

## **Supporting information**

### **Electrodeposition of MoS<sub>2</sub> from Dichloromethane**

Shibin Thomas<sup>1</sup>, Danielle E. Smith<sup>1</sup>, Victoria K. Greenacre<sup>1</sup>, Yasir J. Noori<sup>2</sup>, Andrew L. Hector<sup>1</sup>, C. H. (Kees) de Groot<sup>2</sup>, Gillian Reid<sup>1</sup> and Philip N. Bartlett<sup>1,z</sup>

<sup>1</sup>School of Chemistry, University of Southampton, Southampton, SO17 1BJ, UK

<sup>2</sup>School of Electronics and Computer Science, University of Southampton, Southampton, SO17 1BJ, UK

<sup>z</sup>E-mail: [P.N.Bartlett@soton.ac.uk](mailto:P.N.Bartlett@soton.ac.uk)

#### **Precursor synthesis**

Tetrabutylammonium chloride (Sigma-Aldrich) and ammonium tetrathiomolybdate (Acros Organics) were used as received. Following reported literature,<sup>(1)</sup> ammonium tetrathiomolybdate (367 mg, 1.41 mmol) was dissolved in deionised water (10 mL) to form a red solution. A solution of tetrabutylammonium chloride (0.784 mg, 2.82 mmol) was slowly added dropwise; a red solid immediately precipitated. The red solid was isolated and washed with water (1 x 2 mL) and isopropyl alcohol (1 x 2 mL) before being dried under vacuum and stored under N<sub>2</sub>. Yield: 728 mg, 90%. Required for C<sub>32</sub>H<sub>72</sub>MoN<sub>2</sub>S<sub>4</sub>: C: 54.20, H: 10.23, N: 3.95%. Found: C: 53.92, H: 10.14, N: 4.05%. IR spectrum (Nujol/ cm<sup>-1</sup>): 468sh (Mo=S). <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 0.99 (t, [3H], J<sub>HH</sub> = 7.3 Hz), 1.46 (m, [2H], J<sub>HH</sub> = 1.0 Hz), 1.6 – 1.7 (m, [2H]), 3.34 (m, [2H], J<sub>HH</sub> = 16.9 Hz). <sup>95</sup>Mo NMR (CD<sub>2</sub>Cl<sub>2</sub>): δ = 2213 (s).

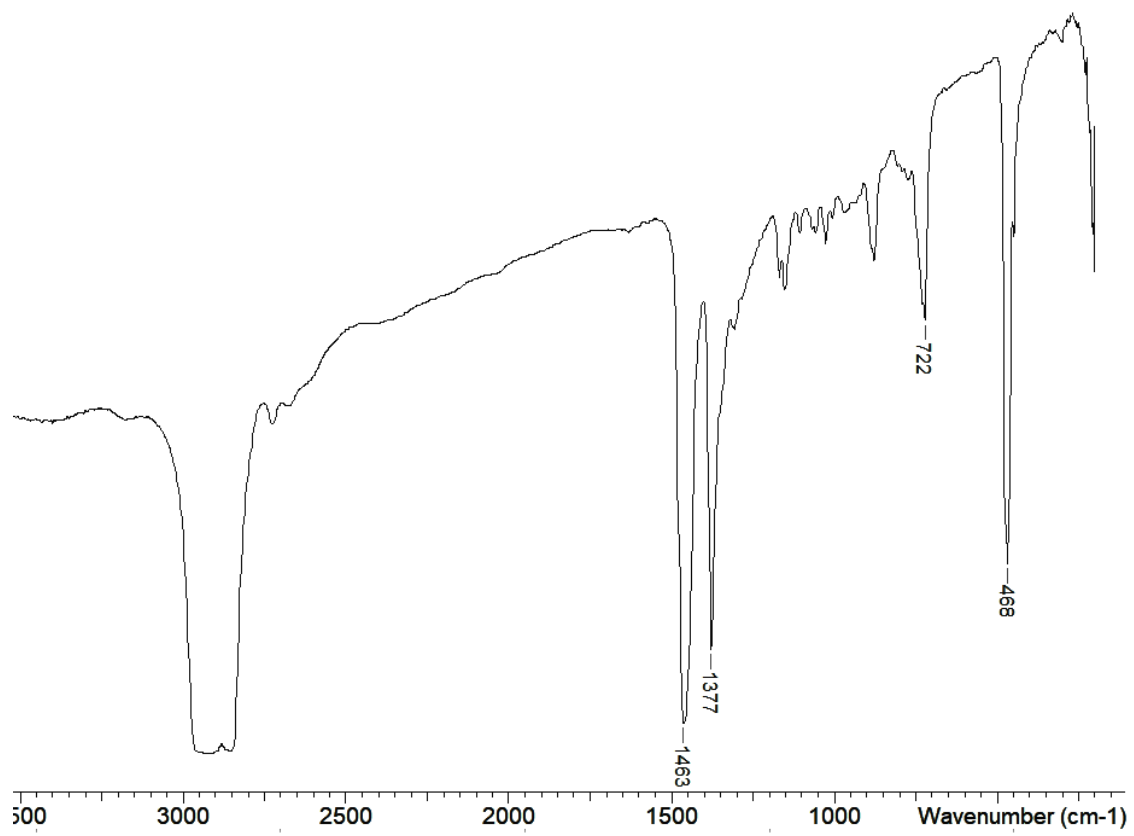


Figure S1: IR spectrum of [N<sup>n</sup>Bu<sub>4</sub>]<sub>2</sub>[MoS<sub>4</sub>].

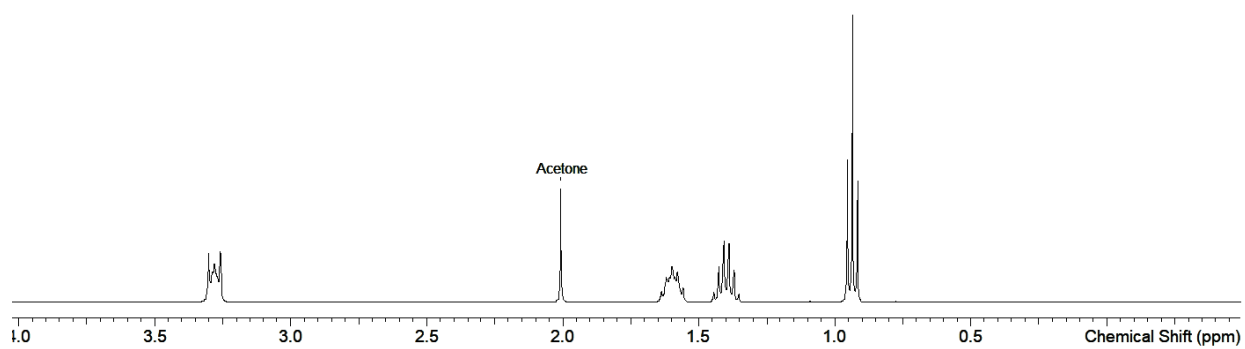


Figure S2: <sup>1</sup>H NMR spectrum of [N<sup>n</sup>Bu<sub>4</sub>]<sub>2</sub>[MoS<sub>4</sub>] in CD<sub>2</sub>Cl<sub>2</sub>.

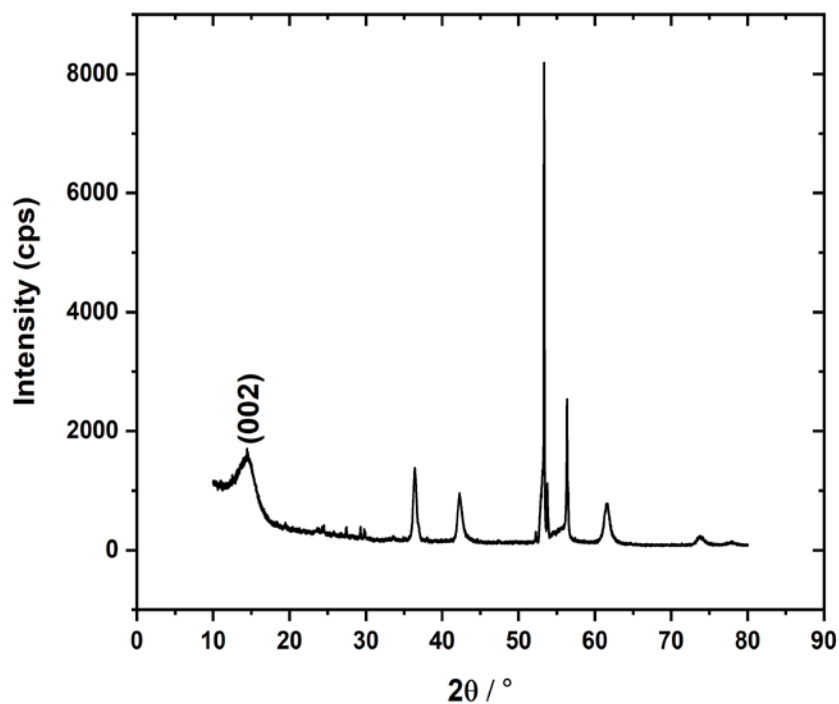


Figure S3: XRD pattern recorded from annealed MoS<sub>2</sub> thin film obtained by applying -0.8 V vs. Ag/AgCl for 1 hour.

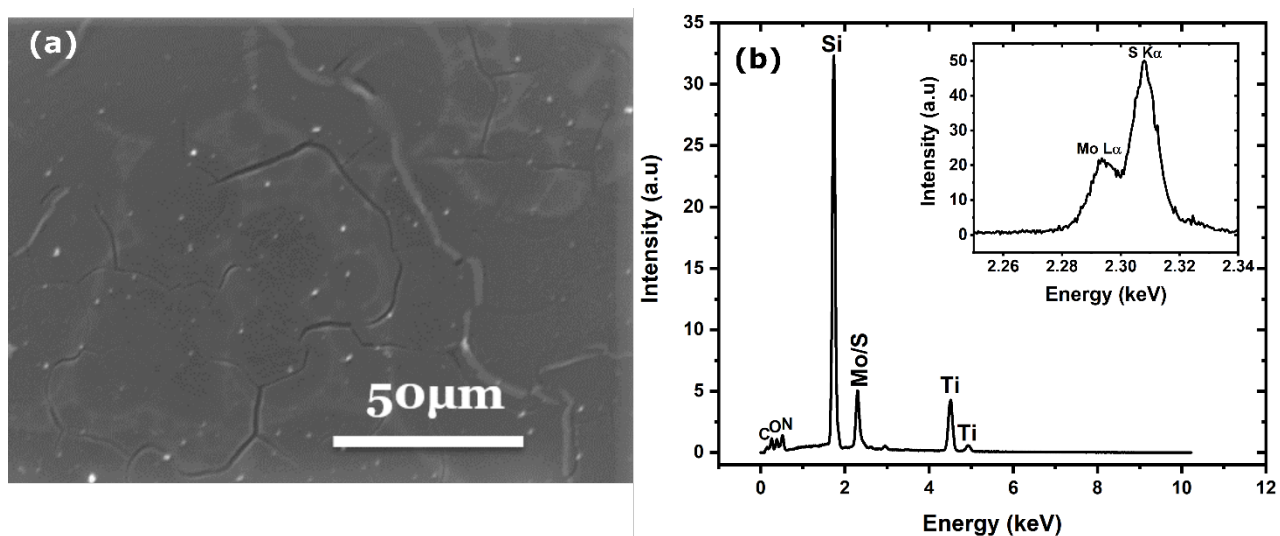


Figure S4: (a) Top-view SEM image of the as-deposited MoS<sub>2</sub> thin film obtained by applying -0.8 V vs. Ag/AgCl for 1 hour. (b) EDX profile of the as-deposited film. The inset shows the WDX spectrum for Mo-L<sub>α</sub> and S-K<sub>α</sub>.

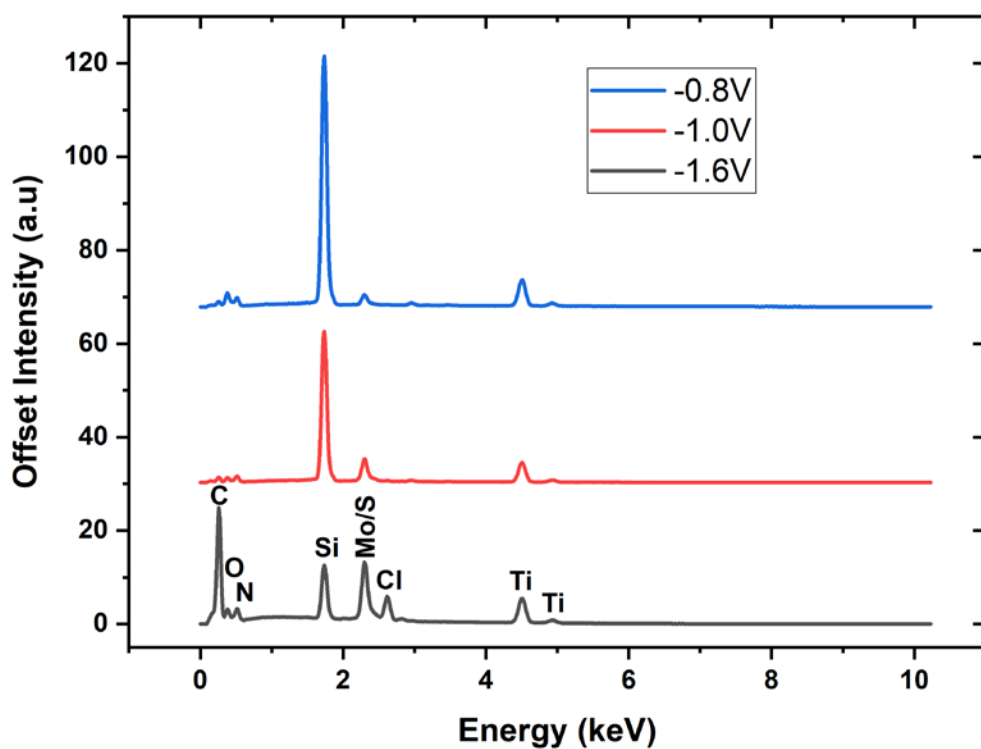


Figure S5: EDX profile of the as-deposited MoS<sub>2</sub> thin films obtained at different applied potentials vs. Ag/AgCl.

#### References

1. G. Alonso, G. Aguirre, I. A. Rivero, and S. Fuentes, *Inorg. Chim. Acta*, **274** (1), 108-110 (1998).