values ranged from -3.9 ‰ to -34.1 ‰. Isotopically lightest carbon was found within  
265 the relatively thick Mg-calcite overgrowths, while the heaviest carbon was found within the  
266 foraminiferal calcite walls where there is only a very small amount of authigenic carbonate in-filling  
267 the test pores. It is likely that the lowest  $\delta^{13}\text{C}$  values obtained from the overgrowths represent the  
268 isotopic composition of pore water DIC from which authigenic precipitates were derived. The  
269 heaviest  $\delta^{13}\text{C}$  values, obtained from the foraminifera wall (-3.9 ‰), represent the carbon isotope  
270 composition closest to the original composition of the foraminiferal calcite. Given the 15 $\mu\text{m}$  spot size  
271 of SIMS analyses, a minor contribution from authigenic Mg-calcite in small pores (with  $\delta^{13}\text{C}$  of ca. -

272 34 ‰) within foraminifera walls can be expected. This contribution from in-filled pores has likely  
273 caused a small shift towards more negative  $\delta^{13}\text{C}$  values from the original foraminiferal calcite  
274 isotopic composition, which is not known, but is expected to be within normal marine values (ca. -1  
275 to 1 ‰). The range of  $\delta^{13}\text{C}$  values in F5 (Fig.2) is mostly due to sputtering of different proportions of  
276 the original biogenic calcite and the authigenic Mg-calcite overgrowth. The foraminiferal specimen  
277 F1, which was visually relatively well preserved, also yields relatively low  $\delta^{13}\text{C}$  values of -7.7 and -  
278 11.3 ‰. This test contains abundant framboidal pyrite within its chambers, indicative of diagenetic  
279 carbon and sulfur cycling, yet there is no obvious morphologic evidence for the presence of  
280 authigenic carbonates. However, Mg/Ca mapping by EPMA reveals the presence of several Mg-rich  
281 interlayers (Fig. 4; Table 3), suggesting that cryptic authigenic Mg-calcite occurs in this test that is  
282 difficult to detect morphologically. Hence the negative  $\delta^{13}\text{C}$  values of F1 also reflect the influence of  
283 methane-derived authigenic Mg-calcite on this specimen.

284

## 285 **5. Implications**

286 Precipitation of authigenic carbonates is expected to occur within sediments at the depth of the  
287 SMTZ at sites of methane seepage. The SMTZ can be located close to the sediment-water interface if  
288 the methane flux from depth is high, or deeper (a few cm to several tens of meters) below the  
289 interface if the methane flux is low. Thus if the methane flux is high, authigenic carbonate  
290 precipitation will be broadly coeval with the depositional age of foraminifera, and under these  
291 conditions foraminifera with authigenic overgrowths can serve as tracers of methane emission  
292 event(s) at the seafloor. On the other hand, precipitation of authigenic carbonates under low methane  
293 flux conditions can occur tens to thousands of years after sediment deposition and is likely to be  
294 decoupled from methane emissions at the seafloor. Predominance of aragonite typically reflects  
295 authigenic precipitation under high methane flux close to the sediment-water interface while Mg-  
296 calcite is associated with lower fluxes and deeper precipitation (Aloisi et al., 2000). Trace element

297 characteristics of authigenic carbonates can also be used for deciphering between precipitation under  
298 relatively open, sea water influenced conditions close to the sediment-water interface and  
299 precipitation in subsurface sediments isolated from the overlying water column (Crémière et al.,  
300 2016a). Precipitation ages of authigenic carbonates can be determined by U-Th dating (Teichert et al.,  
301 2003; Bayon et al., 2009; Crémière et al., 2016b), although this has not yet been attempted on  
302 authigenic overgrowths on foraminifera.

303 Our analyses reveal the presence of authigenic carbonate overgrowths with low  $\delta^{13}\text{C}$  on foraminiferal  
304 tests recovered from within sediments located both above (102 and 132 cm bsf) and below (210 cm  
305 bsf) the present-day depth of the SMTZ (184 cm bsf) in core PC06. This indicates that the depth of  
306 the SMTZ has varied over time, presumably in response to variations in methane fluxes or lateral  
307 shifts in conduit positions (e.g. Sauer et al., 2015), and the overgrowths might represent a  
308 precipitation “event” within the sediments in response to a fluctuating SMTZ. However, with our  
309 available data, we are unable to confirm whether any of the overgrowths were deposited close to the  
310 seafloor (although the presence of Mg-calcite points to precipitation below the sediment-seawater  
311 interface), and without reliable age constraints it is not possible to correlate possible methane  
312 emission and authigenic carbonate precipitation events with changes in climate parameters. For  
313 example, the authigenic overgrowths that we observe at ~200 cm bsf could have precipitated any  
314 time between the deposition of this sediment interval (> 20 000 ka; Panieri et al., 2016) and the  
315 present day.

316 Our findings demonstrate that foraminiferal tests incorporate methane-derived, low  $\delta^{13}\text{C}$  authigenic  
317 Mg-calcite that occurs as morphologically distinct overgrowths and as cryptic pore in-fill. The  
318 ubiquitous presence of such authigenic precipitates on foraminifera in sediment records at sites of  
319 methane seepage complicates the usage of foraminifera as archives of methane emissions at the  
320 seafloor and as tracers of methane-derived DIC in the biomineralization environment because carbon  
321 isotope signatures can be modified by incorporation of light carbon at the SMTZ below the sediment-

322 water interface. Chemical cleaning protocols to remove adhering clays, organic material and Fe-Mn  
323 oxyhydroxide phases from foraminifera have been routinely applied in trace element studies (Boyle  
324 and Keigwin, 1985; Rosenthal et al., 1999; Martin and Lea, 2002) and a weak acid leaching protocol  
325 has also been designed to chemically remove diagenetic carbonates attached to the foraminifera shell  
326 (Pena et al., 2008; Panieri et al., 2012; 2014; Consolaro et al., 2015). Isotope data obtained on such  
327 chemically cleaned foraminifera from methane seep sites can show very low  $\delta^{13}\text{C}$  values (as low as -  
328 17.4 ‰ at Vestnesa Ridge) that have been interpreted to reflect the presence of methane-derived DIC  
329 in the biomineralization environment (Panieri et al., 2014). Our observations of the tight intergrown  
330 nature of biogenic calcite and authigenic Mg-calcite however draws into question the effectiveness of  
331 these cleaning procedures removing the authigenic carbonate precipitates including Mg-calcite. It is  
332 likely that authigenic carbonates, particularly the Mg-rich calcite that is notoriously difficult to  
333 remove, remain after cleaning by weak acid leaching and cause anomalous secondary  $^{13}\text{C}$ -depletions  
334 and elevated Mg/Ca in chemically cleaned foraminifera. Moreover, these authigenic precipitates have  
335 high Mg/Ca that precludes reconstruction of past seawater temperature and thus assessment of the  
336 linkages between seafloor methane emissions and climate change. We suggest that high-resolution  
337 petrographic-elemental observations are therefore essential to ensure the authenticity of putative  
338 biogenic calcite at methane seepage sites.

339 Our work reveals that even foraminiferal tests that appear well-preserved under optical microscope  
340 and SEM may be diagenetically altered as a result of precipitation of authigenic phases at the depth  
341 of the SMTZ at sites of methane seepage. This potentially compromises the value of foraminiferal  
342  $\delta^{13}\text{C}$  and Mg/Ca data as records of past environmental conditions at these locations influenced by  
343 methane seepage both now and in the past. We show that fine scale analyses at sub- $\mu\text{m}$  resolution  
344 allow the unaltered biogenic calcite to be targeted and thus provide the framework to overcome the  
345 challenges identified for the use of foraminifera as paleoceanographic proxies for past episodes of  
346 seafloor methane seepage and climate reconstruction.

347

## 348 **6. Conclusions**

349 Our work on individual tests of benthic foraminifera in a sediment core from an area of methane  
350 seepage offshore western Svalbard highlights the complexity of distinguishing between primary and  
351 secondary signals. We demonstrate that diagenetic overprints on foraminifera can be extensive, but  
352 not always evident using traditional methods such as optical microscopy. Moreover we show, for the  
353 first time, that foraminiferal calcite and authigenic Mg-calcite overgrowths have identical crystal  
354 lattice fringes, meaning they are structurally identical, such that foraminifera serve as preferred  
355 nucleation templates for authigenic Mg-calcite.

356 High spatial resolution analyses utilizing advanced technologies show that the authigenic Mg-calcite  
357 has much lower  $\delta^{13}\text{C}$  and higher Mg/Ca than primary biogenic calcite. The low  $\delta^{13}\text{C}$  values indicate  
358 the methane-derived origin of authigenic carbonate precipitating at the depth of the SMTZ. If the  
359 SMTZ is located close to the seafloor, then authigenic Mg-calcite and biogenic calcite are broadly  
360 coeval and provide a record of methane seepage at the seafloor. However, we caution that the depth  
361 of the SMTZ can be located up to several meters below the sediment-seawater interface; in these  
362 circumstances the authigenic Mg-calcite is younger than the biogenic calcite and low  $\delta^{13}\text{C}$  values  
363 cannot be interpreted in terms of methane emissions at the seafloor.

364

365

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371 The Arctic University of Norway in Tromsø for technical support.

372

373 **FIGURE CAPTIONS**

374 Figure 1. Location of core PC06 offshore western Svalbard in the Arctic Ocean.

375

376 Figure 2. SEM images of foraminifera F1-F5 (Table 1) from core PC06 that show different styles of  
377 diagenetic alteration. In the images of foraminifera F1 and F5, red circles and associated numbers  
378 show spots of SIMS analyses and obtained  $\delta^{13}\text{C}$  values (‰ VPDB; listed in Table 2). Red squares on  
379 F1, F2, F4 and F5 indicate the areas where Mg mapping and Mg/Ca point analyses were undertaken  
380 (Fig. 3a, b, c, d respectively, and Table 3).

381 (a) Well preserved tests of *N. labradorica* (F1, from 132 cm bsf) exhibiting typical granular  
382 microstructure and wall pores without obvious evidence of authigenic carbonate precipitates; infilling  
383 of pyrite as framboids and small aggregates (ca. 3  $\mu\text{m}$ ) occurs inside the chambers (whitish areas).

384 (b) *N. labradorica* (F2, from 132 cm bsf); with relatively well preserved tests exhibiting open wall  
385 pores and the systematic banding of the wall tests. White blebs are framboidal pyrite aggregates.

386 (c) *M. barleeanus* (F3, from 102 cm bsf). (d) Inner side of the test wall showing the foraminifera wall  
387 pores rimmed by triangular crystals of Mg-calcite. (e) Carbonate precipitation has occurred on the  
388 whole inner side of the test and the microcrystalline overgrowths completely mask the original  
389 surface features of the test. The wall pores are apparently acting as nuclei for precipitation of  
390 authigenic carbonate.

391 (f) *N. labradorica* (F4, from 210 cm bsf). (g) The whole inner side of the test is completely covered  
392 with microcrystalline Mg-calcite overgrowths, masking the original surface features of the test. The  
393 tops of the crystals appear as triangular prisms.

394 (h) *N. labradorica* (F5, from 132 cm bsf). The original part of the test (light grey areas in the  
395 backscatter image) has been covered, both on the outside and on the inside, by elongated  
396 rhombohedra crystals with sharp edges (i). SIMS data (Table 2) indicate that the secondary Mg-



397 calcite overgrowths have low  $\delta^{13}\text{C}$  values.

398

399 Figure 3. Electron backscattered images and corresponding Mg distribution maps of cross sections of  
400 *N. labradorica* and *M. barleeanus*, for images of entire foraminifera tests see Fig. 2. Red circles show  
401 locations of quantitative Mg/Ca measurements (beam diameter: ca. 2  $\mu\text{m}$ ) reported in Table 3.

402

403 Figure 4. Electron backscattered images and relative Mg and Ca distribution maps (middle column  
404 images where blue colors indicate low concentrations, green-red-pink colors indicate higher  
405 concentrations) from cross sections of *N. labradorica* and *M. barleeanus*, for images of entire  
406 foraminifera tests see Fig. 2. Triple images are the same place in the same individual. Red circles on  
407 the images in the first column show locations of quantitative Mg/Ca measurements (beam diameter:  
408 ca. 2  $\mu\text{m}$ ) reported in Table 3 and Fig. 3. The images in the last column indicate Ca maps.

409

410 Figure 5. (a) Secondary electron microscope – backscatter electron (SEM-BSE) image of a fragment  
411 of foraminifer F5 showing two locations (orange rectangles) where focused ion beam (FIB) foils,  
412 shown on panels b and d, were prepared. (b) High-angle annular dark-field (HAADF) image of entire  
413 FIB foil milled from the inner side of the foraminifera test wall. Dotted line marks the contact  
414 between foraminiferal calcite and the authigenic Mg-calcite overgrowth. Rectangles e and f give the  
415 positions of EDAX analyses shown on panels e and f. (c) HAADF close up of the contact (dotted  
416 line) between the foraminifera and authigenic overgrowth. Orange circle h gives the position of the  
417 selected area electron diffractogram (SAED) shown on panel h, and rectangle j gives the position of  
418 high-resolution transmission electron microscope (HRTEM) image shown on panel j. (d) HAADF  
419 image of entire FIB foil milled from the divider between the foraminifera chambers. Dotted line  
420 marks the contact between foraminiferal calcite and the authigenic Mg-calcite overgrowth occurring  
421 at both sides. Orange circle g gives the position of the SAED shown on panel g. (e-f) EDAX spectra

422 obtained from foil b showing elevated Mg content in the authigenic overgrowth compared to  
423 foraminiferal calcite where Mg is not detectable. (g-h) indexed SAED patterns from contact zones of  
424 both foils showing a consistent diffraction pattern through contacts indicating structurally identical  
425 phases with matching crystallographic orientations on both sides of contact. (i) HRTEM image from  
426 the contact zone demonstrating identical lattice fringes across the contact.

427

## 428 **TABLE CAPTIONS**

429 Table 1. Bulk stable carbon isotope values of foraminifera from core PC06. Data are from Panieri et  
430 al. (2016).

431

432 Table 2. Stable carbon isotope values of single foraminiferal tests (*N. labradorica*, specimen 1 and 2)  
433 analyzed with secondary ion mass spectrometry (SIMS). Values are expressed in ‰ VPDB.  
434 Locations of measurement spots are indicated in Fig. 2.

435

436 Table 3. Mg/Ca ratios measured in the inner and outer foraminiferal test walls. Locations of  
437 measurement spots are represented in Fig. 4 and indicated in the foraminiferal tests in Fig. 5.

438

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563 available online at [www.geosociety.org/pubs/ft20XX.htm](http://www.geosociety.org/pubs/ft20XX.htm), or on request

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Figure 1  
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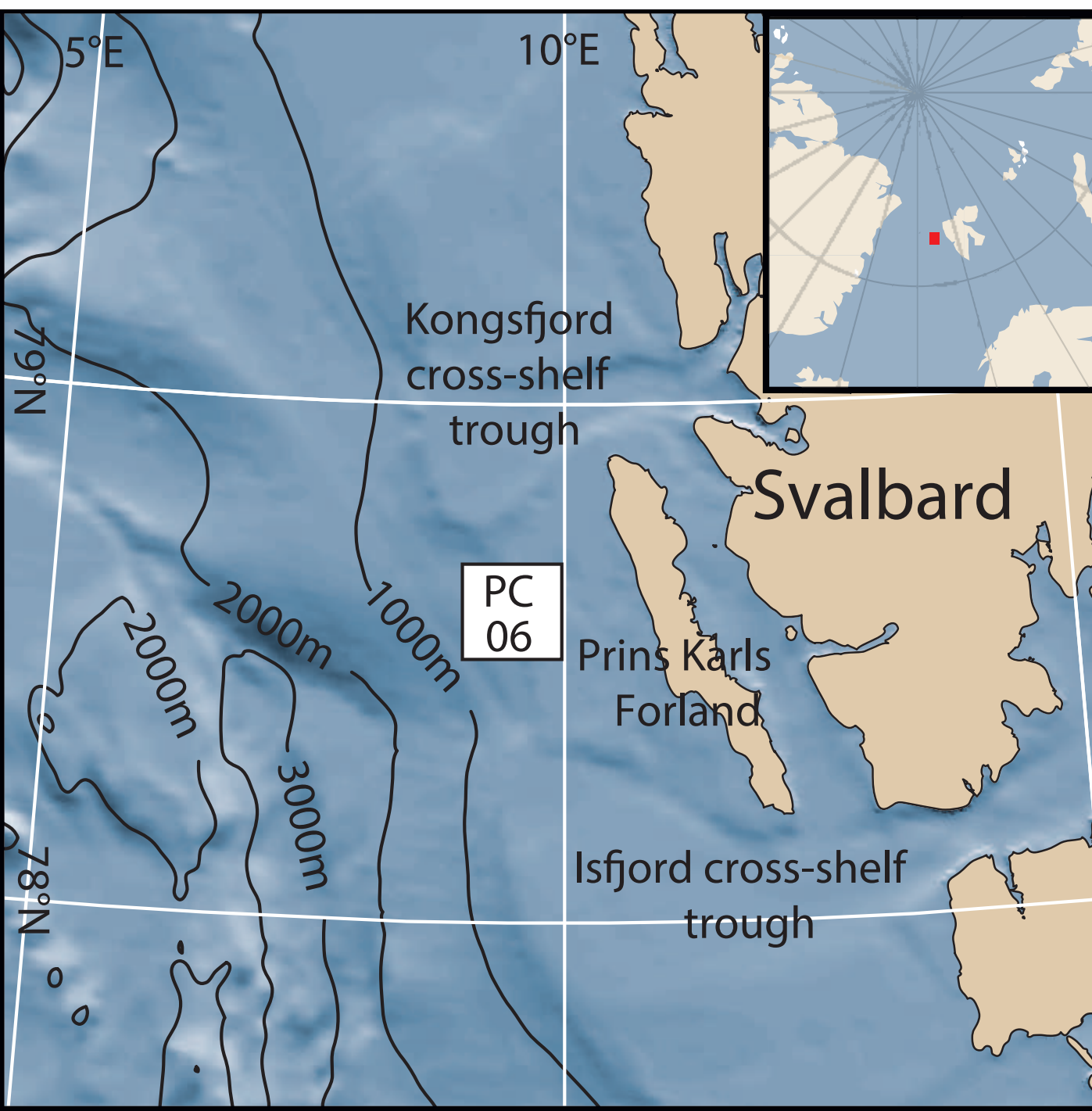


Figure 2

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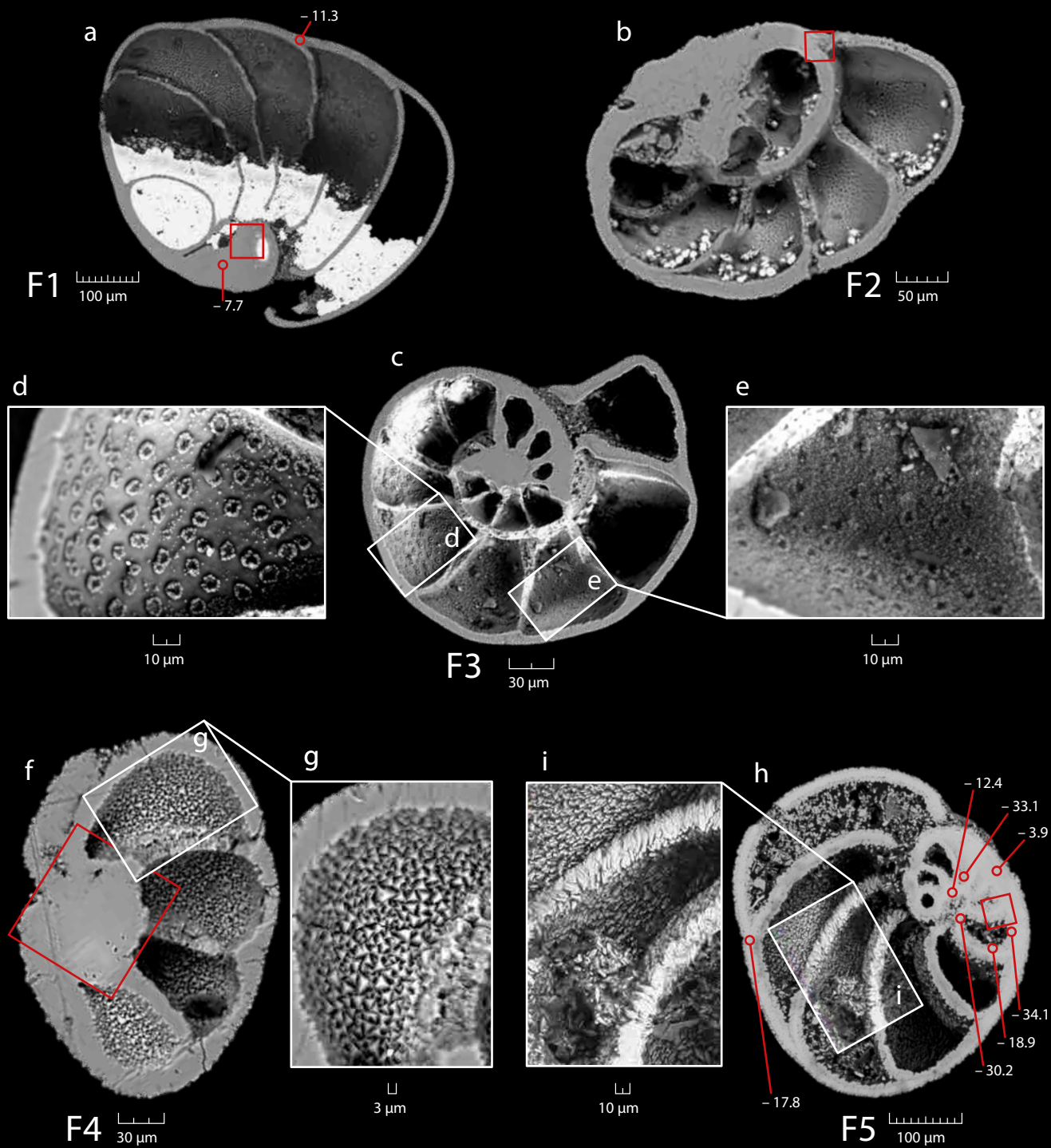


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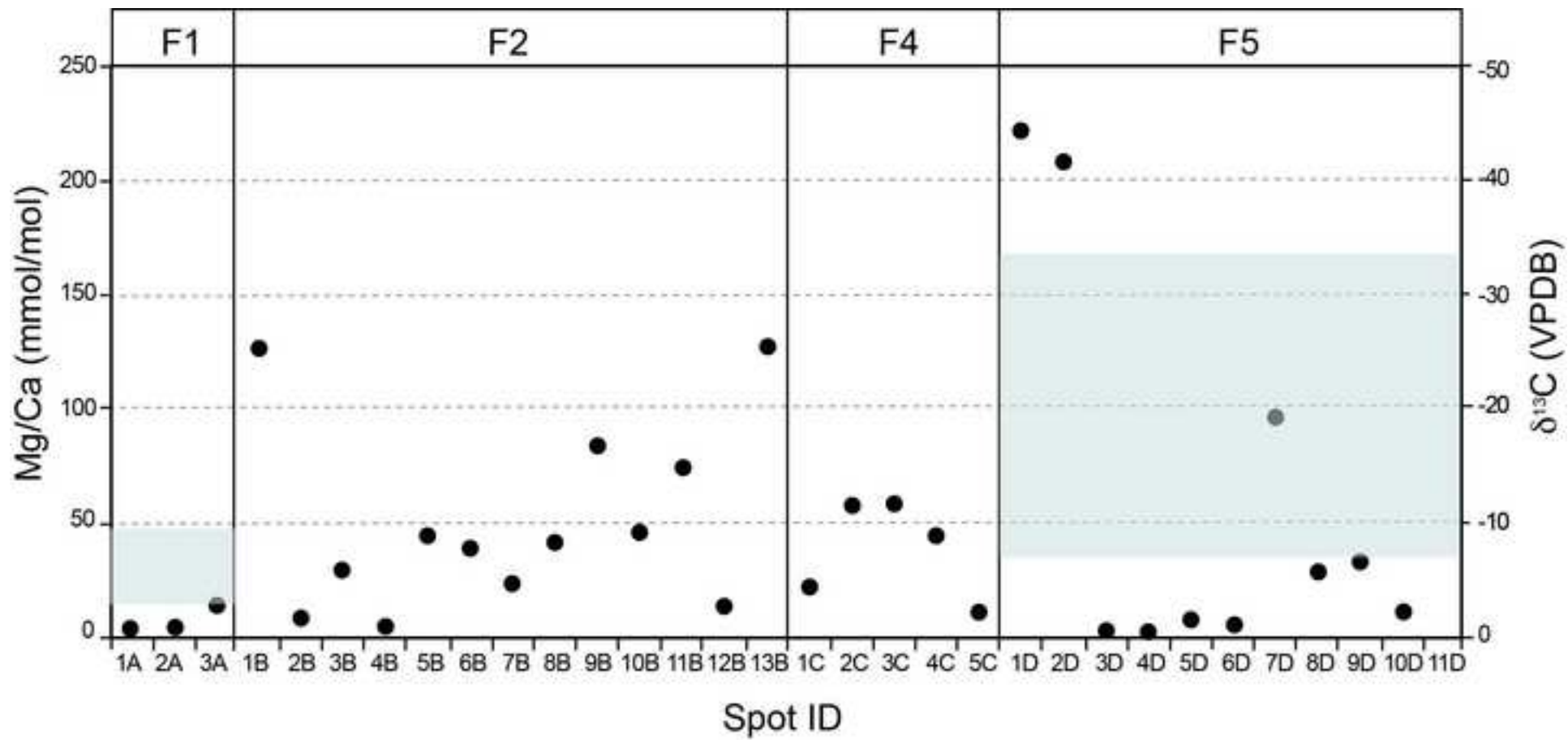




Figure 4

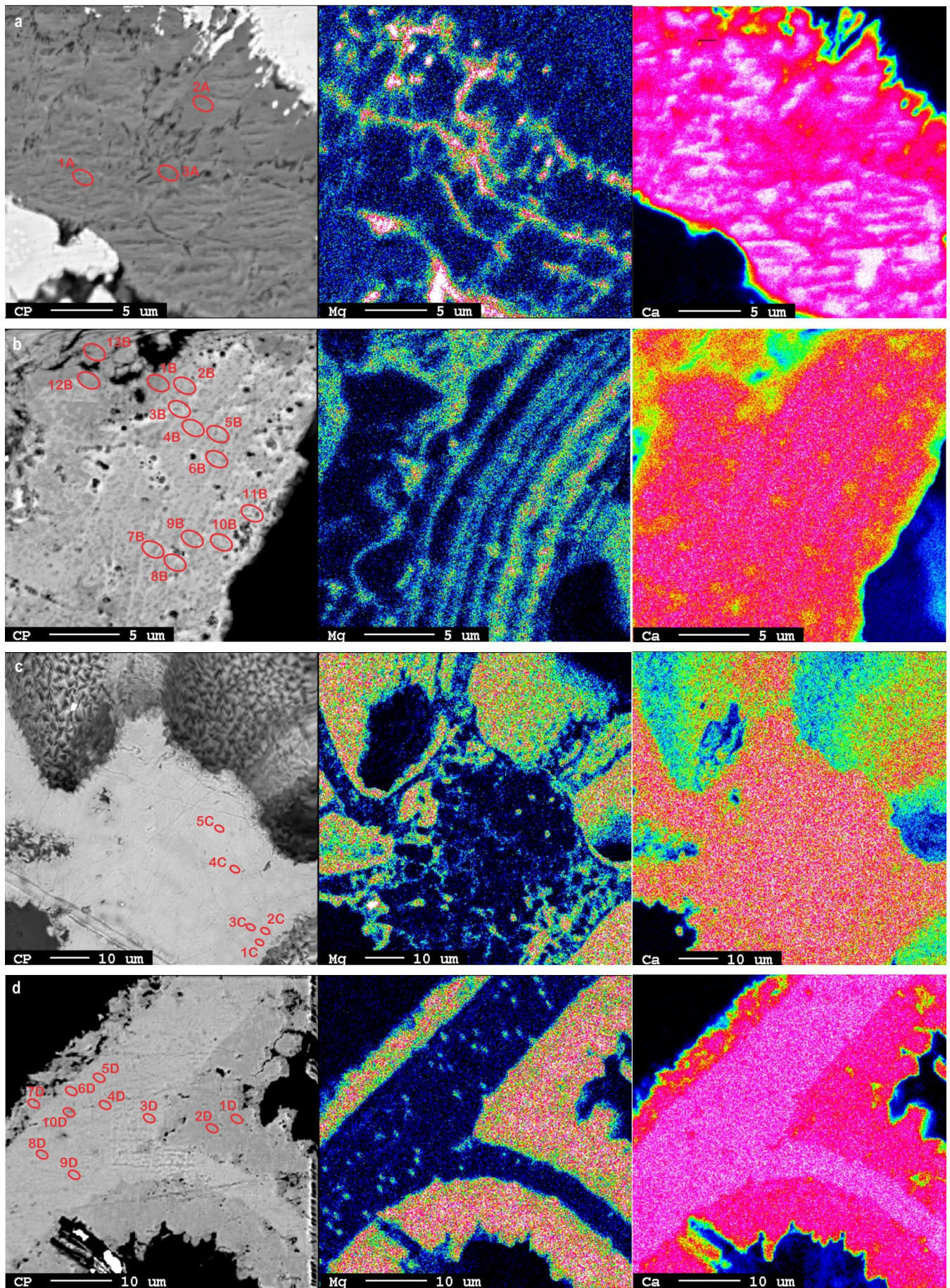
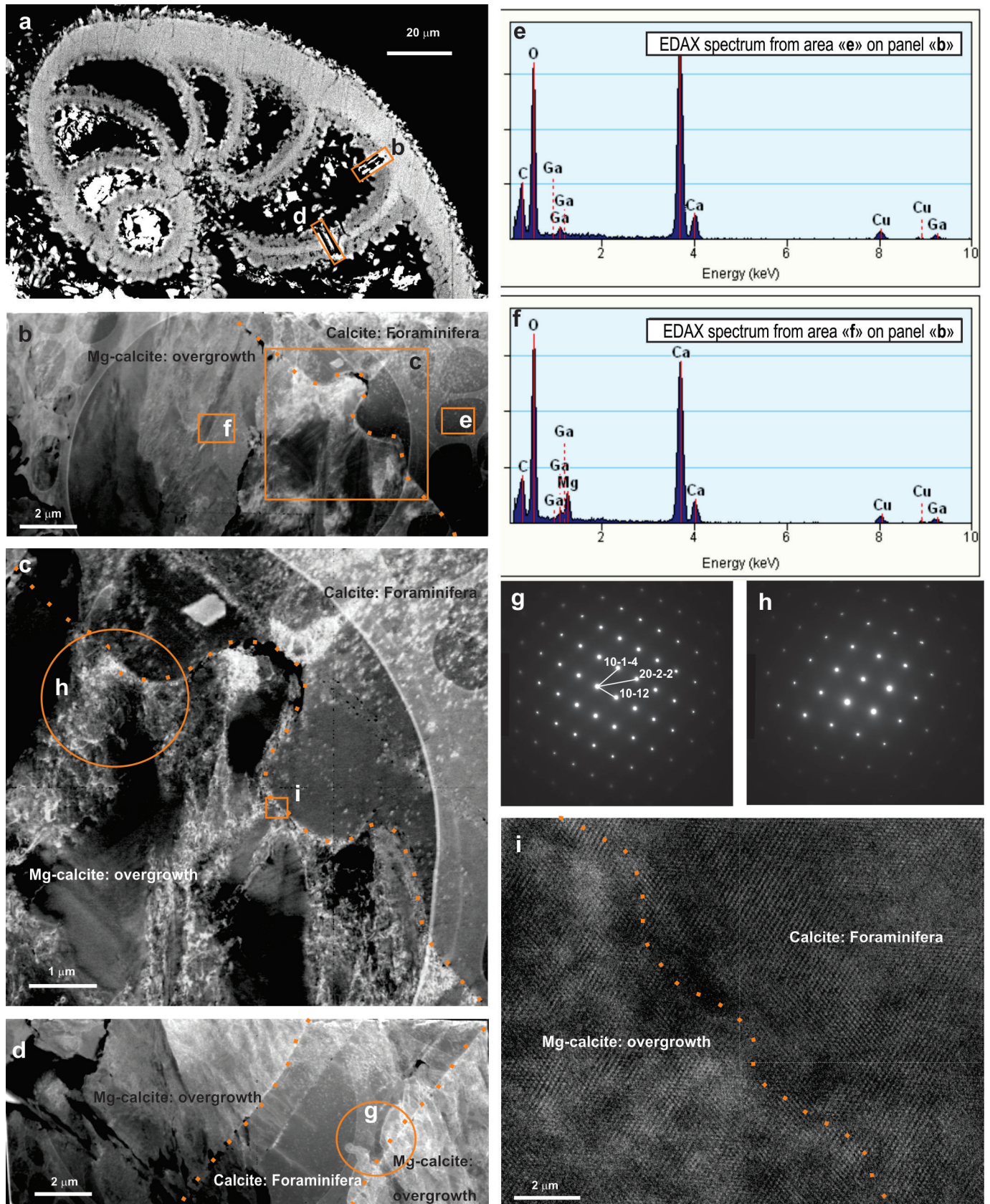




Figure 5



**Table 1-3**[Click here to download Table: Tables 1-3 revised.xlsx](#)

<b>Foraminifera</b>	<b>Species</b>	<b>Depth (cm bsf)</b>	<b><math>\delta^{13}\text{C}</math></b>
F1	<i>Nonionella labradorica</i> (specimen 2)	132	-10.6
F2	<i>Nonionella labradorica</i>	102	-4.7
F3	<i>Melonis barleenaum</i>	102	-5.1
F4	<i>Nonionella labradorica</i>	210	-11.3
F5	<i>Nonionella labradorica</i> (specimen 1)	132	-10.6