

Deoxyribonucleic Acid Encoded and Size-Defined π -Stacking of Perylene Diimides

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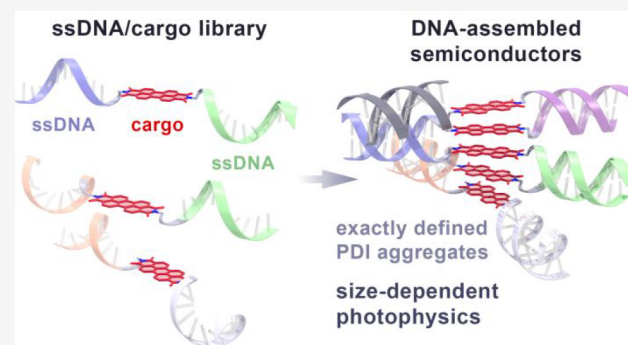


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ABSTRACT: Natural photosystems use protein scaffolds to control intermolecular interactions that enable exciton flow, charge generation, and long-range charge separation. In contrast, there is limited structural control in current organic electronic devices such as OLEDs and solar cells. We report here the DNA-encoded assembly of π -conjugated perylene diimides (PDIs) with deterministic control over the number of electronically coupled molecules. The PDIs are integrated within DNA chains using phosphoramidite coupling chemistry, allowing selection of the DNA sequence to either side, and specification of intermolecular DNA hybridization. In this way, we have developed a “toolbox” for construction of any stacking sequence of these semiconducting molecules. We have discovered that we need to use a full hierarchy of interactions: DNA guides the semiconductors into specified close proximity, hydrophobic–hydrophilic differentiation drives aggregation of the semiconductor moieties, and local geometry and electrostatic interactions define intermolecular positioning. As a result, the PDIs pack to give substantial intermolecular π wave function overlap, leading to an evolution of singlet excited states from localized excitons in the PDI monomer to excimers with wave functions delocalized over all five PDIs in the pentamer. This is accompanied by a change in the dominant triplet forming mechanism from localized spin–orbit charge transfer mediated intersystem crossing for the monomer toward a delocalized excimer process for the pentamer. Our modular DNA-based assembly reveals real opportunities for the rapid development of bespoke semiconductor architectures with molecule-by-molecule precision.



INTRODUCTION

Plants and photosynthetic bacteria employ protein scaffolds to impose spatial control over a multitude of chlorophylls and carotenoids, dictating the energetic landscape and fate of photogenerated excitons and separated charge carriers.^{1–3} However, such deterministic morphology control remains inaccessible for current organic electronic materials as used in organic light-emitting diodes, transistors, and photovoltaics.

Controlling the delocalization of excited states in artificial organic semiconductors is a key, unmet challenge. With classical self-assembly approaches, unrestricted self-aggregation of semiconductors precludes deterministic size-control of π -stacked molecules, leading to a distribution of molecular agglomerates instead.^{4,5} While alternative covalent methods can confine two (or more) semiconductors into π -interacting arrangements,^{6–8} their low synthetic yield and poor solubility of the products render these *de novo* methods inadequate for generating larger systems.

DNA nanotechnology provides a powerful means to repurpose double-helix formation for nanoconstruction.^{9,10}

Base pairing between complementary single-stranded DNA (ssDNA) is stable, predictable, and high yielding, leading to programmable supramolecular architectures.^{11,12} While nucleobases have optical band gaps too large for technological implementation as electronically active components, they are excellent scaffolds that can direct the assembly of attached cargos such as molecular semiconductors.^{13–18}

DNA-mediated assembly of organic semiconductors, however, can be challenging because their hydrophobic aromatic backbone encourages self-aggregation.¹⁹ This has limited semiconductor/DNA constructs mostly to weakly aggregating dyes such as cyanines at the cost of reduced electronic

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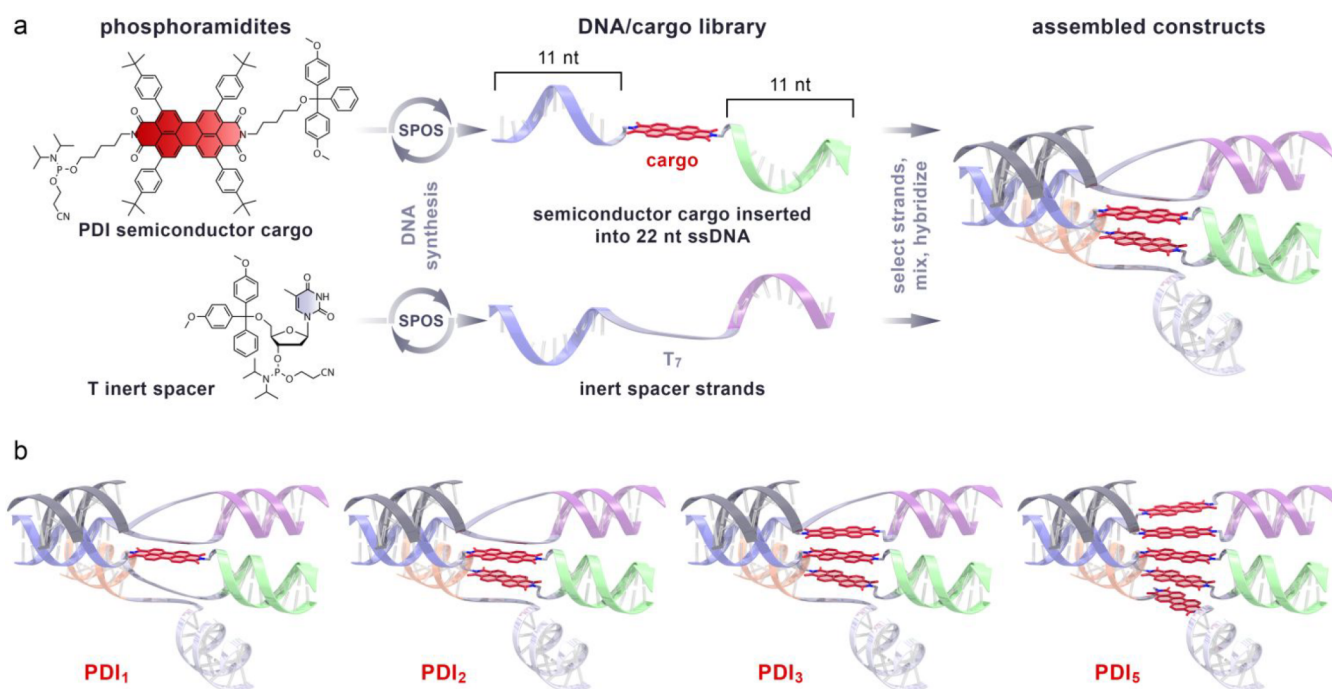


Figure 1. Assembling organic semiconductors via a DNA scaffold. (a) Asymmetric perylene diimide (PDI) phosphoramidites are conjugated to 22 nucleotide (nt) single-stranded DNA (ssDNA) by solid phase oligonucleotide synthesis (SPOS). Alternatively, heptathymine is employed as an electronically inert spacer. Repeating this process with different DNA sequences enables us to set up a cargo/DNA library with partially complementary ssDNA (color coding represents pairwise complementary base sequences). The final semiconductor/DNA constructs are assembled by selecting the desired components from this library, followed by hybridization to form rigid double-stranded DNA (dsDNA). (b) Schematic representation of the semiconductor/DNA constructs used in this study.

coupling.^{20,21} However, these charged dyes are very different from the aromatic organic semiconductors employed in optoelectronic devices. To control the number of interacting dyes, previous approaches have mostly confined all dyes to a single DNA double helix.^{22,23} However, if the dyes are incorporated instead of nucleobases, inserting several unnatural modifications can disrupt local base pairing and the double-helix structure.^{24,25} Hence, examples of more than two π -stacked dyes assembled within one DNA double helix are rare. The other common strategy is the covalent attachment of dyes to the nucleobases. Here, the base pairing and duplex stability are less disrupted, enabling modifications of several adjacent base pairs, especially when a zipper-like arrangement is chosen.²⁶ However, since the dyes are positioned on the outside of the DNA double helix and are constrained to the DNA rotational pitch, their π -orbital overlap and electronic communication across multiple chromophores are limited.

We report here the DNA-encoded assembly of up to five π -conjugated perylene diimides (PDIs). These or an electronically inert spacer are integrated within DNA chains using phosphoramidite coupling chemistry, allowing selection of the DNA sequence to either side, and specification of intermolecular DNA hybridization. In this way, we have developed a “toolbox” for constructing bespoke semiconductor stacks with tailored electronic properties. Our approach facilitates extended PDI stacks with minimal base-pair disruption, compared to previous hairpin approaches.^{19,27} We use a broad range of spectroscopic tools to track the temporal evolution of photogenerated excitons and their subsequent conversion into excimers and triplet states. Our modular DNA-based assembly reveals real opportunities for the rapid

development of customized semiconductor architectures with molecule-by-molecule precision.

RESULTS AND DISCUSSION

The specificity of DNA base pairing enables the exact recognition between two complementary base sequences. ssDNA can share partial base-complementarity with several separate ssDNA strands, forming interconnected double stranded DNA (dsDNA) helices upon hybridization. This method of interconnecting dsDNA allows a combinatorial set of ssDNA to assemble into preprogrammed structures with nanometer resolution.

We adapted this method to generate precisely defined semiconductor/DNA constructs. Each construct consists of six zigzag interconnected dsDNA helices, formed by two 11-nucleotide (nt) “end-caps” and five partially complementary 22 nt cargo-modified ssDNA (Figure 1a and Supporting Information, Section C). The cargo is either a molecular semiconductor or heptathymine (T_7) as an optically and electronically inert spacer. Importantly, the base sequences are specific to a position within the constructs, ensuring that the location of each cargo is programmable, and cargos are fully exchangeable. Perylene diimide (PDI) was chosen as a well-studied model semiconductor due to its chemical stability and strong tendency to π -stack. This approach overcomes previous limitations of dye–DNA base pair disruption by balancing duplex hybridization and hydrophobicity-driven dye assembly. Deploying a strategy where only one PDI molecule is attached per DNA strand, we leave base-pairing and duplex formation comparatively unimpeded. While the strong hydrophobicity of the PDIs is required as a driving force for structure assembly, aggregate size control is defined by two flanking DNA helices.

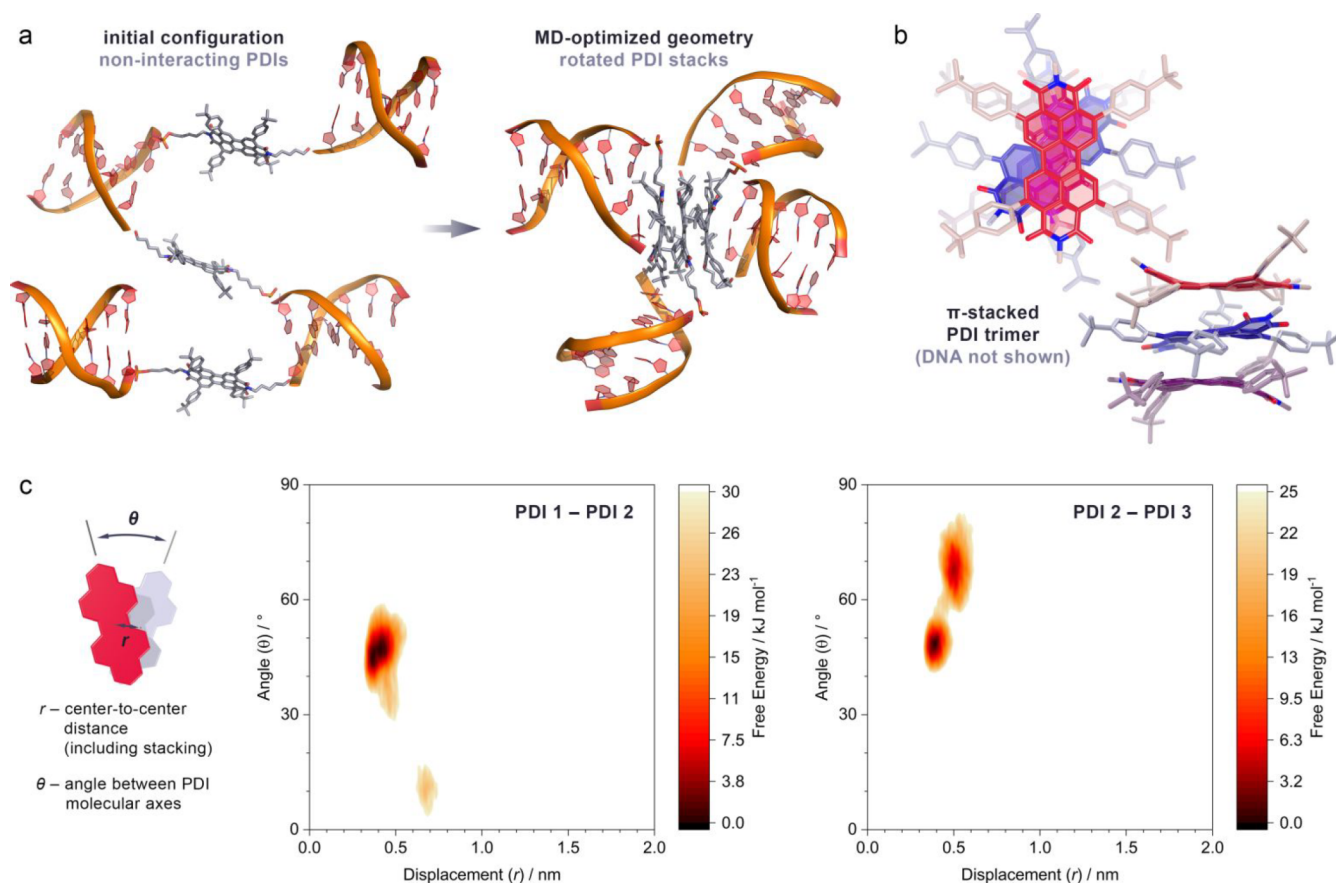


Figure 2. Structure simulations of a PDI trimer using atomistic molecular dynamics (MD) simulations with metadynamics sampling. (a) Comparison of the initial (left) and the minimum-energy configuration (right). (b) The optimized geometry features closely stacked PDIs where the middle one (blue) is rotated by about 45 deg. This configuration maximizes the overlap between the PDI cores while balancing the steric constraints due to the *tert*-butylphenyl substituents and the size mismatch between the dsDNA and the considerably smaller PDIs. Pentyl linkers and DNA are omitted for clarity. (c) MD-calculated free energy maps probing the angle θ and the center-to-center distance r between the adjacent PDIs in the trimer. The energy scale is relative to the energy minimum. Despite different trajectories during the simulation where aggregation of one pair influences the other, both PDI pairs converge to the same global minimum around $r = 0.4$ nm (i.e., π -stacked with only small lateral offset) and $\theta = 45^\circ$.

Cargo-modified ssDNAs were prepared by sequential growth using phosphoramidite-based solid-phase oligonucleotide synthesis (SPOS) (SI, Section D). We chose phosphoramidite coupling over other bioconjugation strategies such as amide formation or copper(I)-catalyzed alkyne–azide cycloaddition (CuAAC), as the phosphoramidite route provides a short and geometrically simple connection between the cargo and DNA, and is compatible with strongly hydrophobic semiconductor cargos. Moreover, the sequential growth via SPOS ensures that the directionality of the DNA is maintained throughout each ssDNA-cargo-ssDNA strand. This is vital for the assembly of our semiconductor/DNA constructs, but very challenging to achieve via alternative amide or CuAAC approaches where the respective 3' and 5' modifiers are geometrically different.

After the synthesis of an initial 11 nucleotide (nt) ssDNA, the PDI cargo was inserted by a modified SPOS procedure, followed by another 11 nt ssDNA. Pentyl chains were employed as linkers to bridge the size disparity between the PDIs (π -stacking distance ≈ 0.4 nm) and the larger dsDNA helix diameter (≈ 2 nm). In order to generate optically and electronically inert spacer strands, seven thymidine units were appended instead of the PDI cargo.

A cargo/DNA library was set up by generating the PDI- and T₇-modified ssDNA strands for each position in the stack. To

generate the PDI/DNA constructs, the desired strands (one for each position) were selected from this library, followed by hybridization into dsDNA. Successful hybridization was confirmed by native polyacrylamide gel electrophoresis (PAGE) and temperature-dependent UV–vis spectroscopy (SI, Sections E, F). The constructs we chose for this study are shown in Figure 1b and range from the monomeric PDI₁ to an extended stack of PDIs in PDI₅. The steric bulk and charge of the surrounding DNA ensure that in dilute solution all relevant electronic interactions are confined to the stacks with negligible crosstalk across neighboring constructs (SI, Section G).

While the DNA defines the number of PDIs per construct and dictates their general position within the stacks, the subnanometer scale arrangement is dominated by local interactions. In the highly polar aqueous environment, the PDIs aggregate due to hydrophobic/hydrophilic differentiation. The local geometry inside these aggregates is governed by electrostatic and dipole interactions and the 3D shape of the functionalized PDIs.

We developed structural models of the aggregated PDIs to both inform our construct design and support later photo-physical analysis. To characterize the conformational ensembles of cargo/DNA constructs, we performed atomistic

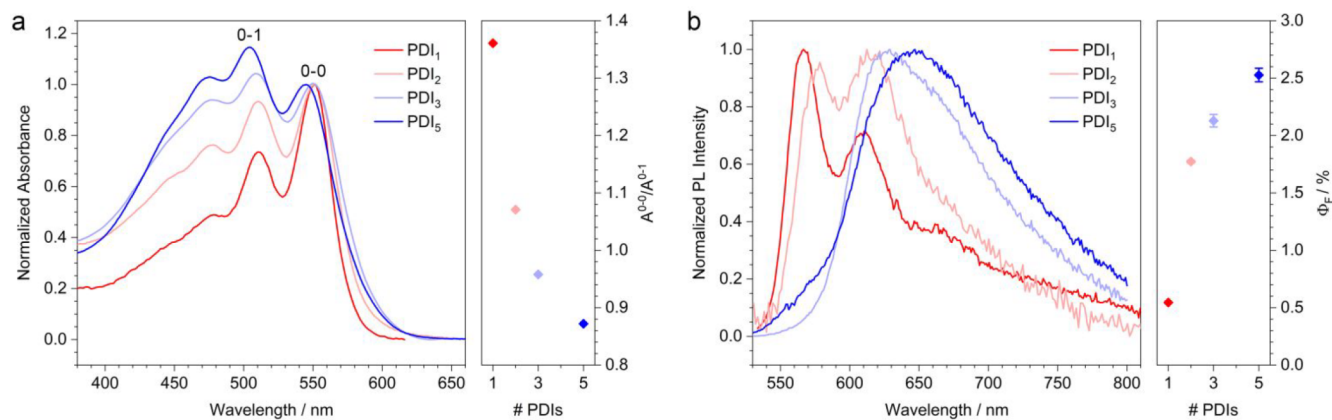


Figure 3. Steady-state optical characterization of the PDI/DNA constructs. (a) Absorption spectra of PDI₁–PDI₅ (10–30 μ M in PBS), normalized to the absorbance of the 0–0 vibrational band. Right panel: Progressive H-aggregation with increasing number of PDIs per construct is evident from the absorption ratio between the 0–0 and 0–1 vibrational bands. (b) Photoluminescence (PL) emission spectra of the PDI/DNA constructs normalized to their respective maxima. The emission profiles change gradually from monomeric to excimer-like with increasing number of coupled PDIs. Right panel: The PL quantum efficiency (ϕ_F) increases with the number of PDIs due to suppressed competing relaxation pathways.

molecular dynamics (MD) simulations in combination with a well-tempered metadynamics algorithm for enhanced sampling.^{28,29} We selected a PDI trimer with four 6 nt dsDNA as our main model system, since it allows us to probe cross-correlation of aggregated PDIs beyond a dimer but is sufficiently small to be simulated at the required level of theory (Figure 2a). See the SI, Section H for details and further simulations.

MD simulations were initialized with noninteracting PDIs (≥ 1.5 nm separation) and proceed in 2 fs steps for 100 ns, driven by a history-dependent biasing potential to escape local minima. Prior MD simulations of similar-sized PDI-DNA base surrogates have demonstrated that the configurations typically converge in <10 ns and well before the 100 ns sampled here.^{30,31} During the simulations, the PDIs and DNA are free to sample a wide range of configurations, including dsDNA dehybridization and variation of the PDI–PDI separation.

The PDI trimer converges to a cofacially stacked configuration which maximizes the electronic overlap between the PDIs (Figure 2a, b). The DNA helices remain fully hybridized.

In order to assess the stability of this arrangement, we computed the free energy maps for the center-to-center distance (r) and the rotational offset (θ) between adjacent PDIs in the trimer (Figure 2c). In this simulation, the energy landscape is multidimensional as the aggregation of two PDIs influences the stacking interactions with the third PDI. Moreover, due to the directionality of the DNA, the trimer is not symmetric with PDI 1 being attached at the 5' end of the connecting dsDNA and PDI 3 at the 3' end. Hence, the energy maps are different for PDI 1–PDI 2 and PDI 2–PDI 3, respectively. The geometries corresponding to each of the observed minima are shown in the SI, Figure S7-2.

PDI 1–PDI 2 shows a clear preference for a 45° rotated arrangement with a center-to-center distance of about 0.4 nm, which approximates the typical distance between π -stacked PDIs with only minimal lateral offset. This rotated stacking motif is known from the solid-state structures of PDIs with bulky imide substituents.³² A shallow local minimum is observed at $r = 0.7$ nm/ $\theta = 10^\circ$. This configuration resembles the slip-stacked solid-state structure of *ortho*-functionalized PDIs such as our monomer,³³ but appears significantly

destabilized in the PDI/DNA constructs due to the boundary conditions imposed by the sterically demanding DNA.

The PDI 2–PDI 3 pair assumes the same 45°-rotated, π -stacked arrangement as its most favorable geometry. A secondary minimum corresponds to a laterally displaced rotated stack with slightly reduced overlap between the PDI cores. This arrangement converts into the closely stacked configuration as the simulation proceeds.

Despite the inherent asymmetry of the PDI trimer due to the DNA directionality, both PDI pairs converge to the same geometry. This confirms that the above-described hierarchy of interactions leads to a specific and stable local geometry.

MD simulations carried out for a T₇-flanked PDI dimer confirm that the PDI aggregation is not significantly altered by the addition of inert spacer strands (SI, Figure S7-1). Moreover, we simulated the geometry of the larger PDI₅ pentamer via computationally less expensive force-field methods (SI, Figure S7-3). The simulation reveals that PDI stacking is not hindered by the steric bulk of the six dsDNA. Indeed, the optimized configuration is very similar to the model PDI trimer with the PDIs adopting a 45° rotated, closely π -stacked arrangement.

The stacking geometry can have significant influence on the electronic communication across the PDI stacks. We quantified the electronic coupling by computing the singlet excitation energy transfer (SEET) integrals (t_s) between adjacent PDIs via time-dependent density functional theory (TD-DFT), using the MD-optimized rotated stack or slip-stacked geometries. We find that $t_s = 112$ meV for the rotated stack, which is comparable to the 80–150 meV reported for self-assembled PDIs,^{34,35} confirming that the PDI/DNA constructs enable strong π - π coupling. In contrast, t_s of the energetically less favorable (and experimentally not observed) slip-stacked geometry is only 53 meV, indicating significantly weaker electronic coupling.

The optical absorption spectra of PDIs are highly sensitive to aggregation effects, rendering these semiconductors an excellent probe for analyzing the DNA-encoded assembly. Modulation of the vibronic band intensities arises from a complex interplay between Coulomb and charge-transfer-mediated interactions that depend on the PDI–PDI spacing and local geometry.^{36,37}

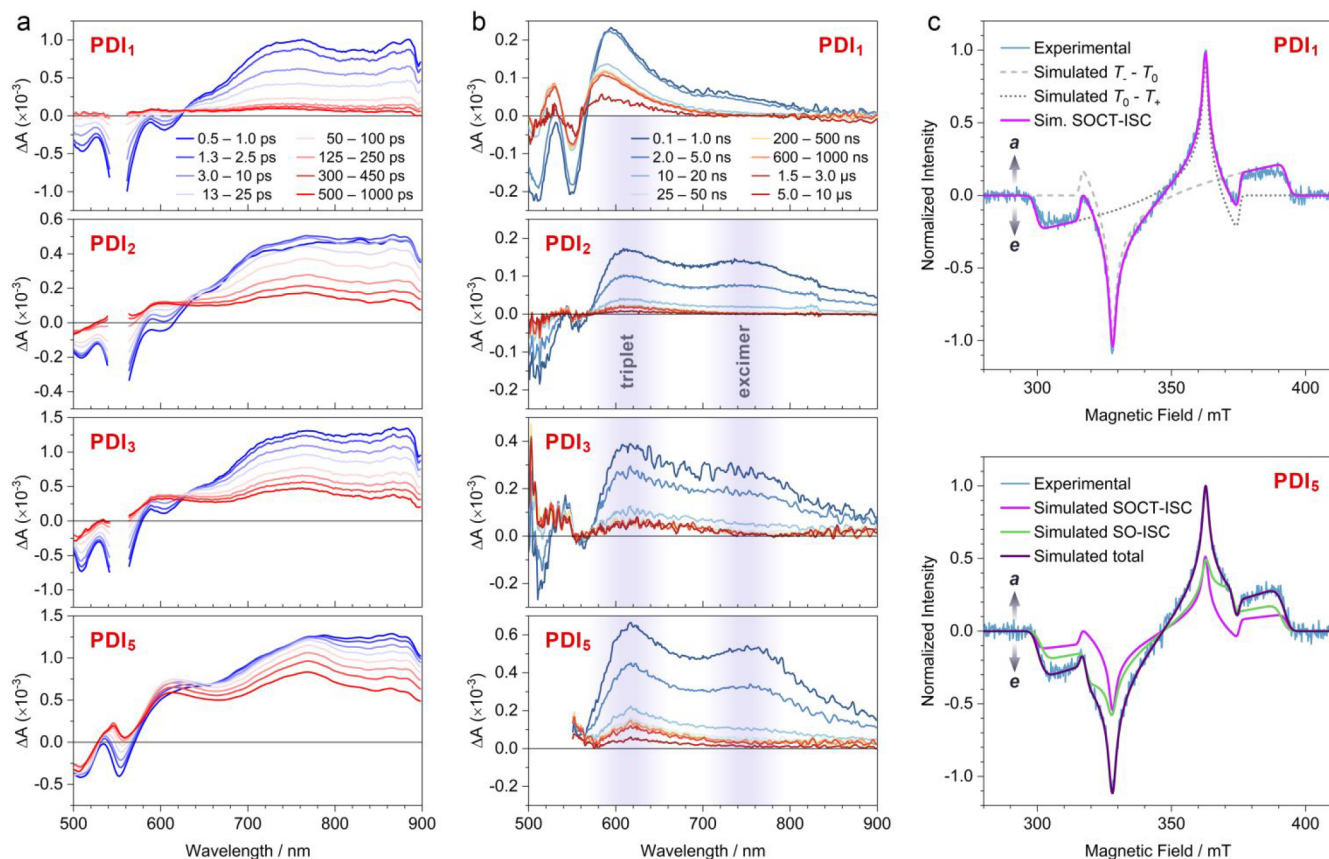


Figure 4. Time-resolved spectroscopy. (a) Femto-/picosecond transient absorption spectra of the PDI/DNA constructs (10–30 μM in PBS). Excitation 550 nm, 24 $\mu\text{J cm}^{-2}$. 540–560 nm removed due to pump scatter. (b) The corresponding nanosecond transient absorption spectra. Excitation 555 nm, 100 $\mu\text{J cm}^{-2}$. (c) Transient EPR spectra of **PDI₁** and **PDI₅**, recorded at 100 K, time averaged 0.5–4 μs after optical excitation at 532 nm. The trEPR spectrum of **PDI₁** is well described by a triplet that is formed via a SOCT-ISC mechanism, giving rise to an *aeae/aea* ESP pattern (purple line). The dotted and dashed lines represent the simulated spectra of the $T_- - T_0$ and $T_0 - T_+$ components. The trEPR spectrum of **PDI₅** requires two triplet species with different ESP patterns. The first one (purple line; 31% contribution) is a SOCT-ISC triplet with the same parameters as in **PDI₁**, whereas the second one (green; 69% contribution) is formed via an excimer-based SO-ISC mechanism, which gives rise to an *eee/aaa* ESP pattern.

PDI₁ displays the typical vibronic progression of monomeric PDIs with a peak ratio between the 0–0 and 0–1 vibronic bands (A^{0-0}/A^{0-1}) of >1.3 (Figure 3a).³⁸ The decreased vibronic peak ratio of **PDI₂** indicates (relatively weak) H-coupling between the two PDIs. **PDI₃** and **PDI₅** show stronger H-aggregation with decreasing A^{0-0}/A^{0-1} peak ratio and blue-shifted absorption peaks. These observations support the anticipated (rotated) side-by-side stacking (as opposed to staircase arrangement) in our PDI/DNA constructs.³⁹ Moreover, the progressive aggregation effects in **PDI₃** and **PDI₅** confirm electronic coupling across the entire stacks.

The photoluminescence (PL) emission spectrum of **PDI₁** mirrors its absorption spectrum, confirming the monomer-like properties of this construct (Figure 3b). The PL quantum efficiency (φ_F) is about 0.6%. Low φ_F values are frequently observed for *ortho*-substituted PDIs and have been attributed to charge transfer (CT) from the phenylenes and increased out-of-plane vibrations promoting nonradiative decay pathways such as CT-mediated intersystem crossing (ISC) to triplets.^{40,41} Consistent with this explanation, the φ_F of the non-nucleic PDI diol 3 increases from 2.6% in solution to 23% in the solid state (SI, Section K).

With increasing number of coupled PDIs, the emission spectra become broader, featureless, and Stokes-shifted. This is

characteristic of PDI excimer emission (1^{*ex}PDI), where the initial photoexcitation delocalizes over multiple chromophores, rearranging them into a lower-energy geometry, before decaying back to the ground state structure.^{42,43} PDI excimers conventionally act as fluorescence traps. However, in our constructs, the rapid nonradiative losses that limit φ_F in **PDI₁** are outcompeted by the excimer formation, causing an overall increase in φ_F from 0.6% (**PDI₁**) to 2.5% (**PDI₅**).

In order to track the evolution of excited states and investigate the impact of the PDI aggregate size, we employed a range of time-resolved spectroscopic techniques, specifically femto-/picosecond and nanosecond transient absorption spectroscopy (fs-TAS and ns-TAS, respectively), and transient electron paramagnetic resonance (trEPR) spectroscopy.

In TAS, photoexcitation of **PDI₁–PDI₅** generates ground state bleach (GSB) signals at 510 and 550 nm, and a short-lived stimulated emission (SE) signal at 605 nm (Figure 4a). Photoinduced absorption bands (PIA) are most prominent in the 630–900 nm range but can partially overlap with the GSB and SE at shorter wavelengths. While the GSB and SE signals are similar across the range of PDI/DNA constructs, the PIA bands and lifetimes of their components reveal significant differences.

Photoexcitation of PDI_1 populates the singlet excited state (^1PDI) with two overlapping PIA signals at 730 and 760 nm and an absorption band at 850–880 nm. The ^1PDI signal decays within 500 ps and the GSB decays with approximately the same rate. Since the SE signal is small and significantly shorter-lived (<25 ps), we identify nonradiative internal conversion (IC) as the dominant decay pathway to the ground state. This is in line with the known tendency of *ortho*-phenyl substituents on PDIs to enhance out-of-plane vibrations, boosting nonradiative relaxation pathways and limiting φ_F as discussed above.

Concomitant with the ^1PDI decay, we observe the appearance of a weak PIA band centered at 600 nm (partially overlapping with the remaining GSB), which decays slowly over $>5 \mu\text{s}$ (Figure 4b). We assign this new species to a small population of nonemissive triplet states (^3PDI), which are formed via intersystem crossing (ISC). This assignment is confirmed by triplet sensitization experiments (SI, Figure S9) and EPR (see below). Monomeric *ortho*-phenyl PDIs have previously been reported to increase the ISC rate through intramolecular charge-transfer interactions.⁴⁴ The triplet formation is also observed in the fs-TAS data of the PDI diol **3** in chloroform solution (SI, Section K).

In contrast to literature reports,^{27,45} we do not observe the (in our case undesired) formation of a radical anion in PDI_1 via photoinduced hole transfer to the purine nucleobases (A and G). We attribute this to the different geometry and linkage of our constructs, steric and electronic effects of the PDI *ortho*-substitution, and the short ^1PDI lifetime (SI, Section L).

In the constructs that contain at least two electronically coupled PDIs the lifetime of excited states is significantly extended (Figure 4a). After 500 ps, about 25% of the PDI_2 and PDI_3 PIA, and more than 50% of the PDI_5 PIA band intensity are still present. Moreover, changes in the PIA spectra reveal the population of different excited states when at least two PDIs are electronically coupled.

In PDI_5 , where these differences are most prominent, a new state with a characteristic absorption band around 770 nm evolves within 1 ps (Figure 4a, bottom panel). The new state overlaps significantly with the GSB and SE bands at shorter wavelengths. In line with the findings from the steady-state PL spectroscopy, we assign this PIA to a singlet excimer state ($^1\text{exPDI}$)—the delocalization of the excited state over several coupled PDIs, which is accompanied by a structural relaxation to a new lower-energy excited state geometry. This assignment is further confirmed by EPR (see below). Tracking the evolution of the ^1PDI 850–880 nm band, we observe an interconversion of the short-lived ^1PDI Frenkel excitons into the $^1\text{exPDI}$. The 770 nm excimer absorption persists for tens of nanoseconds (Figure 4b, bottom panel). Excimer formation is also observed in PDI_2 and PDI_3 with comparable lifetimes but lower ΔA . The fs-TA spectra suggest the coexistence of ^1PDI and $^1\text{exPDI}$ at early times and incomplete conversion to the excimer (Figure 4a, middle panels). These findings are consistent with the trend observed in the steady-state PL spectra and the lower excimer yield according to the ns-TAS.

PDI_2 – PDI_5 display greatly enhanced triplet yields compared to PDI_1 , as evident from the pronounced 600 nm PIA (Figure 4a,b). Higher triplet yields can be explained by the longer availability of singlet (^1PDI and $^1\text{exPDI}$) states due to slower vibrational relaxation to ground state and hence more time for ISC, and a more efficient ISC mechanism via the excimer (see below).

We sought further confirmation of our above assignments from trEPR spectroscopy (Figure 4c). The trEPR spectrum of PDI_1 shows a broad triplet feature with zero-field splitting (ZFS) parameters of $D = 1314 \text{ MHz}$ and $E = -115 \text{ MHz}$, which are consistent with literature values for similar PDI systems.²⁷ The electron spin polarization (ESP) pattern is *eaе/aеа*, where *e* = enhanced emission and *a* = enhanced absorption. Simulations of the triplet spectrum using the EasySpin software package⁴⁶ yield relative triplet sublevel populations of $p_x, p_y, p_z = 0.18, 0.82, 0$, indicating an overpopulation of the T_y zero-field eigenstate.

While the direct spin–orbit coupling mediated intersystem crossing (SO-ISC) from ^1PDI to ^3PDI is negligible in unsubstituted monomeric PDIs,⁴⁷ the intramolecular charge transfer (CT) interactions in *ortho*-substituted PDIs causes increased ISC rates and thus reduced φ_F as discussed above.^{40,41,44} Consistent with this intramolecular CT character, the *eaе/aеа* ESP pattern of PDI_1 and overpopulation of the T_y sublevel suggest that the triplet formation proceeds through a spin–orbit charge transfer intersystem crossing (SOCT-ISC) mechanism.^{48–50} SOCT-ISC leads to an overpopulation of T_y when $D > 0$, which is common in planar aromatic molecules, and $E < 0$. Furthermore, the absence of a clear CT state signal in the PDI_1 trEPR spectrum indicates strong coupling between the donor and acceptor moieties in the singlet CT (^1CT) state, which is known to facilitate SOCT-ISC.⁵¹ Alternative triplet formation mechanisms such as radical pair ISC⁵² or singlet fission⁵³ can be ruled out, since both mechanisms directly populate the high-field eigenstates, leading to, e.g., overpopulation of the T_0 sublevel and giving rise to an *aeе/aeе* ESP pattern, for $D > 0$.

For the analysis of the PDI_5 trEPR spectrum, we assume the same sign for the ZFS parameters as in PDI_1 . However, a single triplet species does not fit the experimental data well. Besides the SOCT-ISC species (with the same ZFS parameters as for PDI_1) we observe a second species with $D = 1260 \text{ MHz}$, $E = -93 \text{ MHz}$, and $p_x, p_y, p_z = 0.46, 0.54, 0$. In this species, the more equal relative populations of the T_x and T_y sublevels and zero population of T_z leads to an *eee/aaa* ESP pattern. Cofacial PDI aggregates have been reported to show triplet state generation through an excimer state with significant CT character ($\text{PDI } \delta^-/\text{PDI } \delta^+$).⁴³ In these systems, the dominant triplet forming mechanism is an excimer-based spin–orbit coupling induced intersystem crossing (SO-ISC), which leads to the *eee/aaa* ESP pattern observed in PDI_5 .²⁷ The ZFS parameters of the SO-ISC species are also slightly smaller than those of the SOCT-ISC triplet, suggesting a more delocalized triplet state.

In PDI_5 , the relative contributions of the excimer-based SO-ISC triplet and intramolecular SOCT-ISC triplet to the observed spectrum are 69% and 31%, respectively. A modified PDI_2 displays the same two triplet species, but with 39% and 61% for the excimer-based SO-ISC and SOCT-ISC triplets, respectively (SI, Section M). This trend of excimer formation outcompeting intramolecular electronic processes as the number of electronically coupled PDIs increases is fully consistent with the progression observed in the PL and TAS spectra.

Upon further cooling, we observe a CT state (a narrow *a/e* signal around $g = 2$) in the PDI_2 and PDI_5 samples, which we attribute to the partial CT character of the excimers (see the SI, Section M).

Overall, the exciton delocalization and excimer formation processes of PDI₂ align with previous PDI dimers restricted in DNA hairpins,²⁷ while our larger-order structures approximate the photophysical characteristics of their solid-state analogues.

CONCLUSION

This work presents a method for rapid development of bespoke semiconductor architectures with molecular precision via DNA-encoded assembly. Using perylene diimide (PDI) as a representative example of strongly packing molecular semiconductors, we have discovered that we can achieve well-controlled aggregation by employing a hierarchy of interactions. While the DNA defines the overall layout on the nanometer scale, the exact intermolecular positioning is achieved via electrostatic and dipole interactions and the 3D geometry of the substituted PDIs. Our semiconductor/DNA constructs provide facile access to semiconductor stacks beyond the size limitations of conventional covalent chemistry approaches and enable systematic exploration of the size dependence of electronic processes. We illustrate the strength of this approach by studying the extension of singlet excited states from monomeric PDI to stacked PDI pentamers. With an increasing number of coupled PDIs, we observe a continuing spatial extension from localized excitons to excimers in which wave functions are delocalized over at least five PDI units. Our balanced hydrophobic-directed approach to assembly will be broadly extendable to most state-of-the-art optoelectronic materials, which share a lack of water-solubilizing moieties similar to PDI. The DNA-based assembly method will open up greater chemical design space to interrogate the size dependence of technologically important phenomena including singlet fission, triplet–triplet annihilation, and symmetry-breaking charge transfer in the future.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/jacs.1c10241>.

Experimental methods, synthetic procedures, structure analysis and additional spectroscopic data (PDF)

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Notes

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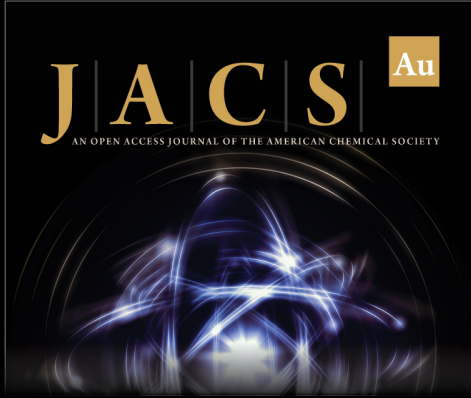
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