Sonogashira Cross-Coupling Reaction of Bromocyanofluoro Pyridine Nuclei: Access to 5- and 6-Alkynylfluoropyridinamidoximes Scaffolds

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Abstract: We disclose a general two-step procedure to access hitherto unknown and under explored 5- and 6-alkynyl-3-fluoro-2-pyridinamidoximes from 5- and 6-bromo-3-fluoro-2-cyanopyridines and a wide range of easily available and bench-stable terminal alkynes, using Sonogashira cross-coupling, as the first step. The generation of the polar amidoxime group is realized at a late-stage upon treatment of the alkynylfluorocyanopyridine by hydroxylamine. This mild and operationally simple two-step room temperature process is compatible with enantiopure chiral substrates and various functionality including free alcohols, unprotected and CBz-protected amines, acetonides, benzyl ethers, amide, imide, di-subsituted alkynes and strained saturated heterocycles.

Introduction

COMMUNICATION

The amidoxime functional group is recognized as a priviledged fragment since it represents a unique binucleophilic group that possesses both the amino and the hydroxyimino functions at the same carbon.¹ These particular features have been advantageously exploited for the synthesis of relevant heterocycles of medicinal interest, such as imidazoles,² isoxazoles,³ 1,2,4-oxadiazoles,³ 1,2,4-oxadiazol-5-ones,⁴ 1,2,4thiadiazoles,⁵ triazole,⁶ and pyrimidines.⁷ Amidoximes have also been reported to be powerful pharmacophores in the identification of lead compounds towards the discovery of new synthetic drugs and prodrugs, with significant biological activity.8 In particular, the use of amidoximes as antioxidant,⁹ antimycotic,¹⁰ antibacterial,¹¹ antimicrobial,12 antileishmanial,13 antihypertensive,14 and antithrombotic¹⁵ agents has become widespread. On the other hand, amidoxime derivatives also exhibit promising histone deacetylase (HDACs) inhibitory activity for cancer therapy,¹⁶ as well as anesthetic properties.¹⁷ Besides this impressive spectrum of biological activities, amidoximes are of interest in materials and polymer chemistry for the preparation of adsorbents, such as fibers and resins for chelation and removal of heavy metal ions.¹⁸ In the course of our research towards original reactivators of organophosphorous (OP) inhibited human acetylcholinesterase (hAChE), we recently found that 3-hydroxy-2-pyridinamidoxime 1 as well as 3-hydroxy-2-pyridinaldoxime 2 (Figure 1) exhibited unprecedented efficacy for phenylphosphonothioate (PhX) hydrolysis, a less toxic analogue of the nerve agent VX. $^{19,20}\,$



Figure 1. : Chemical structures of 3-hydroxy- and 3-fluoropyridin- amidoximes and -aldoximes 1, 2, 5 and 6, and 6-alkynyl-3-fluoro-2-pyridin-aldoximes 3 and amidoximes 4.

We also demonstrated that these oxime-containing scaffolds showed promise as reactivators of VX-inhibited *h*AChE *in vitro*.²⁰ More recently, further developments in that field led us, to discover and highlight the benefit of 6-alkynyl-3-fluoro-2-pyridinaldoximes **3** (Figure 1), as precursor of 6-alkanylfluoropyridinaldoximes, used for the efficient reactivation and resurrection of sarin-inhibited human acetylcholinesterase.²¹ We then hypothesized that unknown alkynylfluoropyridinamidoximes such as **4**



Scheme 1. Two-step pathways syntheses of 6-alkynyl-3-fluoro-2-pyridinamidoximes 4 from fluorocyanopyridine 8.

could find novel applications as precursors of innovative reactivators of OP-inhibited *h*AChE. While halogenated aromatic amidoximes have been reported for use in medicinal chemistry,⁴ it is noteworthy that molecules bearing fluoropyridinamidoxime scaffolds are rare and remain elusive intermediates in synthetic organic and in bioorganic chemistry. Such original structures are appealing for applications in medicinal chemistry projects.

Based on our recent reports on the late stage Sonogashira cross-coupling reactions of 6-bromo-3-fluoro-2-pyridinaldoxime **5** with various alkynes,²² we envisaged that the analogous reaction performed with the unknown 6-bromo-3-fluoro-2-pyridinamidoxime **6**, and suitable alkynes **7** (Scheme 1, path a), could enable the synthesis of previously unexplored 6-alkynyl-3-fluoro-2-pyridinamidoximes **4** as potential reactivators or precursors of reactivators of OP-inhibited *h*AChE. Successful implementation of this methodology will also enlarge the molecular diversity of fluoropyridine libraries, known to be pharmaceutically valuable scaffolds for drug discovery.²³

Herein, we disclose a simple two step protocol for the synthesis of new 6-alkynyl-3-fluoro-2-pyridinamidoximes **4**, using the Sonogashira cross-coupling of 6-bromo-3-fluoro-2-cyanopyridine **8** with alkyne **7**, followed by a late-stage selective amidoximation (Scheme 1, path b). This room temperature sequence is simple and provides underexplored highly versatile chemical scaffolds, with potential uses as reactivators of OP-inhibited *h*AChE.²⁴

Results and Discussion

direct Τo investigate the Sonogashira reaction of bromofluoropyridinamidoxime 6 with alkynes 7 (Scheme 1, path a), we first required access to suitable starting pyridinamidoxime derivatives. Practically, our identified prototypical compound was commercially available 6-bromo-3-fluoro-2-pyridinecarbonitrile 8. Initially, generation of the amidoxime 6 was achieved in high isolated yield (82%) upon treatment of the cyanopyridine 8 with hydroxylamine hydrochloride in EtOH at room temperature, in the presence of sodium carbonate.²⁵ Remarkably, Sonogashira cross-coupling reactions using either the starting cyanopyridine 8 (path b, Scheme 1) or the corresponding amidoxime 6 (path a, Figure 1) have not been reported previously, to the best of our knowledge. We therefore initiated our study by examining Sonogashira cross-coupling of 6 with 1-ethyl-4-ethynylbenzene 7a as our model alkyne, using reported conditions for the C-C cross-coupling of bromopyridine derivatives.²⁶ Surprisingly, the reaction did not perform well, and provided the expected product 4a in a low isolated yield of 25%, along with the alkyne homocoupling byproduct in about 10%. Unfortunately, these unsatisfactory results were invariably replicated during our attempts to optimize the process by screening different conditions, such as reaction temperature, solvent and catalyst loading (data not shown). A common trend was low conversion of the starting bromofluoroamidoxime 6, suggesting that the efficiency of the Pdcatalysed pathway was impacted due to the known propensity of amidoximes to complex and sequester metals.²⁷ In particular, amidoximes are reported to form stable complexes with Pd, and Cu ions.^{28,29} Subsequently, we hypotesized that a combination of electronic and steric factors may govern the overall properties of the palladium catalyst, and hence the overall chemical efficacy of the Sonogashira cross-coupling reaction.

Taking these considerations into account, we set out to overcome this obstacle and access our targeted alkynyl fluoroamidoximes 4 by reversal of the reaction steps as envisioned in path b (Scheme 1). To our delight the reaction of cyanopyridine 8 with 1-ethyl-4ethynylbenzene 7a performed well and vielded to the coupled product 9a in excellent isolated yield of 92% (Table 1). The 6alkynyl-3-fluoro-2-cyanopyridine 9a was next subjected to the optimised reaction conditions for amidoxime formation,²⁵ using hydroxylamine hydrochloride in ethanol and in the presence of sodium carbonate. Gratifyingly, the desired 6-alkynyl-3-fluoro-2pyridinamidoxime 4a was formed and isolated in 91% yield (Table 2). No evidence was found for side reactions such as nucleophilic aromatic substitution³⁰ of the 3-fluoro group by hydroxylamine, or formation of undesired amide byproduct.8 Having validated the synthetic efficiency of the two step procedure (path b, Scheme 1) to access of 6-alkynyl-3-fluoro-2-pyridinamidoxime 4a, we investigated the scope for production of diversely functionalized fluoropyridinamidoximes. We first prepared the corresponding 6alkynyl-3-fluoro-2-cyanopyridines 9b-9k by Sonogashira C-C cross-coupling of 8 and alkynes 7b-7k (Table 1). It is noteworthy that the reaction could be run with unfunctionalized alkynes such as 7b and 7c to afford the desired 6-alkynyl-3-fluoro-2cyanopyridines 9b and 9c in high yields, 93% and 85%, respectively. A range of functionalized alkynes 7d-7k also underwent coupling with 8, with high efficiency. The reaction tolerated free alcohols and amines, and alkynes (7d, 7e) bearing sterically demanding substitution, providing the respective products 9d and 9e in 90% yield each.



Table 1. Sonogashira cross-coupling of alkyne 7a-7j with fluorocyanopyridine 8

Pyridine and morpholine are common structural motifs found in numerous biologically active compounds. Under our conditions, alkynes 7f, 7g, and 7h bearing these heterocycles also coupled readily with the fluorocyanopyridine 8 and afforded synthetically useful to excellent yields of the corresponding products 9f (64%), 9g (43%) and 9h (97%). Functionalities such as O-benzyl protected alcohol 7i, phthalimide 7j, acetonide, amide and free aromatic amine of 7k were well tolerated, under the crosscoupling conditions, and gave the expected alkynyl products 9i (64%), 9j (84%) and 9k (94%) in high isolated yields. It is important to highlight that the chiral nucleoside alkyne 7k provided the stereochemically pure coupling product 9k without any sign of epimerization. Next we turned our attention to the conversion of alkynylfluorocyanopyridines 9b-9k to the corresponding alkynylfluoropyridinamidoximes, by treatment with hydroxylamine at room temperature (Table 2). Pleasingly, the tested nitriles 9a-9k smoothly provided the desired amidoximes 4a-4k at room temperature, in high to excellent isolated yield, after purification by column chromatography. The sole exception was substrate 9g that gave 4g in low yield. The obtained results demonstrate that a wide array of functional groups, including imide, alcohol, aliphatic and aromatic amines, acetonides, aromatic and saturated heterocycles, as well as strained heterocycles such as oxetane, are well tolerated and compatible with the mild conditions of amidoximation, as shown in Table 2.



 Table 2. Amidoximation of 6-alkynyl-3-fluoro-2-cyanopyridine 9a-9k to produce

 6-alkynyl-3-fluoro-2-pyridinamidoximes 4a-4k.

We further illustrated the synthetic potential of this new two-step process for preparation of fluoropyridinamidoxime scaffolds, by application to an isomeric cyanofluoropyridine substrate. Commercially available 5-bromo-3-fluoro-2cyanopyridine 11 was successfully converted in two-steps to the expected ¹⁵N labelled 6-alkynyl-3-fluropyridinamidoxime 13 in 60% yield (Scheme 2), after Sonogashira reaction with alkyne 10, and conversion of nitrile 12 upon treatment with ¹⁵NH₂OH in ethanol, at 80°C.^{22 15}N labelled fluoropyridinamidoximes may be useful for non-radioactive isotopic metabolic labelling, ¹⁵N-tracing and for quantification purposes.³¹ In addition, the value of alkynylfluoropyridinamidoximes as synthetic intermediates towards alkyl-linked hybrid reactivators was demonstrated by the one-pot selective hydrogenation/hydrogenolysis of the internal alkyne and CBz group present in 4j. Exposure of 4j to hydrogen (1 atm) over a Pd/C catalyst smoothly delivered the alkanylfluoropyridinamidoxime 14 in excellent 92% yield, without disturbing the potentially reducible amidoxime functionality, or the C-F bond.21



Scheme 2. Synthesis of 5-alkynyl-3-fluoro-2-pyridinamidoximes 13, and selective hydrogenation of 4j to produce 6-alkanyl-3-fluoro-2-pyridinamidoxime

In summary, an operationally simple two-step synthesis of alkynylfluoropyridinamidoximes is reported. The reaction conditions are compatible with a range of functional groups, giving efficient access to diverse amidoximes. Hydrogenation of the alkynyl linkage was effected in the presence of the amidoxime functionality, giving the alkyl-linked analogue **14**.

Procedure for the Sonogashira cross-coupling reaction for fluorocyanopyridine **8**, synthesis of 6-((4-ethylphenyl)ethynyl)-3-fluoropicolinonitrile (compound **9a**): To a degassed solution of picolinonitrile **8** (68 mg, 0.337 mmol, 1.1 equiv) in THF/Et₃N (4 mL/ 2 mL), Pd[PPh₃]₄ (53 mg, 0.046 mmol, 0.15 equiv) and Cul (18 mg, 0.092 mmol, 0.3 equiv) were added. After degassing the reaction mixture for 5 min at room temperature, the alkyne **7a** (40 mg, 0.307 mmol, 1 equiv) was added dropwise and the reaction mixture was stirred at the room temperature for 16 h. After completion (TLC), the reaction mixture was purified by column chromatography (EtOAc/PE 1:9) to afford the desired coupled picolinonitrile **9a** as a white solid (71 mg, 92%).

Procedure for amidoximation reaction of 6-alkynyl-3-fluoro-2cyanopyridine **9a**, synthesis of 6-((4-ethylphenyl)ethynyl)-3fluoro-*N'*-hydroxypicolinimidamide (compound **4a**): A solution of picolinonitrile **9a** (50 mg, 0.20 mmol, 1 equiv), hydroxylamine hydrochloride (21 mg, 0.30 mmol, 1.5 equiv), and Na₂CO₃ (32 g, 0.30 mmol, 1.5 equiv) in dry ethanol (5 mL) was stirred at reflux over 16 h. After completion (TLC), the reaction mixture was filtered through a small celite pad. The filtrate was concentrated under reduced pressure and the residue was purified by column chromatography (EtOAc/PE 1:3) to afford the amidoxime **4a** as a white solid (72 mg, 91%).

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Keywords: Fluorocyanopyridine • fluoroamidoxime • Sonogashira cross-coupling • alkynyl-fluoropyridinamidoxime • ¹⁵N labelled amidoxime

- a) F. Eloy, R. Lenaers, *Chem. Rev.* **1962**, *62*, 155–183; b) T. Sahyoun,
 A. Arrault, R. Schneider, *Molecules* **2019**, *24*, 2470-2489.
- [2] J. Zaman, K. Mitsuru, A. D. Abell, Org. Lett. 2005, 7, 609–611.
- [3] A. Kivrak, M. A Zora, Tetrahedron 2014, 70, 817–831.
- [4] W. Phakhodee, C. Duangkamol, N. Wiriya, M. Pattarawarapan, R. S. C. Adv. 2018, 8, 38281-38288.
- [5] Y. Kohara, K. Kubo, E. Imarniya, T. Naka, J. Heterocycl. Chem. 2000, 37, 1419–1423.
- [6] a) H. Xu, S.Ma, Y. Xu, L. Bian, T. Ding, X. Fang, W. Zhang, Y. Ren, J. Org. Chem. 2015, 80, 3, 1789–1794.
- [7] M. Kocevar, B. Mlakar, M. Perdih, A. Petric, S. Polanc, B. Vercek, *Tetrahedron Lett.* 1992, 33, 2195–2198.
- [8] M. R. Albayati, M. F. A. Mohamed, A. H. Moustafa, Synth. Commun. 2020, 50, 1217-1231.
- [9] I. Doulou, C. Kontogiorgis, A. Koumbis, E. Evgenidou, D. Hadjipavlou-Litina, K. C. Fylaktakidou, *Eur. J. Med. Chem.* 2014, *80*, 145–153.
- [10] Umesha, K. B. Doddamani, D. S. Srikantamurthy, N. J. Chem. Pharm. Sci. 2018, 1, 58–61.
- [11] D. Pancechowska-Ksepko, K. Spalinska, H. Foks, A. Ke Rdzia, M. Wierzbowska, E. Kwapisz, M. Janowiec, Z. Zwolska, E. Augustynowicz-Kopec, *Phosphorus, Sulfur Silicon Relat. Elem.* **2008**, *183*, 1252–1263.
- [12] J. Y. Saulter, J. R. Kurian, L. A. Trepanier, R. R. Tidwell, A. S. Bridges, D. W. Boykin, C. E. Stephens, M. Anbazhagan, J. E. Hall, *Drug Metab. Dispos.* **2005**, 33, 1886–1893.
- [13] L. Paloque, A. Bouhlel, C. Curti, A. Dumetre, P. Verhaeghe, N. Azas, P. Vanelle, *Eur. J. Med. Chem.* 2011, *46*, 2984–2991.
- [14] H. E. Lape, J. O. Hoppe, M. R. Bell, D. Wood, W. H. Selberis, A. Arnold, Arch. Int. Pharmacodyn. Ther. 1968, 171, 394–414.
- [15] M. Stasevych, V. Zvarych, V. Novikov, M. Vovk, Chem. Chem. Technol., 2019, 13, 4, 417-423.
- [16] P. Jiao, P. Jin, C. Li, L. Cui, L. Dong, B. Pan, W.Song, L. Ma, J. Dong, L. Song, *Bioorg. Med. Chem. Lett.* **2016**, *26*, 4679–4683.
- [17] L. A. Kayukova, K. D. Praliev, A. L. Akhelova, U. S. Kemel'bekov, G. M. Pichkhadze, G. S. Mukhamedzhanova, D. M. Kadyrova, S. R. Nasyrova, *Pharm. Chem. J.* 2011, *45*, 468–471.
- [18] Y. Chen, H. Zhao, Y. Li, W. Zhao, X. Yang, X. Meng, H. Wang, J. Chem. Eng. Data, 2019, 64, 4037-4045.

- [19] L. Louis-Leriche, E. Paunescu, G. Saint-André, R. Baati, A. Romieu, A. Wagner, P. Y. Renard, *Chem. Eur. J.* 2010, *16*, 3510-3523.
- [20] G. Saint-André, M. Kliachyna, S. Kodepelly, L. Louise-Leriche, Emilie Gillon, P.-Y. Renard, F. Nachon, R. Baati, A. Wagner. *Tetrahedron* 2011, 67, 6352-6361.
- [21] J. Yerri, J. Dias, M. Reddy Nimmakayala, F. Razafindrainibe, C. Courageux, A.-J. Gastellier, J. Jegoux, C. Coisne, C. Landry, Fabien Gosselet, J. Hachani, J. F. Goossens, M.-P. Dehouck, F. Nachon, R. Baati, *Chem. Eur. J.* **2020**, 26, 15035-15044.
- [22] Y. Jagadeesh, R. Baati, Eur. J. Org. Chem. 2018, 4161-4165.
- [23] T. Katoh, Y. Tomata, T. Tsukamoto, Y. Nakada, *Tetrahedron Lett.* 2015, 56, 6043-6046.
- [24] G. Mercey, T. Verdelet, J. Renou, M. Kliachyna, R. Baati, F. Nachon, L. Jean, P.-Y. Renard, Acc. Chem. Res., 2012, 756–766.
- [25] M. Kliachyna, V. Nussbaum, J. Renou, G. Santoni, B. Sanson, J.-P. Colletier, M. Arboléas, M. Loiodice, M. Weik, L. Jean, P.-Y. Renard, F. Nachon, R. Baati, *Eur. J. Med. Chem.* 2014, 78, 455-467
- [26] a) G. Mercey, T. Verdelet, G. Saint-André, E. Gillon, A. Wagner, R. Baati, L. Jean, F. Nachon and P.-Y. Renard *Chem. Commun.* 2011, *47*, 5295-5297; b) G. Santoni, J. de Sousa, E. de la Mora, J. Dias, L. Jean, J. L. Sussman, I. Silman, P.-Y. Renard, R. C. D. Brown, M. Weik, R. Baati, F. Nachon. *J. Med. Chem.* 2018, *61*, 7630–7639
- [27] R. Li, X. Feng, M. Zhang, Z. Xing, G. Wu, ACS Omega, 2021, 6, 1894-1900.
- [28] P. R. Sruthi, V. Sarika, A. Suku, A. Krishnan, S. Anas, *Inorganica Chim. Acta*, **2020**, 119305.
- [29] Y. He, S. Gou, L. Zhou, L. Tang, T. Liu, L. Liu, M. Duan, Carbohydr. Polym., 2021, 252, 117160.
- J. M. Ellis, M. D. Altman, A. Bass, J. W. Butcher, A. J. Byfors, A. Donofrio, S. Galloway, A. M. Haidle, J. Jewell, N. Kelly, E. K. Leccese, S. Lee, M. Maddess, J. R. Miller, L. Y. Moy, E. Osimboni, R. D. Otte, M. V. Reddy, K. Spencer, B. Sun, S. H. Vincent, G. J. Ward, G. H. C. Woo, C. Yang, H. Houshyar, A. B. Northrup. *J. Med. Chem.* **2015**, *58*, 1929-1939.
- [31] M.-L. A. Sauer, B. Xu, F. Sutton Proteome Sci. 2014, 12, 1-9.

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A practical and straightforward two-step protocol, to access highly valuable and unknown alkynyl-fluoro-pyridinamidoximes compounds.

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