1 Mechanisms for a nutrient-conserving carbon pump in a seasonally

2 stratified, temperate continental shelf sea

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Abstract

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Continental shelf seas may have a significant role in oceanic uptake and storage of carbon dioxide (CO₂) from the atmosphere, through a 'continental shelf pump' mechanism. The northwest European continental shelf, in particular the Celtic Sea (50°N 8°W), was the target of extensive biogeochemical sampling from March 2014 to September 2015, as part of the UK Shelf Sea Biogeochemistry research programme (UK-SSB). Here, we use the UK-SSB carbonate chemistry and macronutrient measurements to investigate the biogeochemical seasonality in this temperate, seasonally stratified system. Following the onset of stratification, near-surface biological primary production during spring and summer removed dissolved inorganic carbon and nutrients, and a fraction of the sinking particulate organic matter was subsequently remineralised beneath the thermocline. Water column inventories of these variables throughout 1.5 seasonal cycles, corrected for air-sea CO₂ exchange and sedimentary denitrification and anammox, isolated the combined effect of net community production (NCP) and remineralisation on the inorganic macronutrient inventories. Overall inorganic inventory changes suggested that a significant fraction (>50%) of the annual NCP of around 3 mol-C m⁻² yr⁻¹ appeared to be stored within a long-lived organic matter (OM) pool with a lifetime of several months or more. Moreover, transfers into and out of this pool appeared not to be in steady state over the one full seasonal cycle sampled. Accumulation of such a long-lived and potentially C-rich OM pool is suggested to be at least partially responsible for the estimated net air-to-sea CO₂ flux of ~1.3 mol-C m⁻² yr⁻¹ at our study site, while providing a mechanism through which a nutrient-conserving continental shelf pump for CO₂ could potentially operate in this and other similar regions.

1. Introduction

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46 The ocean is an important buffer for atmospheric carbon dioxide (CO₂), naturally storing a large pool of dissolved inorganic carbon (C_T), and also absorbing about a quarter of annual 47 48 anthropogenic CO₂ emissions (Le Quéré et al., 2009) and thereby mitigating their climatic 49 impacts (IPCC, 2013). The timescale on which the ocean's surface mixed layer equilibrates 50 CO₂ with the atmosphere varies spatially, but is typically from a few months to a year 51 (Broecker and Peng, 1974; Jones et al., 2014). Transfer from the surface to the deep interior 52 is a much slower rate-limiting step, which is performed by the solubility, soft tissue, 53 carbonate and mixed-layer 'pumps' in the open ocean (Volk and Hoffert, 1985; Dall'Olmo et 54 al., 2016). Tsunogai et al. (1999) first proposed the existence of an additional 'continental shelf pump' (CSP) that facilitates C_T transfer from shallow coastal surface waters into the 55 56 interior ocean, and thereby enhances local oceanic CO₂ uptake. The CSP operates through 57 autotrophic conversion of C_T into organic matter (OM) in a continental shelf sea, which 58 drives a compensatory net air-to-sea CO₂ flux (Chen and Borges, 2009; Chen et al., 2013; 59 Laruelle et al., 2014). In order to prevent the fixed CO₂ from being quickly released back into 60 the atmosphere, the carbon-enriched shelf waters need to be transported into the open ocean, 61 beneath the seasonal thermocline (Tsunogai et al., 1999; Thomas et al., 2004). 62 The Celtic Sea is a seasonally stratified, temperate sea that forms part of the northwest 63 European continental shelf (Fig. 1). Previous studies of its surface waters have shown that it 64 acts as a net sink of atmospheric CO₂ (Frankignoulle and Borges, 2001; Marrec et al., 2015), 65 which is typical for this type of sea (Borges, 2005; Laruelle et al., 2014). There must 66 therefore be a net flux of carbon (C) out of the shelf sea water column, for example through 67 OM export to sediments, or advective exchange with the open ocean. Celtic Sea sediments 68 could form part of this C sink, as although they predominantly contain sandy material there is 69 some organic matter accumulation (de Haas et al., 2002; Suykens et al., 2011; Diesing et al., 70 2017). Physical mechanisms that could transport shelf waters into the open ocean have also 71 been identified here (Cooper and Vaux, 1949; Shapiro et al., 2003; Ivanov et al., 2004) and in 72 other similar shelf environments (Álvarez-Salgado et al., 2001). In either case, the question 73 arises of how the nutrient supply is sustained that drives the net C uptake. As essential 74 nutrients such as dissolved inorganic nitrogen and soluble reactive phosphorus (DIN and 75 SRP) are also incorporated into OM, but not replenished through air-sea exchange, the net 76 removal of C_T should be accompanied by a corresponding loss of DIN and SRP from the 77 shelf waters (Thomas et al., 2004; Bozec et al., 2006). Further DIN may be lost from the shelf

78 waters through denitrification and/or anammox in the shelf sediments (Hydes et al., 2004; 79 Kitidis et al., 2017). Thus, in order to sustain the shelf sea nutrient inventories in the presence 80 of net CO₂ uptake from the atmosphere, any incoming nutrient supply would therefore need 81 to be accompanied by a stoichiometric deficit of carbon relative to outgoing waters. The 82 mechanism by which this occurs remains uncertain. One possibility is that riverine inputs 83 provide high-nutrient, low- $C_{\rm T}$ waters to the shelf sea. There are also high-nutrient waters in 84 the Irish Sea to the northeast of our study region (Gowen et al., 2002). However, these 85 influences are not thought to be substantial for the Celtic Sea, as on-shelf lateral circulation is 86 relatively slow (Pingree and Le Cann, 1989), and most of the riverine nutrient inputs are 87 quickly removed by biogeochemical processes close to land (Ruiz-Castillo et al., 2017, this 88 issue). Moreover, riverine nutrient inputs to shelf seas tend to be small relative to inputs at the 89 continental margin, for example through upwelling and internal waves at the shelf break 90 (Wollast, 1998). Rivers flowing into the Celtic Sea (e.g. from southern Ireland) also typically 91 have high total alkalinity (A_T) due to chalk and limestone bedrock in their catchments 92 (McGrath et al., 2015). This high A_T supports high C_T concentrations, so the required C 93 deficit relative to nitrogen (N) and phosphorus (P) is unlikely to occur in these waters. 94 Here, we investigate the hypothesis that OM storage coupled with differences in the 95 stoichiometry of OM remineralisation relative to its production could provide the 96 stoichiometric inconsistency required to sustain the nutrient supply in the Celtic Sea, as has been suggested to occur in similar marine systems (e.g. Álvarez-Salgado et al., 2001; Bozec 97 98 et al., 2006). This would require the stoichiometric C:N:P ratio for remineralisation to have a 99 lower C coefficient than in the equivalent ratio for net community production (NCP). The 100 shelf nutrient inventory could therefore be sustained through some combination of enhanced 101 remineralisation of non-C nutrients, or the build-up of a stock of C-rich OM that could be 102 transported into the adjacent open ocean prior to complete remineralisation, thereby acting as 103 a nutrient-conserving CO₂ sink. Our approach is to investigate the distributions and 104 inventories of $C_{\rm T}$ and dissolved inorganic nutrients, which record the integrated effects of 105 NCP and OM remineralisation, along with other processes such as air-sea CO₂ exchange. We 106 use new seasonally resolved C_T and nutrient observations in this way to estimate the annual 107 magnitude and stoichiometry of NCP and remineralisation at a site in the central Celtic Sea. 108 We subsequently infer seasonal and interannual changes in OM inventories and 109 stoichiometry, and consider mechanisms that could support C export while conserving 110 nutrients, and therefore a CSP.

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2. Methods

113	2.1.	Research	cruises
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- Seawater samples were collected during 10 research cruises to the Celtic Sea on the UK
- research ships RRS *Discovery* and RRS *James Cook*, from March 2014 to August 2015
- (Supp. Table S1). All of our datasets are freely available from the British Oceanographic
- 117 Data Centre (www.bodc.ac.uk).

118 2.1.1. Marine carbonate system

- Samples for C_T and A_T were collected via silicone tubing into 250 ml borosilicate glass
- bottles either from the samplers (Ocean Test Equipment) positioned on a CTD rosette frame,
- or from the ship's underway seawater supply, following an internationally established
- protocol (Dickson et al., 2007). Each bottle was sealed shut with a greased ground glass
- stopper after introducing a 2.5 ml air headspace and sterilising the sample with 50 µl of
- saturated mercuric chloride solution. All samples were stored in the dark until analysis.
- Measurements of $C_{\rm T}$ and $A_{\rm T}$ were carried out at the University of Southampton between June
- 126 2014 and January 2016, using several different instruments. The results were calibrated using
- measurements of batches 128, 135, 136, 138, 141, 142, 144, 146 and 148 of seawater
- certified reference material (CRM) obtained from A. G. Dickson at Scripps Institution of
- Oceanography, San Diego, USA (Dickson et al., 2003).
- Seawater $C_{\rm T}$ was measured using either the Versatile INstrument for the Determination of
- Total inorganic carbon and titration Alkalinity (VINDTA 3C, #024 and #038, Marianda,
- Germany) or the Dissolved Inorganic Carbon Analyzer AS-C3 (Apollo SciTech Inc., USA).
- Both of these instruments first acidify a seawater subsample with excess 10% phosphoric
- acid, to convert all C_T into aqueous CO_2 . Nitrogen gas is then bubbled through to carry the
- 135 CO₂ to a detector. In the VINDTA 3C, the detection is by coulometric titration (CM5014 CO₂
- coulometer, UIC Inc., USA), while the AS-C3 uses infrared absorption (LI-7000 CO₂/H₂O
- Analyzer, LI-COR, USA). Based on CRM measurements throughout the entire UK-SSB
- programme, the 1σ precision for $C_{\rm T}$ was \pm 2.6 μ mol kg⁻¹ for the VINDTA 3C (number of
- 139 CRM measurements (n) = 547), and \pm 4.0 μ mol kg⁻¹ for the Apollo AS-C3 (n = 135).

- 140 The A_T was measured using either the VINDTA 3C or the Total Alkalinity Titrator AS-ALK2
- 141 (Apollo SciTech Inc., USA). Both instruments determine $A_{\rm T}$ by an open-cell, potentiometric
- titration of a seawater subsample with 0.1 M hydrochloric acid. The A_T values from the
- 143 VINDTA 3C measurements were recalculated using a modified Gran plot approach
- 144 (Humphreys, 2015a) as implemented by Calkulate v0.1.2 (freely available from
- 145 https://github.com/mvdh7/calkulate). The CRM measurements through the entire UK-SSB
- programme indicated a 1σ precision for $A_{\rm T}$ of \pm 2.7 µmol kg⁻¹ for the VINDTA 3C (n=543),
- and $\pm 3.9 \,\mu\text{mol kg}^{-1}$ for the Apollo AS-ALK2 (n = 109).
- We performed a cross-over analysis using XOVER v1.0.1 (Humphreys, 2015b), which is
- freely available from https://github.com/mvdh7/xover, to assess our C_T and A_T measurement
- accuracy relative to historical data in GLODAPv2 (Olsen et al., 2016). This used only data
- from deeper than 250 m (i.e. off the shelf, beneath the base of the thermocline, and a
- maximum cross-over distance of 200 km). For C_T , there was a mean \pm standard deviation
- 153 (SD) offset of $13.6 \pm 7.6 \,\mu\text{mol kg}^{-1}$ between the combined SSB dataset and nine GLODAPv2
- cruises from 1994 to 2008. Despite the large size of this offset, it does not indicate any
- problem with our measurements, because an ordinary least squares linear regression between
- the offsets and the sampling date reveal a decrease in offset size at a rate of about 1 µmol kg⁻¹
- yr⁻¹ ($r^2 = 0.66$), corresponding to the anthropogenic C_T increase, and reaching a value of 0
- close to the date of the SSB cruises. For $A_{\rm T}$, the equivalent offset was $6.8 \pm 7.1 \,\mu{\rm mol~kg^{-1}}$ for
- nine cruises from 1983 to 2008, which is close to the minimum adjustment limit of 6 µmol
- 160 kg⁻¹ for this variable in GLODAPv2 (Olsen et al., 2016). In general, cruises closer in time to
- 161 SSB sampling had a smaller A_T offset. We conclude that our measurements are accurate to
- within their precision relative to historical observations. Indeed, the internal consistency of
- data within the GLODAPv2 compilation is reported as ± 4 and ± 6 µmol kg⁻¹ for C_T and A_T
- respectively (Olsen et al., 2016).
- 165 2.1.2. Other variables
- Macronutrient samples were collected into 60 ml high density polyethylene (HDPE, Nalgene)
- bottles that had been aged and acid washed before sampling. The measurements of nitrate,
- nitrite, ammonium, and soluble reactive phosphorus (SRP) were carried out at sea during the
- 169 UK-SSB cruises, immediately after sample collection, using the Plymouth Marine Laboratory
- 5-channel Bran and Luebbe AAIII system, following Woodward and Rees (2001). The
- instrument was calibrated using in-house nutrient standards, and accuracy was monitored by

- 172 regular comparisons with Nutrient Reference Material obtained from KANSO Technos
- 173 (Japan). We define dissolved inorganic nitrogen (DIN) as the sum of the concentrations of
- nitrate, nitrite and ammonium. The 1σ precisions were ± 0.13 µmol kg⁻¹ and
- \pm 0.0073 μmol kg⁻¹ for DIN and SRP respectively.
- Water column hydrography was measured by Sea-Bird conductivity-temperature-depth
- 177 (CTD) sensors attached to the sampling rosette frame, to determine the temperature (T in °C),
- practical salinity (S) and depth for each discrete biogeochemical sample. The sea surface
- hydrography (*T* and *S*) was measured continuously by Sea-Bird sensors through the underway
- seawater supply at a nominal depth of 6 m. The T and S sensor data for both the underway
- and CTD systems were calibrated using a set of discrete samples collected during each cruise.
- We consider the 1σ uncertainty in S to be \pm 0.002 (Le Menn, 2011). Further information on
- the sensor measurements and processing can be found in the cruise reports.
- 184 2.1.3. Unit conversions
- The C_T and A_T measurements were determined in units of umol kg⁻¹, while DIN and SRP
- were reported in mmol m⁻³ under analysis conditions (i.e. at 20°C). For the column inventory
- analysis, we converted the C_T , DIN and SRP measurements into units of mmol m⁻³ at the *in*
- 188 situ temperature. We calculated all densities for these conversions using the Gibbs-SeaWater
- 189 (GSW) Oceanographic Toolbox v3.05.5 (McDougall and Barker, 2011).

190 2.2. Central Celtic Sea mooring

- 191 A temperature-salinity (TS) chain (a series of sensors positioned vertically along a cable
- moored to the seafloor) was also deployed on a Cefas SmartBuoy mooring at the Central
- 193 Celtic Sea site (CCS) from March 2014 to July 2015 (Wihsgott et al., 2016). This measured
- seawater T and S at a vertical resolution of 2.5 m in the pycnocline and 5-20 m resolution in
- the surface and bottom layers. The temporal resolution was 5 minutes. A combination of
- sensor types were used: Star Oddi DST Centi temperature loggers, Star Oddi Starmon mini-
- underwater temperature recorder and RBRsolo T temperature logger, along with SBE16+
- 198 SeaCAT CTD and SBE37 MicroCAT CTD sensors. We calculated mixed layer depth (z_{ml} in
- m) as the shallowest depth where the potential density increased by 0.02 kg m⁻³ from its value
- 200 at 10 m.

201 **2.3.** Auxiliary datasets

- We obtained 0.75°×0.75° gridded atmospheric/surface ocean reanalysis data at 3-hourly
- temporal resolution from the ERA-Interim data product (Dee et al., 2011), which is produced
- by the European Centre for Medium-Range Weather Forecasts (ECMWF). Data were
- downloaded from ECMWF on 1st August 2016 for the following variables, over the period
- 206 from 1st March 2014 to 31st December 2015: sea surface temperature (SST in °C), mean sea
- level pressure (P_{baro} in bar), 10 metre U wind component (U_{10u} in m s⁻¹), 10 metre V wind
- component (U_{10v} in m s⁻¹), and 2 metre dewpoint temperature (T_{dp} in °C). We calculated the
- wind speed at 10 m above the sea surface (U_{10} , in m s⁻¹) from its components U_{10u} and U_{10v} .

2.4. Central Celtic Sea site

- 211 Most of the analysis presented here was carried out at CCS, which was the most frequently-
- occupied location during the UK-SSB sampling campaign. All samples collected within a
- 213 12 km radius of 49.4°N, 8.54°W were considered to represent CCS (Fig. 1). This radius was
- selected such that sufficient data were captured to constrain the interior biogeochemical
- seasonality, whilst deviating as little as possible from the site itself. We then manually
- 216 grouped the CCS observational dataset based on the sampling date: we selected a set of date
- boundaries that minimised the width of each time interval, while ensuring that there were
- enough data within each interval to determine the vertical profile of each biogeochemical
- variable, resulting in 15 time intervals (Table 1).
- Vertical profiles of biogeochemical variables at each site were typically homogeneous during
- the winter months, and had a two-layer shape during the stratified summer. As such, it was
- 222 possible to fit error functions to many variables, in order to objectively determine values for
- 223 each variable in the deep and surface layers, along with the depth of the divide. We thus
- generated a fit to all data for each variable at each time interval at each site using an equation
- of the form:

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$$v = v_0 + v_1 \operatorname{erf}\left(\frac{z - z_0}{z_1}\right)$$
 (1)

- where v and z are the values and depths respectively of the measurements of each variable,
- 228 erf is the Gauss error function, and v_0 , v_1 , z_0 and z_1 are coefficients that were adjusted to find
- the least-squares best fit to these data. These coefficients quantify useful properties: z_0 is the
- depth of the boundary between the surface and deep layers; z_1 indicates how sharp or diffuse
- 231 the boundary is, with a smaller value indicating a transition over a narrower depth range; and

- 232 the surface and deep layer values of the variable v are given by $(v_0 v_1)$ and $(v_0 + v_1)$
- respectively (see Supp. Table S2 and Supp. Figs. S1–S4).

234 **2.5.** Air-sea CO₂ flux

- We calculated pCO_2^{sw} from the A_T , C_T , S, T, pressure, silicate and SRP observations using
- version 1.1 of the CO₂SYS program for MATLAB (van Heuven et al., 2011). We calculated
- Free scale pH in the same way (Supp. Fig. S5). We used the carbonic acid dissociation
- constants of Lueker et al. (2000), the bisulfate dissociation constant of Dickson (1990) and
- the boron:chlorinity ratio of Lee et al. (2010). The uncertainties in the input variables resulted
- in a 1σ precision of \pm 8.1 µatm for calculated pCO_2^{sw} . Statistical interpolations of these
- calculated pCO_2^{sw} values were limited by low temporal sampling resolution, resulting in key
- 242 features such as the 2014 spring bloom being partly missed, and therefore inaccurate
- estimates of the time-integrated air-sea CO₂ flux. We therefore created an empirical
- 244 algorithm to predict the surface pCO_2^{sw} time series at CCS from the reanalysis SST data. The
- 245 algorithm used different equations for different time periods and SST ranges (Table 2) to
- 246 account for temporal variation in the surface pCO_2^{sw} -SST relationship resulting from changes
- in biological production and vertical mixing. The mean \pm SD difference between the
- 248 algorithm-predicted and measured pCO_2^{sw} was -4 ± 19 µatm across all 71 measurements
- above 25 m depth at CCS, which is similar to the uncertainties quoted for published pCO_2^{sw}
- algorithms in this region (e.g. Marrec et al., 2015).
- We obtained measurements of the atmospheric CO_2 dry air mole fraction (xCO_2) for the
- 252 Greenhouse Gases Reference Network Site at Mace Head, County Galway, Ireland
- 253 (Dlugokencky et al., 2015). Geographically, this is the closest such observation site to our
- study area. We applied a piecewise cubic Hermite interpolating polynomial (PCHIP) function
- 255 (Fritsch and Carlson, 1980; Kahaner et al., 1988) to their monthly mean data to predict xCO₂
- in parts per million (ppm) for any given date. We estimated the uncertainty in xCO_2 values
- interpolated from the monthly means, ± 2.8 ppm, as the standard deviation of the residuals
- between all of the original individual measurements and the interpolated xCO_2 values at the
- same time points. We then converted xCO_2 into atmospheric CO_2 partial pressure (pCO_2 ^{atm})
- using data from the ERA-Interim reanalysis (Dee et al., 2011) at CCS. First, we estimated the
- water vapour pressure (P_v in bar) from T_{dp} (Alduchov and Eskridge, 1996; Lawrence, 2005):

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$$P_{\rm v} = 610.94 \exp[17.625 \, T_{\rm dp} / (243.04 + T_{\rm dp})] \, 10^{-5}$$
 (2)

The atmospheric partial pressure of CO_2 (pCO_2 ^{atm} in μ atm) was then given by:

$$264 pCO_2^{\text{atm}} = xCO_2 (P_{\text{baro}} - P_{\text{v}}) / 1.01325 (3)$$

- We calculated the air-sea CO₂ flux (*F*, in μmol m⁻² hr⁻¹) following Wanninkhof (2014), using
- 3-hourly resolution data from ECMWF, the algorithm-predicted pCO_2^{sw} (Table 2), and
- 267 pCO_2^{atm} calculated from xCO_2 (Eq. 3). The flux equation is:

$$F = k \alpha \left(p \operatorname{CO}_{2}^{\text{sw}} - p \operatorname{CO}_{2}^{\text{atm}} \right) \tag{4}$$

- where α is the solubility of CO₂, evaluated following Weiss (1974) and converted to units of
- 270 mol m⁻³ atm⁻¹ using the GSW toolbox (McDougall and Barker, 2011), and k is the gas
- 271 transfer velocity in m hr⁻¹:

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$$k = 0.00251 U_{10}^2 (660/D)^{0.5}$$
 (5)

- where D is the dimensionless Schmidt number for CO_2 at the surface layer seawater
- temperature (Wanninkhof, 2014). The flux calculations were carried out in MATLAB
- 275 (MathWorks) using the 'co2flux' program available from https://github.com/mvdh7/co2flux.

276 **2.6.** Column inventories

- 277 The column inventory of each biogeochemical variable was determined at each time interval
- by integrating the fitted profile for the variable (in mmol m⁻³) from the surface to the seafloor
- depth, taken as the mean seafloor depth across all CCS sampling stations (Z = 146.9 m). We
- 280 then attempted to remove the influence of processes other than NCP and remineralisation on
- 281 the column inventories. To correct for air-sea gas exchange, we subtracted the air-sea CO₂
- 282 flux integrated between each time point and winter 2014 (C_{atm}) from each time point's raw C_{T}
- inventory. Hydes et al. (2004) suggest a constant denitrification rate of 0.1 mmol-N m⁻² day⁻¹
- for the Celtic Sea. We therefore subtracted the product of this rate and the number of days
- between each time point and winter 2014 from the DIN inventory at each time point to
- 286 correct for the potential loss of DIN through denitrification (DIN_{dnf}). Kitidis et al. (2017)
- 287 more recently reported a similar mean flux of DIN from sediments into the seawater due to
- denitrification/anammox processes at a site near CCS. Although these authors identified some
- seasonal variability in this value, but we did not include this in our analysis as it was too
- small to alter our interpretation and conclusions.

2.7. Uncertainty analysis

The uncertainties in all of the final results, reported as 95% confidence intervals, were calculated using Monte Carlo simulations. Explicitly, we replicated the column inventory analysis 1000 times, each time adding different random offsets to each individual measurement. These offsets were normally distributed, with a SD equal to the 1σ analytical uncertainty for each measurement. The random noise was added both to the measurements of each variable. We report the 95% uncertainty in each column inventory as double the SD of all of its replicates.

2.8. Salinity correlations

We quantified the seasonally changing patterns in C_T , DIN and SRP remineralisation across the Celtic Sea by determining their correlations with practical salinity in the deep layer. For these correlations, we used only data satisfying the following conditions: (1) salinity > 35.2, to exclude near-coastal waters with nutrient-salinity relationships different from the majority of the shelf; (2a) where seafloor depth < 200 m (i.e. on-shelf), sample depth > 60 m, to exclude the surface layer where DIN and SRP are reduced to virtually zero in the summer months; and (2b) where seafloor depth > 200 m (i.e. off-shelf), 150 m < sample depth < 200 m, again to exclude the surface layer (which was deeper off-shelf) and also deeper waters that were less strongly affected by seasonal remineralisation. The potential density of the off-shelf 150 to 200 m layer throughout the year was similar to the on-shelf winter potential density, supporting our choice of this as the off-shelf endmember for the correlations. Summary statistics for these correlations are provided in Table 3, and scatter plots are shown by Supp. Figs. S12–S14.

3. Results and discussion

3.1. Central Celtic Sea site

316 3.1.1. Air-sea CO₂ flux

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317 Few studies have examined the Celtic Sea inorganic carbon cycle in detail, but those that have were mostly focused on the surface layer and air-sea CO₂ exchange. Frankignoulle and 318 Borges (2001) used a compilation of surface ocean pCO_2^{sw} data to demonstrate that the 319 northwest European continental shelf seas are a net sink for atmospheric CO₂, at a rate 320 321 between 1.8 and 2.9 mol m⁻² yr⁻¹. This result was quantitatively supported by a more recent model study (Wakelin et al., 2012). However, other studies have found smaller net air-sea 322 323 CO₂ fluxes for this region. Borges et al. (2006) compiled published datasets across the entire 324 northwest European shelf, and Marrec et al. (2015) generated predictive algorithms for Celtic Sea pCO₂^{sw} from variables including sea surface temperature, mixed layer depth and 325 326 chlorophyll-a concentration. These studies agree that the Celtic Sea is a net sink for atmospheric CO₂, but they suggest a typical rate a little under 1 mol m⁻² yr⁻¹. Our analysis 327 328 also indicated that the CCS site acted as a net atmospheric CO₂ sink: the mean air-to-sea CO₂ 329 flux at CCS from 21st March 2014 (i.e. the start of the UK-SSB sampling period) to the same date in 2015 was 1.3 ± 0.3 mol m⁻² yr⁻¹ (Fig. 2). The magnitude of this flux is consistent with 330 331 previous Celtic Sea studies, although it is towards their lower end. A separate analysis of in 332 situ CO₂ uptake (photosynthesis) and production (plankton respiration) data also found that 333 CCS was a net atmospheric CO₂ sink, as primary production was typically greater than 334 community respiration (García-Martín et al., 2017, this issue). 335 The Δp CO₂ was vertically uniform and close to atmospheric equilibrium during March-April. Its seasonal pattern was similar to C_T , with CO_2 undersaturation in the surface layer during 336 the spring-summer. The surface pCO_2^{sw} variability was dominated by biological C_T uptake 337 338 and release, rather than seasonal seawater heating and cooling, which would have been 339 expected to induce an increase in pCO_2^{sw} (and therefore ΔpCO_2) during the warmer summer months. Indeed, the amplitude of the SST seasonal cycle was about 10°C. Alone, this would 340 have driven a winter-summer pCO_2^{sw} increase of about 200 µatm (Takahashi et al., 2009), 341 342 whereas we observed a c. 85 µatm decline during this period (Fig. 2a). Other factors that 343 could affect the $\Delta p CO_2$ cycle are atmospheric pCO_2 , and seawater A_T . The amplitude of seasonal atmospheric pCO_2 cycle was an order of magnitude smaller than that of pCO_2 ^{sw}, 344 345 while A_T was not noticeably affected by seasonal stratification, remaining relatively

346 homogeneous both laterally and vertically throughout the Celtic Sea for the duration of the SSB sampling campaign (Supp. Fig. S8). The seasonal variability of atmospheric pCO₂ and 347 348 $A_{\rm T}$ therefore did not significantly influence that of $\Delta p{\rm CO}_2$. 349 The algorithm that we created to predict surface pCO_2^{sw} (Table 2) produced an annual range 350 of about 85 µatm, similar to previous studies (e.g. Marrec et al., 2015). However, there were 351 differences in the seasonal pattern. Firstly, we found that the pCO_2^{sw} had two distinct minima 352 during 2014, first in May-June and then a weaker minimum in November-December 353 associated with an autumn phytoplankton bloom (Wihsgott et al., 2017, this issue). Our 2015 354 data end too early in the seasonal cycle to determine whether these dual minima were 355 repeated. Secondly, Marrec et al. (2015) projected that ΔpCO_2 should be positive for 356 approximately a third of each year, during the winter months, reaching values up to about 357 30 µatm. However, neither our algorithm nor our observations indicate such consistently high 358 Δp CO₂ values. We did not collect any carbonate chemistry samples between November and 359 March, so we cannot be certain which is correct. However, the observational pCO_2^{sw} data 360 from SOCAT (Bakker et al., 2016) used by Marrec et al. (2015) to test their algorithm were 361 in atmospheric equilibrium during this season, rather than supersaturated as predicted, for 362 about half of the years that they considered. This winter supersaturation may therefore be an 363 intermittent phenomenon, but our dataset is unable to identify whether it occurred between 364 2014 and 2015. 365 The most uncertain part of our air-sea CO₂ flux algorithm was between autumn 2014 and 366 winter 2015, due to a lack of observations during that period. Comparison of the pattern of 367 pCO₂^{sw} (Fig. 2a) with other studies (e.g. Marrec et al., 2015) and our understanding of marine carbonate system drivers suggests that the pCO_2^{sw} predicted by the algorithm for this period 368 369 is likely to underestimate the true value. Consequently, we consider the air to sea CO₂ flux 370 calculated for this period as a maximum estimate. 371 3.1.2. Inorganic inventory changes 372 Vertical profiles of C_T , DIN, SRP and related hydrographic variables were constrained at two 373 roughly equivalent time points both in 2014 and 2015, referred to as 'winter' (i.e. March-374 April, pre-spring bloom) and 'summer' (August). Between these, there was a single 'late 375 spring' profile in June 2014, and a series of 'in-bloom' profiles (April-May) captured the 376 progression of the 2015 spring bloom at a relatively high temporal resolution, followed by an 377 'early spring' profile in May 2015. There was an 'autumn' (November) profile in 2014 only.

378 Direct comparisons between the two years can therefore be drawn from comparing winter 379 and summer, while the additional points can be used to more finely resolve the seasonal 380 changes. 381 The TS chain measurements clearly illustrate the seasonal cycle of surface warming and 382 stratification followed by winter mixing (Fig. 3a). The CCS site was vertically mixed during 383 the winter, but physical and biogeochemical variables exhibited a two-layer stratified 384 structure during the spring and summer. Most variables were vertically homogeneous in winter each year, after which C_T , DIN and SRP declined strongly in the surface layer (i.e. 385 386 shallower than 40-50 m) through the spring and summer (Fig. 3b-d), through drawdown 387 associated with NCP. DIN reached virtually zero (i.e. beneath the detection limit of ~0.02 mmol m⁻³), while SRP declined to a minimum of about 0.05 mmol m⁻³, consistent with 388 389 previous studies that have identified DIN as the limiting nutrient in this region (e.g. Davis et 390 al., 2014). The concentrations of these variables increased in the deep layer over the same 391 period, consistent with remineralisation of sinking OM. Stratification began to break down 392 towards the autumn, and the water column was reset to a vertically homogeneous state by the 393 following winter. 394 The dominant control on the vertical density structure in the Celtic Sea is temperature 395 (Pingree et al., 1976). Salinity thus only constitutes a second-order control, and it was less 396 variable throughout the seasonal cycle. However, the vertical partitioning of the water 397 column due to seasonal stratification did facilitate a notable increase in salinity from August 398 to December 2014 in the bottom mixed layer only (Fig. 3e). This suggests that open ocean 399 waters increased in influence relative to on-shelf waters during this time, as no in situ 400 processes can modify deep layer S to this extent. The DIN and SRP increased 401 disproportionately in the deep layer (i.e. beneath c. 50 m) during the same period (Fig. 3c-d), 402 suggesting that increasing nutrient concentrations relative to C_T at this time may have been 403 associated with the advective flux that elevated S, rather than in situ OM remineralisation. In 404 support, Ruiz et al. (2017, this issue) describe a shelf-wide estuarine-type circulation that 405 advects saline, nutrient-rich waters on to the shelf during the stratified summer months. They 406 estimate that 70-80% of the increase in deep layer DIN observed at CCS since the fully 407 mixed winter state could be supplied by this physical transport mechanism.

3.1.3. Production and remineralisation

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409 During the first half of each year, when stratification was increasing, changes in C_T , DIN and SRP profiles between successive time points can reveal the depth ranges over which NCP and 410 net remineralisation occurred (e.g. Fig. 4). For intervals when surface layer concentrations 411 412 declined, the average NCP rate can therefore be estimated by integrating the change in C_T 413 only over the depth range where C_T declined and correcting the result for air-sea CO_2 414 exchange, which can be assumed to affect only the surface layer when the water column is 415 stratified. Similarly, deep layer increases in C_T can be attributed to remineralisation. The 416 same procedure can be carried out for DIN and SRP, with the remineralisation calculation for 417 DIN corrected for denitrification and/or anammox (Hydes et al., 2004; Kitidis et al., 2017), 418 and no gas exchange correction (Table 4). The main spring bloom in 2014 resulted in a mean NCP of 35.9 \pm 2.3 mmol-C m⁻² d⁻¹ from winter to late spring, which is consistent with an 419 NCP estimate for the seasonally-stratified North Sea of 16-46 mmol $\mathrm{m}^{-2}~\mathrm{d}^{-1}$ for this season 420 (Bozec et al., 2006). This represents a total of 3.08 ± 0.20 mol-C m⁻² C_T conversion into OM 421 422 since the mixed winter state, integrated over the period between our winter and late spring 423 2014 time points (i.e. 86 days; Table 1). The equivalent DIN and SRP drawdowns were 342 \pm 7 mmol m⁻² d⁻¹ (Fig. 4a) and 20.74 \pm 0.36 mmol m⁻² d⁻¹ respectively, resulting in an overall 424 C:N:P for NCP of about 117:13.0:0.79. Corresponding remineralisation of C_T , DIN and SRP 425 within the subsurface layer was 0.78 ± 0.50 mol m⁻², 85 ± 73 mmol m⁻² and 8 ± 99 mmol m⁻², 426 427 with C:N:P therefore 117:12.7:1.24. This indicates that 20-40% of the NCP was 428 remineralised within the sub-surface (presumably following sinking of the corresponding 429 OM), while 60–80% must have remained within some OM pool, assuming negligible net 430 advection (Pingree and Le Cann, 1989; Brown et al., 2003). The spring bloom period was 431 followed by much weaker NCP and net water column remineralisation from late spring to 432 summer, and then vertical mixing into the autumn and winter (Fig. 4b-c), during which period the decline in deep layer C_T was due to mixing with low- C_T surface waters, not NCP. 433 434 The spring bloom of the following year was sampled at higher temporal resolution, during 435 cruise DY029 in April 2015. From winter to early spring 2015 (i.e. 61 days; Table 1), the 436 mean NCP determined from C_T profiles and air-sea CO₂ exchange was 30.5 \pm 2.3 mmol-C m⁻² d⁻¹, comparable to the previous year, estimates from nearby similar regions (Bozec et al., 437 2006), and the mean NCP of 31 ± 24 mmol-C m⁻² d⁻¹ (photosynthetic quotient of 1.4) 438 439 determined at CCS from oxygen-to-argon ratio (O₂/Ar) measurements during this cruise 440 (Seguro et al., 2017, this issue). This NCP integrated from the winter to early spring 2015

- 441 time points (i.e. 139 days; Table 1) represented 1.86 ± 0.14 mol-C m⁻² of OM production,
- compared with 3.08 ± 0.20 mol-C m⁻² for the roughly equivalent period (winter to *late*
- spring) in 2014. However, further NCP from early spring to summer in 2015 brought the total
- NCP since winter to 3.30 ± 0.13 mol-C m⁻², similar to the total value for the part of 2014 with
- positive NCP. The corresponding DIN- and SRP-derived values (from winter to summer
- 446 2015) were 343.4 ± 3.9 and 20.86 ± 0.22 mmol m⁻² respectively, resulting in a C:N:P for
- NCP of about 117:12.2:0.74, similar to the previous year (i.e. 117:13.0:0.79), but with lower
- N and P coefficients in 2014, or equivalently more C-rich OM production relative to typically
- assumed (Redfield, 1934; Redfield et al., 1963) and measured (Anderson and Sarmiento,
- 450 1994) ratios. Therefore the total amount of OM generated through NCP appeared to be
- similar during the spring bloom each year, although this may have occurred more rapidly
- during 2014. Consequently, although the winter and summer time points were at a very
- similar time for both years, it may be more appropriate to compare 'winter to late spring
- 454 2014' with 'winter to summer 2015'. Remineralisation from winter to summer 2015 occurred
- 455 at a C:N:P of 117:11.7:0.88, and represented 24–35% of NCP, in both ways similar to 2014.
- Both the production and remineralisation of OM were C-rich relative to the likely limiting
- nutrient, with C:N ranging from 9 to 10 compared with typical values of 6.6 to 7.3 (Redfield,
- 458 1934; Redfield et al., 1963; Anderson and Sarmiento, 1994).
- 459 3.1.4. Total column inventories
- The preceding section's approach of separating surface NCP from deep remineralisation was
- less useful from summer onwards, once stratification had begun to break down. At this time,
- both surface and deep layer $C_{\rm T}$ and nutrient inventories were influenced by vertical mixing,
- along with remineralisation throughout the water column. It is therefore more appropriate to
- consider changes in the total inventories integrated vertically across the entire water column.
- Remineralisation of both $C_{\rm T}$ and nutrients continued after summer 2014, first with apparent
- excess DIN generation until autumn 2014, and then excess $C_{\rm T}$ until winter 2015 (Fig. 5a-b).
- We note that the excess N period coincides with the previously-noted apparent advective flux
- of high salinity to CCS, high DIN waters in the deep layer, so this may have driven the
- unexpected and transient high DIN, rather than remineralisation (Fig. 5c). However, this high
- 470 S water had disappeared by winter 2015, due to vertical mixing and shut-down of the summer
- estuarine-type circulation (Ruiz-Castillo et al., 2017, this issue). The overall trajectory from
- summer 2014 to winter 2015 is therefore probably representative of net remineralisation,
- presumably ongoing from some long-lived OM pool.

474 From late spring 2014 to winter 2015, remineralisation returned a further 2.17 \pm 0.43 mol-C m^{-2} C_T , 169 \pm 50 mmol-N m^{-2} DIN and 3.27 \pm 0.88 mmol-P m^{-2} SRP to the water column, 475 476 corresponding to a C:N:P ratio of 117:9.1:0.2, with substantially lower N and P coefficients 477 and hence higher C:N and C:P than the NCP stoichiometry. Adding the additional apparent 478 late year remineralisation to that experienced in the deep layer from winter to late spring 479 2014 resulted in an overall remineralisation C:N:P of 117:10.0:0.5. This corresponded to 96 \pm 480 22 %, 72 ± 26 % and 56 ± 479 % of the OM production of C_T , DIN and SRP respectively in 481 the 2014 spring bloom being remineralised by winter 2015. Thus, acknowledging the 482 potential for advection to complicate our interpretation (see Section 3.2.2), approximately the 483 same amount of $C_{\rm T}$ that was biologically fixed in the 2014 spring bloom appeared to have 484 been remineralised by winter 2015, while only about three-quarters of the fixed DIN and half 485 of the fixed SRP were remineralised over the same period, noting that the uncertainty in the 486 SRP value is too great for us to speculate about it further. Such a result is difficult to 487 reconcile with the expected behaviours of C, N and P during OM remineralisation 488 (Hopkinson et al., 1997), thus we speculate that this may actually indicate the 489 remineralisation of a C-rich OM pool carried over from before our first W14 sampling point. 490 Furthermore, the air-sea CO₂ flux value that was used to correct these inventories was likely 491 an overestimate from autumn 2014 to winter 2015 (Section 3.1.1). The C_T result is therefore a 492 minimum estimate, and it is possible that more C_T was remineralised than was produced by 493 NCP that year at CCS. 494 The total water column inventories indicate that the shelf was not operating in steady state 495 from winter 2014 to 2015. The winter inventories of DIN and SRP were 10.3 ± 1.4 % and 496 10.6 ± 1.1 % lower respectively in 2015 compared to 2014, while that of C_T was 0.36 ± 0.18 % higher (Fig. 5a). The equal (or slightly greater) OM formation during the 2015 497 498 spring-summer relative to 2014 was therefore unexpected, because lower nutrient 499 concentrations at the end of the winter mixed state should have supported less NCP. As NCP 500 was the ultimate driver of CO₂ uptake, we would also have expected a lower air-to-sea CO₂ 501 flux following these lower winter nutrient concentrations. This did not occur, because the 502 nutricline was about 10-15 m deeper in 2015 (Fig. 6a), resulting in a greater total $N_{\rm T}$ 503 conversion into OM despite its lower initial concentration, and matching the behaviour of the 504 thermocline (Fig. 6b). There does not appear to be an obvious mechanism that would drive a 505 deeper nutricline at times of lower surface nutrient concentrations, which would lead summer

506 NCP to be generally consistent across different years. Rather, it appears to be a coincidence 507 that these phenomena balanced each other out for the two years that we observed. 508 3.1.5. Organic matter stoichiometry 509 Overall mass balance dictates that changes in the OM composition should mirror the changes 510 in the inorganic inventories (corrected for gas exchange and denitrification/anammox). In 511 Fig. 5b, trajectories between adjacent sampling points that trend towards the bottom right 512 (low C_T , high DIN or SRP) thus indicate the on-shelf generation of relatively C-rich OM, 513 while trajectories that tend towards the top left (high C_T, low DIN or SRP) indicated more N-514 rich (and/or P-rich) OM generation, relative to an assumed 'typical' C:N:P of 117:16:1 515 (Anderson and Sarmiento, 1994). The high resolution samples from during the 2015 spring 516 bloom therefore suggest that NCP may have begun with N- and P-rich OM formation, before 517 switching to more C-rich production once surface layer DIN concentrations fell close to zero 518 (Fig. 5b). As mentioned above, the overall change from spring to summer 2015 was similar to 519 the C:N:P uptake stoichiometry observed for the same period in 2014. However, from early 520 spring to summer 2015 we saw the opposite pattern than during the similar period in 2014: 521 the OM became more C-rich. This change to seawater C_T loss with no nutrient loss (as 522 opposed to loss of both to OM following a constant stoichiometry) coincided approximately 523 with CCS surface DIN concentrations approaching zero. We suggest that the C_T loss was 524 therefore a result of continued NCP and OM formation, but that this was using regenerated 525 DIN within the surface layer as its N source rather than drawing from the inorganic pool, 526 which was by then locked away by stratification in the deep layer, as has been observed in the 527 North Sea (Bozec et al., 2006). Although this process should therefore have generated an 528 increasingly C-rich OM pool, presumably associated with more rapid N than C 529 remineralisation from the OM pool, it was not sufficient to return to the original winter 2014 530 C:N:P stoichiometry. These predicted changes in the OM composition are broadly in 531 agreement with the pattern observed by direct OM stoichiometric measurements at CCS (Davis et al., 2017, this issue), although our approach is not able to distinguish between the 532

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dissolved and particulate OM pools.

535 3.2.1. Stratification across the Celtic Sea In addition to the CCS site, we sampled a northeast-southwest transect across the Celtic Sea 536 537 during each UK-SSB cruise (Fig. 1). Analysis of these measurements adds a third dimension 538 (i.e. distance from the open ocean) to our so far two-dimensional analysis (i.e. depth and 539 time), highlighting some additional caveats, sources of uncertainty and broader context. 540 Hartman et al. (2017, this issue) describe surface layer UK-SSB measurements from across 541 the entire northwest European continental shelf, thus providing further context. 542 The pattern of winter mixing followed by surface layer uptake of C_T , DIN and SRP and their 543 increase in the deep layer was consistent across the entire Celtic Sea transect (Fig. 7). 544 However, the deep layer experienced greater increases in these variables (and others such as 545 Δp CO₂) further away from the open ocean, although this was reduced for DIN by 546 denitrification/anammox in the seafloor sediments (Hydes et al., 2004; Kitidis et al., 2017). 547 The depth of the surface layer (as defined by the nutricline, 40-50 m) remained relatively 548 constant across the transect while the seafloor became shallower towards the land, so the 549 greater deep layer concentrations could simply indicate that the same amount of NCP had 550 been remineralised, but spread over a narrower deep layer, consistent with the limiting factor 551 for total remineralisation within the deep layer being some function of the OM supply. In 552 addition, the increasing deep layer C_T and nutrients away from the open ocean could reflect 553 the fact that deep waters further from the open ocean had experienced a longer transit time 554 across the shelf within the estuarine-type circulation (Ruiz-Castillo et al., 2017, this issue), 555 and thus greater accumulation of sinking organic matter from the surface layer, in a manner 556 analogous to how nutrient concentrations tend to increase along isopycnals away from the 557 surface in the open ocean (Takahashi et al., 1985). 558 The increasing strength of biogeochemical stratification moving away from the open ocean 559 has important implications for interpreting our results from CCS, which lay towards the open 560 ocean end of the Celtic Sea transect. For example, the surface layer air-sea CO₂ 561 disequilibrium during the stratified period increased away from the open ocean. We would 562 therefore expect the Celtic Sea as a whole to exhibit a greater net air-sea CO₂ flux than that 563 calculated for the CCS site, which was relatively close to the open ocean end of the transect.

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3.2. Celtic Sea transect

This is consistent with our CCS air-sea CO₂ flux estimate falling towards the lower end of

previously published results (Section 3.1.1). A second example concerns potential advective fluxes within the Celtic Sea, as discussed in the following section.

3.2.2. Biogeochemical relationships with salinity

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In the Celtic Sea, S acts as a tracer of the relative influence of open ocean and shelf waters, particularly in the deep mixed layer beneath the pycnocline, where no other processes can modify it. Throughout the year, S decreases moving on to the shelf, ultimately as a result of dilution by riverine inputs. During the winter, S was vertically homogeneous at each point across the shelf, while the isohalines were tilted away from vertical towards the open ocean in the surface during the summer (i.e. increasing S with depth, Fig. 7h). This suggests that there was likely a net transport of relatively fresh waters towards the open ocean in the surface layer, and a net transport of salt towards the land within the bottom mixed layer at CCS during the summer. During the winter, while C_T and SRP were laterally homogeneous, DIN was correlated with salinity (Table 3) and declined away from the open ocean (Figs. 7d and 8b). There were no relationships between S and any of C_T , DIN or SRP in the surface layer during the summer months, when NCP was the dominant control on these biogeochemical variables. However, correlations could be identified in the deep layer throughout the seasonal cycle, which were variable through time, but fairly consistent with each other (Fig. 8, Table 3). These relationships with S indicated that C_T and SRP were enriched on the shelf relative to the open ocean during the summer months, and returned close to open-ocean values during the winter, consistent with a seasonal cycle of remineralisation and vertical mixing. Although the DIN-S relationship had a similar pattern to those for C_T and SRP, there was a significant offset from the values predicted from the SRP relationship and an assumed N:P stoichiometry of 16:1 (Fig. 8b). This offset was positive, with a mean value of 7.5 ± 2.1 mmol m⁻³ per unit practical salinity, indicating that the shelf waters were virtually always depleted in DIN relative to the open ocean, likely as the result of denitrification and/or anammox in the shelf sediments (Hydes et al., 2004; Kitidis et al., 2017). The preceding interpretation of the changing C_T and nutrient vertical profiles at CCS mostly ignored the possibility of advective influences. However, given the changing deep layer $C_{\rm T}$ and nutrient concentrations in the stratified period, advection of water from another part of the shelf could generate an apparent increase or decrease in remineralisation at CCS. As well as directly changing the inorganic inventories, this could also supply OM to CCS that had originally formed elsewhere on the shelf, particularly given our evidence that a significant

597 fraction of NCP (>50%) remained within the OM pool, as opposed to the deep layer 598 remineralised pool, over seasonal timescales (Fig. 5). Significant advection of a cross-shelf 599 OM gradient would hence introduce uncertainties into our one-dimensional estimates of the 600 fraction of NCP that had been remineralised. Indeed, production and advection of a long-601 lived OM pool could provide a source for the possible excess of remineralisation over NCP 602 that we observed from winter 2014 to winter 2015 at CCS. It is also important to note that we 603 cannot determine how CCS could have been affected by any advective fluxes of material 604 perpendicular to the Celtic Sea transect that we sampled, which could have a different 605 relationship with S than observed along the UK-SSB transect. However, in general we do not 606 expect this to have significantly affected our results, because previous studies have found that 607 water exchange between the Celtic Sea and adjacent open ocean is relatively slow (Holt et al., 608 2009; Huthnance et al., 2009), as is net lateral circulation on the shelf itself (Pingree and Le 609 Cann, 1989), with water taking about 2 years to reach the central Celtic Sea after entering the 610 shelf from the open ocean (Hydes et al., 2004). 611 The Celtic Sea transect dataset also indicates that the increase in C_T observed at CCS from 612 winter 2014 to winter 2015, and the simultaneous declines in DIN and SRP, were not a result 613 of spatial variability. As these biogeochemical variables appeared to diverge from their open 614 ocean values moving into the shelf sea (Fig. 8), apparent changes at a static site such as CCS 615 could be caused by advective reorganisation of the water on the shelf, with a greater or lesser 616 open ocean influence at CCS in different years. If this were the case, then the apparent 617 nutrient loss that we observed could simply be an artefact of the spatial variability, rather than 618 a temporal biogeochemical trend. However, it appears that the changes in C_T , DIN and SRP 619 that we observed at CCS over this time period were consistent throughout most of S range 620 observed across the Celtic Sea transect (Fig. 9). This therefore supports our earlier 621 interpretation that the seasonal biogeochemical changes observed at CCS were representative 622 of processes operating across the wider Celtic Sea. 623 Finally, our transects indicated that the shelf $A_{\rm T}$ was relatively homogeneous both spatially 624 and temporally throughout the UK-SSB sampling campaign (Fig. 7i; Supp. Fig. S8), with a 625 weakly positive relationship with salinity. At CCS, we observed no low A_T excursions from 626 the $A_{\rm T}$ -salinity relationship that could be evidence for pelagic calcification, for example by 627 coccolithophores (Suykens et al., 2010), confirming that our calculated water column 628 inventories of C_T at CCS were not significantly affected by such processes during the UK-629 SSB sampling period.

Our estimated net oceanic CO₂ uptake of >1 mol-C m⁻² yr⁻¹ within the Celtic Sea suggests 631 that an active CSP operates within this system. Given the preceding presented data and 632 analysis, we suggest two mechanisms through which such a CSP could operate in a nutrient-633 634 conserving manner, i.e. without a significant external nutrient-rich source from rivers, and 635 without the shelf becoming depleted in these nutrients over time, as follows. 636 Firstly, we saw an increase in vertically-integrated C_T at CCS from winter 2014 to winter 637 2015, along with a simultaneous decline in DIN and SRP (Fig. 5a). The initial winter 2014 638 values of $C_{\rm T}$, DIN and SRP were relatively close to the base of the thermocline in the 639 adjacent open ocean, suggesting that there may recently have been relatively strong water 640 exchange across the shelf break, which did not occur again during our sampling period. If the 641 shelf sea were to follow a cycle of increasing C_T while DIN and SRP declined between these 642 irregular 'flushing' events, then the exchange of shelf waters with the open ocean would 643 constitute a net export of C_T while DIN and SRP were replenished. 644 Secondly, as previously identified in similar environments (e.g. Butler et al., 1979; Johnson et 645 al., 2013), it appears that a substantial fraction (>50%) of the OM produced during the spring 646 bloom each year was only slowly remineralised over many months (Fig. 5). Indeed, evidence 647 that deep layer advective transport could have accounted for 70-80% of the DIN increase at CCS from winter to summer each year (Ruiz-Castillo et al., 2017, this issue) suggests that our 648 649 calculated fraction of OM that was not remineralised is a minimum estimate, so the long-650 lived OM pool could be greater still. Moreover, the non-steady state situation that we 651 observed over the 1.5 sampled years further suggests that some fraction of this 652 (semi-)refractory OM pool was potentially not remineralised within the same year and hence 653 may play a role in redistributing C and nutrients within the Celtic Sea itself (Fig. 7). We 654 therefore speculate that this long-lived OM pool could also function to export C from the 655 shelf sea into the open ocean, for example during exchange with off-shelf waters with a lower 656 OM concentration. 657 Future synthesis of the inorganic measurements presented here along with OM concentration 658 and stoichiometry datasets collected during the UK-SSB research programme in the context 659 of a system which likely operates on a multi-annual timescale will be required to assess the 660 capacity of these possible mechanisms to sustain a nutrient-conserving CSP in temperate, 661 seasonally stratified continental shelf seas like the Celtic Sea.

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3.2.3. Nutrient-conserving mechanism

4. Conclusions

The Celtic Sea experiences seasonal physical and biogeochemical stratification with low $C_{\rm T}$ and nutrient concentrations driving a net air to sea ${\rm CO_2}$ flux, while remineralisation increases these concentrations in the deep layer. Repeated observations at the CCS site suggest that the inventories of DIN and SRP declined by about 10% from winter 2014 to winter 2015, while $C_{\rm T}$ increased. Transect data indicate that these temporal changes observed at CCS were consistent across much of the Celtic Sea, rather than being an artefact of advection and spatial variability. Lower availability of nutrients for NCP in 2015 was compensated by a deeper nutricline, so there was no reduction in either the total amount of NCP or air-sea ${\rm CO_2}$ exchange during the subsequent spring bloom. Analysis of stoichiometry during nutrient uptake for NCP and remineralisation indicated the production of C-rich OM. However, some fraction of the OM generated on the shelf appears to not have been remineralised within the year it was created, suggested that a long-lived stock of OM could accumulate in the shelf waters. This, and the overall changes in inorganic inventories that we observed, could provide alternative mechanisms through which a nutrient-conserving CSP could operate, both in the Celtic Sea and in other seasonally stratified, temperate continental shelf seas.

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6. References

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7. Tables

Table 1. Summary of the time intervals used in the CCS site analysis.

Name of time	Year	Interval	Sampling date range	Cruise(s)
interval		code	(dd/mm/yyyy)	
Winter	2014	W14	26/03/2014 - 27/03/2014	DY008
Late Spring	2014	lSp14	19/06/2014 - 22/06/2014	JC105
Summer	2014	Su14	05/08/2014 - 23/08/2014	DY026A,
				DY026B
Autumn	2014	A14	10/11/2014 - 29/11/2014	DY018
Winter	2015	W14	22/03/2015	DY021
In-bloom 1	2015	IB1	03/04/2015 - 06/04/2015	DY029
In-bloom 2	2015	IB2	11/04/2015 — 12/04/2015	DY029
In-bloom 3	2015	IB3	15/04/2015	DY029
In-bloom 4	2015	IB4	16/04/2015	DY029
In-bloom 5	2015	IB5	20/04/2015	DY029
In-bloom 6	2015	IB6	21/04/2015	DY029
In-bloom 7	2015	IB7	25/04/2015 - 26/04/2015	DY029
In-bloom 8	2015	IB8	28/04/2015	DY029
Early Spring	2015	eSp15	21/05/2015 - 23/05/2015	DY030
Summer	2015	Su15	24/07/2015 - 23/08/2015	DY033, DY034

Table 2. Details of the empirical algorithm used to predict surface pCO_2^{sw} at CCS.

Day of year (DoY)	SST range / °C	pCO ₂ sw / µatm					
range							
80 < DoY < 232	SST < 10.5	$1.547 \times 10^{-3} \exp(SST) + 372.7$					
80 < DoY < 232	10.5 ≤ SST ≤	$324.5 + 6.596 \times 10^{24} \exp[-5.000(SST)]$					
	11.0						
80 < DoY < 232	11.0 < SST	$-0.557(SST)^2 + 23.57(SST) + 141.1$					
$DoY \le 80 \text{ or } 233 \le DoY$	All	$-0.2479(SST)^3 + 10.846(SST)^2$					
		- 151.42(SST) + 1050.3					

Table 3. Correlation statistics between biogeochemical variables and salinity. Columns headed dX/dS contain gradient \pm its standard error in μ mol kg⁻¹ (Figure 8); N is the number of observations used in each regression.

Cruise	Interval code	C_{T}			DIN			SRP					
		$\mathrm{d}C_{\mathrm{T}}/\mathrm{d}S$	r^2	p	N	dDIN/dS	r^2	p	N	dSRP/dS	r^2	p	N
DY008	W14	-10.6 ± 8.9	0.11	0.26	13	7.6 ± 0.6	0.86	0.00	25	-0.23 ± 0.09	0.21	0.02	25
JC105	lSp14	-109.6 ± 10.0	0.86	0.00	21	1.0 ± 0.8	0.07	0.24	21	-0.46 ± 0.06	0.75	0.00	21
DY026*	Su14	-60.6 ± 10.8	0.78	0.00	11	4.1 ± 1.2	0.36	0.00	23	-0.29 ± 0.06	0.47	0.00	26
DY018	A14	-99.5 ± 10.6	0.49	0.00	92	-0.2 ± 1.1	0.00	0.89	183	-0.46 ± 0.07	0.18	0.00	202
DY021	W15	-19.6 ± 9.1	0.28	0.05	14	5.9 ± 0.7	0.59	0.00	47	0.01 ± 0.06	0.00	0.87	47
DY029	IB1–IB8	-39.5 ± 7.3	0.54	0.00	27	7.4 ± 0.4	0.65	0.00	192	0.11 ± 0.02	0.12	0.00	202
DY030	eS15	-26.7 ± 13.4	0.50	0.12	6	9.5 ± 1.2	0.85	0.00	13	0.32 ± 0.10	0.48	0.01	13
DY033	Su15	-63.1 ± 8.1	0.68	0.00	31	4.2 ± 0.4	0.42	0.00	182	-0.21 ± 0.02	0.30	0.00	186
DY034	Su15	-120.9 ± 24.0	0.78	0.00	9	0.4 ± 1.8	0.00	0.84	44	-0.51 ± 0.12	0.31	0.00	44

^{*}DY026 includes all data from both DY026A and DY026B.

8. Figures

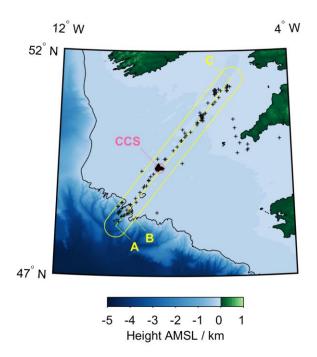


Figure 1. The UK-SSB sampling locations in the Celtic Sea. The pink circle contains the data considered to represent the Central Celtic Sea site (i.e. within 12 km of 49.4°N, 8.54°W), while the yellow lines indicate the transect route for Fig. 7 and enclose the data included in those plots; the points labelled A, B and C correspond to the equivalent points on Fig. 7. The 200 m bathymetry contour, traditionally used to define the edge of the shelf, is shown as a black line. Topography data are from the GEBCO_2014 30 arc-second grid (version 20150318, http://www.gebco.net).

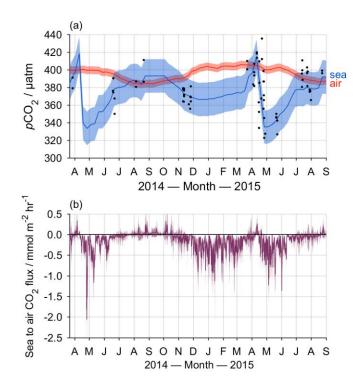


Figure 2. Time-series of air-sea CO_2 disequilibrium and flux at CCS. (a) Atmospheric (red) and sea surface (blue) pCO_2 at CCS. The black points show the actual measurements of surface pCO_2^{sw} , while the blue line shows our algorithm-predicted interpolation of these data (Table 2) that was used to calculate ΔpCO_2 and then the air-sea CO_2 flux. (b) The sea to air CO_2 flux. The shaded areas show the 95% confidence intervals.

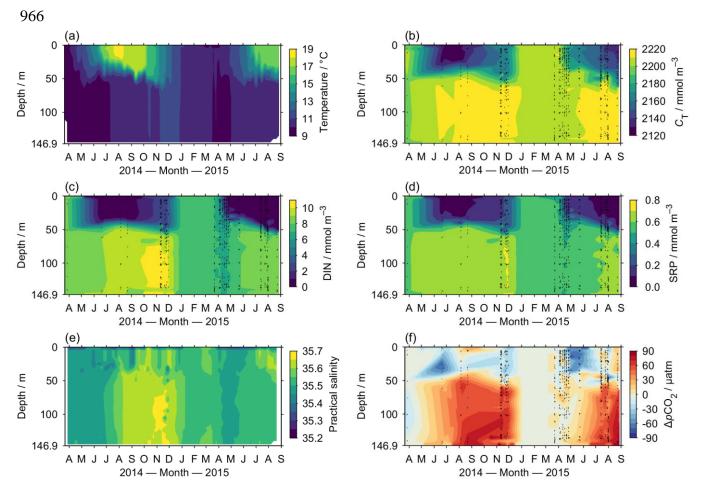
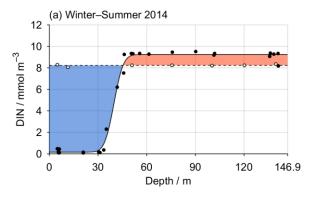
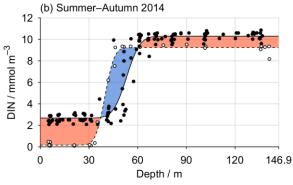


Figure 3. Hovmöller diagrams of CCS water column (a) temperature, (b) dissolved inorganic carbon, (c) dissolved inorganic nitrogen (i.e. nitrate + nitrate + ammonium), (d) soluble reactive phosphorus, (e) practical salinity, and (f) air-sea CO₂ disequilibrium. Where shown, sampling locations are indicated as black points. The cycle of winter mixing and summer stratification is clear for most variables, while *S*, DIN and SRP show an anomalous increase in the deep layer during late 2014, indicating an influx of open ocean waters.





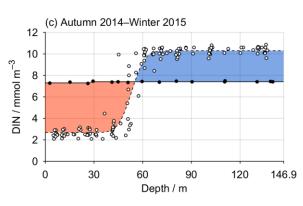


Figure 4. Changes in the DIN profile at CCS for selected time intervals. In each panel, the earlier profile is shown as a dashed line, while the later one is solid. The blue shaded areas therefore indicate decreases in DIN, while orange shows increases. (a) Primary production in the surface layer drew down DIN, while remineralisation at depth increased the DIN concentration. (b) The onset of vertical mixing eroded the surface-deep DIN gradient, overlaid by a remineralisation-driven increase in the total DIN inventory. (c) The water column returned to a vertically homogeneous state following further vertical mixing.

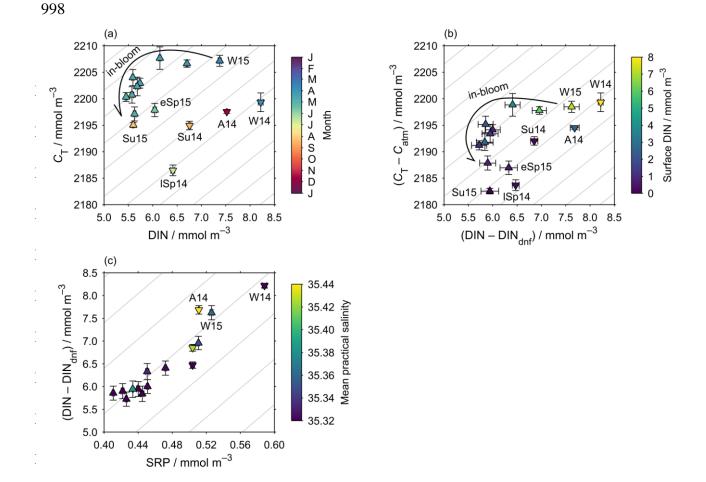
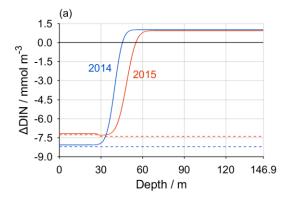


Figure 5. Water column inventories at CCS throughout the SSB study period. The inventories have been divided by the water depth, so the values shown are the vertical mean at each time point. Downwards/upwards pointing triangles represent 2014/2015 data respectively, and the grey diagonal lines indicate a C:N:P stoichiometry of 117:16:1 (Anderson and Sarmiento, 1994). W = winter; eSp = early spring; lSp = late spring; Su = summer; A = autumn; 14/15 indicates the year. (a) Raw inventories of C_T and N_T measured at CCS, coloured by the sampling date. (b) Here, the C_T inventory is corrected for air-sea gas exchange relative to the first sampling point, while the DINis corrected for denitrification/anammox. The colour shows the surface layer DIN concentration. (c) Relative changes in DIN (corrected for denitrification/anammox) and SRP, coloured by mean salinity at CCS at each time point. Higher salinity values indicate the influence of open ocean waters.



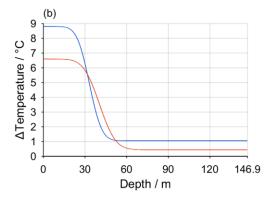


Figure 6. Winter to summer changes in (a) DIN and (b) seawater temperature for 2014 (blue) and 2015 (red). In (a), the dashed lines show the lowest possible Δ DIN, i.e. its value if the entire winter DIN inventory had been converted into OM. All available DIN in the surface layer was converted to OM in both years. This represented a greater change in DIN in 2014 due to the higher initial DIN concentration, but the deeper nutricline in 2015 led to greater NCP overall. Remineralisation increased the deep layer DIN concentration by the same amount in both years, but this represented a smaller absolute DIN addition in 2015 owing to the deeper nutricline.

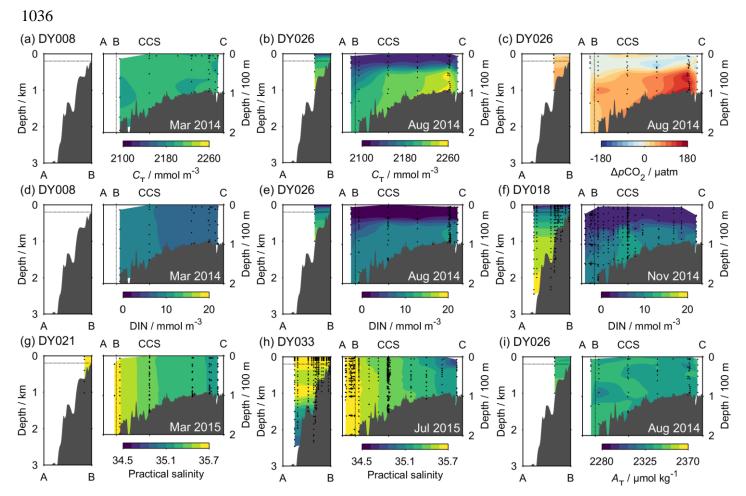


Figure 7. Transects across the Celtic Sea for selected variables and UK-SSB cruises. In each panel, the area above the horizontal dotted line in the left plot (shelf break) is the same as that to the left of the vertical dotted line in the right plot (on the shelf), and the black points indicate where samples were collected. Points A, B, CCS and C are located on Fig. 1. (a) and (b) show C_T in winter and summer 2014 respectively, while (c) shows $\Delta p CO_2$ in summer 2014. (d), (e) and (f) show the evolution of DIN from spring through summer to autumn 2014. (g) and (h) illustrate changes in salinity between winter and summer 2015, while (i) shows A_T in summer 2014. A full set of transects for every variable and UK-SSB cruise is provided in Supp. Figs. S6–S11.

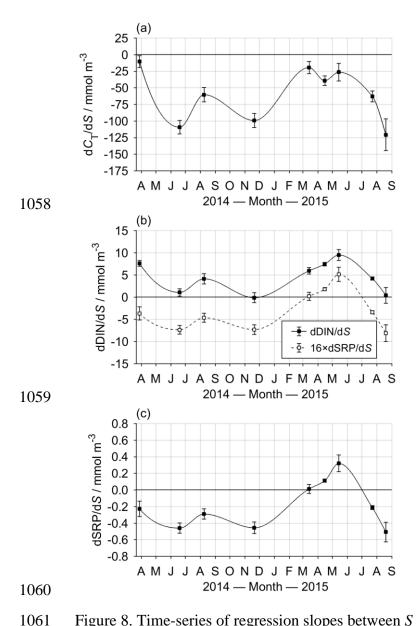


Figure 8. Time-series of regression slopes between S and (a) C_T , (b) DIN, and (c) SRP. Negative values indicate that the on-shelf deep layer is enriched in the nutrient relative to the open ocean, while positive values indicate on-shelf depletion. Error bars show the standard error in the gradient at each point (Table 3). In (b), the predicted relationship based on dSRP/dS is also shown, considered to represent the component driven by OM remineralisation. The offset between this and the solid line (mean \pm SD difference = 7.5 ± 2.1 mmol m⁻³) was likely driven by denitrification/anammox in the Celtic Sea; note that this offset value is reported per unit practical salinity (which is itself dimensionless).

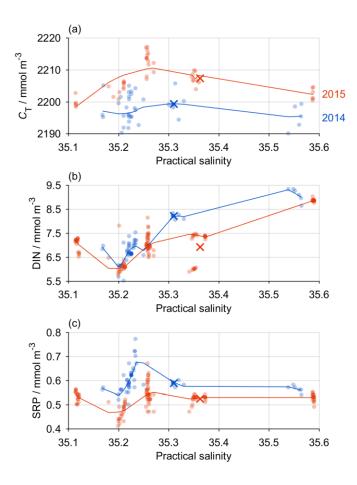


Figure 9. Changes in (a) C_T , (b) DIN and (c) SRP, across the entire Celtic Sea transect from winter 2014 (blue) to winter 2015 (red). The \times symbols show the mean conditions at CCS each winter. Salinity values increase towards the open ocean. The increase in C_T observed at CCS from winter 2014 to winter 2015, and corresponding declines in DIN and SRP, appear to be consistent across much of the Celtic Sea, rather than being a feature of spatial variability local to CCS.

9. Supplementary information

9.1. Supplementary tables

Table S1. Summary of the UK-SSB research cruises.

Cruise code ^a	Cruise sampling	Cruise Principle	C _T /A _T sample	No. of	
	dates (dd/mm/yy) ^b	Investigator	collectors	$C_{\mathrm{T}}/A_{\mathrm{T}}$	
				samples	
DY008	21/03—04/04/14	H. A. Ruhl	L. Darroch	45	
JC105	16/06—22/06/14	J. E. Hopkins	J. Fox, R. Houlding,	105	
			P. Nelson		
DY026A	04/08—13/08/14	R. Sanders	R. Sims	60	
DY026B	17/08—22/08/14	D. Sivyer	R. Sims	16	
DY018	10/1101/12/14	J. Sharples	L. Darroch,	301	
			J. E. Hopkins		
DY021	03/03—24/03/15	E. M. S.	N. Hicks	47	
		Woodward			
DY029	02/04—28/04/15	A. Poulton	A. Poulton,	214	
			K. Mayers		
DY030	05/05—23/05/15	G. Fones	R. Sims	72	
DY033	13/07—01/08/15	C. M. Moore	R. Sims	201	
DY034	07/08—31/08/15	H. A. Ruhl		52	

^aCruise codes beginning with DY were on the RRS *Discovery*, while JC stands for the RRS *James Cook*. ^bCruise sampling dates indicate the time period for collection of samples used in this study, not necessarily the entire duration of the cruises.

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	C_{T}			DIN				
Interval code	v_0	v_1	<i>Z</i> 0	<i>Z</i> 1	v_0	v_1	<i>Z</i> 0	<i>Z</i> 1
W14	2199.3	0	0	0	8.22	0	0	0
lSp14	2168.9	38.70	40.0	9.64	4.64	4.391	43.8	6.80
Su14	2173.2	42.87	36.2	8.36	4.68	4.542	39.9	6.48
A14	2190.3	27.88	54.4	9.99	6.48	3.808	53.3	9.24
W15	2207.1	-1.08	80.0	12.38	7.34	0.056	28.0	0.69
IB1	2206.6	2.04	70.7	0.68	6.56	0.529	52.9	15.83
IB2	2203.6	7.86	35.8	13.93	5.37	1.465	34.3	17.85
IB3	2192.8	15.97	26.8	19.54	3.21	3.549	19.7	26.09
IB4	2197.8	12.38	36.5	15.30	4.37	2.385	35.8	18.15
IB5	2195.1	13.27	42.3	7.71	4.70	2.027	41.7	8.91
IB6	2195.5	11.54	30.2	13.78	4.06	2.897	32.4	8.04
IB7	2191.3	17.68	35.9	9.39	3.92	2.944	35.2	10.53
IB8	2186.6	21.58	37.7	3.21	4.03	3.247	37.2	4.89
eSp15	2186.4	21.92	35.1	4.04	4.17	3.634	35.6	5.34
Su15	2180.1	37.27	44.1	12.17	4.20	4.122	48.6	7.78
	SRP			Practical salinity				
Interval code	v_0	v_1	<i>Z</i> 0	<i>Z</i> 1	v_0	<i>v</i> ₁	<i>Z</i> 0	<i>Z</i> 1
W14	0.59	0	0	0	35.310	0	0	0
lSp14	0.37	0.303	42.1	10.80	35.324	-0.0101	18.8	2.33
Su14	0.36	0.302	38.8	7.01	35.395	0.0429	21.8	17.17
A14	0.45	0.248	54.2	9.41	35.437	0.0444	58.1	2.45
W15	0.49	0.040	0.4	29.42	35.356	0.0096	15.0	3.33
IB1	0.50	0.025	51.6	12.65	35.343	0.0222	54.8	14.72
IB2	0.43	0.089	36.5	14.87	35.338	-0.0251	23.7	52.69
IB3	0.27	0.250	17.4	30.97	35.326	-0.0097	43.7	18.90
IB4	0.34	0.146	37.0	20.29	35.321	-0.0106	31.8	6.99
IB5	0.38	0.143	42.1	8.16	35.326	-0.0174	42.4	6.72

IB6	0.34	0.189	33.1	8.33	35.322	-0.0168	34.9	8.36
IB7	0.33	0.184	36.5	11.04	35.312	-0.0058	42.3	0.90
IB8	0.34	36.363	0.16	5.02	35.311	0.0041	35.3	2.01
eSp15	0.35	35.486	0.2	6.03	35.322	0.0218	35.9	7.20
Su15	0.34	48.082	0.27	9.08	35.398	-0.0266	35.1	0.03

9.2. Supplementary figures

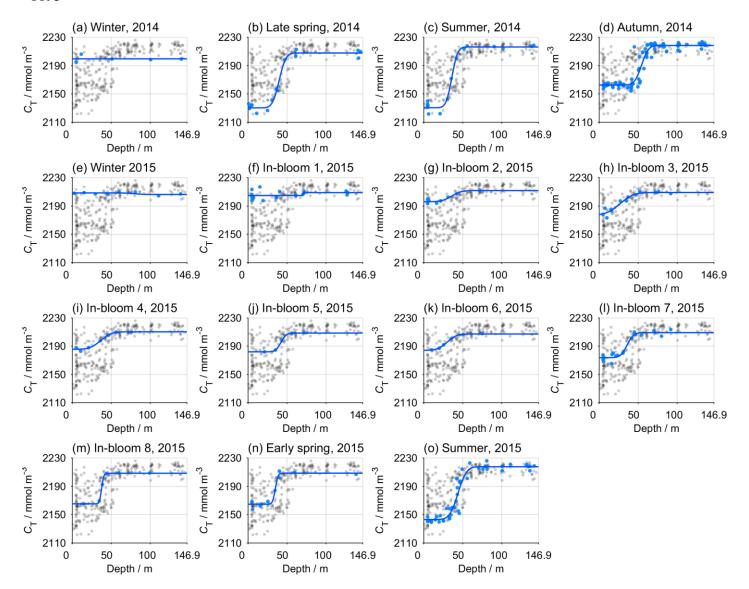


Figure S1. Dissolved inorganic carbon (C_T) profiles at each time point at CCS (Table 1). The grey circles show the entire CCS dataset, while the blue circles highlight the data at each time point. The blue line shows the final vertical profile used for inventory calculations (Supp. Table S2).

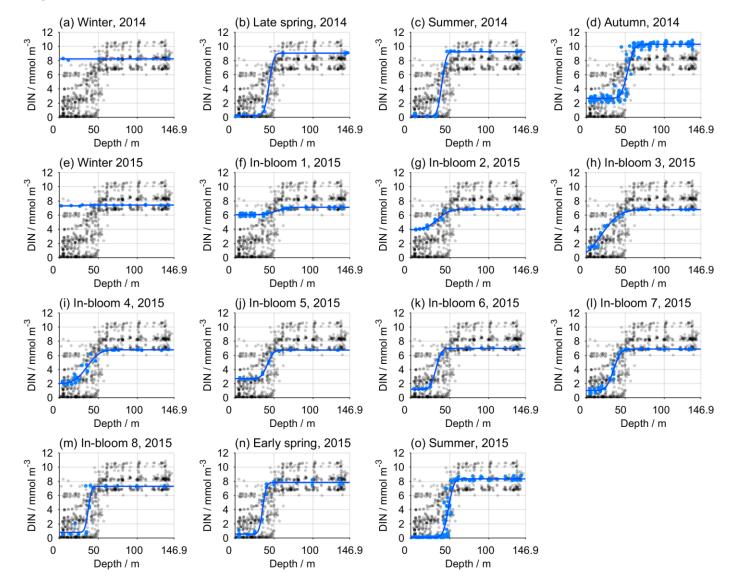


Figure S2. Dissolved inorganic nitrogen (DIN, nitrate + nitrite + ammonium) profiles at each time point at CCS (Table 1). The grey circles show the entire CCS dataset, while the blue circles show the data at each time point. The blue line shows the final vertical profile used for inventory calculations (Supp. Table S2).



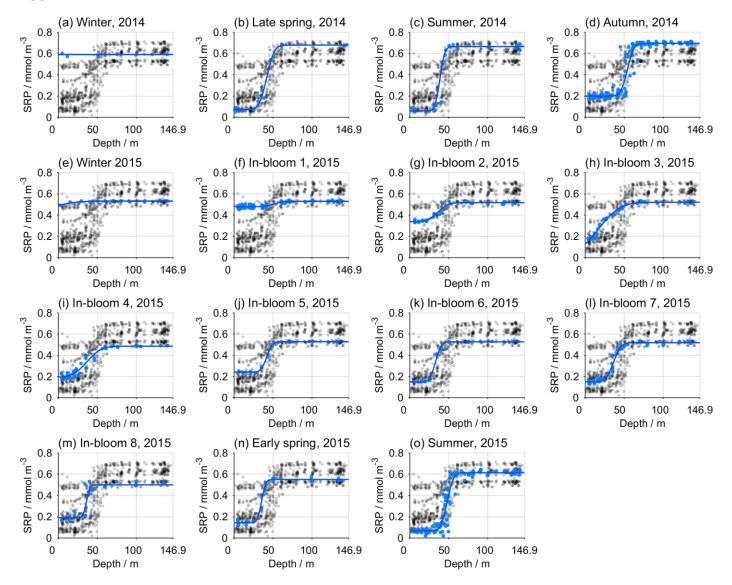


Figure S3. Soluble reactive phosphorus (SRP) profiles at each time point at CCS (Table 1). The grey circles show the entire CCS dataset, while the blue circles show the data at each time point. The blue line shows the final vertical profile used for inventory calculations (Supp. Table S2).



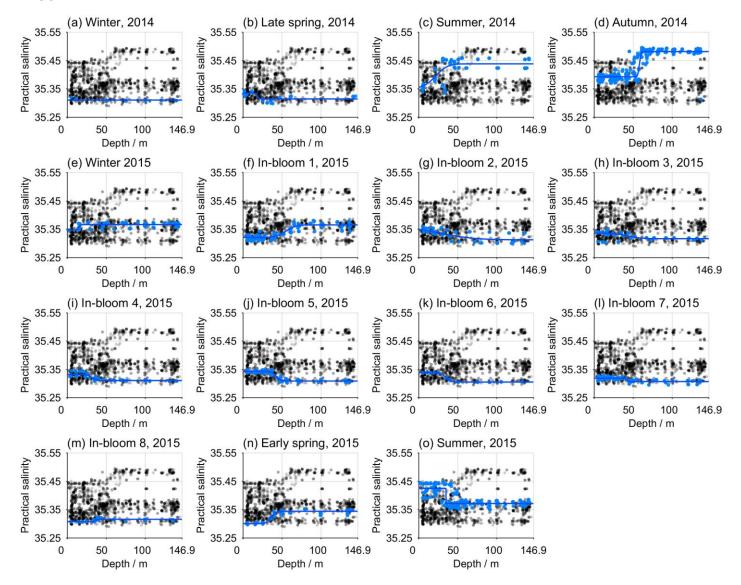


Figure S4. Practical salinity profiles at each time point at CCS (Table 1). The grey circles show the entire CCS dataset, while the blue circles show the data at each time point. The blue line shows the final vertical profile used for inventory calculations (Supp. Table S2).

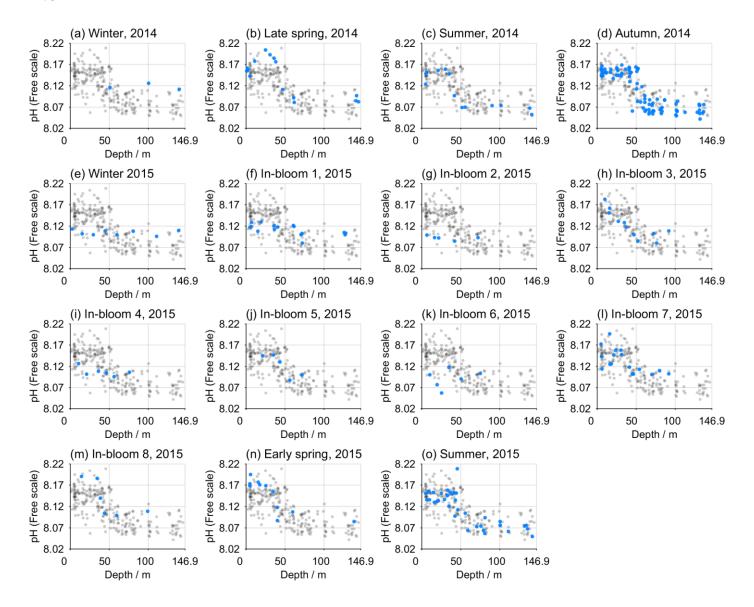


Figure S5. Seawater pH on the Free scale at each time point at CCS (Table 1), calculated from A_T and C_T measurements (van Heuven et al., 2011). The grey circles show the entire CCS dataset, while the blue circles show the data at each time point.

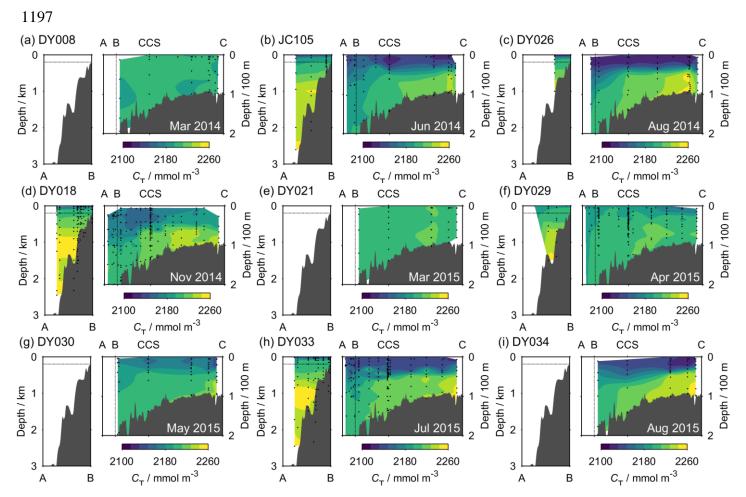


Figure S6. Transects of dissolved inorganic carbon (C_T) measured across the Celtic Sea for all of the SSB cruises. In each panel, the area above the horizontal dotted line in the left plot is the same as that to the left of the vertical dotted line in the right plot. Black points indicate the sample locations. The geographical locations of points A, B, CCS and C are shown by Fig. 1.

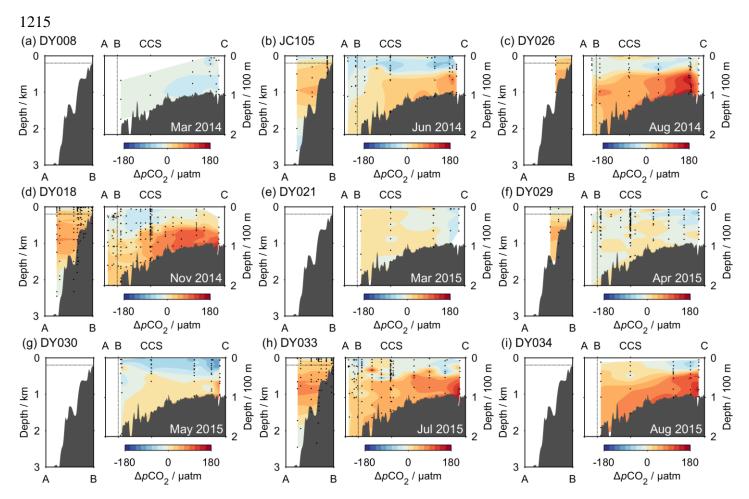


Figure S7. Transects of Δp CO₂, calculated from C_T and A_T measurements, across the Celtic Sea for each SSB cruise. The layout is as described for Fig. 7.

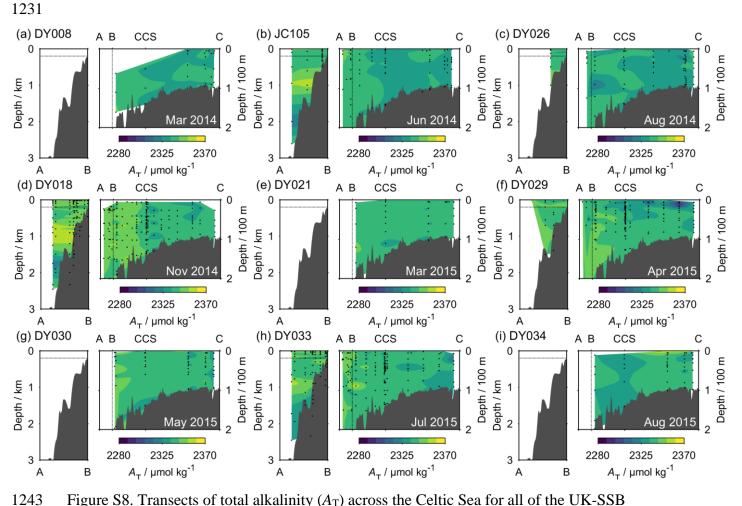


Figure S8. Transects of total alkalinity (A_T) across the Celtic Sea for all of the UK-SSB cruises. In each panel, the area above the horizontal dotted line in the left plot is the same as that to the left of the vertical dotted line in the right plot. Black circles indicate sample locations.



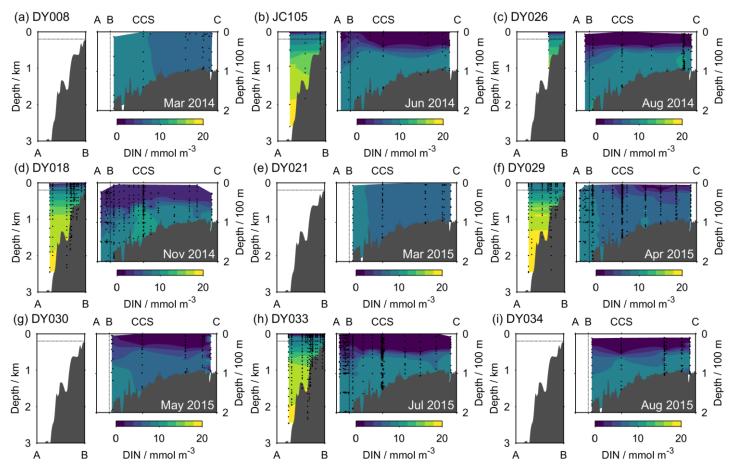


Figure S9. Transects of dissolved inorganic nitrogen (DIN, nitrate + nitrite + ammonium) across the Celtic Sea for all of the UK-SSB cruises. In each panel, the area above the horizontal dotted line in the left plot is the same as that to the left of the vertical dotted line in the right plot. Black circles indicate sample locations.

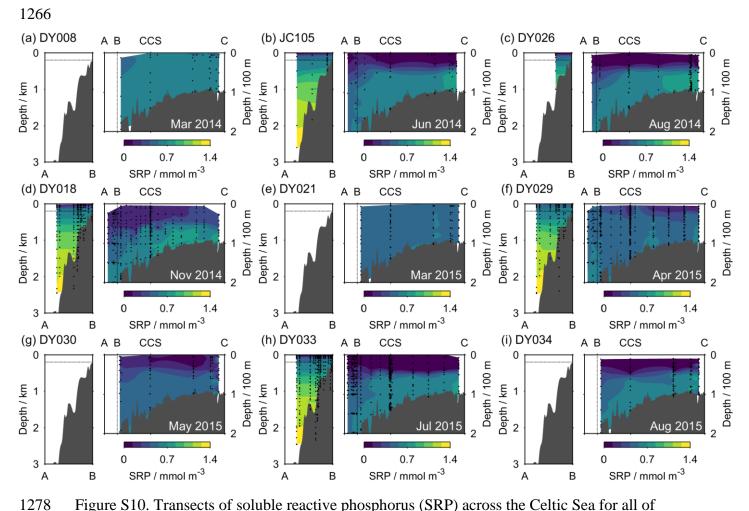


Figure S10. Transects of soluble reactive phosphorus (SRP) across the Celtic Sea for all of the UK-SSB cruises. In each panel, the area above the horizontal dotted line in the left plot is the same as that to the left of the vertical dotted line in the right plot. Black circles indicate sample locations.

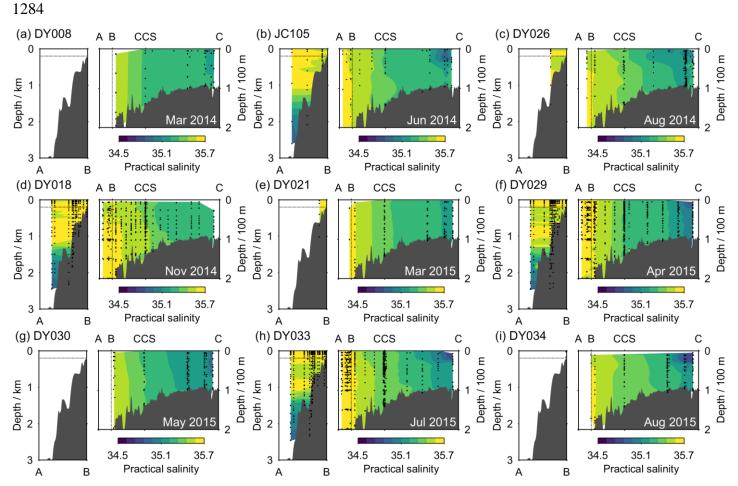


Figure S11. Transects of practical salinity across the Celtic Sea for all of the UK-SSB cruises. In each panel, the area above the horizontal dotted line in the left plot is the same as that to the left of the vertical dotted line in the right plot. Black circles indicate sample locations.



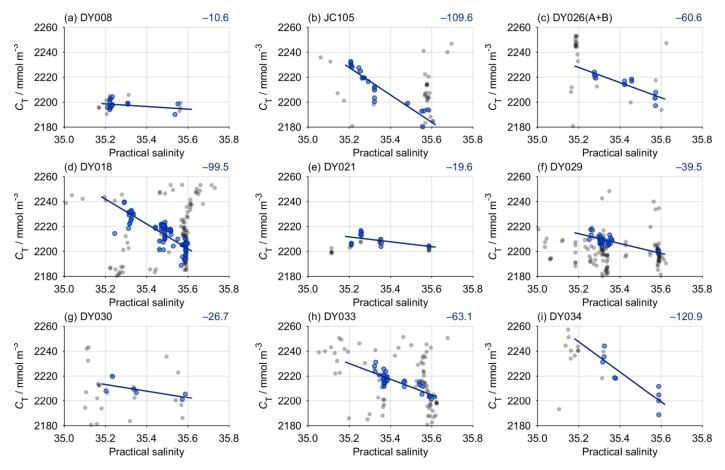


Figure S12. Relationship between dissolved inorganic carbon (C_T) and practical salinity for each UK-SSB cruise. The points used to generate each regression are shown in blue (Section 2.8), and the slope is indicated towards the top right of each panel (see also Fig. 7 and Table 3).

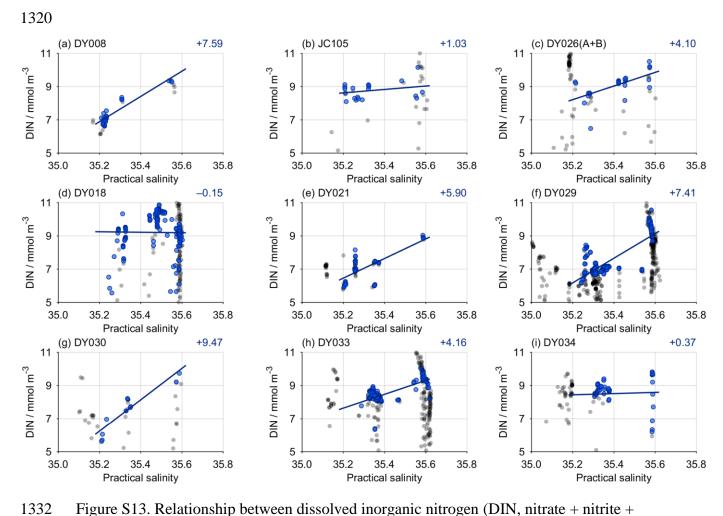


Figure S13. Relationship between dissolved inorganic nitrogen (DIN, nitrate + nitrite + ammonium) and practical salinity for each UK-SSB cruise. The points used to generate each regression are shown in blue (Section 2.8), and the slope is indicated towards the top right of each panel (see also Fig. 7 and Table 3).



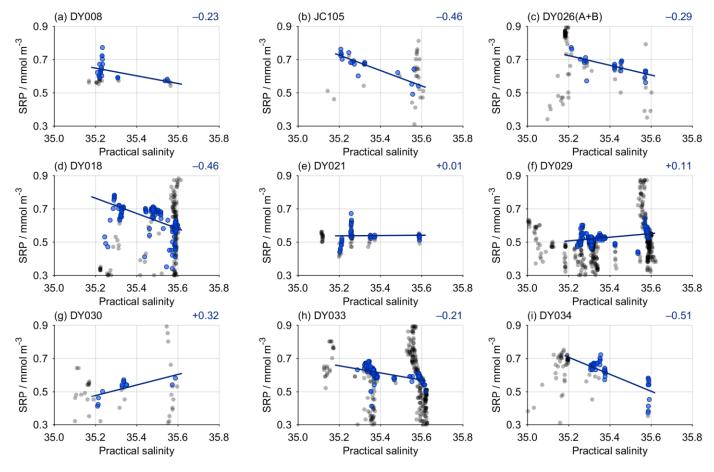


Figure S14. Relationship between soluble reactive phosphorus (SRP) and practical salinity for each UK-SSB cruise. The points used to generate each regression are shown in blue (Section 2.8), and the slope is indicated towards the top right of each panel (see also Fig. 7 and Table 3).