## Supplementary Information for:

## 2 A cryogel-based bioreactor for water treatment applications

- 3 Dmitriy Berillo\*<sup>1</sup>, Jonathan L.Caplin<sup>2</sup>, and Andrew Cundy<sup>3</sup> Irina Savina<sup>1</sup>
- <sup>4</sup> School of Pharmacy and Biomolecular Sciences, University of Brighton, Brighton, UK.
- <sup>2</sup> School of Environment and Technology, University of Brighton, Brighton, UK.
- 6 <sup>3</sup> School of Ocean and Earth Science, University of Southampton, Southampton, UK.
- 7 \*Corresponding author e-mail: dmitriychemist@gmail.com

8

1

9 **Summary**: this supplementary information contains 9 figures and 2 tables in 19 pages.

10

11

### 2. Materials and Methods.

Two known strains of phenol-degrading bacteria Pseudomonas mendocina (P. mendocina or 12 Pse) (NCIMB 13264) and Rhodococcus koreensis (R. koreensis or Rho) (NCIMB 13709) were 13 purchased from the National Collection of Industrial, Food and Marine Bacteria, UK. 14 Acinetobacter radioresistens (A. radioresistens or Acn) was obtained from the culture 15 collection PABS, UoB (Acinetobacter 16S rRNA analysis is given in supplementary 16 17 information). Tryptone soya broth (TSB), tryptone soya agar (TSA) and phosphate buffered saline (PBS) were obtained from Oxoid Ltd, UK. Live/Dead Bac Light kit (L7007) sodium 18 chloride, phenol, ammonium chloride, Gellan Gum (Gel)(technical grade) were purchased 19 from Fisher Scientific Co, UK. Fluorescein isothiocyanate (FITC) and rhodamine B were 20 purchased from Sigma Aldrich, UK. 4-aminoantipyrine, glutaraldehyde (GA) solution 50 w/v 21 %, polyvinyl alcohol (PVA) (deacetylation degree 86.7-88.7%, Mw 67,000 Da), 22 polyethyleneimine, linear (Mw 423 Da), chitosan (CHI) (medium viscosity), Pur-A-Lyzer TM 23 Mega 1000 Dialysis Kit (1000 Da) and (14 kDa) pore size were purchased from Sigma-Aldrich. 24 Potassium ferricyanide and ammonium hydroxide (28-30 wt/v % solution of NH<sub>4</sub>OH in water) 25 Gellan Gum were obtained from Acros Organics, UK. 26

### 2.1 Acinetobacter 16S rRNA analysis

The presumptive *Acinetobacter* isolates were characterised biochemically by Gram staining, growth on differential and enriched media, the oxidase test, the catalase test, the motility test, (MacFaddin et al., 2000 and 2000) the Phenylalanine Deaminase test, (MacFaddin et al., 2000) and with API 20E and API 20NE kits (BioMerieux, France). Isolates identified as presumptive *Acinetobacter* species were further identified using 16S rRNA analysis.

Molecular identification was performed via the polymerase chain reaction (PCR), using the 16S rRNA universal primers (27F: 5'-AGAGTTTGATCMTGGCTCAG-3' and 1492R: 5'-ACCTTGTTACGACTT-3') for DNA amplification.( Weisburg et al., 1991; Jiang et.al 2013) DNeasy TM Blood & Tissue kit (Qiagen, UK) was used to extract and purify DNA following the manufacturer's instructions. The extracted DNA was diluted with biologically sterilised water at a ratio of 1:10 and mixed using a Mega Mix- Royal solution kit (Microzone, UK), containing Taq polymerase, 2x enhancing buffer (6 mM MgCl2) with 400 μM dNTPs and blue MiZN loading dye and stabiliser, prior to PCR. The stock DNA extractions were stored at -20oC for further use.

A 12.5μl volume of Mega Mix- Royal solution was added to an equal volume of DNA, primer and water to give 25 μl total volume to activate the tag polymerase. Following initial denaturation at 95°C for 5 minutes the PCR thermocycler (G-Storm, LabTech International, UK) was run for 30 cycles of 95°C for 1 minute, 50°C for 45 seconds, with extension at 72°C for 1.5 minutes and a final extension of 72°C for 10 minutes.

Separation and visualisation of the amplified DNA fragments was achieved through agarose gel electrophoresis of 10 µl of amplification products with an agarose concentration of 1.5% agarose. A size ladder up to 4500pb was used with distilled water as a control, with electrophoresed Tris-Acetic-EDTA (TAE) buffer for three hours at 100 volts on a HU15 horizontal gel units (LabTech International, UK).

After electrophoresis the gels were stained using ethidium bromide at a concentration 0.5% for 15 minutes then, after washing in water, imaged and documented using the SynGene programme and SynGene INGENIUS UV imager(Fig. S1). Identification of DNA sequences was performed on 25  $\mu$ l aliquots of amplification products at a concentration of 20ng/ $\mu$ l by the Eurofins Genomics DNA sequencing service.

The API20NE biochemical identification was confirmed by comparison of the identity code (0000032) with the API database online; this resulted in a 96.7% similarity to *Acinetobacter radioresistens*.

The result of 16s rRNA sequences of *Acinetobacter radioresistens* were compared with those in the Genebank database through BLAST analysis, using the Greengen's website (http://greengenes.lbl.gov/cgi-bin/JD\_Tutorial/nph-16S.cgi). From this the identity of the presumptive phenol-degrading *Acinetobacter* isolates were identified as *Acinetobacter radioresistens* strains SK82 at a probability of 99.80% and 100% for the forward and reverse primers.

2.2 Estimation of toxicity of cross-linking polymers.

Cultured media (80 mL) was centrifuged at 10 000 rpm for 10 minutes at 4°C. The obtained pellet was re-suspended in PBS buffer. The 0.5 mL suspension of bacteria in PBS buffer was used as a positive control. 0.5 mL of the suspension was mixed with PEI-al, PVA-al, and PVA polymers, and a combination PEI-al & PVA-al. GA (0.25 and 0.5 v/v %) was used as a negative control. Part of the samples were kept at 4 °C for 24h, while the other part of the same composition solution was frozen at -12 °C and kept frozen for 24h. Then, the frozen samples were defrosted at room temperature and the number of live bacteria estimated using MTT assay.

The viability of cross-linked bacteria within the cryogel structure was evaluated using the LIVE/DEAD® bacterial viability kit and confocal laser scanning microscopy (CLSM, Leica TCS SP5). Hydrogel-Rho-Gellan0.5% Hydrogel-Rho-Gellan(1.0%) and Cryo-Rho-

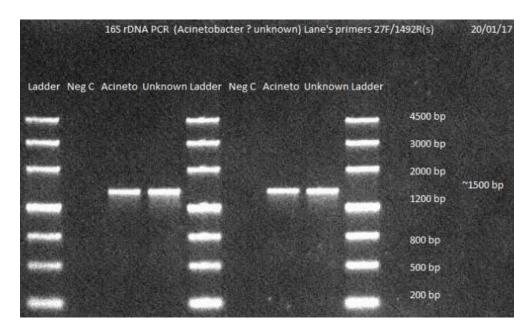
Gellan(0.5%) PVAal (0.3%) cryogels were cut into slices with thickness of 1 mm. The samples were rinsed twice with 0.9% NaCl buffer and stained with SYTO 9 stain (wavelength 480/500 nm) and propidium iodide (wavelength 490/635 nm) for 15 minutes at room temperature under dark conditions according to the protocol of the LIVE/DEAD ® Bacterial viability kit.

## 2.3 Monitoring of phenol derivative concentration

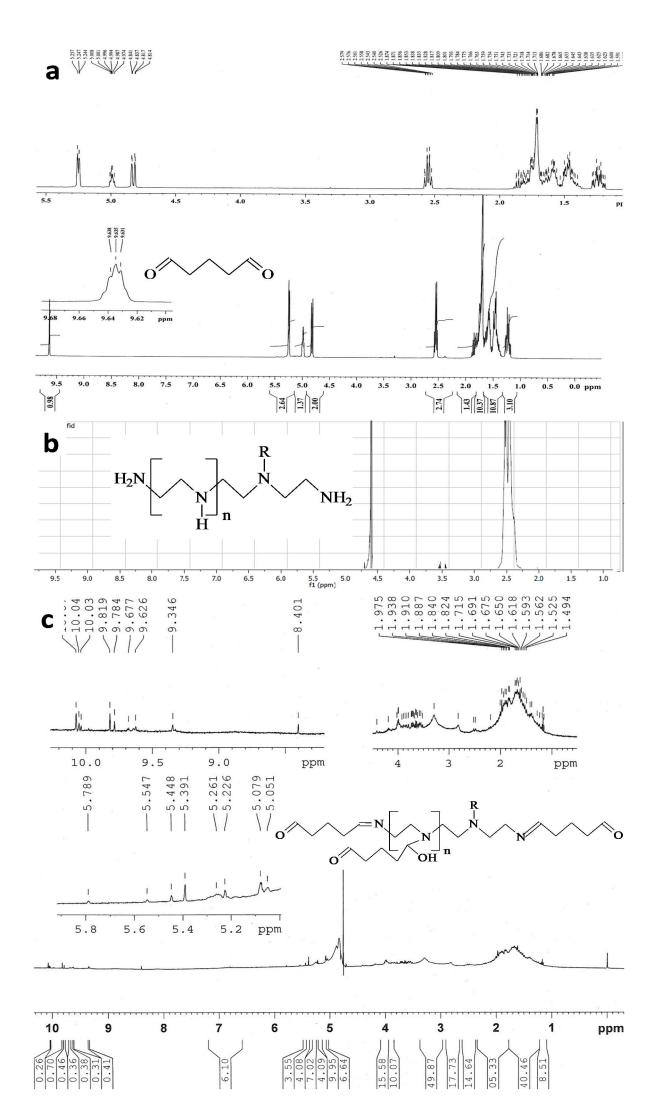
For assessment of phenol, m-cresol, 2-CP and 4-CP concentration, samples of 100 μL volume were transferred into an eppendorf tube (1.5 mL) and diluted with water (0.8 mL). Fifty μL of ammonium buffer was added to adjust pH to ten. 4-aminoantipyrine (25 μL, 20 mg/mL) solution followed by 25 μL potassium ferricyanide (0.242 mol/L) was added. The mixtures were vortexed and centrifuged at 13 000 rpm and absorbance at 510 nm was measured. The final concentrations of phenol derivatives and their possible by-products in water after degradation by the 3D, cryostructured, macroporous bioreactors (henceforth termed cryobacteria reactors, or CBRs) were estimated using an Agilent 1100 HPLC. All samples were diluted 1:1 with HPLC grade methanol and frozen at -20 °C for 2-3h to precipitate unwanted compounds. Samples were centrifuged at 10 000 rpm for five minutes and filtered through 0.22-μm filter membranes before HPLC analysis using a UV detector at 280 nm. A reverse phase column C18 was used, and the flow rate of the mobile phase was 1 mL/min.

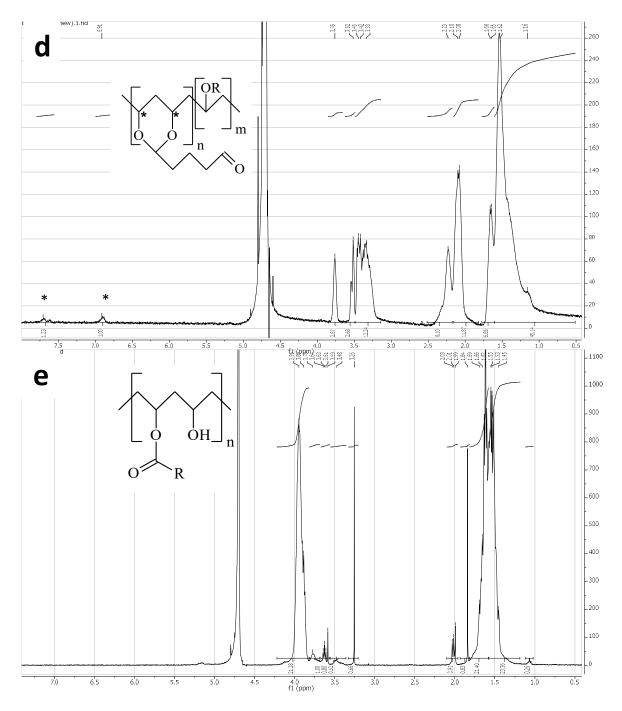
HPLC method. Mobile phase Acetic acid 1 % v/v in water (C) and methanol (D) was used. Column and sample temperatures were 45 °C and 25 °C, respectively; C) gradient start at 75.0, End at 44.0 %; D) gradient start at 25.0, end at 56.0 % for 9.5 min, and at composition 44 % - 56 % C-D for 5 min; C) gradient start at 44.0, end at 0 % with duration of 1.0 min and D) gradient start at 56.0 %, end at 100.0 %, for 1.0 min. 5 minutes at 100 % of CH<sub>3</sub>OH & 1 % HAc. C) gradient start at 0, end at 75.0 % for 1.0 min; D) gradient start at 100.0, end at 25.0 % with duration of 1.0 min. Retention time (min) used: phenol (5.58-6.6), p-cresol (8.88), m-cresol (8.99), p-hydroxybenzoic acid (3.4), hydroquinone (1.97), protocatechuate (2.396), 2-chlorophenol (9.44), 4-chlorophenol(10.75).

# **3. Results**



**Fig. S1**. Electrophoresis of 16S rRNA *Acinetobacter* unknown(Gramm negative, coccus/coccobacillus, oxidise negative, catalase positive, non-fermentative) and comparison with *Acinetobacter radioresistence*.





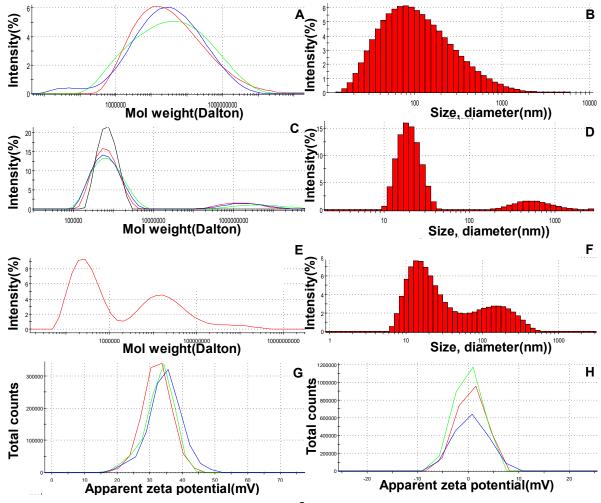
**Fig. S2.** H-NMR-spectra recorded in D<sub>2</sub>O of: a) glutaraldehyde, b) PEI, c) PEI-al, d) PVA-al, e) PVA.

## Characterisation of new polymers and morphology of materials

A group of multiplets in the range of 3.55-3.70 ppm in the spectra (Fig. S1a) corresponds to backbone protons of PEI, which shifted after the modification of the polymer. The condensation reaction of the primary and secondary amino groups of PEI with GA leads to the disappearance of amine functionality at 2.6 ppm (Fig. S1b). The backbone protons of GA

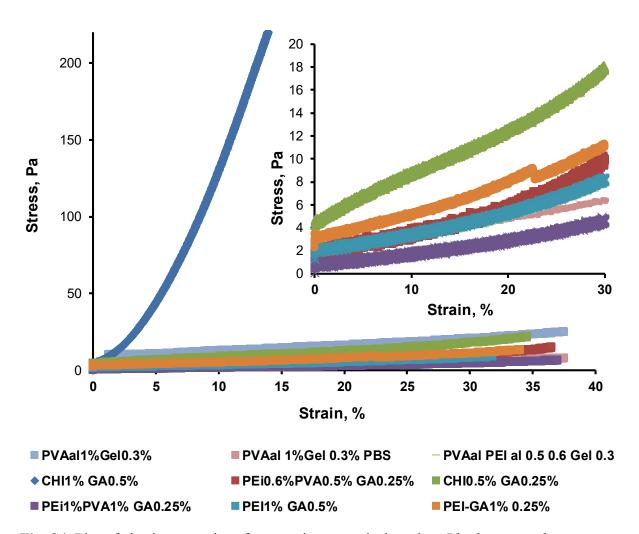
moiety appeared at 1.25-1.75 ppm and in the field of 5ppm attributed to hemiacetal form of GA (Fig. S1d)( Wang et al. 2014). Aldehyde groups were observed at 9.6 ppm (Fig. S1a). The proton of Schiff's base in the <sup>1</sup>H-NMR spectrum of PEI-al appeared at 8.36 ppm (Ederer et al., 2007; Boghaei et al., 2008; Makhubela et al., 2011). PVA-al has two additional signals at 7.7 and 6.3 ppm which are attributed to presence of protons of the cyclic acetal form and multiplets at 2.23 ppm related to glutaraldehyde moieties(CH<sub>2</sub>(4H)) and the singlets at 2.1ppm attributed to acethyl groups(CH<sub>3</sub>) of PVA (Fig. S1c). Taking into account the relation of integrals of peaks for backbone -(CH<sub>2</sub>-CH(R))n- of PVA at 1.6ppm and the characteristic signal of glutaraldehyde (CH<sub>2</sub>(4H)) one can conclude that the ration between structural unit of PVA to GA was 14.8: 1(Fig. S1c). FTIR spectra of PEI-al and PVA-al showed carbonyl groups at 1714 cm<sup>-1</sup> and 1717 cm<sup>-1</sup>, respectively, which indicated the presence of free aldehyde groups in the polymer structure.





**Fig. S3**. Hydrodynamic weight distribution of polymers: a & b) PEI-al; c & d) PVA; e & f) PVA-al. Zeta potential of polymers in water: g) PEI-al; h) PVA-al.

The concentration of aldehyde groups in PVA-al was estimated using a known reaction with hydroxylamine and subsequent titration of released hydrochloric acid(Zhao et al., 1991). Thus, the 2% PVA-al solution contained 6.25 mM/L of aldehyde groups, while the 1% PEI-al solution contained approximately 57 mM/L of aldehyde groups, calculated via measuring the 2.4-dinitrophenylhydrazine assay, which was used previously for estimation of amount of aldehyde groups in oxidised dextran(Berillo et al. 2012). PVA-al and PEI-al contained 0.3125 mM/g<sub>dry polymer</sub> and 5.7 mM/g<sub>dry polymer</sub> of aldehyde groups, respectively.



**Fig. S4.** Plot of elastic properties of composite cryogels based on *Rhodoccoccus korensis* using different polymer compositions and cross-linking agents.

**Table S1.** Evaluation of toxicity of polymers at different compositions and temperature, prior to MTT assay incubation of bacteria (at  $4 \, ^{\circ}$ C for 24h, otherwise specified) of polymers with PVA-al and PEI-al 0.6:0.6% (1:1).

Sample name	% of viable	Number of viable
~ ····································		
	cells	cells, CFU xE6
Pse & PBS (positive control)	50.0	62.5±1.4
Pse & PBS and 2% glucose.	43.0	53.9±1.15
Cryo-Pse-PEI-al:PVA-al-glucose 0.6:0.6:2.0(%); -12°C	13.7	17.2±5.4
Cryo-Pse-PEI-al:PVA-al 0.6:0.6(%); -12°C	12.0	15.0±0.25
Pse & PEI-al:PVAal 0.6:0.6(%)	34.4	43.2±5.4
Pse & PEI-al:PVA-al-glucose 0.6:0.6:2.0(%)	32.6	40.9±4.58
Pse & PBS initial (positive control)	100.0	125.53±2.8
Pse & PBS and 2% glucose initial	86.2	108.25±2.3
Rho & PBS (positive control)	100.0	802±15
Rho & PEI-al:PVA-al 0.6:0.6% and 0.9% NaCl	96.2	772.3±41
Rho & PVA-al 2.4%	76.5	613.8±14
Rho & PVA-al 3%	72.4	581±7
Rho & PEI 1.0% GA 0.5%(6000 CFU initially)	20.3	163±3.4
Rho & 0.5% GA in PBS (6000 CFU initially)	17.1	$137.4 \pm 5$
Acn & PEI-al 3%	75.2	44.6 ±1.1
Acn & PVA-al 2.4%	69.5	41.2 ±0.41
Acn & PEI 1% PBS	49.7	$29.5 \pm 0.86$
Acn & PEI 1% PBS 0.25% GA	13.7	8.12 ±0.15
Acn & GA 0.25% PBS negative control	7.6	$4.54 \pm 0.22$
Acn & PBS buffer positive control	100.0	59.3 ±1.2
Acn & PEI-al 0.365%	25.12	$14.9 \pm 0.3$
Acn & PVA-al 1.2%	92.4	54.8 ±0.23

Acn & PEI-al 0.5% PBS	13.8	$8.2 \pm 0.2$	
Cryo-Acn-PEI-al-PVA-al 0.6:0.6%; -12C	24.3	$14.4 \pm 1.9$	
Cryo-Acn-PEI-al-PVA-al-glucose 0.6:0.5:2.0%; -12C	17.4	$10.3 \pm 0.74$	
Cryo-Acn-PVA-al-glucose 1.2:2.0(%) PBS; -12C	109.1	$64.7 \pm 0.97$	
Acn & PVA-al-glucose 1.2:2.0%, PBS	54.13	32.1±7.9	
Acn & PEI-al: PVA-al0.6:0.6%, PBS	44.5	26.4±0.9	

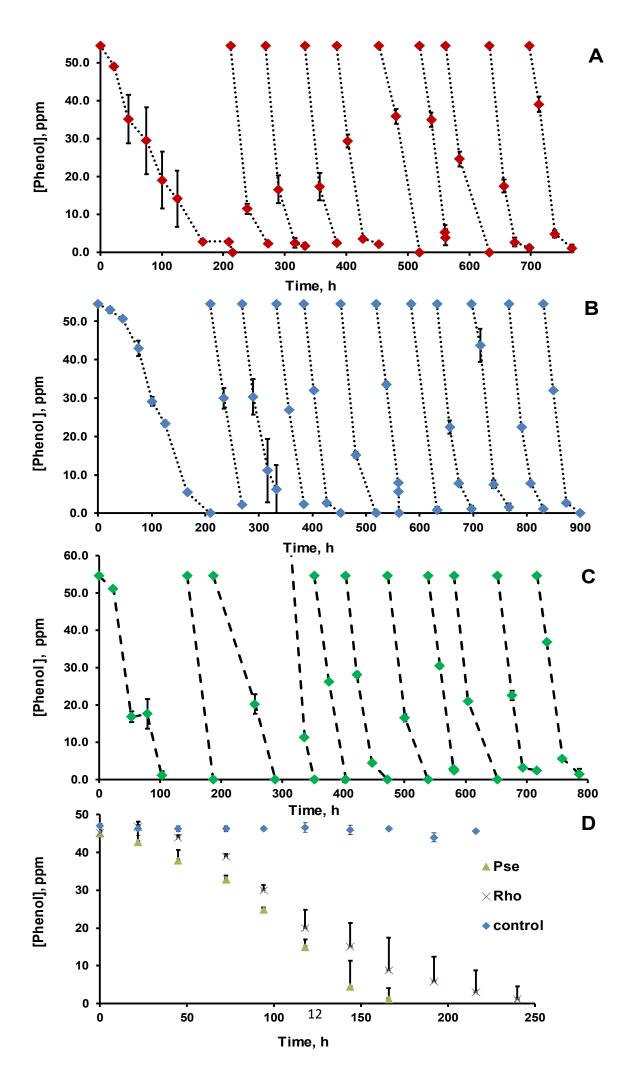
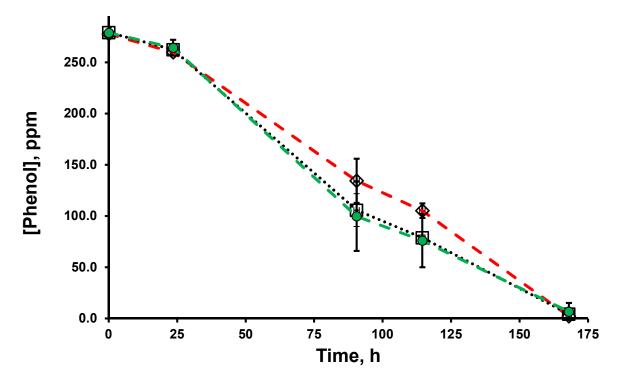
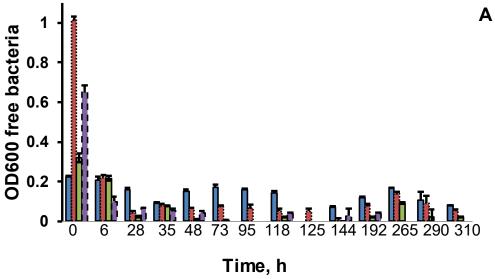


Fig. S5. Degradation of phenol by: A) CBR-Acn, B) CBR-Pse, C) CBR-Rho (2 plastic carriers (12.2±3.6x10<sup>6</sup> CFU) per each bottle V 200 mL; 54 ppm for 10 cycles, (n=3).

D) Control experiment bioremediation of phenol(50ppm) in presence of suspension of bacteria Pseudomonas mendocina (▲), Rhodococcus korensis (x), solution of phenol without bacteria(♦), for 1<sup>st</sup> cycle (amount of bacteria comparable to amount of bacteria used in cryogels)(n=2).



**Fig. S6.** Degradation of phenol in MSM vs time by: CBR-*Pse*  $(38\pm8.3x10^6 \text{ CFU})(\bullet)$ ; CBR-Rho  $(576\pm52x10^6 \text{ CFU})(\diamond)$ ; CBR-Acn  $(17.6\pm0.96x10^6 \text{ CFU})(\Box)$ , for 5<sup>th</sup> cycle, 2 plastic carriers per bottle V 200 mL, 280ppm phenol;



B

O.6

O.5

O.4

O.2

O.1

O.2

O.2

O.1

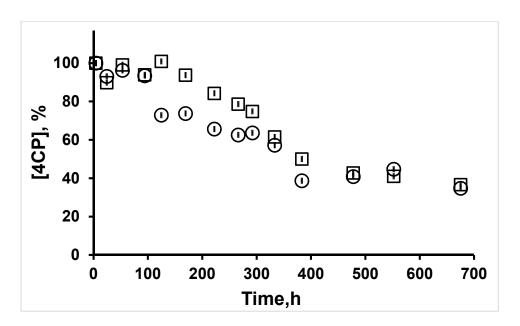
Time, h

**Fig. S7**. Dynamics of change of viable bacteria of *Pse* (blue and green) and *Rho* (red and violet) in the 50 ppm 4-CP solution in a dynamic mode (shaking at 150 rpm) in buffer: A) 25 mM carbonate buffer; B) in MSM, (n=3).

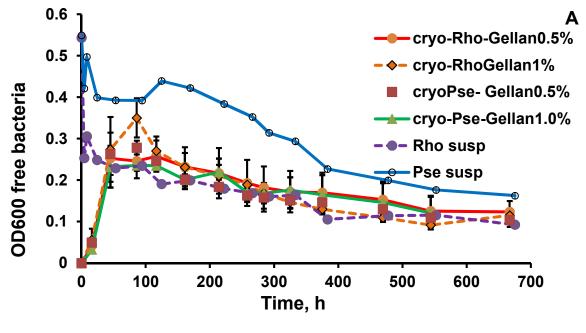
**Table S2.** Concentration of 4-CP estimated using HPLC after the bioremediation period.

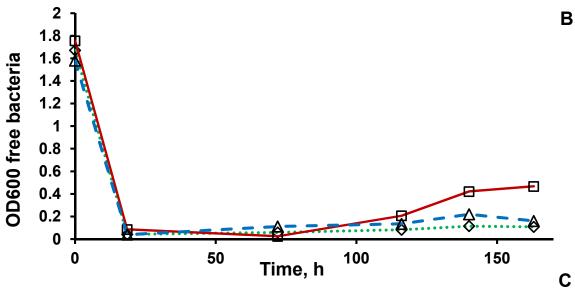
Initial concentration of the 4-CP was 62 ppm, volume 40 mL.

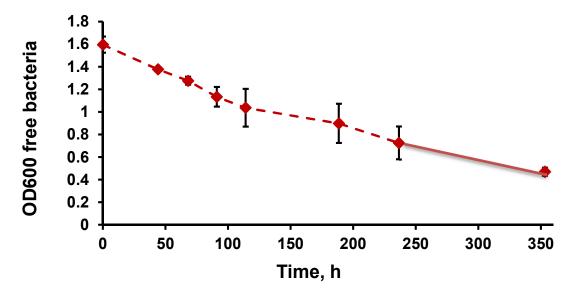
Name of the sample	Remaining phenol after	remaining
	one month	phenol from
	bioremediation, ppm	initial, %
Rho suspension, MSM buffer	45.2	72.9
Control PEIal-PVA-al cryogel (no bacteria) MSM	59.0	95.1
buffer		
Control PVA-al cryogel (no bacteria) MSM buffer	61.31	98.9
Cryogel <i>Pse</i> PVA-al PEI-al 0.5% 0.6% <b>MSM</b>	34.84	56.2
Cryogel <i>Pse</i> PVA-al PEI-al 1.0%: 0.25% <b>MSM</b>	31.85	51.3
Cryogel Pse PVA-al 1% MSM buffer	25.95	41.8
Rho suspension, carbonate buffer	59.26	95.5
Rho suspension, carbonate buffer	61.07	98.3
Rho suspension, carbonate buffer	59.63	96.2
Pse suspension(commercial) carbonate buffer	55.76	89.9



**Fig. S8.** Degradation of 4-CP (V 40 ml, 50 ppm) in MSM by suspension of free *Pse* 4-CP adapted ( $\square$ ), *Rho* 4-CP adapted (O) vs time in a dynamic mode (shaking at 150 rpm), number of cell  $0.6 \times 10^8$  per sample (n=2), concentration estimated using aminoantipyrine assay.







- 180 Fig. S9. Dynamics of change of the population of free bacteria during the bioremediation
- process in a dynamic mode (shaking 150 rpm): A) 50ppm of 4-CP solution in MSM;
- B) nonadapted on the agar plate, cultivated suspension *Rho* in MSM: 2-CP 60 ppm (▲); 2-
- 183 CP 20 ppm ( $\stackrel{\blacksquare}{}$ ); 4-CP 10 ppm ( $\stackrel{\bullet}{}$ ) (n=3). C) Rho adapted to CP 25 ppm) during the
- bioremediation process of 2-CP in MSM in a shaking mode (n=2).

186

### References:

- 187 Berillo D., Elowsson L., Kirsebom H. Oxidized dextran as cross-linker for chitosan cryogel
- 188 scaffolds and formation of polyelectrolyte complexes between chitosan and gelatin
- 189 Macromolecular Bioscience 2012, V.12, I.8, P.1090–1099
- 190 https://doi.org/10.1002/mabi.201200023
- 191 Jiang, L., Ruan, Q., Li, R., & Li, T. (2013). Biodegradation of phenol by using free and
- immobilized cells of Acinetobacter sp. BS8Y. Journal of basic microbiology, 53(3), 224-230.
- 193 https://doi.org/10.1002/jobm.201100460
- Ederer, T., Herrick, R. S., & Beck, W. (2007). Metal Complexes with Biologically Important
- Ligands. CLXVI Tetracarbonyl Complexes of Chromium (0) and Molybdenum (0) with Schiff
- 196 Bases from Pyridine-2-carboxaldehyde and α-Amino Acid Esters as N, N-Chelate Ligands.
- 197 Zeitschrift für anorganische und allgemeine Chemie, 633(2), 235-238.Boghaei, D. M., &
- 198 Askarizadeh, E. (2008). https://doi.org/10.1002/zaac.200600252
- 199 MacFaddin, J. F., (2000) Catalase-Peroxidasetest In Biochemical tests for identification of
- 200 medical bacteria, Lippincott Williams & Wilkins. p.78-97: London; Philadelphia; ,; V. 3rd.
- 201 MacFaddin, J. F., (2000) Motility test In Biochemical tests for identification of medical
- bacteria, Lippincott Williams & Wilkins. p. 327-332: London; Philadelphia; V. 3rd.
- 203 MacFaddin, J. F., (2000) Phenylalanine Deaminase Test. In Biochemical tests for identification
- of medical bacteria, Lippincott Williams & Wilkins. p.388-393: London; Philadelphia; V. 3rd.

- 205 DM Boghaei, E Askarizadeh Synthesis and characterization of water-soluble zinc (II) Schiff-
- 206 base complexes derived from amino acids and 3-formyl-4-hydroxybenzyl-
- triphenylphosphonium chloride. Journal of Coordination Chemistry, 61(12), 1917-1926.
- 208 https://doi.org/10.1080/00958970701792794
- 209 Makhubela, B. C., Jardine, A., & Smith, G. S. (2011). Pd nanosized particles supported on
- 210 chitosan and 6-deoxy-6-amino chitosan as recyclable catalysts for Suzuki–Miyaura and Heck
- 211 cross-coupling reactions. Applied Catalysis A: General, 393(1-2), 231-241.
- 212 https://doi.org/10.1016/j.apcata.2010.12.002
- 213 Wang, L. F., Sung, K. H., & Yu-Lun, L. O. (2014). U.S. Patent No. 8,716,399. Washington,
- 214 DC: U.S. Patent and Trademark Office.
- Weisburg, W. G., Barns, S. M., Pelletier, D. A., & Lane, D. J. (1991). 16S ribosomal DNA
- amplification for phylogenetic study. Journal of bacteriology, 173(2), 697-703.
- 217 https://doi.org/10.1128/jb.173.2.697-703.1991
- Zaushitsyna, O., D. Berillo, H. Kirsebom and B. Mattiasson (2014). "Cryostructured and
- 219 crosslinked viable cells forming monoliths suitable for bioreactor applications." Topics in
- 220 Catalysis 57(5): 339-348. https://doi.org/10.1007/s11244-013-0189-9
- Zhao, H., & Heindel, N. D. (1991). Determination of degree of substitution of formyl groups
- 222 in polyaldehyde dextran by the hydroxylamine hydrochloride method. Pharmaceutical
- research, 8(3), 400-402. https://doi.org/10.1023/A:1015866104055