

# Measurements and models of the temperature change of water samples in sea-surface temperature buckets

G. Carella,<sup>a</sup> A. K. R. Morris,<sup>a</sup> R. W. Pascal,<sup>a</sup> M. J. Yelland,<sup>a</sup> D. I. Berry,<sup>a</sup> S. Morak-Bozzo,<sup>b</sup> C. J. Merchant<sup>b,c</sup> and E. C. Kent<sup>a\*</sup>

<sup>a</sup>National Oceanography Centre, Southampton, UK

<sup>b</sup>Department of Meteorology, University of Reading, UK

<sup>c</sup>National Centre for Earth Observation, Reading, UK

\*Correspondence to: E. C. Kent, National Oceanography Centre European Way, Southampton SO14 3ZH, UK. E-mail: eck@noc.ac.uk

Uncertainty in the bias adjustments applied to historical sea-surface temperature (SST) measurements made using buckets are thought to make the largest contribution to uncertainty in global surface temperature trends. Measurements of the change in temperature of water samples in wooden and canvas buckets are compared with the predictions of models that have been used to estimate bias adjustments applied in widely used gridded analyses of SST. The results show that the models are broadly able to predict the dependence of the temperature change of the water over time on the thermal forcing and the bucket characteristics: volume and geometry; structure and material. Both the models and the observations indicate that the most important environmental parameter driving temperature biases in historical bucket measurements is the difference between the water and wet-bulb temperatures. However, assumptions inherent in the derivation of the models are likely to affect their applicability. We observed that the water sample needed to be vigorously stirred to agree with results from the model, which assumes well-mixed conditions. There were inconsistencies between the model results and previous measurements made in a wind tunnel in 1951. The model assumes non-turbulent incident flow and consequently predicts an approximately square-root dependence on airflow speed. The wind tunnel measurements, taken over a wide range of airflows, showed a much stronger dependence. In the presence of turbulence the heat transfer will increase with the turbulent intensity; for measurements made on ships the incident airflow is likely to be turbulent and the intensity of the turbulence is always unknown. Taken together, uncertainties due to the effects of turbulence and the assumption of well-mixed water samples are expected to be substantial and may represent the limiting factor for the direct application of these models to adjust historical SST observations.

**Key Words:** sea-surface temperature; climate change; observation bias; error model

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## 1. Introduction

Global average surface temperature is the primary metric used to summarize the changing climate and underpins international policy to reduce carbon emissions (Rockström *et al.*, 2009; UNFCCC, 2015). Also, it is well understood that to quantify, mitigate, and adapt to the many impacts of climate change, a range of measures of environmental change is needed (Briggs *et al.*, 2015). The long observational record of surface temperature remains an indispensable indicator of climate change, and a measure of direct relevance to societal interests via

temperature impacts on health, food production and economies. Moreover, the ability of climate models to reproduce observed changes enables evaluation of climate model predictions; surface temperature, covering the past *ca.* 150 years, is the longest available observational record for such assessments (IPCC, 2013). Global surface temperature (GST) is usually constructed from near-surface air temperature over land and sea-surface temperature (SST) for the ocean (Kent *et al.*, 2016). Historical SST provides a lower boundary condition for reanalyses of past dynamics of the atmospheric circulation; centennial reanalyses such as the 20th Century Reanalysis (Compo *et al.*, 2011), and ERA-20C

(Poli *et al.*, 2016) provide valuable resources for climate research and understanding the impacts of weather variability and climate change on the biosphere and human societies.

The greatest source of uncertainty in the long-term evolution in global average surface temperature arises from uncertainty in the bias adjustments applied to SST (Jones, 2016). Observations of SST show characteristic biases that depend on measurement method (Kent and Kaplan, 2006; Kent and Taylor, 2006; Kennedy *et al.*, 2011; Kent *et al.*, 2016). Changes in the observing system therefore lead to changing biases in SST regionally, and over time (Kent *et al.*, 2010).

To construct accurate climate records of SST from observations in archives such as the International Comprehensive Ocean–Atmosphere Data Set (ICOADS: Freeman *et al.*, 2017), it is necessary to estimate these biases, make adjustments, and estimate the uncertainty in those adjustments (Kennedy *et al.*, 2011; Hirahara *et al.*, 2014; Huang *et al.*, 2015). The approaches taken vary, but there is agreement that biases, and uncertainties in the bias adjustments, in observations made from ships of the temperature of sea-water samples taken with buckets (Kent *et al.*, 2016) are an important contributor to the overall uncertainty in changes of GST since 1850 (Jones, 2016). The overall bias adjustment required in historic SST datasets therefore evolves as the proportion of observations made using buckets changes over time. Errors in both the bias adjustments and our knowledge of the mix of observations affect estimates of decadal-scale variability through the historic record. The proportion of ships making bucket observations has decreased over time with the introduction of engine-room intake and hull sensor measurements. The design, and therefore thermal properties, of the buckets used have also evolved. Broadly, the evolution over time of the type of buckets used to measure SST on ships was from wooden buckets (partly insulated), to canvas (uninsulated), and then to rubber or plastic buckets (typically well insulated) (Kent *et al.*, 2010).

The most-used historical SST gridded products make these adjustments for bucket bias in two different ways (Kent *et al.*, 2016). HadISST (Rayner *et al.*, 2003), HadSST3 (Kennedy *et al.*, 2011) and COBE-SST2 (Hirahara *et al.*, 2014) construct bias adjustments from weighted climatological monthly fields of estimates of bucket bias based on a physical model (Folland and Parker, 1995). ERSSTv4 (Huang *et al.*, 2015) makes bias adjustments to all ship observations based on night-time marine air temperature (NMAT) from the HadNMAT2 dataset (Kent *et al.*, 2013).

The factors affecting bucket measurements of SST are reasonably well-known (Kent *et al.*, 2016) and have been estimated using physical models developed by Folland and Parker (1995, hereafter FP95). The FP95 models, used in HadISST, HadSST3 and COBE-SST2, simulate the evaporative, direct and radiative heat exchanges experienced by samples of water in buckets as a function of the bucket's structural and thermal characteristics (dimensions and material) as well as the airflow around the bucket. The contribution of each term in the model is expected to vary for different bucket types, and FP95 presents two different formulations designed to estimate heat exchange from wooden and canvas buckets. The FP95 models were coded in BASIC and have been converted to FORTRAN (Kent *et al.*, 2017).

There were few measurements available to FP95 to provide supporting validation for their models. Ashford (1948) compared temperature changes of water samples in seven different types of bucket measured in a wind tunnel at a single wind speed. One of these buckets (the Met. Office Mark II) was a canvas bucket of the same type as that represented in FP95, the others were better-insulated buckets of various designs. FP95 concluded that their model could reproduce the temperature change of the Met. Office Mark II canvas bucket with reasonable accuracy. However, in order to predict the measured temperature change, FP95 adjusted their canvas model, assuming free evaporation from the base and sides only. Moreover, Ashford (1948) only reported the rate of change of water temperature in the first minute, while



**Figure 1.** (a) Wooden bucket; (b) canvas bucket; (c) German scoop (modern version). [Colour figure can be viewed at [wileyonlinelibrary.com](http://wileyonlinelibrary.com)].

it may have taken historical thermometers several minutes to equilibrate (FP95). Ashford (1948) did not make measurements with a wooden bucket. Roll (1951) made measurements in a wind tunnel of the characteristics of a single bucket type, the German scoop thermometer, at a wide range of wind speeds. FP95 did not develop versions of their model based on this type of bucket.

FP95 also described a comparison of their model output with the results of measurements made at sea of the temperature change of water samples in canvas buckets, and again concluded that their model showed reasonable agreement.

The amount of data available to test the canvas FP95 bucket model was limited, and there were no measurements for temperature change for wooden buckets. In this article we therefore compare measurements made in the laboratory of heat exchange from replicas of historical wooden and canvas buckets with the output of the FP95 model. The experimental set-up and the implementation of the FP95 model are described in section 2. The measurements are compared to the model predictions in section 3 and, with insight from these comparisons, we review the wind tunnel results presented by Ashford (1948) and Roll (1951). Section 4 discusses the results and draws conclusions about the wider applicability of our measurements and the FP95 model.

## 2. Materials and methods

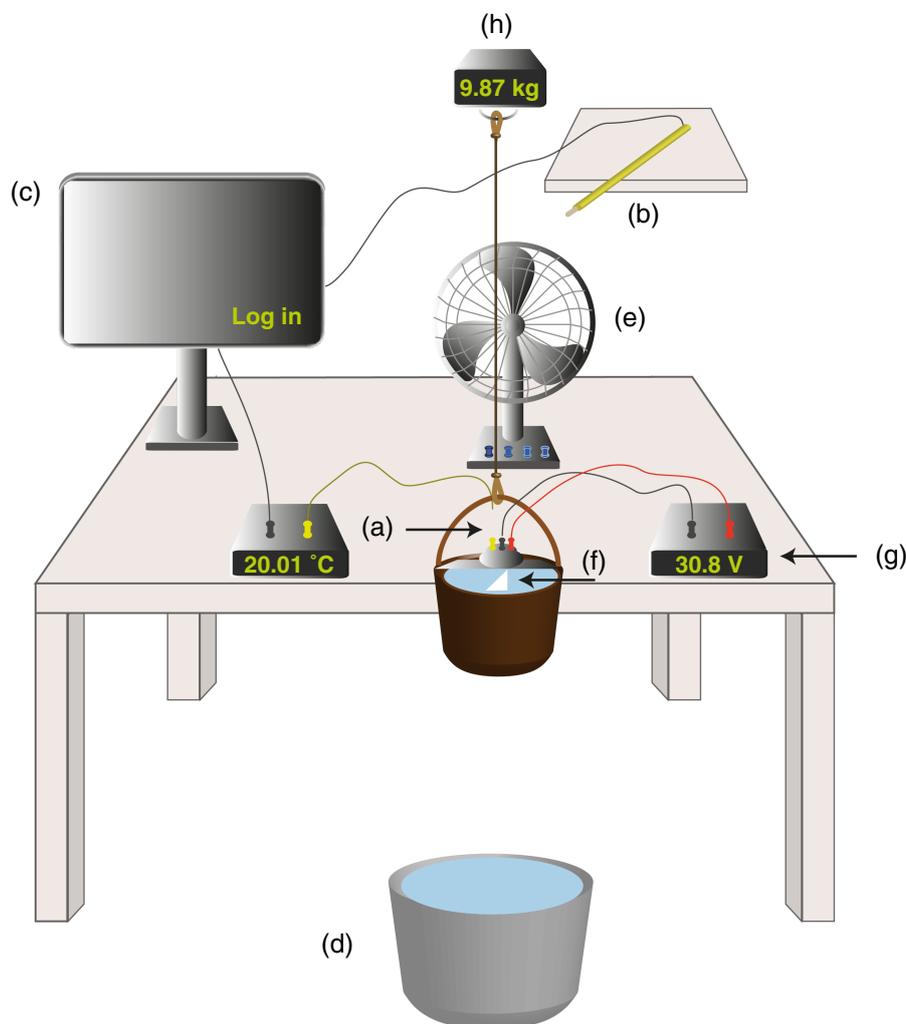
### 2.1. Description of the experimental set-up

The buckets used in this study (Figure 1) are replicas of the Mk II Met. Office canvas bucket and a nineteenth-century wooden bucket similar to that modelled by FP95. Their structural characteristics are listed in Table 1. The two buckets are of similar size (wood: 21.8 cm average inner diameter by 17.6 cm deep, up to a set water level, wider at top than at the bottom and a volumetric capacity  $\sim 6.6$  l; canvas: 17.8 cm inner diameter by 19.4 cm deep, up to a set water level, and volumetric capacity  $\sim 4.8$  l). The wooden bucket is made of oak 16 mm thick reinforced around the outside by two stainless steel bands. Only the sides of the canvas bucket are canvas: the base is wooden with a metal weight inside; the top is wooden with a metal spring-closing lid; the canvas is stitched and the top and base held in place with leather bands and metal pins. The masses of the wooden and canvas buckets when wet are  $\sim 3.3$  and  $\sim 2.9$  kg respectively.

Figure 2 illustrates the experimental set-up. The experiments were performed in the National Oceanography Centre (Southampton, UK) Calibration Laboratory. This is kept at a roughly constant temperature of  $20^\circ\text{C}$ , but the humidity is not controlled. A precision F250 thermometer was used to measure the water temperature ( $t$ ) and a Vaisala probe was used to monitor the ambient air temperature ( $t_a$ ) and the relative humidity ( $R$ ). Data from the probes were logged every 2–3 s (alternate readings). The water temperature probe when not in use was left in a plastic container filled with water approximately in equilibrium

Table 1. Structural characteristics of the buckets discussed in this study.

Bucket type	Bucket material	Water level (m)	Diameter (inner) (m)	Thickness (mm)	Bucket volume (l)	Bucket mass (wet & empty) (kg)
Wooden	Oak	0.176	0.218	16	6.6	3.30
Mk II canvas	Mixed	0.194	0.178	–	4.8	2.91
German scoop	Mixed	0.116	0.097	13	0.9	3.35



**Figure 2.** Illustration of the experimental set-up: (a) precision thermometer (F250); (b) air temperature and relative humidity probe (Vaisala); (c) PC used for logging; (d) plastic bin (containing clean fresh water) used to soak the buckets; (e) fan; (f) automatic stirrer; (g) power generator for the automatic stirrer; (h) hanging scale. [Colour figure can be viewed at [wileyonlinelibrary.com](http://wileyonlinelibrary.com)].

with the ambient air temperature. A plastic bin was used to soak the buckets (the soaking time was about 4 min), which were then hung in front of a fan with three different speed settings (Table 2). The centre of the fan was positioned about 0.5 m from the bucket.

The largest uncertainty in the ambient conditions comes from the airflow around the bucket. Because the bucket was fairly close to the fan relative to the bucket dimensions, the speed was not uniform around the bucket. The airflow was measured using a WindMaster ultrasonic anemometer (Gill Instruments Ltd.) for 30 s at each of six different positions: five positions in the vertical plane where the bucket would hang (centre of the bucket position and 0.5 m above, below, left and right) and at 0.35 m upwind from the centre of the bucket location. The airflow used in the implementation of the FP95 model was that measured where the centre of the bucket would be, with uncertainty derived from the standard deviation of measurements made in these surrounding locations.

Because the FP95 models assume the water sample is stirred, the water was mixed at all times using an automatic stirrer, connected to a power generator. The wooden bucket is open at the top (Figure 1). The top of the canvas bucket is a thick wooden disc with a hole for a metal lid. This lid was pushed

inside the bucket by the plastic support of the stirrer during the measurements. The edge of the lid was in the water, but this is not expected to substantially affect the heat exchange as the metal lid was attached to the wooden top, limiting heat exchange by conduction. A ‘weak stirring’ regime, characterized by a mild but noticeable stirring, was created adopting an L-shaped metal piece as the stirrer; a ‘strong stirring’ regime was also implemented, where some tape was added to produce a sail-shaped stirrer. A hanging scale with precision of 0.01 kg was used to measure the mass of the filled bucket; the water level was also set and marked for each bucket and the bucket filled up to the level indicator. Finally, (clean) fresh water was used instead of salty water. The effect of salinity on latent heat of evaporation is well-known, and the vapour pressure over saline sea-water is typically reduced by 2% compared to fresh water (Zeng *et al.*, 1998).

Figure 3 shows thermal pictures of a replica Mk II Met. Office canvas bucket (the type used by the UK Meteorological Office in the 1930s and 1940s (Ashford, 1948)), filled with water warmer than the ambient air temperature. The bucket is unstirred and the lid is shut. It is clear that the water in the bucket is cooling over time, with the cooling proceeding faster in the area facing the fan (located to the right of the bucket in these pictures). Initially

Table 2. Summary of each experiment.

Experiment type	Bucket type	Stirring type	$t_0 - t_a$ (°C)	Fan speed (m s <sup>-1</sup> )
<i>dt</i> 1	Wooden, Canvas	Strong, Weak	~5	3.53 ± 0.49
<i>dt</i> 2	Wooden, Canvas	Strong, Weak	~-1	3.53 ± 0.49
<i>dt</i> 3	Wooden, Canvas	Strong, Weak	~-5	3.53 ± 0.49
<i>u</i> 0	Wooden, Canvas	Strong	~5	0.05 ± 0.02
<i>u</i> 1	Wooden, Canvas	Strong	~5	2.17 ± 0.37
<i>u</i> 2	Wooden, Canvas	Strong	~5	2.88 ± 0.44
<i>u</i> 3	Wooden, Canvas	Strong	~5	3.53 ± 0.49

The table reports the bucket type, the stirring type, the approximate water temperature at time = 0 min minus the ambient air temperature  $t_0 - t_a$  (°C) and the fan speed (m s<sup>-1</sup>) for each experiment. The uncertainty in the fan speed is reported at one standard deviation. For model simulations when the fan was turned off ( $u0$ ) a small airflow speed (0.05 m s<sup>-1</sup>) was assumed to be consistent with the measured fluctuations (standard deviation 0.02 m s<sup>-1</sup>). For an extended summary see Appendix.

the whole of the bucket is much warmer than the environment having been soaked in the warm water before exposure to the air. The structure of the bucket (rope handle, leather bands at top and bottom, stitched seam) can just be seen as cooler than the canvas body of the bucket containing the water. After 5 min the body of water can clearly be seen at higher temperature than the rest of the bucket, which is now colder than ambient temperature having cooled by evaporation. These images suggest that the non-canvas parts of the bucket are insulating and probably do not contribute strongly to the heat exchange which occurs almost exclusively through the canvas walls of the bucket.

## 2.2. Description of the model – experimental comparison procedure

In order to test FP95 heat exchange models we measured both the time and airflow dependence of the temperature of water in the buckets and compared the results with the model predictions. The FP95 model used is the laboratory version described by Kent *et al.* (2017). This laboratory version is similar to the full version used by FP95, but does not differentiate between the different ambient conditions expected during hauling and on-deck phases of measurement (for more information see Kent *et al.* (2017)). Moreover, it sets the solar term to zero, as our measurements were taken indoors and away from windows. We have also excluded the salinity effect on the estimate of saturation vapour pressure, as we used fresh water. The bucket is modelled by FP95 as a cylinder in an incident airflow that is assumed to be non-turbulent. It is further assumed that the water in the bucket is well mixed (at temperature  $t$  (°C)) and the bucket has been immersed in the sea for long enough to reach equilibrium. Inputs to the models are the airflow around the bucket for each fan speed, the ambient air temperature and humidity as measured, and the initial air–water temperature difference. For the canvas bucket, the rate of change of temperature as modelled by FP95 can be represented as

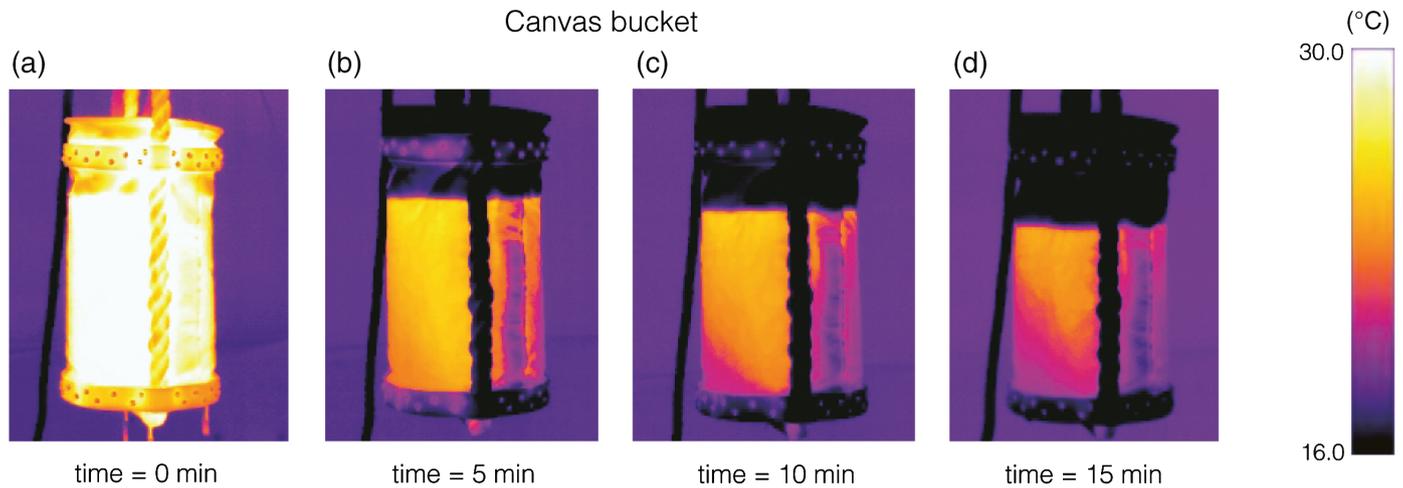
$$\frac{dt}{d\tau} = \frac{A}{c\mu} \{f_r h_r (t_a - t) + f_t h_t (t_a - t) + f_e h_e (e_a - e)\} \quad (1)$$

where the ambient air temperature is  $t_a$  (°C), the ambient vapour pressure is  $e_a$  and  $e$  is the saturation vapour pressure at  $t$  (both in hPa). The transfer coefficients are  $h_r$  for long-wave radiation,  $h_t$  for direct heat transfer and  $h_e$  for evaporative heat transfer (all in W m<sup>-2</sup> K<sup>-1</sup>).  $h_e$  is 1.7  $h_r$ .  $A$  (m<sup>2</sup>) represents the total surface area of the bucket. The fraction of the surface area affected by long-wave heat exchange,  $f_r$ , represents the sides and base. For the direct ( $f_t$ ) and evaporative heat ( $f_e$ ) exchange, the fraction is the same for both components, but is allowed to vary: the sides always contribute (up to the fill level) but the contribution of the top and base may be excluded, or included as required. Each of the transfer coefficients  $h_t$  and  $h_e$  depend on wind speed and the bucket geometry; slightly different values are used for the base and sides. FP95 explore different choices of  $f_t$  and  $f_e$ : heat exchange from the base, the top and the sides; heat exchange from the sides only; and heat exchange from the sides and the

base (or the top), which is the final choice for FP95 as it gave the best agreement with Ashford (1948) results.  $c$  is the specific heat capacity (J kg<sup>-1</sup> K<sup>-1</sup>) and  $\mu$  is the effective mass (kg) of the bucket. In FP95  $\mu$  is the combined mass of the bucket material and the water sample. The wooden bucket model is similar to that for the canvas bucket, but the thermal forcing experienced by the outside of the bucket walls acts to conduct heat through the wooden sides and base. The open top evaporates freely, but is assumed to experience a lower airflow as the water level is below the top of the bucket. More details can be found in FP95 and Kent *et al.* (2017).

Although FP95 have assumed little heat exchange from the top of the canvas bucket because of the lid, our experimental set-up shows that for the water sample to be properly mixed, and for the measurement to be made, it requires the lid open (pushed down), permitting heat exchange from the upper water surface. On the other hand, the thermal images shown in Figure 3 suggest that most of the contribution to the overall heat loss is from the sides of the canvas bucket. The thick wooden base is not expected to make much contribution to the heat loss. However, if the top was open (it is not in these images) then exchange of heat from the open top is expected, although the airflow within the bucket would be rather small, limiting this effect. Therefore, when implementing the FP95 canvas bucket model in this study we have run the model assuming heat exchange from either the top and sides or from the sides only, with each included in the ensemble from which the model uncertainty range is calculated. In our implementation of FP95 we assume no contribution from the bucket material to the effective mass and heat capacity of the canvas bucket. This seems justified by Figure 3, as the images show that the temperature change largely affects only the water sample; the non-canvas parts of the bucket quickly reach ambient temperature, suggesting that the temperature change for the wooden and leather parts is superficial. The choice of the effective mass, which sets the heat capacity, will scale the temperature change but will not affect its functional dependence.

The models were initialized with measured ambient conditions (summarized in Appendix) and the appropriate bucket dimensions (Table 1; other bucket properties are set by the choice of the wooden or canvas model). The probe used to measure the water temperature has a finite response time and typically took between 30 s and 1 min to reach equilibrium, less when the air and water temperatures were similar. Each experiment was considered to start when the recorded water temperature reached a local maximum or minimum (depending on whether the water was warmer or colder than the air). Uncertainty in the equilibration temperature was estimated to be around 0.01 °C (much smaller than, for example, the variation in the air temperature over each experiment) so the estimated uncertainty is not sensitive to the value chosen. For each experiment, the uncertainties in the model outcomes were expressed as an ensemble of 100 realizations. Each realization was obtained by forcing the model with samples of the measured ambient air temperature, relative humidity, wind speed (each randomly generated from a normal distribution with mean and standard deviation as measured) of



**Figure 3.** (a)–(d) Thermal pictures taken at 5 min intervals of the Met. Office Mk II canvas bucket (Figure 1) filled with warm water hung in front of a fan positioned to the right in these images. The bucket is not stirred and the lid is shut.

the water temperature at time = 0 min (randomly generated from a normal distribution with mean as measured, standard deviation of  $0.01\text{ }^{\circ}\text{C}$ ) and of the bucket geometry. For the canvas bucket model, the randomly generated samples of the bucket diameter and water level were drawn from a normal distribution with mean as measured and standard deviation of  $0.5\text{ cm}$ ; the uncertainty in the bucket geometry is included to account for the small variations in the initial mass of the water sample (for both bucket types the standard deviation over all the measurements was about  $0.05\text{ kg}$ ). For the wooden bucket model, the uncertainty in the geometry of the bucket mainly arises because of the uncertainty in the contribution of the bucket mass to the heat exchange of the water sample; the randomly generated samples of the bucket diameter and water level were therefore drawn from a normal distribution with mean as measured at the mid-point of the bucket walls and standard deviation equal to the thickness of the bucket walls. For the wooden bucket, the uncertainty in the factor that accounts for the sheltering of water by the sides of the bucket from the effects of airflow was also considered. The bucket was fairly full, with the water level about  $2\text{ cm}$  below the top, so only a modest sheltering of the airflow would be expected. FP95 also assumed sheltering was modest and used two sheltering factors,  $1.00$  and  $0.75$  ( $1$  represents no sheltering). We used a range similar to that assumed by FP95: the randomly generated samples were drawn from a normal distribution with mean of  $0.875$  and a standard deviation of  $0.125$  (with upper limit of  $1$ ). Additionally, we included in the overall uncertainty for the wooden bucket model the uncertain thermal conductivity of wet oak (with samples randomly generated from a normal distribution with mean of  $0.3\text{ W m}^{-1}\text{ }^{\circ}\text{C}^{-1}$  as assumed in FP95, standard deviation of  $0.2\text{ W m}^{-1}\text{ }^{\circ}\text{C}^{-1}$  and with upper and lower limit defined by the thermal conductivity of dry oak ( $0.17\text{ W m}^{-1}\text{ }^{\circ}\text{C}^{-1}$ ) and water ( $0.6\text{ W m}^{-1}\text{ }^{\circ}\text{C}^{-1}$ ) respectively). Finally, for the canvas bucket, in order to account for the uncertain contribution of evaporation from the bucket top, we generated model outputs with heat exchange from the sides only and also model outputs with heat exchange from the sides and the top.

Leakage, determined by the change in mass, was largest for the canvas bucket, and decreased over time ( $0\text{--}3\text{ min}$ :  $\sim 0.05\text{ kg min}^{-1}$ ,  $4\text{--}20\text{ min}$ :  $\sim 0.04\text{ kg min}^{-1}$  and  $20\text{ min}$  onwards:  $\sim 0.03\text{ kg min}^{-1}$ ). No significant leakage was measured for the wooden bucket. We included the changing mass in the canvas bucket model, but, as noted by Kent *et al.* (2017), the leakage makes very little difference as decreases in the surface area subject to heat exchange affect a decreasing volume of water, with little overall effect as long as the bucket remains fairly full.

Firstly, the evolution of the bucket temperature over time was measured with a set of experiments varying the temperature of

the water in the plastic bin used for soaking the bucket and from which the water sample is taken. The experiments were performed using the two different stirring regimes ('strong' and 'weak') to test how different mixing conditions may affect the heat exchange from the water sample ( $dt1$  to  $dt3$  in Table 2). For each bucket, the water temperature was measured for  $15\text{ min}$  for three air–water temperature regimes and each of these measurements was repeated three times. In the first set of experiments ( $dt1$ ) the initial water temperature ( $t_0$ ) was warmer than the air temperature ( $t_a$ ):  $t_0 - t_a \sim 5\text{ }^{\circ}\text{C}$ . The second set ( $dt2$ ) has  $t_0$  slightly colder than  $t_a$ :  $t_0 - t_a \sim -1\text{ }^{\circ}\text{C}$ . In the third set ( $dt3$ ) the water temperature was colder again:  $t_0 - t_a \sim -5\text{ }^{\circ}\text{C}$ . The fan was at its fastest setting, about  $3.5\text{ m s}^{-1}$ ,  $7\text{ knots}$  ( $u_3$ , see Table 2), for all six experiments (three temperatures and two stirring regimes).

Secondly, we measured the water temperature for  $15\text{ min}$  for each of the four available different airflows ( $u0$  through to  $u4$  in Table 2) for an initial warm-water bucket temperature difference of  $t_0 - t_a \sim 5\text{ }^{\circ}\text{C}$  and under the strong stirring regime. Again, each set of measurements was repeated three times.

### 3. Results and discussion

In this section we describe the results of the comparison of temperature change measured in the laboratory and predicted by the models (section 3.1) for different degrees of mixing of the water sample (sections 3.1.1 and 3.1.2) and for different airflows (section 3.1.3). Also we present here the results of the comparison with historical measurements in wind tunnels (section 3.2) made by Ashford (1948) and by Roll (1951). Analysis was performed using the R software package (R Development Core Team, 2016).

#### 3.1. Comparison of temperature change measured in the laboratory and predicted by the models

##### 3.1.1. Evolution of water temperature under strong stirring

Figure 4 shows the measured and modelled water temperature as a function of time for both the wooden and canvas buckets, for the range of three different initial water temperatures and also for the strong and weak stirring regimes. When the initial water temperature is warmer than ambient air temperature (set of experiments  $dt1$ ) the water is cooled both directly and by evaporation. When the initial water temperature is slightly colder than the air temperature (set of experiments  $dt2$ ) the water is warmed directly and cooled by evaporation. For these conditions the evaporation dominates and the water sample cools.

When the water is significantly colder than ambient air (set of experiments *dt3*), the water is again being warmed directly and cooled by evaporation, this time with a net warming overall. As expected, the canvas bucket cools much more rapidly than the wooden bucket, despite their similar volumes. This feature is well reproduced in the model simulations. For both buckets the contribution of the uncertainty in the airflow (fan) speed explains a large portion of the overall model uncertainty; this is shown for each air-temperature regime by the error bars on the right of the plot, which represent the 95% confidence level uncertainty at time = 15 min computed from the sample quantiles (Hyndman and Fan, 1996) of the model ensemble generated accounting for the wind uncertainty only. For the canvas bucket, the remaining uncertainty is mostly due to the variations in the ambient relative humidity and air temperature; on the other hand, for the wooden bucket the biggest contribution to the remaining model uncertainty is represented by the uncertainty in the thermal conductivity of the bucket walls. The model estimates for the wooden bucket underestimate the observed temperature change for the strong-stirring regime (Figure 4(a)), although the experimental results are close to the limits of the estimated model uncertainty. However, the rate of temperature change increases over the first few minutes of the 15 min sampling period in both the measurements and the model (shown for the model in the inset in Figure 4(a)). A simple picture of temperature change would show a decreasing rate of temperature change over time as the water sample approaches equilibrium with its surroundings (as seen for the canvas bucket in Figure 4(b)). The model reproduces the measured behaviour well, and shows that the initial slow rate of temperature change is caused by the time-scale for the conduction of heat through the walls of the wooden bucket. The water inside the bucket does not respond to the thermal forcing on the outside of the bucket until the temperature gradient within the bucket walls is established; once this occurs the temperature change of the water increases. In the 15 min sampling period this effect dominates over the reduction in thermal forcing over time as the bucket sample reaches its equilibrium temperature.

In contrast, the canvas bucket with strong stirring (Figure 4(b)) shows the expected decrease in the rate of temperature change over time, as already noted, and again the measurements and the model show the same general behaviour, with the modelled and measured temperature change agreeing at the 95% confidence level, although close to the limit of the estimated uncertainty in our experimental set-up. As noted by FP95 and Farmer *et al.* (1989) the temperature in the canvas bucket will eventually asymptotically reach an 'effective wet-bulb temperature' when the evaporative cooling is balanced by the warming from the atmosphere (Folland, 1991).

### 3.1.2. Evolution of water temperature under weak stirring

The effects of weaker stirring are explored in Figures 4(c) and (d). If the water is not well-mixed the largest temperature changes will be expected near the water surface and the bucket walls. The temperature is measured in the centre of the bucket where a smaller temperature change would be expected, and this is what is observed. The observed temperature change under weak stirring is lower than under strong stirring and the measured temperature change for the wooden bucket remains in agreement with the model predictions under both low (set of experiments *dt2* and *dt3*) and high thermal forcing (set of experiments *dt1*: warm initial water temperature), although the model assumes well-mixed conditions. The time evolution of the temperature change is unsteady compared with the better-mixed case (compare Figures 4(a) and (c)). For the canvas bucket, the difference due to reduced stirring is particularly noticeable for the high-forcing case (*dt1*): here, an initial lower rate of temperature change is very obvious, similar to that observed for the wooden bucket and predicted by the wooden bucket model. This can

again be explained by an initial setting up of temperature gradients in the water, in a similar way to the gradients established in the wooden bucket walls. As for the wooden bucket measurements, the weak stirring temperature change is unsteady.

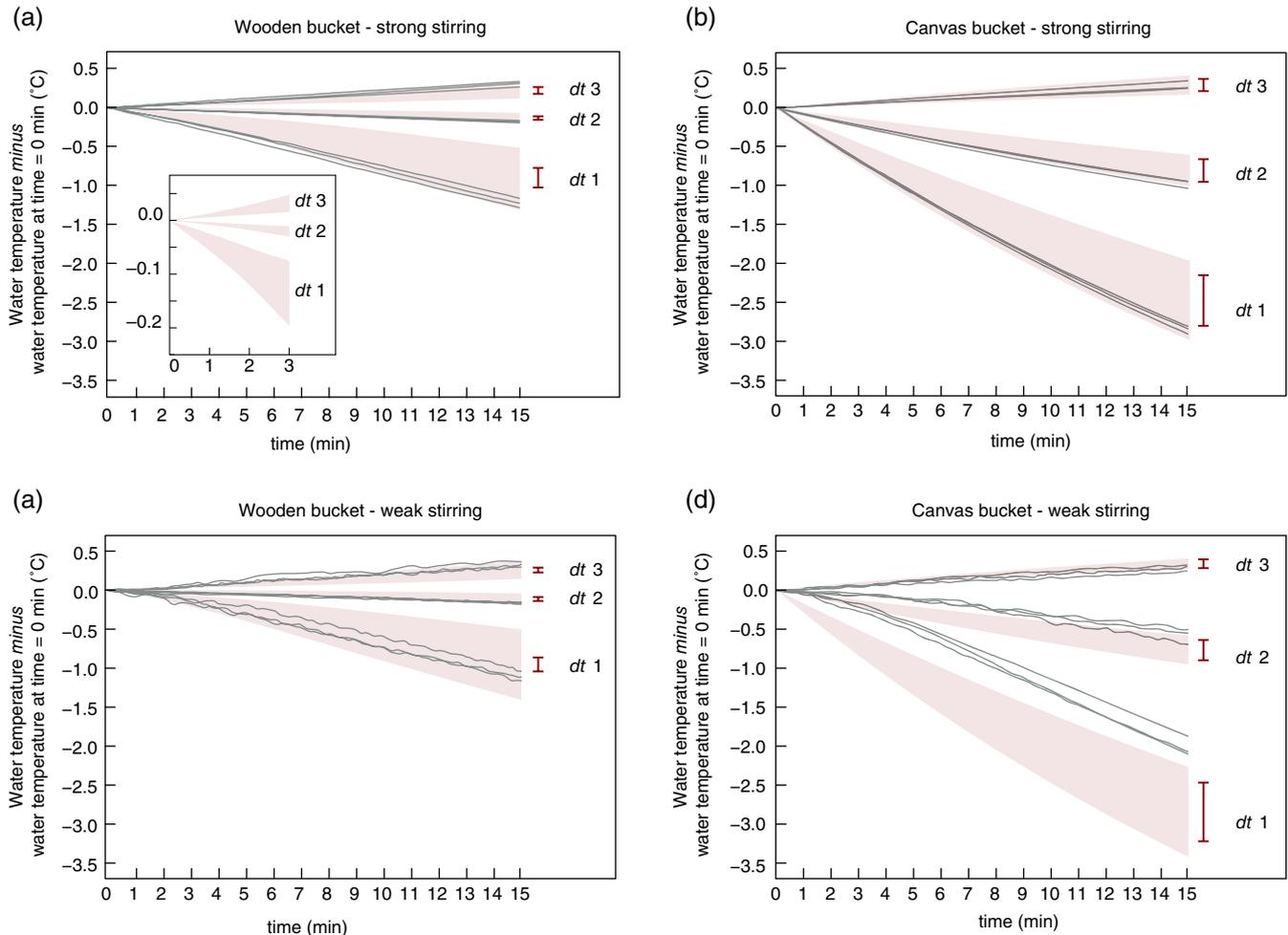
### 3.1.3. Effect of airflow

The model heat exchange coefficients  $h_t$  and  $h_e$  depend approximately on the square root of the airflow, since the incident flow is assumed to be non-turbulent. Figure 5 shows the observed bucket temperature (grey dots) at time = 5 min for the various air (fan) speeds and the values predicted by the model (shading) for the wooden (Figure 5(a)) and the canvas (Figure 5(b)) bucket for water  $\sim 5^\circ\text{C}$  warmer than air temperature. When the fan was turned on ( $u1 - u3$ ), for each bucket, the observed dependence on airflow is similar to that assumed in the model, although for the wooden bucket the observed temperature change is either close to or, for some experiments, lies outside the limits of the estimated uncertainty range, as in Figure 4(a) for *dt1*. On the other hand, when the fan was turned off ( $u0$ ), for both buckets the modelled and the observed temperature change do not agree within the range of the estimated uncertainty (Figure 5). FP95 models assume a Reynolds number always larger than one; this means that the situation when there is no airflow around the bucket is very uncertain but the temperature change will be small in these conditions. Finally, our experimental set-up means that we cannot increase the speed of the airflow around the bucket beyond  $\sim 3.5\text{ m s}^{-1}$ , and the uncertainty in the speed is large.

## 3.2. Comparison with historical measurements in wind tunnels (Ashford, 1948; Roll, 1951)

### 3.2.1. Ashford (1948)

Measurements in a stronger airflow regime, about  $9\text{ m s}^{-1}$ , were made by Ashford (1948), hereafter Ashford) for seven different buckets. The results were presented as the rate of change of water temperature in the first minute plotted as a function of the water temperature minus wet-bulb temperature ( $\Delta t_{\text{wb}}$ ). Plotted in this way buckets that evaporate strongly will show a curved relationship of temperature change with  $\Delta t_{\text{wb}}$  due to the Clausius–Clapeyron relationship. When the water temperature is varied at the same ambient air temperature, as is the case for all the measurements we consider here, the wet-bulb temperature will be constant and  $e_a$  and  $t_a$  (Eq. (1)) are also constant.  $\Delta t_{\text{wb}}$  therefore varies linearly with variations in  $t$ , as does the direct heat exchange. However, variations in  $e$  are nonlinear and the relationship between temperature change and  $\Delta t_{\text{wb}}$  will be nonlinear if the effects of evaporation are important. In contrast, buckets where the direct heat exchange dominates over evaporation, or under conditions where the air is close to saturation, will show a close-to-linear relationship when plotted in this way. Figure 6 shows measured values (from runs *dt1* to *dt3* in Table 2) obtained with strong stirring as a function of  $\Delta t_{\text{wb}}$  (with wet-bulb temperature computed following the approach of Stull (2011)). The change of water temperature over the first minute exhibits different characteristic relationships with the  $\Delta t_{\text{wb}}$  according to the bucket thermal capacity. Both the wooden bucket (Figure 6(a)) and the uninsulated canvas bucket (Figure 6(b)) are characterized by a nonlinear relationship in the model, because of evaporation (through the top for the wooden bucket and through the sides for the canvas bucket). In the first minute the measurements are noisier than our estimates of uncertainty, especially for the wooden bucket (Figure 6(a)). The measurements are fairly consistent with the model results for each bucket type but the nonlinear relationship cannot be confirmed because of the noise. Also plotted in Figure 6(b) are the results from measurements with the same type of bucket by Ashford. The increased temperature



**Figure 4.** Measured (lines) and modelled (shading) evolution of the water temperature over time. Shaded regions represent model uncertainty at 95% confidence level. Also shown is the wind speed only contribution to the model uncertainty at time = 15 min (bars). Each panel shows three sets of measurements with different initial water temperatures. *dt1*: water  $\sim 5^\circ\text{C}$  warmer than ambient air temperature; *dt2*: water  $\sim 1^\circ\text{C}$  colder than ambient air temperature; *dt3*: water  $\sim 5^\circ\text{C}$  colder than ambient air temperature. (a) Time evolution of water temperature – initial water temperature for wooden bucket, strong stirring. Inset shows expansion of first 3 min for *dt1*; (b): as (a) but for canvas bucket, strong stirring; (c) as (a) but for weak stirring; (d): as (b) but for weak stirring. Fan speed of  $\sim 3.5\text{ m s}^{-1}$  throughout. The 95% confidence level was computed from the sample quantiles (Hyndman and Fan, 1996) of the model ensemble. [Colour figure can be viewed at [wileyonlinelibrary.com](http://wileyonlinelibrary.com)].

change in the Ashford results is modest, despite the much greater airflow ( $\sim 9\text{ m s}^{-1}$  cf.  $\sim 3.5\text{ m s}^{-1}$ ), and the measurements agree well with the model. These results extend the range of airflows over which the canvas bucket model has been tested, and suggest that the wind speed dependence in these experiments is reasonably predicted by the model. We note that the Ashford measurements for the canvas bucket were used by FP95 as validation, but that here we have assumed a smaller heat capacity for the bucket (by excluding the contribution of the bucket itself, based on Figure 3 as discussed in section 2.1). However, the modelled rate of temperature change at one minute as a function of  $\Delta t_{\text{wb}}$  for each of these different choices of the effective mass remains consistent with the Ashford measurements under either assumption.

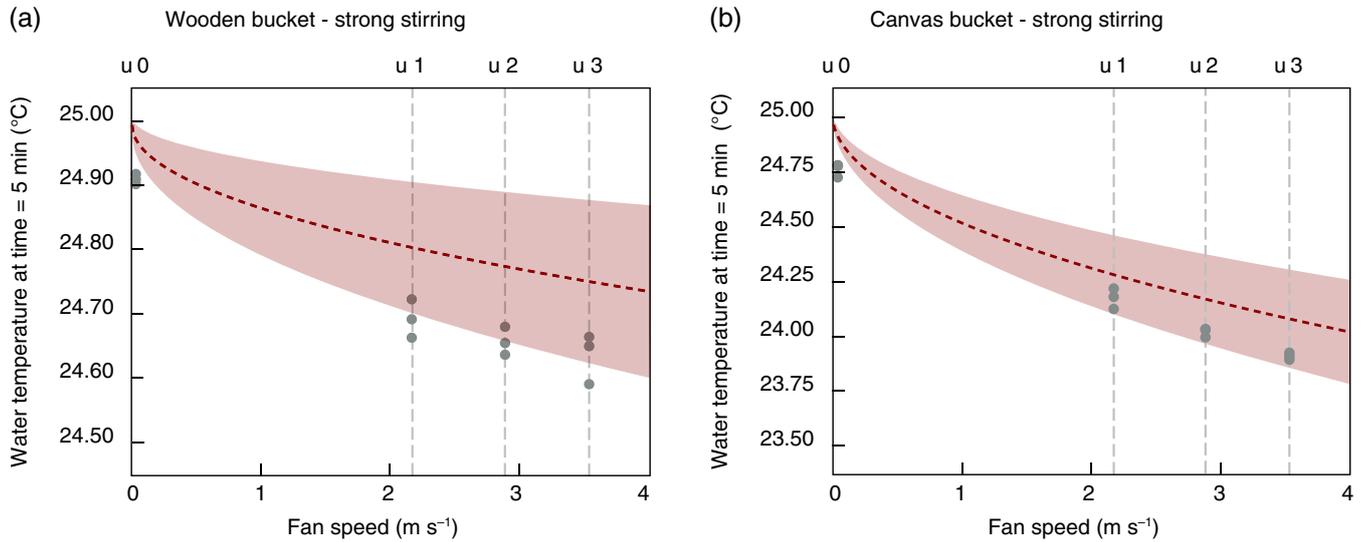
The results presented by Ashford allow a comparison of the characteristics of a range of different bucket types, and Figure 7 shows a selection of measurements reproduced from his Fig. 2. Two types of bucket showed much greater temperature changes than the others: the canvas bucket as tested in the present study (Met. Office Mk II) and the German scoop thermometer. A modern version of the German scoop is shown in Figure 1. The version tested by Roll (1951), and Ashford, is likely to be similar to this modern bucket. The capacity of the scoop is small (Table 1) and it is mostly made of metal. A rubber buffer with an air cushion covers the sides. Older versions had a leather cover with felt filling, but we do not know which type was used by either Ashford or Roll (1951). The base is double-walled with cork insulation between. An integral thermometer, mechanically

isolated to avoid breakage during use, means that the reading can be made immediately after hauling. Also plotted in Figure 7 are results from the new bucket design in versions with, and without, a lid. These new buckets were designed to minimise temperature change and show much lower rates of temperature change for a given water–wet-bulb temperature difference. Ashford describes the new bucket as canvas, but it has a copper vessel inside, which makes it partially insulated. The curvature of the lines becomes much less apparent for these buckets that show progressively smaller temperature change. This would be expected if the new designs were particularly effective at reducing heat loss by evaporation. Ashford reports that the temperature change was little affected if the outside of the bucket was wet or dry (note the Mk II canvas bucket cannot be kept dry). However, it may be that the curvature is simply not visible over the noise in the measurements for buckets with small rates of temperature change.

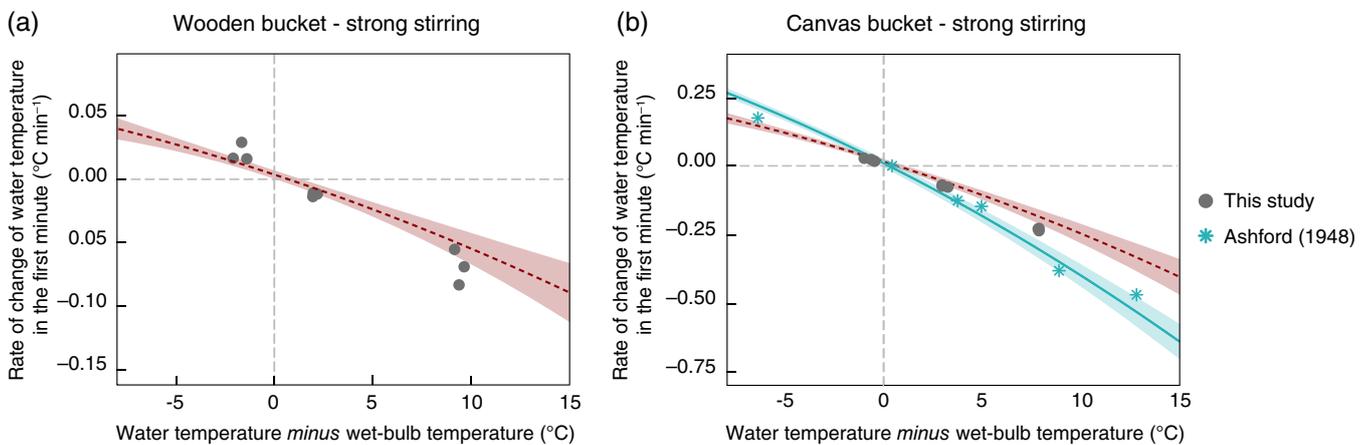
### 3.2.2. Roll (1951)

The German scoop thermometer was also studied in a wind tunnel at a range of wind speeds by Roll (1951, hereafter Roll). The measurements of temperature change after 1 min ( $\Delta t|_{1\text{min}}$ ) are presented in terms of a wind speed-dependent coefficient ( $\beta$ ) and an equivalent air ( $\theta_a$ ) and water temperature ( $\theta_b$ ):

$$\Delta t|_{1\text{min}} = \beta(\theta_a - \theta_b). \quad (2)$$



**Figure 5.** Measured (dots) and modelled (dotted line and shading) water temperature at time = 5 min as a function of the air (fan) speed for  $t_0 - t_a \sim 5^\circ\text{C}$ . Lines represent the median of the model output; shaded regions represent uncertainty at 95% confidence level in the model output. (a) Wooden bucket, strong stirring; (b) canvas bucket, strong stirring. Note change to y-axis scales. The 95% confidence level was computed from the sample quantiles (Hyndman and Fan, 1996) of the model ensemble. [Colour figure can be viewed at wileyonlinelibrary.com].



**Figure 6.** Measured (dots) and modelled (dotted line and shading) rate of change of water temperature at time = 1 min as a function of the water temperature *minus* wet-bulb temperature difference. Also shown is the measured (stars) and modelled (solid line and shading) rate of change for the Ashford (1948) results with the Mk II Met. Office bucket. Lines represent the median of the model output; shaded regions represent uncertainty at 95% confidence level in the model output. (a) Wooden bucket, strong stirring; (b) canvas bucket, strong stirring. Fan speed of  $\sim 3.5\text{ m s}^{-1}$  throughout, except for the experiments in Ashford (1948) for which the speed was  $\sim 9\text{ m s}^{-1}$ . Note change to y-axis scales. The 95% confidence level was computed from the sample quantiles (Hyndman and Fan, 1996) of the model ensemble. [Colour figure can be viewed at wileyonlinelibrary.com].

$\theta$  is defined following Rössler (1948):

$$\theta = t + \alpha \frac{e}{p} \tag{3}$$

to give:

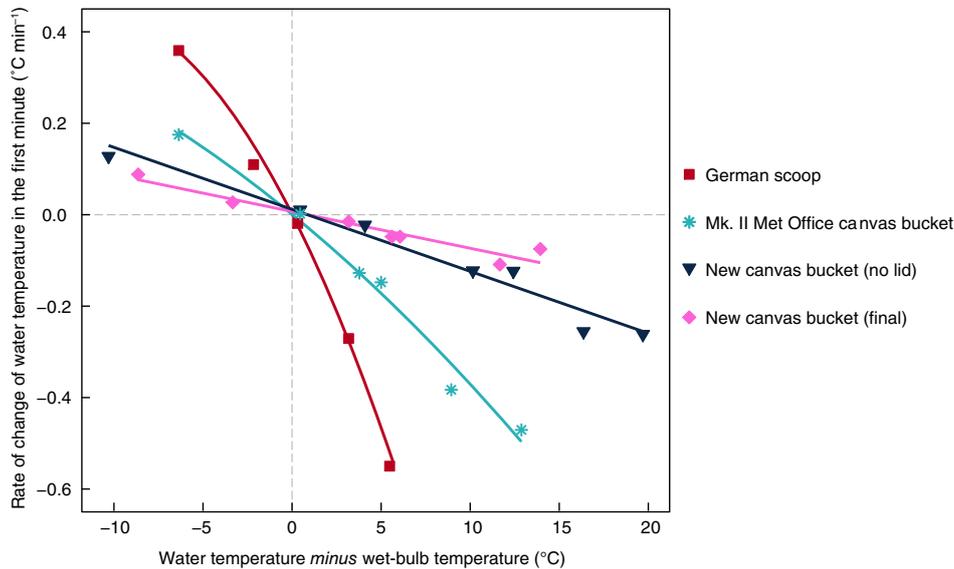
$$\Delta t|_{1min} = \beta \left\{ (t_a - t) + \frac{\alpha}{p} (e_a - e) \right\}, \tag{4}$$

where  $p$  is the atmospheric pressure (hPa) and  $\alpha = 1560\text{ (K)}$ . Equation (4) is of similar form to Eq. (1), if the small term for long-wave radiation is neglected in the latter. We can then interpret the term  $\beta$  as a heat transfer coefficient. Figure 8(a) (measurements read from hand-drawn Fig. 2 in Roll) shows the wind speed dependence of  $\beta$  for the scoop, from 2 to 19  $\text{m s}^{-1}$ . Roll's results show a much stronger airflow dependence of  $\beta$  (a power greater than 1) than that shown by the FP95 model (an approximate square-root dependence) which is tentatively confirmed for the canvas bucket by our measurements and those of Ashford (Figure 6(b)). Either an FP95-type model is not appropriate for the interpretation of Roll's measurements, or these measurements taken at higher wind speeds are indicating a stronger airflow dependence than the model, and also the canvas

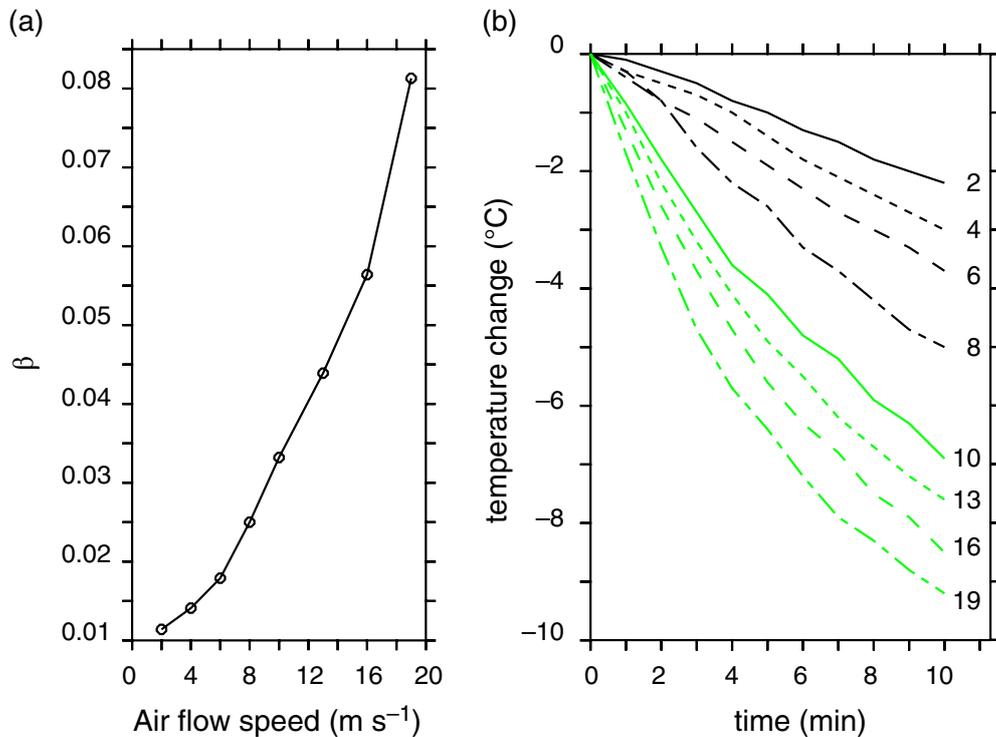
bucket measurements (both those of Ashford and our laboratory measurements).

The time evolution of the water temperature measured by Roll over the first 10 min is shown in Figure 8(b) for each of eight different wind speeds and an air–water temperature difference of  $-10^\circ\text{C}$ . The values plotted were read from Fig. 1 in Roll; the original graph consists of hand-drawn lines. Unfortunately Roll does not provide much information about the conditions under which the measurements were made. A small increase in the rate of temperature change over time is apparent at lower airflow speeds ( $2\text{--}8\text{ m s}^{-1}$ ), as was seen with the wooden bucket, which might indicate that the behaviour of the scoop is comparable to the wooden bucket. There also seems to be some separation between the measurements taken at lower airflow speeds and those at higher speeds ( $10\text{--}19\text{ m s}^{-1}$ ), which might indicate that conditions had changed over the course of the experiment.

One explanation for the stronger wind speed dependence might have been due to  $\beta$  having been estimated from measurements taken after 1 min. At the start of exposure to the atmosphere, partly-insulated buckets take time to establish temperature gradients within the bucket walls (Figure 4(a)) and if the time-scale for this process depends on the airflow, which seems reasonable, then aliasing of this signal might cause an apparent



**Figure 7.** Reproduction of Ashford (1948) results; values have been read from Fig. 2 in Ashford (1948). The plot shows the rate of change of water temperature at the first minute as a function of the initial water temperature *minus* the wet-bulb temperature for the German scoop (squares), the Mk II Met. Office canvas bucket (stars), the new canvas bucket (diamonds) and the new canvas bucket without the lid (triangles). Here the lines represent polynomial fit to the data, while in the original figure the lines were hand drawn. [Colour figure can be viewed at [wileyonlinelibrary.com](http://wileyonlinelibrary.com)].



**Figure 8.** Reproduction of Roll (1951) results; values have been read from Figs 1 and 2 in Roll (1951). (a)  $\beta$ , Eq. (2) as a function of airflow speed. (b) Temperature change over 10 min at eight different airflow speeds (as annotated for each line,  $\text{m s}^{-1}$ ). Values were read from the original figures every minute. [Colour figure can be viewed at [wileyonlinelibrary.com](http://wileyonlinelibrary.com)].

increase in  $\beta$  with airflow. This was investigated (noting that the ambient environmental conditions are uncertain) but  $\beta$  as estimated from Figure 8(b) shows strong wind-speed dependence throughout the first 10 min.

Despite the uncertainties around the Roll measurements it seems clear that the airflow dependence of temperature change measured in the wind tunnel is greater than that predicted by the FP95 model, which predicts an approximate square-root dependence. Whilst the dimensions, design and thermal properties of the scoop are rather different to those of the wooden bucket, all of these differences could be accounted for, and the wind speed functional dependence would remain similar.

FP95 assume that the incident flow is laminar. They note that turbulence in the incident flow would increase the heat transfer coefficient, and further note that turbulent incident flow was

likely for measurements made on a ship. It is also likely for our measurements in the laboratory, and for the two sets of wind tunnel results, but the intensity of turbulence for each of these sets of measurements is unknown. At the higher wind speeds measured by Roll the incident flow would certainly have been turbulent and his stronger speed dependence could potentially be explained by an increasing intensity of turbulence with wind speed giving an increased heat transfer coefficient (Lowery and Vachon, 1975). This means that comparing measurements made in different wind tunnels, and even at different flow speeds within the same wind tunnel, is difficult, and will reduce the confidence with which any derived heat exchange characteristics can be applied to measurements at sea for which the intensity of the turbulence is always unknown. These considerations suggest that the uncertainty in the relationship between the relative wind

speed and the heat transfer coefficient is larger than has been so far assumed (e.g. by Kennedy *et al.* (2011)). This uncertainty arises from at least two components: from the unknown reduction of the ambient relative wind speed by sheltering; and from the unknown intensity of turbulence. Further work is needed to clarify the impact of this additional uncertainty on estimates of global multi-decadal temperature changes.

#### 4. Summary and conclusions

Tests in the laboratory show that the FP95-type models used to estimate the biases in bucket-derived SST measurements work well, when conditions are similar to those assumed in the models. At the range of airflows tested (a maximum of  $\sim 3.5 \text{ m s}^{-1}$ ), the model for the canvas bucket predicted a temperature change within the estimated experimental uncertainty for a range of air–water temperature differences (Figure 4) and airflow speeds (Figure 5). For the wooden bucket, although close to the limit of the estimated uncertainty, the model slightly underestimates the observed temperature change. We conclude that the models are able to reasonably reproduce the temperature change measured for the two buckets. The model simulations helped us to understand an observed initial period of reduced temperature change for the wooden buckets (Figure 4(a)). This was caused by the time taken for heat to be conducted through the bucket walls, an effect included in the wooden bucket model. However, the assumptions made in the model derivation may in practice be rather limiting. Our measurements showed that if the sample is not vigorously stirred, then the temperature change will be much lower than when the water is well-mixed, particularly when the rate of temperature change is large. This was particularly obvious for the canvas bucket filled with water substantially warmer than the ambient air temperature (Figure 4(d)).

We reviewed the results of some previous measurements of temperature change for a range of different bucket types taken in wind tunnels (Ashford, 1948; Roll, 1951). Ashford made measurements using the same canvas bucket used in this study, but at a substantially higher airflow speed ( $\sim 9 \text{ m s}^{-1}$ ). The temperature change for a given thermal forcing (defined as the water temperature *minus* wet-bulb temperature,  $\Delta t_{wb}$ ) was only slightly larger than that measured in the laboratory at  $\sim 3.5 \text{ m s}^{-1}$  (Figure 5(b)), suggesting that the approximate square-root dependence of the heat transfer on airflow speed used by FP95 was reasonable. This modest airflow dependence was however not supported by the results of Roll, who made measurements for a single bucket type (the German scoop, Figure 1) at a wide range of wind speeds. Roll's results showed a much larger increase in heat transfer with airflow, a dependence stronger than linear. A possible reason for such inconsistency in the airflow dependence of heat exchange was suggested by FP95: they note that any turbulence in the incident flow will act to increase their heat exchange coefficient. The strong increase in temperature change observed by Roll with increasing airflow could reasonably be explained by an increase in the turbulent intensity of the incident flow with airflow. This explanation however leads to the problematic conclusion that any estimates of heat transfer coefficients will be affected by the particular circumstances of the experimental, or shipboard, conditions.

Ashford took measurements of temperature change in a range of different bucket types. His results clearly showed a wide range of different heat exchange characteristics (Figure 7), as did our measurements for wooden and canvas buckets (Figure 6). The heat exchange characteristics are broadly predictable for each bucket type and depend on the geometry, size and degree of insulation.

The FP95 formulation is fairly straightforward to adapt for different bucket types. The cylindrical bucket geometry can be specified, as can the degree of insulation of the bucket walls. Modern buckets for which the outer surface would not remain wet could be modelled by setting  $f_e < f_t$  in Eq. (1). The heat transfer coefficients  $h_e$  and  $h_t$  are formulated based on a Nusselt number. There are empirical formulations for the Nusselt number

that are likely to be applicable in a wider range of conditions (e.g. Churchill and Bernstein, 1977). However, the problem of unknown intensity of turbulence in the incident flow, and how that turbulence might depend on local obstacles for any particular measurement, remains. Despite this, the models might be expected to be effective at estimating the relative rates of temperature change for different types of bucket.

We need to consider the impact of our conclusions on the FP95-derived bias adjustments used in HadISST, HadSST3 and COBE-SST2. FP95 were well aware of the difficulties associated with quantifying biases in historical SSTs and attempted to design their bias adjustment methodology to be robust to the uncertainties they identified. FP95 conclude that their bias adjustment fields are 'fairly insensitive to uncertainties such as the size of the bucket or the details of its exposure on deck'. This is because the parameters assumed to be characterized by the largest uncertainty in the model (i.e. the mix of bucket types and the assumed exposure time for uninsulated canvas buckets) are estimated such that the internal consistency of the observations is improved. The mix of bucket types (wooden or canvas) is calculated to improve the agreement between the adjusted SST and NMAT anomalies in the Tropics (FP95) and the exposure time for canvas buckets is adjusted to give more similar seasonal cycles before and after World War II. The resulting adjustment fields are only weakly dependent on the highly uncertain airflow around the bucket, and show a much stronger dependence on the water temperature *minus* wet-bulb temperature (Kent *et al.*, 2017). Constraining the uncertain parameters in FP95 models to improve the internal consistency of the data leads to reasonable large-scale estimates of the biases in historical SST bucket observations (Kent *et al.*, 2016).

We conclude therefore that new measurements of temperature change of water samples in buckets made on board ships at sea would be more valuable than additional measurements made, for example, in wind tunnels. However, it would be challenging to make enough measurements with different types of buckets, in different environmental conditions, and in differently exposed locations on different types of ships to fully explore the dependencies.

A good approach to estimating bias adjustments for historical bucket measurements would be to directly estimate the adjustments from the observations themselves, guided by the dependencies shown by the physically based models. From our results, and those of Ashford and Roll, we conclude that the adjustments are likely to be strongly dependent on  $\Delta t_{wb}$ , as are the FP95-derived fields used by HadISST, HadSST3 and COBE-SST2. The relationship between temperature change and  $\Delta t_{wb}$  will be scaled depending on bucket type and will vary with measurement protocols (relating to the way the measurement was made – including how quickly – and whether the bucket was sheltered from the sun or the wind and whether the sample was well-mixed). On a secondary level, the temperature change will also depend on ambient conditions not related to  $\Delta t_{wb}$  (including airflow speed, the intensity of turbulence in incident flow, and solar radiation). Such approaches have not been explored in the past but are now possible because of a much increased number of observations (Freeman *et al.*, 2017), improved metadata (Carella *et al.*, 2017) and increased computer capacity.

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## Appendix: Extended summary of each experiment

Experiment type	Bucket type	Stirring type	R (%)	$t_a$ (°C)	$t_0$ (°C)	$t_0 - t_a$ (°C)
dt1	Wooden	Strong	60.1 ± 0.1	20.77 ± 0.07	25.09 ± 0.20	4.32 ± 0.21
dt2	Wooden	Strong	71.9 ± 0.2	20.95 ± 0.07	19.38 ± 0.04	-1.57 ± 0.08
dt3	Wooden	Strong	71.1 ± 2.1	21.01 ± 0.15	15.63 ± 0.13	-5.38 ± 0.20
dt1	Wooden	Weak	59.0 ± 0.9	20.85 ± 0.13	25.27 ± 0.14	4.42 ± 0.20
dt2	Wooden	Weak	72.0 ± 0.9	21.14 ± 0.25	19.39 ± 0.00	-1.75 ± 0.25
dt3	Wooden	Weak	72.7 ± 0.3	21.04 ± 0.07	15.69 ± 0.15	-5.35 ± 0.17
u0	Wooden	Strong	56.9 ± 0.4	22.15 ± 0.15	25.04 ± 0.09	2.89 ± 0.18
u1	Wooden	Strong	61.7 ± 0.9	21.07 ± 0.18	25.04 ± 0.05	3.97 ± 0.19
u2	Wooden	Strong	70.9 ± 0.3	20.65 ± 0.12	25.05 ± 0.07	4.40 ± 0.14
u3	Wooden	Strong	60.1 ± 0.1	20.77 ± 0.07	25.09 ± 0.20	4.32 ± 0.21
dt1	Canvas	Strong	69.5 ± 0.9	20.96 ± 0.17	25.00 ± 0.13	4.04 ± 0.22
dt2	Canvas	Strong	64.3 ± 0.6	20.70 ± 0.08	19.27 ± 0.10	-1.43 ± 0.13
dt3	Canvas	Strong	63.5 ± 0.6	20.93 ± 0.21	15.67 ± 0.07	-5.26 ± 0.22
dt1	Canvas	Weak	60.2 ± 0.0	20.75 ± 0.07	25.01 ± 0.05	4.26 ± 0.09
dt2	Canvas	Weak	64.9 ± 0.4	20.77 ± 0.22	19.34 ± 0.07	-1.43 ± 0.23
dt3	Canvas	Weak	64.2 ± 0.6	20.84 ± 0.05	15.56 ± 0.14	-5.28 ± 0.15
u0	Canvas	Strong	68.0 ± 1.3	22.10 ± 0.12	25.08 ± 0.11	2.98 ± 0.17
u1	Canvas	Strong	70.6 ± 0.6	21.31 ± 0.21	24.83 ± 0.11	3.52 ± 0.24
u2	Canvas	Strong	71.1 ± 0.3	20.93 ± 0.08	24.95 ± 0.09	4.02 ± 0.12
u3	Canvas	Strong	69.5 ± 0.9	20.96 ± 0.17	25.00 ± 0.13	4.04 ± 0.22

R: relative humidity (%),  $t_a$ : ambient air temperature,  $t_0$ : water temperature at time = 0 min (°C). The experiment corresponding to each row in the table was repeated three times and was run for 15 min: the relative humidity and the ambient air temperature represents the mean over 15 min and all the repetitions; the water temperature at time = 0 represents the mean over all the repetitions. The uncertainty in each variable is reported at one standard deviation.

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