Heterogenous contribution of the Southern Ocean to deglacial atmospheric CO2 rise

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Glacial-interglacial changes in atmospheric CO₂ are generally attributed to changes in ocean carbon chemistry in response to large-scale shifts in the ocean's biology, alkalinity and physical circulation. The modern ocean plays an important role in regulating CO₂, with the Southern Ocean taking up more CO₂ than any other. It is likely to have played a crucial role in controlling past atmospheric CO₂. However, the physical, biological and chemical variables involved in controlling ocean-atmosphere CO2 exchange during glacial-interglacial cycle are not completely understood. Here we use boron isotopes, carbon isotopes in planktonic foraminifera and an alkenone-based proxy of temperature to reconstruct seawater pH and CO₂ partial pressure in sub-Antarctic surface waters, and investigate the mechanisms regulating surface water CO2 over the last glacialinterglacial cycle. Our reconstruction over the last 25,000 years, from a sediment core recovered south of Tasmania, shows surface waters in that region were a sink for atmospheric CO2 during the Last Glacial Maximum. Our reconstruction suggests changes in the strength of the biological pump and the release of deep-ocean CO2 to surface waters contributed to the last deglacial atmospheric CO2 rise. These findings demonstrate that variations in the distribution of Southern Ocean water masses in this sector and upwelling intensity played a key role in regulating atmospheric CO2 during the last glacial-interglacial cycle.

Atmospheric CO₂ concentrations range from minima of ~180 ppmv during glacial intervals to maxima of ~280-300 ppmv during interglacial intervals¹. The ocean is the largest active carbon reservoir; hence, changes in ocean-atmosphere interactions, ocean circulation and marine biogeochemistry likely played major roles in modulating glacial-interglacial (G-IG) atmospheric CO₂ variability²⁻⁵. Despite this understanding of the ocean's overall role in modulating atmospheric CO₂, information on the time and spatial evolution of ocean sourcesink behaviour remains scarce. Determining the factors controlling ocean-atmosphere CO₂

dynamics requires an understanding of the natural physical and biological processes that affect the state of CO₂ in ocean, and this is of key importance in validating the different mechanisms used in geochemical models to explain G-IG variations in atmospheric CO₂ (e.g. refs 6-10). Changes in these processes in the Southern Ocean are hypothesised to have been crucial¹⁰⁻¹⁹, however, direct proxy histories of variables reflecting past G-IG changes in surface-ocean *p*CO₂ are limited to one record of deglacial change in the Atlantic sector of the Southern Ocean¹⁵ that only captures part of the G-IG change in atmospheric CO₂. Our study focuses on the Indo-Pacific sector of the Southern Ocean south of Australia and, for the first time, we present a Southern Ocean surface-ocean carbonate chemistry reconstruction that allows the quantification of ocean-atmosphere CO₂ exchange accompanying the full G-IG rise in atmospheric CO₂.

We analysed δ¹⁸O, δ¹³C, δ¹¹B, and shell weight for the planktonic foraminifer *Globigerina bulloides* in sediment core MD972106 (45°09'S, 146°17'E, water depth 3310 m; see Methods) to reconstruct surface water pH and CO₂ partial pressure (*p*CO₂) over the last 25 kyr (thousand years) in the Southern Ocean region south of Tasmania. Long-term climatological studies²⁰ show that the core site is located at the southern edge of the Subtropical Front (STF), that marks the northern boundary of the Southern Ocean separating the Subtropical Zone (STZ) from the sub-Antarctic Zone (SAZ) surface waters, and is bathed by Lower Circumpolar Deep Water (Fig. 1 and Supplementary Fig. 1). The core site is also located just north (defined by the STF) of the high nutrient, low chlorophyll (HNLC) area of the Southern Ocean²¹. Sea-surface temperature (SST) reconstructions for the Late Quaternary indicate that the STF migrated north of its present position during glacial periods and was at its most northerly position at the last glaciation²². Thus we infer surface waters over MD972106 during the last glaciation had SAZ properties.

Glacial-interglacial palaeo-proxy record

Down-core variations in *G. bulloides* δ^{18} O exhibit a G-IG amplitude of ~2 ‰ (Fig. 2a). The G-IG change in global ocean δ^{18} O is estimated to have been ~1.1 ‰ (ref. 23), hence the ~2 ‰ shift in the *G. bulloides* δ^{18} O record requires a SST (and a carbonate ion effect²⁴) component, which is quantified here using alkenone derived SSTs. The Holocene and Last Glacial Maximum (LGM, ~18 to 22 ka, thousand years ago) δ^{13} C averages are similar, but the G-IG transition displays a minimum at ~12 ka, and a minimum during the Holocene at ~7.4 ka (Fig. 2b). *Globigerina bulloides* shell weights are heavier during glacial times (Fig. 2c). There is also a clear difference in *G. bulloides* δ^{11} B values, with Holocene δ^{11} B about 2.0 ‰ lower than in the LGM (Fig. 2d). Alkenone-based SST estimates show a G-IG change of ~4.1°C with a minimum of 8.2°C at ~21.4 ka, followed by post-glacial warming to 13.9°C at ~11 kyr and cooling to ~13°C over the past ~6 kyr (Fig. 2e). The most recent samples (~2.6 to 2 ka) show deviations in δ^{18} O, shell weight and δ^{11} B compared to earlier Holocene values. The δ^{18} O and shell weights are confirmed with replicate sample measurements.

Down-core *G. bulloides* δ^{18} O, δ^{11} B and shell weights co-vary with each other (Fig. 2) and with atmospheric CO₂ over the last 25 kyr (Fig. 3b). This co-variation is interpreted here to arise from G-IG evolution of surface water temperature and chemistry²⁴⁻²⁹. Because MD972106 was recovered at a depth above and close to the modern calcite saturation horizon (Supplementary Fig. 2), it is important to consider the possible effects of post-depositional calcite dissolution on shell weights^{28,30}, particularly when considering G-IG lysocline changes versus primary calcification driven by surface water carbonate ion concentration ([CO₃²⁻]). Holocene core-top *G. bulloides* shell weights indicate the foraminiferal lysocline at the South Tasman Rise is at ~3,600m water depth³¹. Carbonate preservation indicators, percent calcium carbonate and percent whole foraminifera (Supplementary Fig. 3), suggest that calcite dissolution did not play a significant role over the past 25 kyr at this site.

Reconstructing seawater pH and pCO₂

Past variations in sea surface water pH at the site of MD972106 are calculated from the δ^{11} B-pH relationship for *G. bulloides*³² and the alkenone SST record (see Methods). Estimated pH shows a decline from LGM values of ~8.38 to Holocene values of ~8.16 (Fig. 3a). By comparison, the modern anthropogenic-influenced surface water pH value at the MD972106 site was 8.11 in 2001 (Supplementary Note 1). Variations in surface water pCO_2 (Fig. 3b) over the past 25 kyr can be estimated using the SST alkenone record, the δ^{11} B surface water pH record and estimates of either alkalinity or DIC (see Methods). There is close agreement between pCO_2 calculated using estimates of pH combined with either; (1) estimates of alkalinity or (2) DIC (Fig. 3b). Methods (1) and (2) both use pH estimated from *G. bulloides* δ^{11} B, and hence any differences result from the assumptions underlying estimates of the alkalinity and DIC parameters, respectively. These calculations show surface water pCO_2 rising from ~165 ppmv at the LGM to ~280 ppmv in the early Holocene (Fig. 3b). The interval ~6.3 to 4.2 ka is marked by elevated pCO_2 of ~315 to 360 ppmv that decreases to ~270 ppmv at ~2 ka (Fig. 3b). The 2001 surface water pCO_2 at this site was ~330 ppmv, broadly consistent with the known anthropogenic rise in atmospheric CO₂ (Supplementary Note 1).

The difference between the $p\text{CO}_2$ value of the surface water and atmosphere ($\Delta p\text{CO}_2$) can be calculated by comparison with ice core CO₂ data from Antarctica³³ (Fig. 3b). Estimated $\Delta p\text{CO}_2$ (Fig. 3c, 5e) indicates that surface waters near MD972106 were a net sink for atmospheric CO₂ during the LGM ($\Delta p\text{CO}_2 = -25 \text{ ppmv}$) and continued to operate in this way during the G-IG transition until ~12 ka, where surface waters became a net source of CO₂ during the early Holocene ($\Delta p\text{CO}_2 = 15 \text{ ppmv}$). Between ~6.3 to 4.2 ka the surface waters were a strong net source for atmospheric CO₂ ($\Delta p\text{CO}_2 \sim 50 \text{ to } 90 \text{ ppmv}$), before becoming again a weak net sink ($\Delta p\text{CO}_2 \sim -6 \text{ ppmv}$) at ~2 ka (Fig. 3c, 5e). In addition, past surface seawater

[CO₃²-] calculated from estimates of pH, SST and alkalinity (and DIC) show surface water [CO₃²-] falling from ~265 μ mol kg⁻¹ in the LGM to ~190 μ mol kg⁻¹ for most of the Holocene (Fig. 4a). The anthropogenically perturbed surface water [CO₃²-] at this site is ~183 μ mol kg⁻¹ whereas the pre-industrial [CO₃²-] is estimated to be ~212 μ mol kg⁻¹ (Supplementary Note 2), which is within the uncertainty of the most recent Holocene [CO₃²-] (Fig. 4a).

The shell weight-[CO₃²⁻] relationship²⁷ was also applied to the MD972106 data to reconstruct [CO₃²⁻] for surface waters over the past 25 kyr (Fig. 4b). The close agreement between estimated surface water [CO₃²⁻] derived from these two independent proxies (*G. bulloides* δ^{11} B values and shell weight) reinforces our findings that δ^{11} B of *G. bulloides* yields reliable reconstructions of seawater pH.

Southern Ocean contribution to atmospheric CO₂ change

Our reconstruction is consistent with the hypothesis that the G-IG modulation of the Southern Ocean carbon sink likely arose from changes in SST, biogeochemistry and ocean circulation, specifically; 1) colder LGM seawater increased CO₂ solubility; 2) a stronger 'biological pump' during the LGM reduced the DIC in surface waters; and 3) surface water pH and carbonate ion concentration [CO₃²⁻] was higher during the LGM.

Two hypotheses involving the Southern Ocean have been offered to explain G-IG cycles in atmospheric CO₂. The first invokes increased strength of the biological (or 'soft-tissue') pump in the high-latitude Southern Ocean during the LGM^{6-8,11} in response to an increase in nutrient utilisation, possibly through iron fertilisation³⁴, in sub-Antarctic surface waters. This is supported by studies indicating that export production in the area between the STF and PF were higher during the LGM than the Holocene^{14,35-37}. The second invokes decreased exchange between surface waters and the deep-ocean south of the Polar Front³⁸⁻⁴⁰. The increased stratification reduces upwelling and exchange of nutrient and CO₂ rich deep-

waters with the surface ocean. This is supported by increased nutrient utilisation as shown by $\delta^{15}N$ records³⁸, reduced ventilation of the deep Southern Ocean⁴¹, and subsequent increase in ocean CO₂ storage^{17,19} during the LGM. No single mechanism⁴² can explain the full glacial draw-down in atmospheric CO₂, although a combination of mechanisms has been proposed^{2,14,16}.

Zonal Asymmetry of deglacial change in the Southern Ocean

Our full deglacial Southern Ocean pH and pCO_2 reconstruction can be compared to a reconstruction of late deglacial pH and pCO_2 in the Atlantic sector of the subantarctic Southern Ocean¹⁵, to provide further insights into the subantarctic role in CO₂ change. A reconstruction of surface ocean pCO_2 , spanning ~16-2 kyr and also utilising $\delta^{11}B$ and $\delta^{13}C$ of G. bulloides¹⁵ was undertaken at a site in the Atlantic-sector SAZ at a similar latitude to MD972106. That study, while not spanning the full deglacial amplitude of CO₂ change, suggested surface waters in that sector were in approximate equilibrium with the atmosphere at ~16 ka and became a strong net source of CO₂ to the atmosphere (ΔpCO_2 of ~50 ppmv; Fig. 5e) by ~15 ka, before declining intermittently to reach approximate atmospheric equilibrium at ~4 ka. That pattern was interpreted to suggest that there was enhanced upwelling of deep waters in the Atlantic sector of the SAZ during the deglaciation.

At our site south of Tasmania, surface waters were a net sink for atmospheric CO_2 ($\Delta pCO_2 = -25$ ppmv) during the LGM (Fig. 3c, 5e). The northward migration of the STF at the LGM²² resulted in an expansion of the Southern Ocean water masses, and coincided with increased delivery of bioavailable dust-borne iron to the ocean³⁴; both of these processes would also lead to the expansion of the zone of high biological production in the Southern Ocean. At the same time, the stratification of the deep Southern Ocean during the LGM reduced the contribution of upwelling of more dissolved CO_2 to surface waters. At the site of MD972106,

G. bulloides δ^{13} C temperature- and [CO₃²⁻]-adjusted (δ^{13} C_{adj}) values (see Methods for details on these adjustments), and the δ^{13} C gradient between planktonic and benthic foraminifera ($\Delta\delta^{13}$ C gradient) were largest during the LGM (Fig. 5a, c), indicating high primary productivity^{43,44}. Similarly, alkenone concentrations (Fig. 5d) were highest during the LGM and provide qualitative information on past productivity of coccolithophorid algae^{45,46}. Comparison of alkenone concentrations (Fig. 5d.) and Δp CO₂ (Fig. 5e.) shows that, when there was maximum biological production in this region, CO₂ degassing was suppressed. Our data suggests the biological pump was stronger during the LGM (i.e. the DIC gradient between surface and deep ocean increased), possibly through iron fertilisation^{34,47}, and steadily decreased over the last deglaciation. Changes in the strength of the biological pump could also be related to changes in circulation⁴⁸, via a more isolated deep ocean, in particular south of the Polar Front³⁸.

The δ^{13} C recorded by *G. bulloides* in the Atlantic sector of the SAZ shows a ~1.5 ‰ decrease from the LGM to the onset of maximum CO₂ degassing (not shown)¹⁵, while the MD972106 site shows a smaller decrease (δ^{13} C decrease of ~0.3 ‰, Fig. 2b; δ^{13} C_{adj} decrease of ~0.1 ‰, Fig. 5a) over the same period. The inferred increase in upwelling in the Atlantic sector is not reflected in the reconstructed Δp CO₂ record at MD972106 (Fig. 3c, 5e). Surface waters at MD972106 during the last deglaciation remained a net sink for CO₂ (average Δp CO₂ = -25 ppmv; Fig. 3c, 5e). The steadily decreasing δ^{13} C_{adj} (Fig. 5a) and $\Delta\delta^{13}$ C gradient (Fig. 5c) during the last deglaciation is consistent with a progressive increase in the contribution of upwelling and advection of 'old' CO₂-rich (and ¹³C-depleted water) CDW to sub-Antarctic surface waters. Nevertheless, surface waters remained a net sink for atmospheric CO₂ during this period, suggesting that productivity (drawdown of CO₂) likely kept in pace with any augmented nutrient supply, and compensated for any increased CO₂ from upwelling in this

subantarctic region. Alkenone concentrations (a proxy for coccolithophorid algae primary/export production; Fig. 5d) shows enhanced biological productivity was maintained during this interval. There is compelling evidence for a build-up of carbon in the deep Southern Ocean towards the LGM^{12,17,19}, and subsequent release and ventilation⁴¹ of this deep carbon store during the last deglaciation which is consistent with the rise in atmospheric CO₂ (refs 17, 19).

The interval between \sim 6.3 to 4.2 ka, when surface waters at MD972106 were a net source of atmospheric CO₂ (Δp CO₂ of \sim 50 to 90 ppmv; Fig. 3c, 5e), coincides with a period of strengthened Southern Hemisphere Westerly Winds (SWW) inferred from analysis of terrestrial records^{49,50}. Changes in the latitude and intensity of the SWW are potential drivers of changes in atmospheric CO₂, as suggested by modelling¹⁰, opal burial rates⁵¹ and is with agreement of equatorial and Antarctic carbon isotope variations⁵². Intensification of SWW increases the contribution of upwelled CO₂-rich CDW (and low δ ¹³C water) to surface waters within the Antarctic Zone. *G. bulloides* δ ¹³C_{adj} at MD972106 reach a minimum at \sim 7.4 ka (Fig. 5a) and $\Delta\delta$ ¹³C gradient falls below the Holocene average gradient (Fig. 5c). Our data are thus consistent with the hypothesis that the net CO₂ source during this period resulted from increased upwelling of CO₂-rich CDW within the Antarctic Zone that advected northward to SAZ surface waters by the strengthened SWW^{49,50}.

The deviations in the most recent G. bulloides samples ($\delta^{18}O$, shell weight and $\delta^{11}B$; Fig 2.) stand in contrast to earlier Holocene values, and reflects changing surface water CO_2 dynamics. Surface waters at ~ 2 ka were a weak net sink for pCO_2 ($\Delta pCO_2 = -6$ ppmv) and likely results from the interplay and change in the balance between a stronger 'biological pump' and a corresponding decrease in the contribution and/or intensity of upwelling and advection of CO_2 -rich CDW. The increasing $\delta^{13}C_{adj}$ and $\Delta\delta^{13}C$ gradient (Fig. 5a, c) and the corresponding

increase in alkenone concentrations (Fig. 5d) from ~4.1 to 2 ka is indicative of increased primary productivity, and corresponds to a decrease in wind strength from ~4 to 1 ka^{49,50}.

Our findings demonstrate the interplay and changing balance between the removal of CO₂ by biological productivity in surface waters and CO₂ outgassing via circulation changes during the last G-IG rise in atmospheric CO₂. The modern Southern Ocean shows considerable zonal variability in annual and seasonal air-sea CO₂ flux⁵³, so such spatial heterogeneity in past ocean-atmosphere CO₂ exchange is not surprising. This study provides evidence that the CO₂ dynamics of Holocene surface waters in the Southern Ocean are dependent on competing factors such as the ocean's biological pump and changes in the location and intensity of SWW. The evolving relationship between atmospheric dynamics and carbon exchange is critically important for understanding the role of Southern Hemisphere winds in modulating the uptake of anthropogenic CO₂ (refs 54, 55).

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Author contributions

A.D.M. and W.R.H. designed the study with input from M.R.P. and J.B. The manuscript was written by A.D.M., W.R.H., M.R.P. and J.B. with contributions from M.J.C., E.C., C.P., M.K.G., and T.B.C. Analysis and interpretation of the measurements was completed by

A.D.M., W.R.H., M.R.P., J.B., M.J.C., E.C., C.P., M.K.G., and T.B.C. All authors contributed towards improving the final manuscript.

Competing Interest

The authors declare no competing interests.

Supplementary Information is available in the online version of the paper.

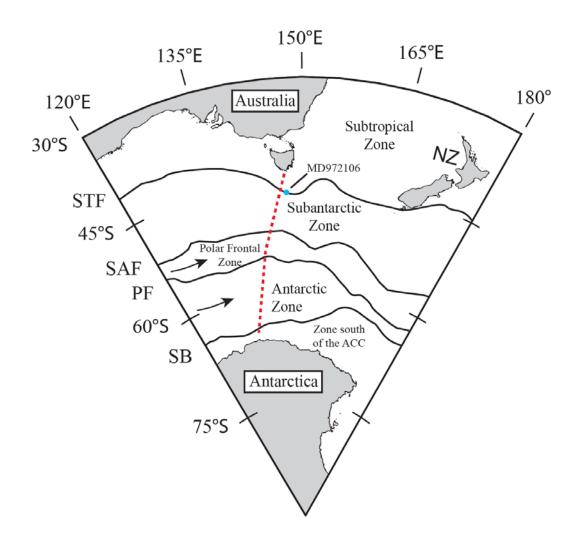


Figure 1. Location of the MD972106 sediment core and modern position of Southern Ocean water masses and fronts. MD972106 is located at 45°09'S, 146°17'E (water depth 3310 m). STF, Sub-Tropical Front; SAF, Sub-Antarctic Front; PF, Polar Front; SB, southern boundary of the Antarctic Circumpolar Current (ACC). Arrows are used to indicate the ACC direction. Dashed red line shows the repeat hydrographic transect between Tasmania and Antarctic along ~140° E (WOCE SR3 section line).

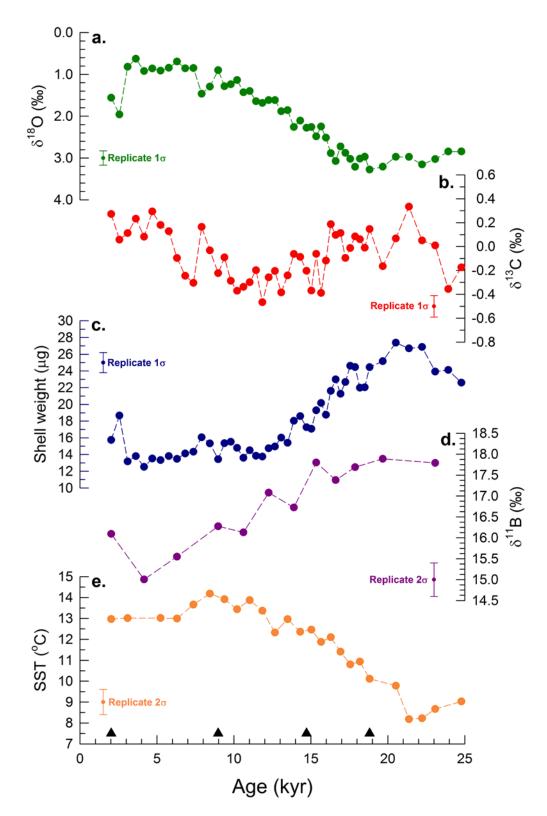


Figure 2. δ^{18} O, δ^{13} C, shell weight and δ^{11} B for the planktonic foraminifer, *Globigerina bulloides*, and palaeo SST estimates from alkenones in sediment core MD972106. Measured *G. bulloides* **a**, δ^{18} O (green circles), **b**, δ^{13} C (red circles), **c**, shell weight (blue circles) and **d**, δ^{11} B (purple circles). **e**, Palaeo SST estimates from MD972106 obtained from alkenones

unsaturation ratio $(U_{37}^{K\prime})$ (orange circles). Age model was constructed using calibrated ^{14}C ages (black triangles; Methods). Error bars represent 1σ uncertainties for $\delta^{18}\text{O}$, $\delta^{13}\text{C}$, and shell weight, and 2σ uncertainties for $\delta^{11}\text{B}$ based on replicate measurements (Methods). Error bars on palaeo SSTs represent the 2σ analytical uncertainty for replicate measurements on a homogenous laboratory (sediment) standard (Methods).

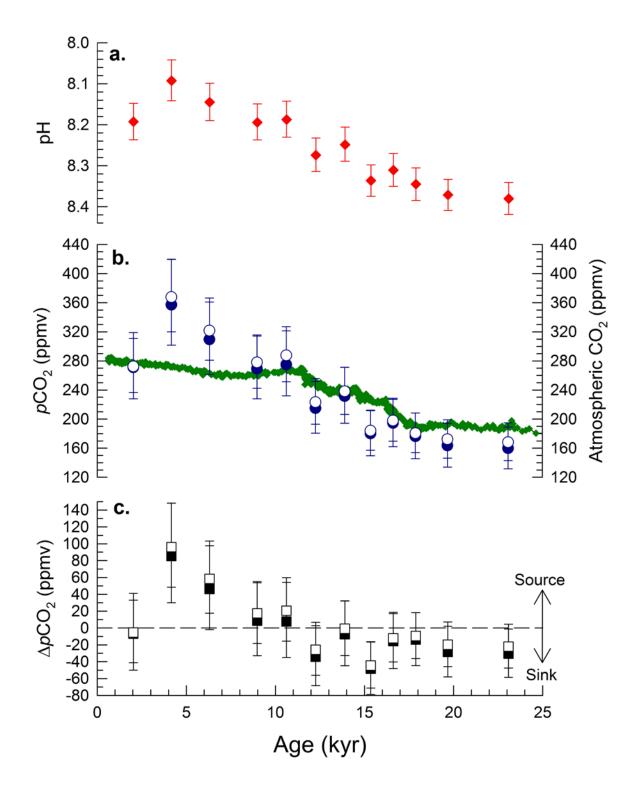


Figure 3. Reconstructed surface water pH and pCO_2 at the MD972106 site and atmospheric CO_2 over the past 25,000 years. a, Reconstructed surface water pH (red diamonds). b, Reconstructed pCO_2 calculated from pH, alkenone SST and alkalinity (Method 1; blue circles) and pCO_2 calculated from pH, alkenone SST and DIC (Method 2; blue open circles). Also shown is the atmospheric CO_2 record (Antarctic ice core CO_2 composite³³; green

diamonds). \mathbf{c} , $\Delta p \mathrm{CO}_2$ is the difference between reconstructed surface water $p \mathrm{CO}_2$ (Method 1 (black squares); and Method 2 (black open squares)) and atmospheric CO_2 (Antarctic ice core CO_2 composite³³). Error bars in \mathbf{a} , \mathbf{b} and \mathbf{c} represent 95% confidence intervals around the Monte Carlo mean, where 2σ uncertainties on the individual input variables and other carbonate system parameters are propagated via the Monte Carlo simulation (n = 10,000) in the statistical analysis program R (Methods).

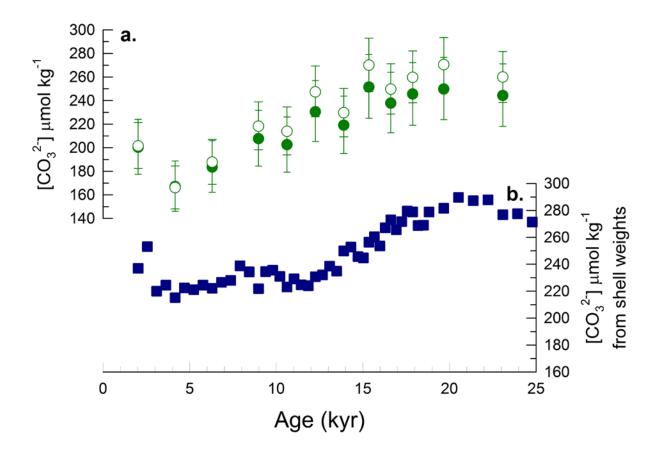


Figure 4. Reconstructed surface water carbonate ion concentration. a, [CO₃²⁻] from δ¹¹B-pH estimates, alkenone SST and alkalinity (Method 1; green circles), and from δ¹¹B-pH estimates, alkenone SST and DIC (Method 2; green open circles). **b**, [CO₃²⁻] from shell weight data (blue squares) using the shell weight-[CO₃²⁻] relationship²⁷. Error bars in **a** represent 95% confidence intervals around the Monte Carlo mean, where 2σ uncertainties on the individual input variables and other carbonate system parameters are propagated via the Monte Carlo simulation (n = 10,000) in the statistical analysis program R (Methods).

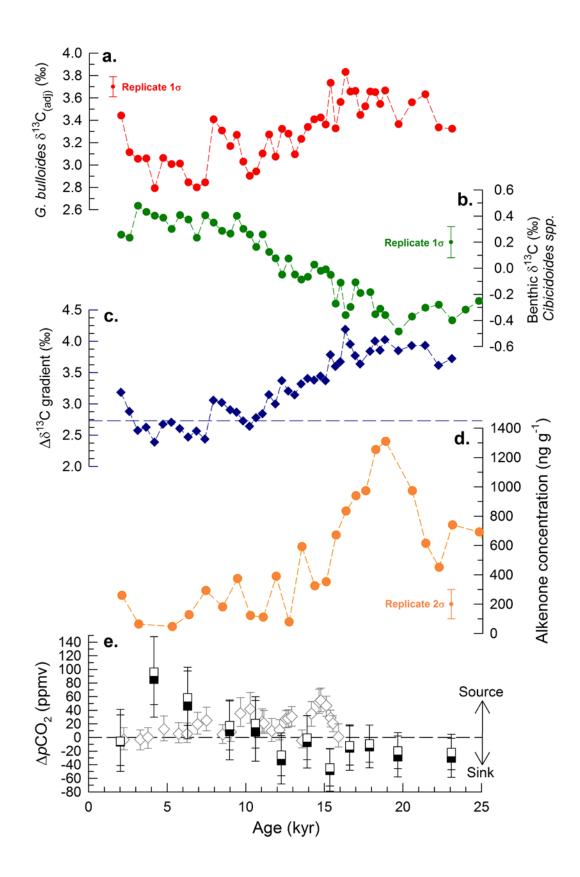


Figure 5. Planktonic and benthic foraminiferal δ^{13} C, $\Delta\delta^{13}$ C gradient, alkenone concentrations and Δp CO₂ over the past 25,000 years. a, G. bulloides δ^{13} C(adj) (red circles), b, benthic δ^{13} C for Cibicidoides spp. (green circles), c, $\Delta\delta^{13}$ C gradient ($\Delta\delta^{13}$ C

planktonic_(adj) – benthic foraminifera); blue diamonds and blue dashed line shows the Holocene $\Delta\delta^{13}$ C gradient average, **d**, alkenone concentrations (orange circles), **e**, Δp CO₂ differences between surface water pCO₂ and atmospheric CO₂ (from Antarctic ice core CO₂ composite³³) as in Figure **3c**. Also shown in **e** is the calculated Δp CO₂ for the Atlantic sector of SAZ¹⁵ (sediment core PS2498-1 at 44.15°, 14.23°, 3,783m water depth; grey open diamonds and error bars are the 95% uncertainty bounds). In **a**, *G. bulloides* δ^{13} C are temperature and [CO₃²⁻] adjusted (Methods). Error bars in **a** and **b** represent 1 σ uncertainties based on replicate measurements; error bars in **d** represent 2 σ for replicate extractions and measurements on a homogenous laboratory (sediment) standard, and error bars in **e** represent 95% confidence intervals around the Monte Carlo mean, based on 2 σ uncertainties on the individual input variables and other carbonate system parameters are propagated via the Monte Carlo simulation (n = 10,000) in the statistical analysis program R (Methods).

Methods

Materials and methods. Sediment core MD972106 (45°09'S, 146°17'E, water depth 3310 m) is a 32 m long piston core collected using the CALYPSO Kullenberg corer on board the RV *Marion Dufresne* in 1997 (in this paper, we present data from the top 2.5m of the core) and was recovered from the northern flank of the South Tasman Rise. The MD972106 site is located at the southern edge of the Subtropical front (STF) (Fig. 1) and modern bottom water covering the South Tasman Rise is Circumpolar Deep Water (CDW), a mixture of high δ^{13} C (low nutrient) NADW (or northern source deep-water) and re-circulated low δ^{13} C (high nutrient) deep-water from the Indian and Pacific Oceans (Supplementary Fig. 1).

MD972106 was sampled every 5 cm on board RV *Marion Dufresne*. Raw sediment samples were oven dried at 60°C, and then disaggregated in distilled water at the ACE CRC, University of Tasmania. Water saturated sediment samples were wet sieved through a >150 μ m mesh. The dried >150 μ m fraction was sieved to isolate the 300-355 μ m fraction from which \geq 50 planktonic foraminifera *Globigerina bulloides* were picked. We selected *G. bulloides* as this species shows a seasonal range in δ^{18} O values in this region of the Southern Ocean that is consistent with calcification in near-surface waters⁵⁶, and comparisons between flux-weighted sediment trap and surface sediment δ^{18} O values indicate the sedimentary record retains the seasonal isotopic imprint for *G. bulloides*⁵⁶.

Each batch of foraminifera was ultrasonically cleaned in methanol and oven dried at 60° C. Any broken tests were discarded. The cleaned samples were used to determine the average shell weights and isotopic compositions (δ^{18} O, δ^{13} C and δ^{11} B), with the shells undergoing further cleaning prior to determination of their δ^{11} B values. Average shell weight, δ^{18} O and δ^{13} C measurements were conducted at 5 cm intervals. Determination of δ^{11} B were made at 20 cm resolution.

Age model. Age control for MD972106 is based on accelerator mass spectrometry (AMS) radiocarbon (14 C) dating of the planktonic foraminifer, *Globorotalia inflata*. Five published AMS 14 C dates for MD972106 (ref. 57) were recalibrated to calendar ages using the Calib 7.1 program⁵⁸ with the Marine13 data set⁵⁹. Ages were corrected for the local 14 C reservoir correction (Δ R) for surface waters in this region of the Southern Ocean using a constant Δ R regional mean value (Δ R = 107±13 yr) calculated for the Chatham Islands from the marine reservoir correction database⁶⁰. The Δ R (107±13 yr) incorporates the measured 14 C age of 560±40 yrBP obtained from gastropods⁶¹. Using a different and older Δ R for the 14 C dates at 16.1 ka and 23.5 ka would slightly change the chronology, but this would not affect the conclusions of our study for the LGM and early deglacial periods. For example, using a Δ R = 900 yr for the 14 C dates at 16.1 ka and 23.5 ka would change the chronology by ~0.8 - 0.9 kyr, and using a Δ R = 900 yr for the 14 C date at 16.1 ka and a Δ R = 1500 yr for 14 C date at 23.5 ka, would change the chronology by ~0.8 to 1.4 ka. The average sedimentation rate at MD972106 was ~9.3 cm kyr 1 0 over the past 25 kyr.

Stable isotope analyses. Oxygen (δ^{18} O) and carbon (δ^{13} C) isotope ratios were determined for the planktonic foraminifera, *G. bulloides* (300–355 μm), and the benthic foraminiferal taxa *Cibicidoides spp.* (from the >150 μm size fraction). Measurements of δ^{18} O and δ^{13} C were conducted on an automated individual–carbonate reaction Kiel Device coupled to a Finnigan MAT-251 isotope ratio mass spectrometer at the Research School of Earth Sciences, ANU. Calcite samples weighing ~200 μg were reacted with 103% phosphoric acid at 90°C to liberate sufficient CO₂ for isotopic analysis. The δ^{18} O and δ^{13} C values are reported as per mil (‰) deviations relative to the Vienna Peedee Belemnite (VPDB) standard. The results have been normalised on the VPDB scale such that the NBS-19 calcite standard yields δ^{18} O VPDB (-2.20 ‰) and δ^{13} C VPDB (+1.95 ‰) and NBS-18 yields δ^{18} O VPDB (-23.0 ‰) and δ^{13} C VPDB

(-5.0 %). The 2σ uncertainties for replicate in-run measurements of NBS-19 (n=197) were ± 0.05 % and ± 0.02 % for $\delta^{18}O$ and $\delta^{13}C$, respectively. The average standard deviation for repeat measurement of two planktonic samples was ± 0.17 % for $\delta^{18}O$ and ± 0.09 % for $\delta^{13}C$, and for benthic samples was ± 0.05 % for $\delta^{18}O$ and ± 0.12 % for $\delta^{13}C$.

Boron isotope (δ^{11} B) values were determined for the planktonic foraminifera, G. bulloides (300-355 µm) at Southampton University using previously described methods⁶²⁻⁶⁴. Oxidative cleaning procedures followed those in Barker *et al.*, (ref. 65). We analysed ten G. bulloides (150–250 µg of calcite) for each in-run measurement and this provided ~2–3 ng of boron. The average 2σ in-run measurement precision was <0.2 ‰. The average agreement of repeat measurements on separate sample aliquots was ± 0.4 ‰ (n=7) and the average external 2σ precision is, at most, ± 0.4 ‰. Foraminiferal cleaning procedures⁶⁵ were verified for all samples by Al/Ca ratios <100 µmol mol⁻¹ and this provides a sufficient screen against clay contamination⁶⁶.

Planktonic foraminifera shell weights. *G. bulloides* shell weights were determined by picking 50 or more whole individual shells from the 300-355 μ m size fraction. Each batch of whole foraminiferal shells were ultrasonically cleaned in methanol and oven dried at 60°C. Any shells broken during the cleaning process were discarded and the remaining shells were counted. The shells were weighed on a microbalance (precision = 0.1 μ g) and average shell weights were calculated by dividing the measured weight by the total number of whole foraminifera. Replicate shell-weight measurements were determined by selecting a separate aliquot of 50 or more shells from the same sample. The mean difference of replicate shell weights is $\pm 1.2 \mu$ g (1 σ ; n = 55).

Palaeo sea surface temperature (SST) and alkenone concentrations. Alkenones were analyzed at Geoscience Australia following the methods of Calvo *et al.*, (ref. 67) and, with

them, SST estimates were obtained from the alkenone unsaturation ratio ($U_{37}^{K\prime}$ index). Alkenone concentrations were quantified by using n-hexatriacontane as an internal standard. The standard deviation for alkenone concentrations is 50 ng g⁻¹ (2 σ ; replicate extractions and measurements of a homogenous laboratory sediment standard). Alkenone-derived SSTs were reconstructed using the $U_{37}^{K\prime}$ -SST relationship of Müller *et al.*, (ref. 68) where $U_{37}^{K\prime}=0.033$ x SST + 0.044 for annually averaged SSTs. $U_{37}^{K\prime}$ measurements were completed at 10 cm sample resolution. The standard deviation for replicate measurements of a homogenous laboratory (sediment) standard is, at the most ± 0.6 °C (2 σ). The standard error of SST estimates from the $U_{37}^{K\prime}$ -SST calibration⁶⁸ is ± 1.5 °C. Long-term climate data from the World Ocean Atlas 2013 (WOA13; 0.25°)⁶⁹ gives a modern annual SST of 12.3 °C at the MD972106 locality, which is close to the most recent core-top sediment $U_{37}^{K\prime}$ derived SST of 13.0 °C.

Reconstruction of surface water pCO_2 . Seawater pCO_2 can be calculated from temperature, salinity and pressure, and knowledge of two out of the six main carbonate system parameters (free aqueous carbon dioxide $(CO_{2(aq)})$, bicarbonate ion (HCO_3^-) , carbonate ion (CO_3^{2-}) , hydrogen ion (H^+) , dissolved inorganic carbon (DIC), and total alkalinity (TA)). Accordingly, past changes in seawater pCO_2 can be calculated using estimates of seawater pH and either DIC or TA. Here, we use an alkenone-based proxy for seawater temperature, planktonic foraminiferal $\delta^{11}B$ to estimate pH, planktonic foraminiferal $\delta^{13}C$ to estimate DIC, and salinity derived seawater alkalinity to constrain major parameters of the seawater carbonate system. Past variations in sea surface water pH at the site of MD972106 are calculated from a recently published $\delta^{11}B$ -pH relationship for G. bulloides⁷⁰ and the alkenone SST record. The $\delta^{11}B$ -pH relationship for the same species⁷¹. The $\delta^{11}B$ -pH relationship⁷⁰ is similar to and within error of a previously published $\delta^{11}B$ -pH relationship⁷¹. Even though the calibration uncertainty for Martínez-Botí

et al., (ref. 71) is considerably lower than Raitzsch et al., (ref. 70), we used the extended δ^{11} B-pH relationship data set for *G. bulloides*⁷⁰.

Variations in surface water pCO_2 (Fig. 3b) over the past 25 kyr are estimated using the SST alkenone record, the $\delta^{11}B$ surface water pH record and estimates of either alkalinity or DIC. One set of estimates is derived from δ^{11} B-based pH and seawater alkalinity, assuming that past variations in alkalinity were proportional to salinity, which can be estimated from past sea level 62,72 (Method 1) (Fig. 3a). An alternative set of estimates is derived from δ^{11} B-based pH, and foraminiferal δ^{13} C-based DIC (Method 2) (Fig. 3a). Both approaches use a Monte Carlo method to calculate pCO₂ from pH, DIC, salinity and temperature. This method generates quasi-normally distributed errors around central values with 10,000 repetitions, and sea water carbonate system calculations using the R-based⁷³ seacarb package⁷⁴. uncertainties on the individual input variables and other carbonate system parameters are included to the end; δ^{11} B (± 0.4 %), $U_{37}^{K'}$ -derived SST (± 1.0 °C), salinity (\pm 3psu), ALK (± 200 μ mol kg⁻¹), and DIC (\pm 76 μ mol kg⁻¹). A disequilibrium uncertainty \pm 20 ppm (2σ) is also included. Although we note the importance of propagating all the uncertainties, we also note that pH is by far the dominant variable on pCO₂, and that pH is well constrained by $\delta^{11}B$ alone⁷⁵. Thus we are very confident that the change in pCO_2 forcing over this time is much more robust than the pCO_2 change in terms of absolute uncertainty⁷⁵. The average pCO_2 difference between calculations using Method (1) and (2) is ~7 ppm.

Surface water pCO_2 and hence ocean circulation and productivity exerts an important control on atmospheric CO_2 levels. However, the difference between the two (ΔpCO_2) is also affected by terrestrial components of the carbon cycle such as (re-) growth of biomass or the dynamics of (coastal) permafrost. Hence, more important than the absolute magnitude of ΔpCO_2 is the change in this value, because this provides direct information as to whether a

particular area of the ocean was more or less important in driving the direction of atmospheric change. Thus, while there are intervals in our record where the analytical and/or age uncertainties preclude us from being certain whether our study area was acting as source or sink of CO_2 to the atmosphere, we can be confident that the change in surface water ΔpCO_2 levels in this area of the Southern Ocean made an important contribution to the last deglacial atmospheric CO_2 rise.

We used the total pH scale, equilibrium constants for the dissociation of carbonic acid from Dickson (ref. 76) and the boron isotope equilibrium constant from Klochko (ref. 77). Modern silicate and phosphate values were applied to the calculations but they had very little effect on the final results⁷⁸. The proxies recorded in *G. bulloides* and in coccolithophorid alkenones predominantly reflect austral spring conditions in the region, because this is the seasonal maximum of production for both $\tan^{56,79}$. Thus, the estimates of pCO_2 likely reflect austral spring conditions.

The calculated pCO_2 from boron isotope-based pH estimates and DIC (Method 2) utilizes an estimate of surface water DIC over time and requires a DIC estimate for the most recent sediment sample. The WOCE SR3 section line between 44° S and 46° S (ref. 80-82) indicates a modern surface water DIC of ~2061 μ eq kg⁻¹ at this site, where *G. bulloides* are known to calcify⁵⁶. Estimates of DIC are then varied over time as a function of *G. bulloides* δ^{13} C, with the difference between modern surface- and deep-water DIC derived via the difference between planktonic and benthic foraminiferal δ^{13} C (i.e. Δ DICsw-Dw is proportional to $\Delta\delta^{13}$ C planktonic-benthic). The biological drivers of surface-water δ^{13} C are mainly modulated by organic carbon production; carbonate precipitation does not significantly fractionate surfacewater DIC. Therefore, any changes in the CaCO₃:Corg rain ratio would have a minor effect on DIC. Studies of the mechanisms of modulation of DIC in surface waters in this region (e.g.

ref. 83) and of the Southern Ocean more broadly (e.g. ref. 84), suggest the CaCO₃:Corg rain ratio is very low, with only a minor effect on surface DIC and surface δ^{13} CDIC. Any changes to the rain ratio would also affect surface-water pH, which we estimate through boron isotopes. Modern surface and deep water DIC are from WOCE line SR3 section between 44° S and 46° S (ref. 80-82), δ^{13} Csw from King and Howard (ref. 85) and δ^{13} CDW are from Moy *et al.*, (ref. 57).

G. bulloides temperature- and [CO₃²-]-adjusted δ^{13} C (δ^{13} C_{adi}). Temperature and carbonate ion concentration ([CO₃²-]) have an additional, often neglected, effect on planktonic for aminiferal δ^{13} C (refs 85,86, 87). For Method 2, temperature and [CO₃²-] adjustments on G. bulloides δ^{13} C used to estimate DIC do not significantly change the pCO₂ estimates. The average difference in pCO₂ estimates calculated from (a) boron isotope-based pH estimates and DIC (where G. bulloides δ^{13} C is not temperature and [CO₃²⁻]-adjusted) and (b) boron isotope-based pH estimates and DIC (where G. bulloides δ^{13} C is temperature- and [CO₃²⁻]adjusted) is ~2 ppm (Supplementary Fig. 4). G. bulloides δ^{13} C values are adjusted for temperature using the δ^{13} C disequilibrium- temperature relation 0.11% °C⁻¹ (ref. 87). The δ^{13} C disequilibrium-temperature⁸⁷ relation we apply here is similar to the relationship estimated from a Southern Ocean sediment trap field study⁸⁵. Temperature adjustments are made using the alkenone-based SST estimates in MD972106. G. bulloides δ^{13} C values are also adjusted for the so called carbonate ion ([CO₃²⁻]) effect using the relationship between δ^{13} C disequilibrium and $[CO_3^{2-}]$ (-0.014% $[CO_3^{2-}]^{-1}$) (ref. 86). Estimates of $[CO_3^{2-}]$ are from surface water pCO₂ calculations that use boron isotope-based pH estimates and estimates of seawater alkalinity (Method 1). The 'close agreement' between the [CO₃²⁻] outputs from Method 1 and 2 (Fig. 4a) would yield a similar adjustment to G. bulloides δ^{13} C in terms of [CO₃²-] adjustments.

Data availability

All data are archived at the Australian Antarctic Data Centre (AADC) and are publicly accessible at AADC reference link.

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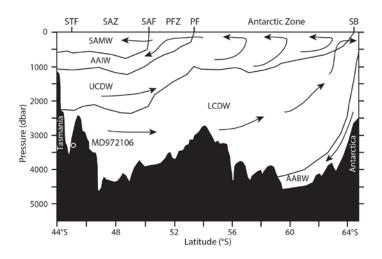
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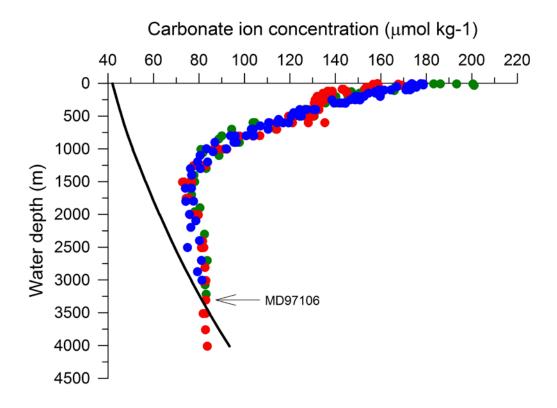
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Supplementary Information: Heterogenous contribution of the Southern Ocean to deglacial atmospheric CO2 rise



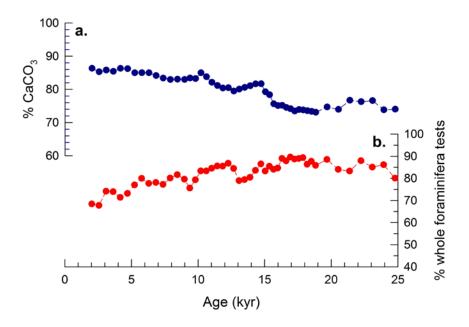
Supplementary Figure 1. Circulation and deep-water masses along WOCE SR3 line transect. MD972106 (45°09'S, 146°17'E, water depth 3310 m) is located at the southern edge of the Subtropical front and bathed in Lower Circumpolar Deep Water. STF = Subtropical Front, SAZ = Subantarctic Zone, SAF = Subantarctic Front, PFZ = Polar Frontal Zone, PF = Polar Front, SB = southern boundary of the ACC, SAMW = Subantarctic Mode Water, AAIW = Antarctic Intermediate Water, UCDW = Upper Circumpolar Deep Water, LCDW = Lower Circumpolar Deep Water and AABW = Antarctic Bottom Water. Southern Ocean frontal locations and water masses in the South Tasman Rise region are well defined through the WOCE SR3 repeat hydrographic section between Tasmania and Antarctica¹⁻⁴. Climatological data from the World Ocean Atlas 2013 (WOA13; 0.25°)⁵, shows modern summer and winter sea surface temperatures (SST) of 14.0°C and 11.0°C at MD972106, respectively, indicating that the core site is located at the southern edge of the STF. Southern Ocean frontal locations and water mass profiles compiled from Rintoul and Bullister (ref. 1) and Sokolov and Rintoul (ref. 4).



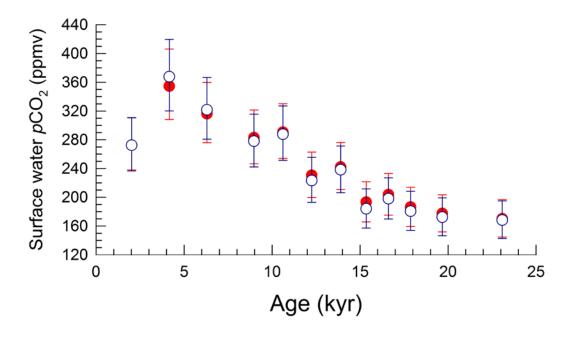
Supplementary Figure 2. WOCE line SR3 vertical profiles of carbonate ion concentration ([CO₃²⁻]) between 44°S and 46°S, and calcite solubility (solid line).

The seawater calcite saturation horizon occurs at ~3,500 m water depth. MD972106 (45°09'S, 146°17'E, water depth 3310 m) is located above the modern calcite saturation horizon and above the foraminiferal lysocline that occurs at ~3,600 m water depth⁶.

[CO₃²⁻] are from WOCE line SR3 section from voyage AA9404 (green circles), AA0301 (blue circles) and AA9706 (red circles). [CO₃²⁻] was calculated from TA, DIC, temperature, pressure, salinity, [PO₄³⁻] and [SiO₄⁴⁻] from the WOCE line SR3 section between 44° S and 46° S (refs 7-9) using the CO2Sys Excel macro (version 2.1) (ref. 10) and constants from Mehrbach *et al.* (ref. 11) refitted by Dickson and Millero (ref. 12). Calcite solubility was calculated from the equation of Mucci (ref. 13) that includes pressure adjustments in K_{SP} following Ingle (ref. 14).



Supplementary Figure 3. Indicators of post-depositional dissolution. Carbonate Preservation Indices (%CaCO₃ and %whole foraminiferal tests) for MD972106, a, %CaCO₃ for MD972106 (blue circles) and **b**, %whole foraminiferal (%WF) (red circles) as a function of age (ka). The timescale for MD972106 is based on radiocarbon dates (see Methods). %CaCO₃ is highest during the Holocene (~85%) and decreases to ~75% during the Last Glacial Maximum (LGM). Similarly, %WF is lowest during Holocene (~76%) and increases during to ~86 % during the LGM. At MD972106 (and other South Tasman Rise sites¹⁵), the pattern of decreasing %CaCO₃ during the LGM likely results in part from the dilution of carbonate by terrigenous sediment¹⁵. MD972106 is located above the modern calcite saturation horizon (at ~3,500m water depth; Supplementary Fig. 2). Holocene core-top G. bulloides (300-355 µm) shell weights indicate the foraminiferal lysocline at the South Tasman Rise occurs at ~3,600m water depth⁶. These indicators of post-depositional dissolution suggest that dissolution did not play an important role over the past 25 kyr at MD972106. %CaCO₃ are from Moy et al., (ref. 16) and %WF were determined using similar methods to those described by Howard and Prell (ref. 17).



Supplementary Figure 4. Reconstructed surface water pCO_2 from pH, alkenone SST and DIC (Method 2) at the MD972106 site over the past 25,000 years. For Method 2, temperature and $[CO_3^{2-}]$ adjustments on G. bulloides $\delta^{13}C$ used to estimate DIC do not significantly change the pCO_2 estimates for this method. The average difference in pCO_2 estimates calculated from (a) boron isotope-based pH estimates and DIC (where G. bulloides $\delta^{13}C$ is not temperature and $[CO_3^{2-}]$ -adjusted; blue open circles) and (b) boron isotope-based pH estimates and DIC (where G. bulloides $\delta^{13}C$ is temperature- and $[CO_3^{2-}]$ -adjusted; red circles) is ~2 ppm. The small average pCO_2 difference of ~2 ppm indicates that the temperature and $[CO_3^{2-}]$ adjustments on G. bulloides $\delta^{13}C$ used to estimate DIC do not play a significant role in pCO_2 estimates for Method 2. G. bulloides $\delta^{13}C$ values were adjusted for temperature and for $[CO_3^{2-}]$ (see Methods). Error bars represent 95% confidence intervals around the Monte Carlo mean, where 2σ uncertainties on the individual input variables and other carbonate system parameters are propagated via the Monte Carlo simulation (n = 10,000) in the statistical analysis program R (Methods).

Supplementary Note 1 Anthropogenic-influenced surface water pH, pCO_2 and $[CO_3^{2-}]$ at the MD972106 site calculated using WOCE line SR3 data from voyage AA0301 (ref. 8; Fig. 1 in main text) and the CO2Sys Excel macro (version 2.1) (ref. 10) and constants from Mehrbach *et al.* (ref. 11) refitted by Dickson and Millero (ref. 12).

Supplementary Note 2 Anthropogenic CO₂ value for surface waters from the WOCE line SR3 close to MD972106 site are ~45 μmol kg⁻¹ (GLODAP v1.1 database¹⁸), which corresponds to a [CO₃²⁻] decrease of ~29 μmol kg⁻¹ with respect to estimated pre-industrial values. Present-day surface [CO₃²⁻] is ~183 μmol kg⁻¹; thus pre-industrial [CO₃²⁻] is estimated to be ~212 μmol kg⁻¹. Calculated [CO₃²⁻] for surface waters are from WOCE line SR3 TA, DIC, temperature, pressure, salinity, [PO₄³⁻] and [SiO₄⁴⁻] from the GLODAP v1.1 database¹⁸ using the CO2Sys Excel macro (version 2.1) (ref. 10) and constants from Mehrbach *et al.* (ref. 11) refitted by Dickson and Millero (ref. 12).

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