

Transient mobilization of subcrustal carbon coincident with Palaeocene-Eocene thermal maximum

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Abstract

Plume magmatism and continental breakup led to the opening of the northeast Atlantic Ocean during the globally warm early Cenozoic. This warmth culminated in a transient (170 thousand year, kyr) hyperthermal event associated with a large, if poorly constrained, emission of carbon called the Palaeocene-Eocene thermal maximum (PETM) 56 million years ago (Ma). Methane from hydrothermal vents in the coeval North Atlantic igneous province (NAIP) has been proposed as the trigger, though isotopic constraints from deep-sea sediments have instead implicated direct volcanic carbon dioxide (CO₂) emissions. Here, we calculate that background levels of volcanic outgassing from mid-ocean ridges and large igneous provinces yield only one-fifth of the carbon required to trigger the hyperthermal. However, geochemical analyses of volcanic sequences spanning the rift-to-drift phase of the NAIP indicate a sudden ~220-kyr-long intensification of magmatic activity coincident with the PETM. This was likely driven by thinning and enhanced decompression melting of the sub-continental lithospheric mantle, which critically contained a high proportion of carbon-rich metasomatic carbonates. Melting models and coupled tectonic–geochemical simulations indicate that >10⁴ gigatons of subcrustal carbon was mobilized into the ocean and atmosphere sufficiently rapidly to explain the scale and pace of the PETM.

During the early Cenozoic, a major episode of continental breakup in the northeast Atlantic (Fig. 1a) resulted in widespread magmatism and the opening of the northeast Atlantic Ocean^{1,2}. Initially, continental extension above the Iceland plume caused rifting and an associated phase of subaerial volcanism lasting ~5 million years (Myr)^{3,4,5,6}. By ~56 Ma, rifting had progressed to near continental breakup¹ (Fig. 1d), coinciding with an order-of-magnitude increase in melt production rates², and accelerated westward migration of Greenland⁷ (Fig. 1e). This transitional phase of volcanism in the NAIP involved peak eruption rates of ~2.4 km³ yr⁻¹ (ref.⁴) and occurred contemporaneously with intense early Eocene warming, including the PETM⁸. Expansion of the northeast Atlantic then continued through mid-ocean ridge spreading^{1,5} (Figs. 1b,d,e).

During the PETM, sea surface temperatures increased by ~5°C (ref.⁹), ocean acidification occurred^{10,11} and there was a geologically abrupt negative δ¹³C carbon isotope excursion that lasted about 170 thousand years (kyr)^{8,12} (Fig. 1c). It is thought that the release of approximately 1–1.5 × 10⁴ gigatons (Gt) of carbon into the ocean-atmosphere system is necessary to account for these changes^{11,13}. This massive carbon flux has been attributed to methane release (δ¹³C –60 ‰) associated with

clathrate destabilization along continental shelves¹⁴ and hydrothermal venting during interaction of magmas and organic-rich mudrocks (δ¹³C –35 to –50 ‰)¹⁵. Simulations of the latter process predict peak emissions of 0.2–0.5 Gt C yr⁻¹, potentially potent enough to drive PETM warming¹⁶. Alternatively, based on paired δ¹¹B–δ¹³C data and carbon-cycle modelling¹¹, and B/Ca in planktic foraminifera¹³, it has been proposed that triggering of the PETM required the injection—over about 50 kyr—of an isotopically heavier source of carbon (δ¹³C –11 ‰), most likely tied to volcanic outgassing across the NAIP¹¹. Under this scenario, considerably more carbon is required from magmatic and volcanic sources to drive the negative δ¹³C excursion than is required from a methane/organic carbon source. Hence, it remains unclear whether NAIP magmatism could contribute enough CO₂ on sufficiently short timescales (10s to 100s kyr) to initiate PETM warming, especially given that this magmatism occurred over several million years^{1,2,11,15}. Critically, how carbon emissions were distributed over this interval is unknown¹¹.

Quantifying background volcanic CO₂ fluxes

We quantitatively estimated the combined CO₂ emissions from incipient mid-ocean ridge volcanism and large igneous

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provinces (LIPs) in the NAIP, parameterised by plate-tectonic reconstructions (Fig. 1b,d) and existing estimates of magmatic productivity⁴ (Methods). In our model, we considered pre-eruptive CO₂ concentrations of 2 wt%, which are typical of flood basalt eruptions¹⁷. The associated CO₂ release from the ocean crust is then calculated from the ratio between the observed levels of magmatic production along the present-day mid-ocean ridge system ($\sim 18 \text{ km}^3 \text{ yr}^{-1}$), and estimated CO₂ degassing fluxes along the modern global ridge system ($7 \times 10^{11} \text{ mol yr}^{-1}$)¹⁸. This calculation suggests that $\sim 10\%$ of the CO₂ in the juvenile ocean crust is degassed at ridges. Using this estimate, the NAIP ridges likely produced $\sim 1.1 \times 10^3 \text{ Gt C}$ during the PETM. This is an order of magnitude lower than the requisite $\sim 1\text{--}1.5 \times 10^4 \text{ Gt C}$ (refs. ^{11,13}). Inclusion of the effects of LIP magmatism—assuming most probable eruption rates⁴ and total degassing—yields less than one-fifth of the carbon needed to drive and sustain the PETM^{11,13}. Furthermore, seafloor isochrons indicate that magmatic productivity along the northeast Atlantic ridge peaked after, not during, early Eocene hyperthermals (Fig. 1d). Therefore, some other major, but transient, source of volcanic carbon appears to be required if the volcanic outgassing hypothesis^{11,13} is correct.

Spatial and temporal characteristics of PETM volcanism

We investigated several volcanic sequences spanning the Palaeocene-Eocene boundary (Fig. 1c). The Deep Sea Drilling Project Leg 81 Site 555 lies on the Rockall Plateau (Fig. 1a), with volcanics extruded just prior to breakup in the proto-northeast Atlantic (Fig. 1b). Here, Phase 1 volcanism¹⁹ (Fig. 2a) is coeval with the Milne Land basalts in East Greenland and the Middle to Upper Series lavas in the Faroe Islands¹ (Fig. 2b-c). In the Rockall sequence, we found a sharp increase in the frequency of volcanic tuffs just below the Palaeocene-Eocene boundary (Fig. 2a) (Methods). The PETM is defined by $\delta^{13}\text{C}$, however the volcanostratigraphy in our study area is not conducive to developing a high-resolution carbon isotope stratigraphy (Methods). Therefore, we rely on a combination of radiometric, magnetostratigraphic and paleontological age constraints, in addition to existing sediment accumulation rate estimates (Methods). Mudstones interbedded with the uppermost tuffs are thought to contain the dinoflagellate cyst, *Apectodinium augustum*²⁰, which signifies a sudden prevalence of tropical sea-surface temperatures linked to PETM warming²¹. Based on sedimentation rates (50 cm ka^{-1})²², this volcanic flare-up lasted for 171–213 kyr, similar to the duration of the coeval PETM^{8,12}, and was followed by a sharp decline in volcanism¹⁹. The tuffs exhibit wide compositional diversity from basanites to dacites (Extended Data Figs. 1, 3a–b; Extended Data Tables 1–3), an increased range of magnesium number (Mg#; ≤ 63), and a marked shift to highly negative ϵNd signatures at ca. 56.03 Ma (Fig. 2a). This activity signals a step change in magmatic processes and volcanic unrest along the rift, as recorded across a major area of the NAIP ($>130,000 \text{ km}^2$)^{1,23}. Sampled PETM-age tuffs are enriched in Rb and Ba and depleted in Nb and Sr (Extended Data Fig. 3c; Extended Data Table 3), and most have similar compositions to some of

the lowermost (‘negative ash series’) tuffs of the Danish Basin, thought to derive from nearby volcanoes along the continental shelf²³. However, the very negative ϵNd values of some of the uppermost tuffs (Fig. 2a; Extended Data Table 2) likely reflect assimilation of ancient continental crust and suggest that they were derived from a Greenland source²⁴. Taken together, this is consistent with the observed intensification of volcanic activity across the wider NAIP during the PETM²⁵ (Fig. 1a).

Interpretation of geochemical data from the Rockall tuffs is complicated because they derive from a range of sources and have experienced varying degrees of seafloor weathering¹⁹. Hence, we next studied the coeval thick basaltic lava sequences that were emplaced subaerially near the rift axis (Fig. 1). The Palaeocene-Eocene lavas of the Faroe Islands and East Greenland (Fig. 1b) are ideally suited because, unlike the contemporaneous lavas in SE Greenland²⁴, they are minimally affected by crustal contamination^{26,27,28,29}. The base of the Milne Land Formation in East Greenland, which correlates with the base of the Middle Lava Series (also known as the Malinstindur Formation³⁰) of the Faroes^{1,26} (Fig. 2b-c; hereafter both referred to as the ‘Middle Lavas’), is dated at $56.1 \pm 0.4 \text{ Ma}$ and the duration of activity is reasonably well constrained^{1,2} (Fig. 2). Thus, the base of the Middle Lavas correlates with (or shortly predates) the intensification of volcanism at Rockall Plateau (Fig. 2a).

We estimate that the 1.25 km-thick Middle Lavas^{1,26} were emplaced over 200–300 kyr (Fig. 2b–c; Methods), yielding an average eruption rate of $\sim 4\text{--}6 \text{ m kyr}^{-1}$, but potentially an order of magnitude higher early in the eruptive cycle¹. At the base of the sequence, thick ($>100 \text{ m}$) pyroclastic deposits, also reported in Greenland²⁶, signal an early volatile-rich explosive phase²⁹. The overlying tholeiitic lavas are characterized by a sharp increase in Mg#, from values of ~ 52 throughout the Lower Lava Formation²⁶ to >80 in the basal part of the Middle Lavas (Fig. 2b-c). The basal lavas are coincident (within dating uncertainties) with the onset of the PETM⁸. Many of these lavas are highly magnesian³¹ (e.g., with MgO = 24%), implying hot liquidus temperatures, and are locally characterized by massive olivine accumulation^{26,29}. They also exhibit high TiO₂ contents (typically 1.5–2.5 wt.%), and enrichment in Light Rare Earth Elements (LREEs), e.g., $(\text{La}/\text{Yb})_n = 2\text{--}3$, and $(\text{Eu}/\text{Yb})_n > 2.5$ (Fig. 2b)^{27,32}. This unusual pulse of high-Ti magmatism, similar to that of the Jurassic Karoo flood basalts in southern Africa³³, generated the long-lived ($\sim 300 \text{ kyr}$)¹ Skaergaard layered igneous intrusion at $55.75 \pm 0.35 \text{ Ma}$ (Fig. 1a)¹. Finally, the Upper Lava Formation (Fig. 2b-c) signifies an abrupt shift to low-Ti depleted MORB-like lavas^{26,27,32} (Fig. 2d-e). Their age and diagnosis as MORB are consistent with independent geotectonic constraints for onset of seafloor production between 56–55 Ma (Fig. 1d-e). Thus, there is evidence that the early volcanism responsible for the PETM-aged high Mg# Middle Lavas was both transient¹ ($\sim 227 \text{ kyr}$) and chemically anomalous (Fig. 2).

The geochemical compositions of these lavas have been attributed to partial melting of the SCLM during asthenospheric upwelling²⁷, a model that is supported by studies of their feeder intrusions^{28,31}. To test this hypothesis further, we applied two models of mantle melting. First, we adopted a non-modal batch

melting model of a garnet-bearing lherzolitic mantle source³³ (Fig. 2d). For the second model, we estimated melt percentage based on average lherzolite mineralogy, average partition coefficients, and a plume composition based on primitive basalt (Fig. 2e) (Methods; Extended Data Table 4). Both models suggest that the Middle Lavas derived from a high degree of melting, up to ~13%, but typically in the range of 4–8%, with their high $(Sm/Yb)_n$ ratios (2–2.5) requiring ~20–>50% of garnet lherzolite in the mantle source (Fig. 2e). Metasomatism in the mantle source has already been documented in this region³⁴ and invoked to explain the compositional characteristics of these high Ti basalts, notably their LREE enrichment and variable Nb and Ta anomalies^{28,31}. The compositions of the Middle Lavas are compatible with enhanced melting of metasomatized SCLM by the Iceland plume³², immediately before and during continental breakup and therefore just prior to seafloor spreading at c. 55.8 Ma (Figs. 2b–c). This short-lived phase of elevated melting—as observed in other rifts globally (e.g., ref.³⁵)—is supported by the peak in crustal P wave velocity (7.5 km/s) across the continental-ocean transition of the northeast Atlantic margins⁶. This peak straddles the seaward side of the continental-ocean transition, and likely results from intrusion of similarly high-Mg mafic melts related to increased melting just prior to continental breakup⁶.

Trace element compositions show that the degree of melting varies systematically through the Faroes sequence (Fig. 2e), with the Lower Lava Formation (Fig. 2b–c) representing a relatively low degree of melting (2–5%; Fig. 2e). To expand the dataset, lavas from farther along the rift axis in northeast Greenland are included (Hold with Hope)³⁶ (Fig. 1b), which have an upper age of ca. 57 Ma (C25n–C24r)⁵. These lavas exhibit geochemical characteristics that have been attributed to the incorporation of SCLM during melting^{5,36}. Our model confirms a low degree of melting (0.5–2%) of a variably garnet-rich lherzolite source (Fig. 2e). Taken together, these observations confirm an up-section increase in the involvement of SCLM in melting during the late Palaeocene, peaking just before the PETM (Fig. 2e), then rapidly declining in intensity prior to the onset of seafloor spreading. There is possible evidence for this peak in Palaeocene–Eocene³⁷ basaltic lavas of the Vandfaldsdalen Formation in central east Greenland³⁸. These lavas exhibit Nd, Sr and Pb isotope signatures indicative of an SCLM component, with limited crustal contamination³⁸. Assimilation of the deep SCLM is particularly important because this zone is metasomatically enriched in carbonates, and is thus a major carbon reservoir^{39,40,41,42}.

Mobilization of lithospheric mantle carbon

Rifting significantly disrupts the >100 km thick cratonic lithosphere⁴³. Because carbonates are only stable in the mantle at high pressures (~3 GPa at a temperature of 1300°C)⁴³, significant carbon release will only occur when extending cratons are 130 km thick or more⁴³. Seismic tomography indicates that the lithosphere in central Greenland (i.e., the locus of the Iceland plume at 56 Ma³) is 180 km thick⁴⁴; firmly meeting this

criterion. Peridotite xenoliths provide evidence for metasomatism in the deep SCLM beneath Greenland³⁴, as does the occurrence of strongly alkaline and carbonatitic magmatism across the wider NAIP in the early Cenozoic⁴⁵. Therefore, breakup of the North Atlantic craton provided a combination of conditions optimal for carbon release: continental rifting of sufficiently thick cratonic lithosphere thermo-mechanically weakened by the Iceland mantle plume³ (Fig. 4a) allowed for the melting of deep, carbonate-rich domains of SCLM^{40,41,43}. This is in some ways analogous to the carbon release scenario proposed for the Siberian Traps, linked to the Permo-Triassic extinction event⁴². Thickening of such carbonate enriched SCLM is thought to occur during supercontinent formation, followed by thermal re-equilibration resulting in warming and conversion to asthenosphere⁴⁶. During continental breakup, this enriched and heated asthenosphere is then channelled into thinner parts of the continental and oceanic lithosphere, where it experiences decompression melting⁴⁶.

Both the thickness and composition of the early Cenozoic NAIP lithospheric mantle make it a viable source of large amounts of carbon from the deep, carbonated SCLM. A key aspect of our hypothesis is that the most extensive period of CO₂ outgassing (~50 Mt C yr⁻¹) occurred over a short period (~200 kyr), and most likely peaked just prior to PETM onset (Fig. 2b–c). In our model, continental rifting of thick lithosphere generates a steep lithosphere-asthenosphere boundary (Fig. 4). This promotes edge-driven convection in the asthenospheric mantle^{40,47,48} that gives rise to lateral advection of metasomatized SCLM^{39,47} into a zone of intense decompression melting that extends deep (~150 km) (Fig. 4b). Numerical models predict this transient phase of deep decompression melting to occur during the final stages of breakup, 6–7 Myr after rifting onset⁴³, which is remarkably consistent with the inferred increase in deep melting just prior to breakup⁶ (Figs. 2d–e), and its timing relative to NAIP onset⁵. Because alteration zones permeate the lowermost SCLM, it is weak and prone to thermal erosion and delamination at the step change in cratonic lithosphere thickness⁴⁰ (Fig. 4). This facilitates detachment of metasomatic domains and entrainment into the melt zone beneath the evolving rift (Fig. 4c) where pressures are lower, temperatures are higher, and metasomatic carbonates are thermodynamically unstable, releasing CO₂^{39,43}. Indeed, it has been suggested that relics of such metasomatic material that delaminated during northeast Atlantic rifting can explain residual enriched ‘EM1’ components still present in the Icelandic mantle today^{49,50,51}. The question is then whether this process can contribute enough carbon to help explain the PETM carbon isotope excursion.

Recent estimates suggest that the SCLM may contain 5–8% CO₂, or possibly more⁴¹. To quantify the potential influence of rift-related SCLM disturbance, we performed 10,000 Monte Carlo calculations using CO₂ contents ranging from 1–8%, and length-scales of the carbonated SCLM in the narrow melting zone below the rift near breakup⁴³ based on tectonic reconstructions, chemical tomography and lithospheric models (Methods; Extended Data Tables 5–6). These calculations suggest that only 4–8% of this zone needs to melt to exceed the requisite >10⁴ Gt C for PETM warming^{11,13} (Fig. 3). This is con-

sistent with the modelled 5–8% melting of a garnet-rich lherzolite during the PETM (Fig. 2d–e), and earlier estimates that the intrusive feeders of these Ti-rich basalts formed by 4%³¹ to 7%²⁸ batch melting of fertile, metasomatized lherzolites in the SCLM²⁸. In summary, our data suggest that rift evolution enabled the distinct peak in melt supply and volcanism just prior to breakup (evident in the compositional characteristics of thick lava piles; Fig. 2), with vigorous decompression melting of entrained SCLM and subsequent upward migration of CO₂ (Fig. 4). This CO₂ was delivered to the atmosphere via degassing from active volcanic sources (including subaerial flood basalts and along the rift axis), crustal magmatism, and deep extensional fault systems that define active rifts⁵² (Fig. 4d).

Northeast Atlantic volcanism and climate change

The step change in NAIP magmatic productivity² immediately prior to the PETM gave rise to widespread magmatic and volcanic unrest (Figs. 2, 4). While some volcanic tuffs at Rockall Plateau are likely linked to the Middle Lavas (especially those with high Mg#), many of these tuffs are likely genetically unrelated, but are rather a manifestation of the general increase in mantle melting, and fluxes of magma and volatiles through magmatic systems along the continental shelf and Greenland continental margin at this time²³. The Rockall sequence places firm constraints on both the timing and duration of this climate-altering phase of volcanism at ~171–213 kyr, which is similar to the duration of the lower, high Mg#, part of the Middle Lavas (Fig. 2). This surge in regional volcanic activity could explain the wide variability in the composition of tuffs belonging to the Danish negative ash series²³. Indeed, the chemical weathering of extensive tephra blankets may also explain the large decrease in ¹⁸⁷Os/¹⁸⁸Os ratios observed in sediments just prior to PETM onset at Svalbard⁵³.

Enhanced melting of the SCLM during continental breakup resulted from an unusual combination of conditions in the NAIP (Fig. 4) and provides a direct mechanism to dramatically increase the outgassing of deep carbon. Our data suggest that enhanced mantle melting, and the release of deep carbon, most likely reached peak intensity just prior to PETM onset (Fig. 2), which is consistent with other independent evidence for a massive surge of $\delta^{13}\text{C}$ depleted carbon^{11,13} (Fig. 3). Melting of deep, carbonate-rich domains of the SCLM can result in a fivefold increase in magmatic CO₂ output, providing a powerful mechanism to boost short-term CO₂ degassing, and reconciling an apparent major deficit of carbon from ‘background’ ridge and LIP magmatism (Fig. 3). This enhanced mantle melting occurred contemporaneously with methane release from hydrothermal vents related to the same phases of magmatism^{15,16,54}, which potentially supplied an isotopically lighter endmember to the PETM carbon cycle¹¹. Our study supports the proposal that large-scale lithospheric melting can induce global warming⁴², if the tectonic setting is primed to facilitate intensive magmatic CO₂ release (Fig. 4). This highlights the critical role that solid Earth degassing plays in driving abrupt hyperthermal events such as the PETM, promoting

fundamental reorganization of Earth’s surface environment and biosphere.

Online content

Any methods, additional references, Nature Research reporting summaries, source data, extended data, supplementary information, acknowledgements, peer review information; details of author contributions and competing interests; and statements of data and code availability are available at <https://doi.org/10.1038/s12345-111-2222-3>.

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Author contributions

T.G. conceived the idea, led the study, interpreted the data and prepared the manuscript and figures. T.H. performed the modelling, with input from T.G. R.N.M. assisted with tectonic and geodynamic interpretation, and G.F., J.L., R.B. and M.P. provided support with geochemical analysis and interpretation. G.F. carried out the melt modelling. A.M. calculated the seafloor production rates and provided support with *GPlates* and *pyGPlates*. D.K. contributed to tectonic interpretations. T.G. wrote the manuscript with input from all co-authors.

Competing interests:

The authors declare no competing interests.

Figure Captions

Figure 1: Early Cenozoic tectonic and magmatic evolution of the northeast Atlantic | **a**, Map of the present-day northeast Atlantic region showing the distribution of Palaeocene–Eocene lava flows and intrusives, with dated volcanics denoted by coloured symbols (modified after ref.³) and the locations of major hydrothermal vent complexes in the Vøring and

Møre Basins¹⁵ and offshore Northeast Greenland⁵⁴ shown as stars (SDRs = seaward-dipping reflectors; COB = continent-ocean boundary). **b**, Plate tectonic reconstruction showing nascent ridge systems developing along the Labrador Sea and northeast Atlantic. **c**, Ages of the volcanic sections discussed (Up=Upper; VFF=Vandfaldsdalen Fm), defined by radiometric dates^{1,2}, magnetostratigraphy and nannofossil zonation^{22,37,25}, and corresponding carbon and oxygen isotope records showing the PETM isotope excursions (solid and faint lines show 1 Myr and 20 kyr locally weighted functions, respectively)⁵⁵. **d**, Seafloor production rates for the Labrador Sea and northeast Atlantic, derived from GPlates (Methods), shown alongside the timing of Eocene hyperthermals. **e**, Palaeolongitude of Greenland⁷ indicating the onset of ocean crustal production in the Northeast Atlantic at 56 Ma..

Figure 2: Palaeocene-Eocene volcanostratigraphy and geochemistry of the proto-northeast Atlantic ridge. **a**, Simplified log of the Rockall ‘Phase 1’ sequence¹⁹ showing lithologies, Mg# (i.e., $100 \times \text{molecular MgO}/(\text{MgO} + \text{FeO})$, where FeO is assumed to be 0.9FeOT), and ϵNd (Extended Data Fig. 1). **b**, Simplified log of the Faroes Basalt Formations²⁶, with Mg# and (Eu/Yb)*n* (chondrite-normalized⁵⁶); Mg# data are from ref.²⁶ and (Eu/Yb)*n* are from refs.^{27,32} (see Extended Data Fig. 2); note the sharp transition to high Mg# (and enriched REE contents) at ca. 56.1 Ma¹, which is also observed in **c**, east Greenland (Milne Land Formation)²⁶. **d**, (La/Yb)*n* vs (Eu/Yb)*n* of the Faroes and Hold with Hope (HwH) lavas (chondrite-normalized⁵⁶) and modelled non-modal batch melting of a lherzolitic mantle source, adopted from³³, showing different degrees of melting of a garnet lherzolite (green, blue and red curves). **e**, (Sm/Yb)*n* vs (Ce/Sm)*n* and an REE melting model (Methods), showing percentage melt along the top and the relative proportions of garnet- and spinel-lherzolites from 100% gt-lherzolite (red curve) to 100% sp-lherzolite (green curve). Both models indicate that the Faroes Middle Lava Formation (i.e., high Mg# basalts in the lower 500 m of the Middle Lavas; see **b**), that erupted immediately prior to and during the PETM, experienced the highest degrees of melting of a mantle source containing $\gtrsim 10\%$ garnet.

Figure 3: Simulations of volcanic carbon release during the PETM. Results are plotted as cumulative distribution functions (CDFs). The gray lines show the estimated carbon output from ridge volcanism and LIPs alone; with S1 and S2 showing low ($0.6 \text{ km}^3 \text{ yr}^{-1}$) and high ($2.4 \text{ km}^3 \text{ yr}^{-1}$) LIP eruption rate scenarios⁴, respectively (Extended Data Fig. 4) (see Methods). The coloured lines show the effects of adding 4% to 8% carbonated (c-) SCLM melt along the incipient ridge during breakup. The gray vertical bars denote the carbon output necessary to drive and sustain PETM warming estimated by Gutjahr et al.¹¹ (labelled G; 10,200–12,200 Gt C), and Haynes & Hönisch¹³ (labelled H; 14,900 Gt C).

Figure 4: Deep carbon mobilization and release in the northeast Atlantic during the PETM. Model stages: **a**,

thermo-mechanical weakening of the sub-continental lithospheric mantle (SCLM) by the Iceland plume; **b**, thermal removal, delamination and lateral advection of metasomatized SCLM (shown as the white stippled pattern) by edge-driven convection in the asthenospheric mantle; **c**, carbonate-rich domains are entrained into the central melt zone beneath the evolving rift, where they are thermodynamically unstable and involved in transient decompression melting just prior to continental breakup. **d**, massive outgassing of CO₂ at Earth’s surface from the northeast Atlantic rift and associated volcanic and tectono-magmatic systems.

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Methods

Calculating seafloor production rates

Seafloor production (*SP*) rates (Fig. 1d), which were utilized in our volcanic CO₂ flux calculations, were calculated using the plate model of Müller et al. (2016)⁵⁷. We used the open-source python library, *pyGPlates* (<https://www.gplates.org/docs/pygplates/>) in order to filter and extract the data from the plate model. For the target time-steps (*t*, in Myr), we broke the complete mid-ocean ridge system into a series of spreading and transform segments in order to isolate the spreading segments (*j*) where new oceanic crust is formed. This approach has recently become common in analysing tectonic scenarios in ocean basins^{58,59,60,61}. At each segment, for each time-step, we extracted the full spreading rate (*u*, in km Myr⁻¹) and the length of the spreading segment (*L*, in km). The full spreading rate was calculated by summing the half spreading rate of each individual flank of each ridge segment, thereby sidestepping any issues pertaining to asymmetric spreading. We did this because, for the purposes of our analysis, we were only interested in the total amount of new seafloor generated. We then took the product of the spreading segment length and full spreading rate to calculate the seafloor production rate (in km² Myr⁻¹), and then summed the area of all segments per time-step, to obtain a total seafloor production per Myr (equation 1), as follows:

$$SP(t) = \sum_j uL \quad (1)$$

Geochemical analysis of the volcanic tuffs

We carried out major, trace and isotopic analysis (variously) on 20 tuff layers sampled from DSDP Site 555 on the Rockall Plateau. X-ray fluorescence (XRF) analyses were carried out in the Grant Institute of Earth Science at the University of Edinburgh, using procedures described in refs.^{62,63}. Major-element concentrations (Extended Data Table 1) were determined after fusion with a lithium borate flux containing La₂O₃ as a heavy absorber, using an existing method⁶⁴. Rock powder was dried at 110°C for at least 1 hour, and a precisely weighed 1 g aliquot ignited at 1100°C to determine loss on ignition (LOI). The residue was then mixed with Johnson Matthey Spectroflux 105 in a sample:flux ratio of 1:5, based on the unignited sample mass, and fused in a muffle furnace in a Pt5%Au crucible. After the initial fusion, the crucible was reweighed, and any flux weight loss was made up with extra flux. After a second fusion over a Meker burner, the molten mixture was swirled several times to ensure homogeneity, cast onto a graphite mold, and flattened with an aluminium plunger into a thin disk. The mold and plunger were maintained at a temperature of 220°C on a hotplate.

Trace-element concentrations (Extended Data Table 1) were determined on pressed-powder samples. Eight grams of rock powder were mixed thoroughly with eight drops of a 2% aqueous solution of polyvinyl alcohol. The mixture was loaded into a 40-mm diameter aluminium cup in a stainless-steel die and

compressed against a polished tungsten carbide disc in a hydraulic press at 0.6 tons cm⁻². The fused and pressed samples were analysed using a PANalytical PW 2404 automatic X-ray fluorescence spectrometer with a Rh-anode X-ray tube. Trace-element background positions were placed as close as possible to peaks, and long count times were used at both peak and background positions. Where background count rates were measured on either side of the peak, as in most trace-element determinations, the count time was divided between the two positions. Analytical conditions are given in refs.^{62,63}.

Corrections for matrix effects on the intensities of major-element lines were made using theoretical alpha coefficients calculated on-line using the PANalytical software. The coefficients were calculated to allow for the amount of extra flux replacing volatile components in the sample so that analytical totals should be 100% less than the measured LOI. Intensities of the longer wavelength trace-element lines (La, Ce, Nd, Cu, Ni, Co, Cr, V, Ba, and Sc) were corrected for matrix effects using alpha coefficients based on major-element concentrations measured at the same time on the powder samples. Matrix corrections were applied to the intensities of the other trace-element lines by using the count rate from the RhK_α Compton scatter line as an internal standard⁶⁵. Line-overlap corrections were applied using synthetic standards.

The spectrometer was calibrated against USGS and CRPG geochemical reference standards using the values given by Govindaraju (1994)⁶⁶, except that the values of Jochum et al. (1990)⁶⁷ were used for Nb and Zr in BCR-1 and BHVO-1. Excellent calibration lines were obtained using these standards. Analytical precision and accuracy are comparable to the values reported in refs.^{62,63}. Four USGS geostandards were analysed along with the samples and the data from these (Supplementary Table 1) were used to assess accuracy and precision.

Trace element analysis was performed on a representative selection of four of the tuffs using an Inductively Coupled Plasma Mass Spectrometer (ICP-MS). Samples were homogenized using a unidirectional crushing machine prior to dissolution via a three-stage, mixed acid (HF-HNO₃-HCl) closed-vessel approach. After digestion, samples were dried down prior to resuspension in 2% HNO₃ for analysis, carried out using a Thermo Scientific X-Series-2 at the University of Southampton. Alongside samples (Extended Data Table 3), blanks and International standard JA-2 (ref.⁶⁸) and Southampton internal basalt standard BRR-1 (ref.⁶⁹) were prepared and analysed: recoveries of JA-2 and BRR-1 are shown in Supplementary Table 2.

Finally, we analysed the ¹⁴³Nd/¹⁴⁴Nd isotope composition of several samples of tuff from Site 555 (Fig. 2a; Extended Data Table 2). Here, ~25 mg of freeze-dried sediment was homogenized using a pestle and mortar prior to leaching in 6M HCl for 2 hours to remove any diagenetic, alteration-related material. Samples were then digested via a mixed acid, closed vessel HNO₃-HF approach on a hotplate at 120°C. Digests were then taken to incipient dryness prior to resuspension in 2% HNO₃ for analysis. All acids were sub-boiled to ensure purity and prepared at the University of Southampton. Aliquots were then purified via first cation (AG50-X8 200-400 mesh resin) and

then reverse phase (LN Spec, Eichrom Industries) chromatography to remove isobaric interferences. All measurements were made on a Thermo Fisher Neptune Multi-Collector-ICP-MS at the University of Southampton. Mass bias corrections were performed using repeat measurements of the standard JNdi-1 (ref.⁷⁰) with a $^{143}\text{Nd}/^{144}\text{Nd}$ value of 0.512115. Nd isotope ratios are presented using epsilon notation as follows:

$$\epsilon\text{Nd} = \left(\frac{^{143}\text{Nd}/^{144}\text{Nd}_{\text{sample}}}{^{143}\text{Nd}/^{144}\text{Nd}_{\text{CHUR}}} - 1 \right) \times 10^4 \quad (2)$$

where $^{143}\text{Nd}/^{144}\text{Nd}_{\text{CHUR}}$ is the Nd isotopic composition of the chondrite uniform reservoir, with a value of 0.512638 (ref.⁷¹). Repeat analysis of two separately digested JNdi-1 samples ($n=8$, 2SD) yielded $^{143}\text{Nd}/^{144}\text{Nd}$ accuracy and precision values of 0.5121124 ± 0.000006 , corresponding to ϵNd of 10.25 ± 0.12 . This is in close agreement with literature values of 0.512115 ± 0.000007 (ref.⁷⁰). Procedural blanks were between 45 and 50 pg Nd, comprising <1% of the sample Nd in all cases. Present-day ϵNd values were corrected for radioactive ingrowth from ^{147}Sm following its formation using a radioactive decay constant for ^{147}Sm of 6.54×10^{-12} yrs, a sample age of 55 Ma, and the ratio of $^{147}\text{Sm}/^{144}\text{Nd}$ as measured in pre-column chemistry solutions.

Establishing a chronostratigraphic framework

We established a chronostratigraphic framework for Site 555 on the Rockall Plateau, building on several existing constraints at this site (Fig. 1a). The PETM as geochemically defined could not be identified at this site, because of the high proportion of volcanic and siliciclastic material (Fig. 2a), intruded at many levels by dolerite dykes and sills with emplacement temperatures likely in the range $\sim 1000^\circ$ to 1250° . Thus, the adjacent sediments likely experienced temperatures $>650^\circ$, thermal conditions which are known to seriously affect $\delta^{13}\text{C}$ composition in mudrocks and carbonates⁷². Accordingly, we rely on a combination of radiometric, magnetostratigraphic and palaeontological age constraints, in addition to well-defined sediment accumulation rate estimates. The most reliable radiometric date in this sequence, from an olivine phyric basalt near the Palaeocene-Eocene boundary, yielded a K-Ar age of 54.5 ± 2 Ma (ref.⁷³). Although uncertain, this can be narrowed further by considering the distribution of calcareous nannoplankton, including *Fasciculithus*, which disappears in the mudstones below the hyaloclastites (top NP9), with its last occurrence dated to ~ 55.6 Ma (<http://www.mikrotax.org>). This suggests that the age of the lava is most likely in the range 56.5–55.6 Ma, supporting the original position of the Palaeocene-Eocene boundary⁷⁴. Further, the overlying mudstones interbedded with the tuffs are thought to contain the dinoflagellate cyst species, *Apectodinium augustum*²⁰, characteristic of the TP5a palynozone⁷⁵. Therefore, most of the tuff layers at Site 555 (Fig. 2) were emplaced prior to, and during the PETM. We estimate the duration of this eruptive phase by calculating the thickness of strata between the bottom and upper tuffs in the phase (i.e., 691 and 605.5 mbsl, respectively), and using the sediment accumulation rates (50 cm kyr^{-1}) at these levels at Site 555²².

This gives a duration of 171 kyr, which is near-identical to that of the contemporaneous PETM interval^{8,12}. However, given that the earliest tuffs occur just below the Palaeocene-Eocene boundary, the true duration is more likely to be of the order of 213 kyr, assuming that the PETM section represents the 170 kyr interval^{8,12}. This interpretation is consistent with the occurrence of tuffs in the upper part of the sequence (Extended Data Figs. 1, 3) that are geochemically similar to those reported in the PETM recovery phase elsewhere in the NAIP²². We therefore consider a duration range of 171–213 kyr to be most reasonable for this eruptive phase. Given the above observations, we estimate the age of basal strata in this hole (~ 950 m) at 56.6 Ma, which is consistent with biostratigraphy and magnetostratigraphy²².

The base of the Milne Land Formation (Lower), which is correlated with the base of the Faeroes Middle Lava Formation^{1,26} has been well-dated at 56.1 ± 0.4 Ma. Storey et al. (2007)¹ use the relative age of the well-dated Danish Ash-17 to firmly place the PETM interval after the onset of Middle Lavas volcanism. A late Palaeocene age is also supported by palynologic studies of the underlying (non-marine) coal-bearing Prestfjall Formation, and its position near the boundary between magnetic polarity chrons c24r and c25n⁷⁶ (i.e., 57.1 Ma; ref.⁷⁷). For the Upper Lava Formation (ULF), we assumed an upper age of 55.5 Ma; this is based on an average of three radiometric ages, including one obtained from the upper ULF in the Faeroes (55.1 ± 0.5 Ma; ref.¹). We used two radiometric ages obtained from the Skaergaard intrusion (55.75 ± 0.3 Ma, and 55.65 ± 0.3 Ma; ref.⁷⁸), where the parental magma has been genetically linked to the Milne Land Formation⁷⁹, and accordingly its crystallization is considered to provide an upper age constraint for Milne Land Formation volcanism¹ (and thus, the lower Geikie Plateau Formation²⁶). Using this age model, we calculate the duration of the lowermost high Mg# part of the Middle Lavas (i.e., the lowermost 800 m; Fig. 2b) to be 227 kyr, similar to that of the volcanic phase at Rockall.

Establishing a chronostratigraphic framework for the Faroese Lower Lava Formation (correlative with the Nansen Fjord Formation of east Greenland) is more challenging because of the large uncertainties associated with radiometric ages through this succession². However, this inherent uncertainty does not impact our results in any way; data for the lower formations are shown only for reference purposes on our stratigraphic section (Fig. 2b–c). As a reasonable solution, we used magnetic polarity chrons for the Faroese sequence⁸⁰, combined with the chemical stratigraphy of Larsen et al. (1999)²⁶ and using a standard geomagnetic polarity time scale⁷⁷. We calibrated the ages at certain depths in the succession using the ages of chron boundaries as tie points and applying a linear interpolation between these known points (which assumes constant lava accumulation rates). We found that this age model was broadly consistent with existing radiometric ages (i.e., in most cases within uncertainty)².

REE melting model

A simple batch melting equation was used in the construction of Figure 2e:

$$C_L/C_0 = \frac{1}{D + F - PF} \quad (3)$$

where C_0 is the initial concentration of some element in the mantle source, C_L is its concentrations in the liquid, F is the melt fraction, D is the average distribution coefficient for the mantle phases weighted by their respective mass fractions before the onset of melting, and P is the average distribution coefficient for the mantle phases weighted by their respective contribution to the melt (see ref.⁸¹). We use a peridotite mantle source with a mineralogical composition given by McKenzie and O’Nions (1991)⁸²: (1) 57.8% olivine, 27.0% orthopyroxene, 11.9% clinopyroxene and 3.3% spinel (spinel lherzolite); and (2) 59.8% olivine, 21.1% orthopyroxene, 7.6% clinopyroxene and 11.5% garnet (garnet lherzolite). Partition coefficients (D) vary significantly with pressure, temperature and liquid composition, and this limits the reliability of partial melting models since the choice of values is always somewhat subjective. Here we have attempted to be more objective by using averages of all appropriate published D values (Extended Data Table 4). We use the following mantle melting proportions for spinel lherzolite (equation 4; ref.⁸³) and garnet lherzolite (equation 5; ref.⁸⁴):

$$\text{liquid} = -0.22 \text{ ol} + 0.38 \text{ opx} + 0.71 \text{ cpx} + 0.13 \text{ sp} \quad (4)$$

$$\text{liquid} = 0.08 \text{ ol} - 0.19 \text{ opx} + 0.81 \text{ cpx} + 0.30 \text{ gt} \quad (5)$$

where ol is olivine, opx is orthopyroxene, sp is spinel, cpx is clinopyroxene and gt is garnet.

The concentrations of REE in our mantle source (Ce 1.236 ppm, Sm 0.346 ppm, Yb 0.432 ppm) are calculated from the composition of primitive basalt from the Ontong Java Plateau and are thought to be typical of mantle plumes⁸⁵.

Please note that Extended Data Table 4 cites additional references^{86,87,88,89,90,91,92,93}.

Modelling of CO₂ fluxes

We perform a simple Monte Carlo simulation, sampling from probability distributions for six uncertain parameters (Extended Data Tables 5–6) to estimate the probable combined CO₂ emissions from incipient ridge volcanism and active large igneous provinces (LIPs). This enables us to evaluate the relative contributions to carbon release during the PETM due to ridge production and LIP formation. We also calculate the effect of incorporating carbonated SCLM melt in different amounts during enhanced melting along the ridge axis. In our calculations, we assumed a PETM duration of 170 kyr (refs.^{8,12}).

We use Beta distributions to represent uncertainty in the parameters (Extended Data Table 6). The Beta distribution is a continuous distribution over a fixed interval [0, 1], but can be rescaled to any desired range. It is defined by two shape parameters α and β , which can be estimated from the distribution mean (μ) and variance (σ^2):

$$\alpha = \mu \left(\mu \frac{(1 - \mu)}{\sigma^2} - 1 \right) \quad (6)$$

$$\beta = \alpha \left(\frac{1}{\mu} - 1 \right) \quad (7)$$

where μ = the mean, σ = standard deviation, and σ^2 = the variance. To estimate Beta distribution parameters, we use best estimates (from published data and observations) of the minimum, mean and maximum values for each variable (Extended Data Table 6, and discussed below), and apply these to equations 6 and 7 above (using re-scaled values for μ and σ). For simplicity we assume that the standard deviation for a given variable is 20% of the range, $\sigma_v = 0.2$ (max–min). The inputs to the calculations (sampled, constant and fixed) and outputs are listed in Extended Data Tables 5–6 and the corresponding histograms for each of the sampled variables are shown in Extended Data Fig. 4.

To quantify the CO₂ degassing flux from ridges in the NAIP (i.e., the northeast Atlantic and the Labrador Sea), we used seafloor production rates parameterized from plate-tectonic reconstructions (see above) for the ridges at 55 Ma—the first time step at which seafloor generation is recorded in the northeast Atlantic. In these calculations we assumed an oceanic crustal thickness of 6 km, the global average⁹⁴. Only a small fraction of the total amount of CO₂ available in the basaltic oceanic crust is degassed at ridges. To estimate this, we used the ratio between the oceanic crustal productivity (i.e., the total amount of crust formed, which is $\sim 18 \text{ km}^3 \text{ yr}^{-1}$) and the estimated CO₂ fluxes from the present-day global ridge system ($7 \times 10^{11} \text{ mol yr}^{-1}$; a middle value of ref.¹⁸). This analysis indicates that about 10% of the CO₂ available in the ocean crust is degassed at ridges (see main text), corresponding to the upper 600 m. Thus, in our simulations we used a Beta distribution with a mean value of 10% (i.e., 10% of the total available CO₂ is lost from the crust), and minimum and maximum values of 5% and 15%, respectively. For the pre-eruptive CO₂ content of basaltic magmas, we used a Beta distribution with a mean value of 0.5 wt%, and minimum and maximum values of 0.2 wt% and 2 wt%, considered reasonable for flood basalts¹⁷ and ocean crust at this time⁹⁵.

To estimate the CO₂ fluxes from LIPs, we used existing eruptive rate estimates for the NAIP⁴. We consider two LIP eruptive rate scenarios: (S1) low flood basalt productivity ($0.6 \text{ km}^3 \text{ yr}^{-1}$); and (S2) high productivity ($2.4 \text{ km}^3 \text{ yr}^{-1}$; ref.⁹). The combined fluxes from incipient ridge volcanism and each of these eruptive scenarios for LIPs are shown as the grey lines on our cumulative distribution functions (Fig. 3). We assumed near-total loss of CO₂ from flood basalt volcanism (mean 95%, minimum 90%, maximum 100% loss), based on observations from fresh basaltic glass in flood basalts that show most of the CO₂ is lost to degassing¹⁷. Ridge and LIP CO₂ outgassing was calculated using equation 8:

$$W_{CO_{\text{Ocean}}} = \frac{12}{44} \frac{t_{\text{rift}}}{10^{12}} (O_{\text{Floss}} \cdot r_{\text{basaltprod}} \cdot \rho_{\text{basalt}} \cdot B_{\text{FractCO}_2}) + (C_{\text{FractLIP}} \cdot r_{\text{LIPprod}} \cdot \rho_{\text{basalt}} \cdot B_{\text{FractCO}_2}) \quad (8)$$

where $W_{CO_{ocean}}$ = total weight of C released from the ocean crust and LIPs; t_{rift} is the duration of extension in years; O_{loss} = fraction of CO_2 lost from the ocean crust; $r_{basaltprod}$ = volumetric rate of basalt production from the NAIP ridges (see equation 1); ρ_{basalt} = density of basalt (3000 kg m^{-3}); $B_{FractCO_2}$ = CO_2 content (weight percent) of basalt, expressed as a fraction; $C_{FractLIP}$ = fraction of LIPs that is fully degassed of CO_2 ; and $r_{LIPprod}$ = volumetric rate of basalt production from LIPs. Note that a factor of 12/44 converts from mass of CO_2 to mass of C.

The combined carbon flux estimates show that outgassing from ridges and LIPs alone are insufficient to trigger PETM warming, unless they are augmented by a transient increase in CO_2 content (as we propose, via the incorporation of carbonated SCLM during mantle melting). To account for this apparent deficit, we estimated the potential influence of SCLM melting contributions, using scenarios S1 and S2 (above) as a baseline (i.e., the background volcanic carbon flux; Fig. 3). During continental extension (as in the early Cenozoic northeast Atlantic) the deep metasomatized lithospheric mantle is stretched and exhumed along the length of the rift³⁹ (Fig. 4). The lithospheric extension raises the cratonic lithosphere, exposing metastable metasomatized domains to anomalously hot asthenospheric melts beneath the rift axis. This process is expected to occur along the length of the evolving rift. We estimated the total volume of carbonated SCLM (c-SCLM) from the length of the rift segments active between 56–55 Ma (Fig. 1d), assuming a plausible range of thicknesses and melting widths for this layer (Extended Data Table 6).

The precise thickness of the c-SCLM of the North Atlantic craton at 55 Ma is poorly constrained and is likely to have been spatially heterogeneous, but is considered to lie in the range 10 km (ref.⁴³) to ~30 km thick (based on chemical tomography of cratonic regions⁹⁶), which is broadly consistent with geochemical and tectono-magmatic models of the North Atlantic craton^{97,98}. We assume a mean c-SCLM thickness of 20 km, with minimum and maximum values of 5 and 25 km. We estimated the width of SCLM involved in melting, using petrological-thermomechanical models of lithospheric rifting processes⁴³. In cases where the lithosphere is about 200 km thick, the width of the melting zone is most likely to be of the order of 25 km (ref.⁴³). We therefore used a Beta distribution with a mean width of 25 km and range of 5 to 30 km. Melting of metasomatic material occurs on both sides of the stretching lithosphere, doubling the total contribution to CO_2 output. c-SCLM is expected to have a high CO_2 content, of the order 5–8 wt%, but potentially even higher⁴¹. We therefore use a mean value of 5 wt%, minimum of 1 wt% and maximum of 10 wt%. The total contribution of SCLM melting to fluxes was estimated using equation 9:

$$W_{CSCLM} = \frac{l_{SCLM} \cdot w_{SCLM} \cdot h_{SCLM} \cdot \rho_{SCLM}}{10^{12}} \quad (9)$$

where W_{CSCLM} = total weight of carbon in the SCLM (Gt); l_{SCLM} = length of SCLM involved in melting along the northeast Atlantic rift system; w_{SCLM} = width of the SCLM zone (km); h_{SCLM} = thickness of the SCLM (km), and ρ_{SCLM} = density of SCLM (lherzolite; 3200 kg m^{-3}). A factor of $1/10^{12}$

converts weight from kg to Gt. Finally, we calculated the total weight of carbon (W_{TotalC}) produced from the ocean crust, LIPs and enhanced melting of the c-SCLM using equation 10:

$$W_{TotalC} = W_{CO_{ocean}} + F_{CSCLM} W_{CSCLM} \quad (10)$$

where F_{CSCLM} is the fraction of c-SCLM melted during extension (calculated with 0.04, 0.05 and 0.08; see Fig. 3).

Please note that additional references are cited in the Extended Data Figures and Tables^{99,100,101,102,103,104,105}.

Data availability

All data generated or analysed during this study are provided in the online version of this article (Supplementary Data File S1) and in Extended Data Tables 1–6. The map in Fig. 1b was plotted with open-source plate tectonic application software GPlates (<https://www.gplates.org/>; licensed for distribution under a GNU General Public License). Source data are provided with this paper. Any new geochemical data generated in this study are also available to download via the figshare repository, at the following DOI address:

S1: 10.6084/m9.figshare.19732948

Supplementary information

Supplementary information:
Supplementary tables 1–2.

Supplementary Data File S1:

Major and trace element compositions of volcanic tuffs from DSDP Site 555 in the northeast Atlantic (for stratigraphic context, see Fig. 2a and Extended Data Fig. 1). Note that $Mg\# = 100 \times \text{molecular MgO}/(\text{MgO} + \text{FeO})$, where FeO is assumed to be 0.9FeOT.

Supplementary Data File S2:

$^{143}\text{Nd}/^{144}\text{Nd}$ and associated ϵNd measurements of tuffs, lavas and hyaloclastites from DSDP Leg 81 Site 555. The sample ID number includes the site number (555), core box reference (e.g., 65-1), and the depth from the top of a given core (in cm). The $^{143}\text{Nd}/^{144}\text{Nd}$ ratios and associated ϵNd values are corrected to an age of 55 Ma. Also provided are published $^{143}\text{Nd}/^{144}\text{Nd}$ and associated ϵNd measurements from Site 555 lavas⁷³. Errors on discrete measurements are 2 and 1 standard error (SE).

Source data

Source Data Fig. 1:

Numerical data values as plotted on Fig. 1c–e.

Source Data Fig. 2:

Numerical data values as plotted on Fig. 2a–e.

Source Data Extended Data Fig. 2:

Numerical data values as plotted on Extended Data Fig. 2.

Code availability

More details on the computational methods and tools used for this study are available from the corresponding author (Thomas.Gernon@noc.soton.ac.uk) upon reasonable request.

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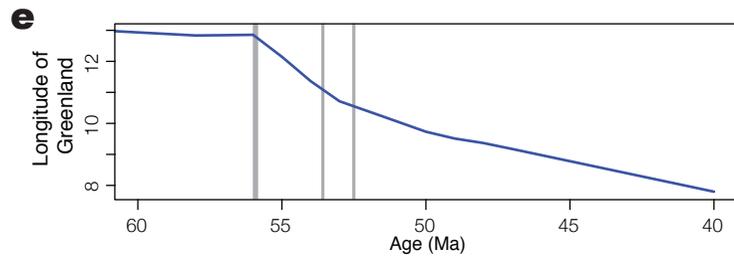
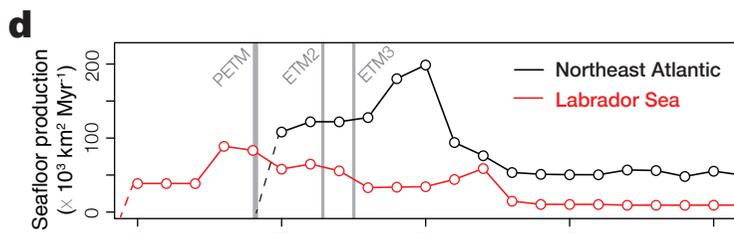
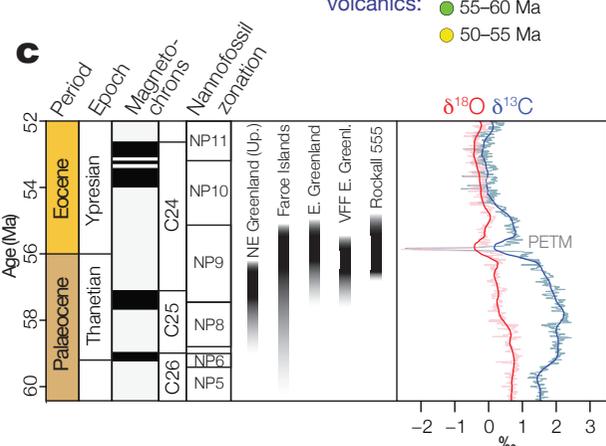
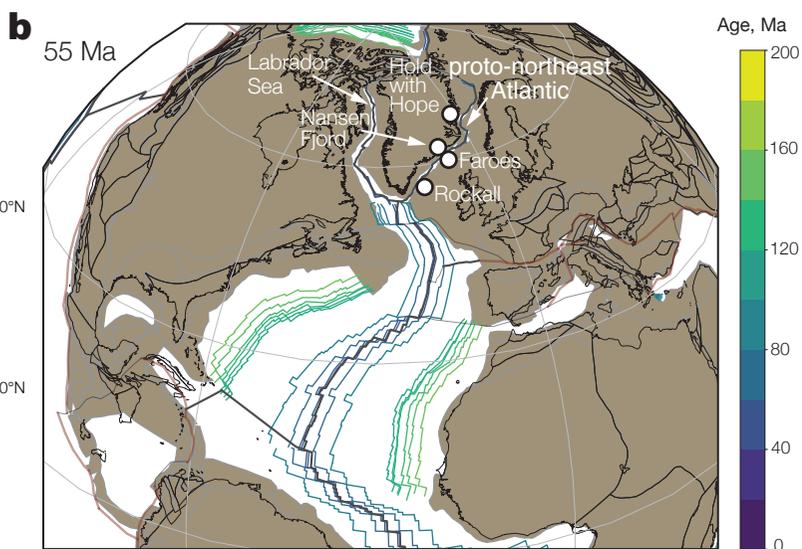
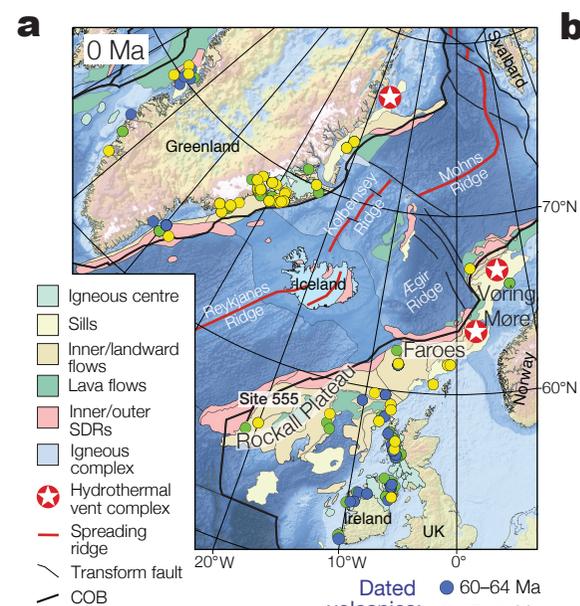
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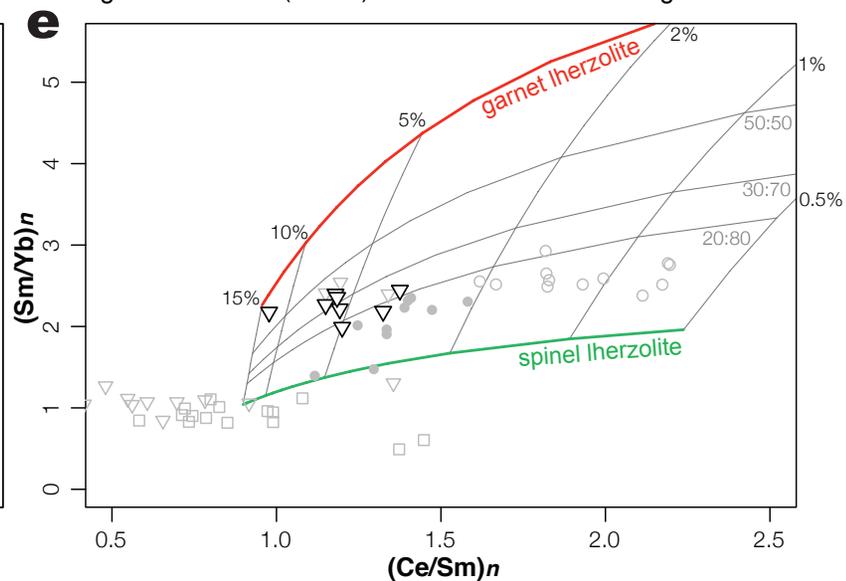
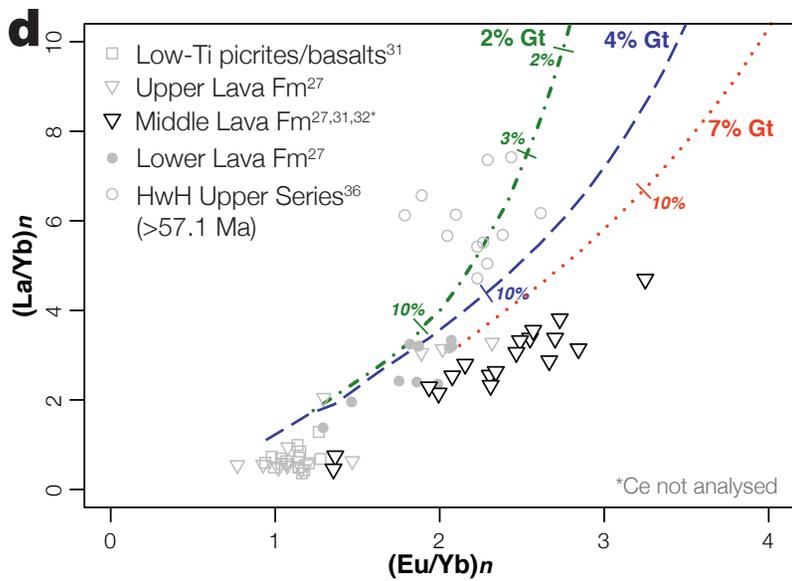
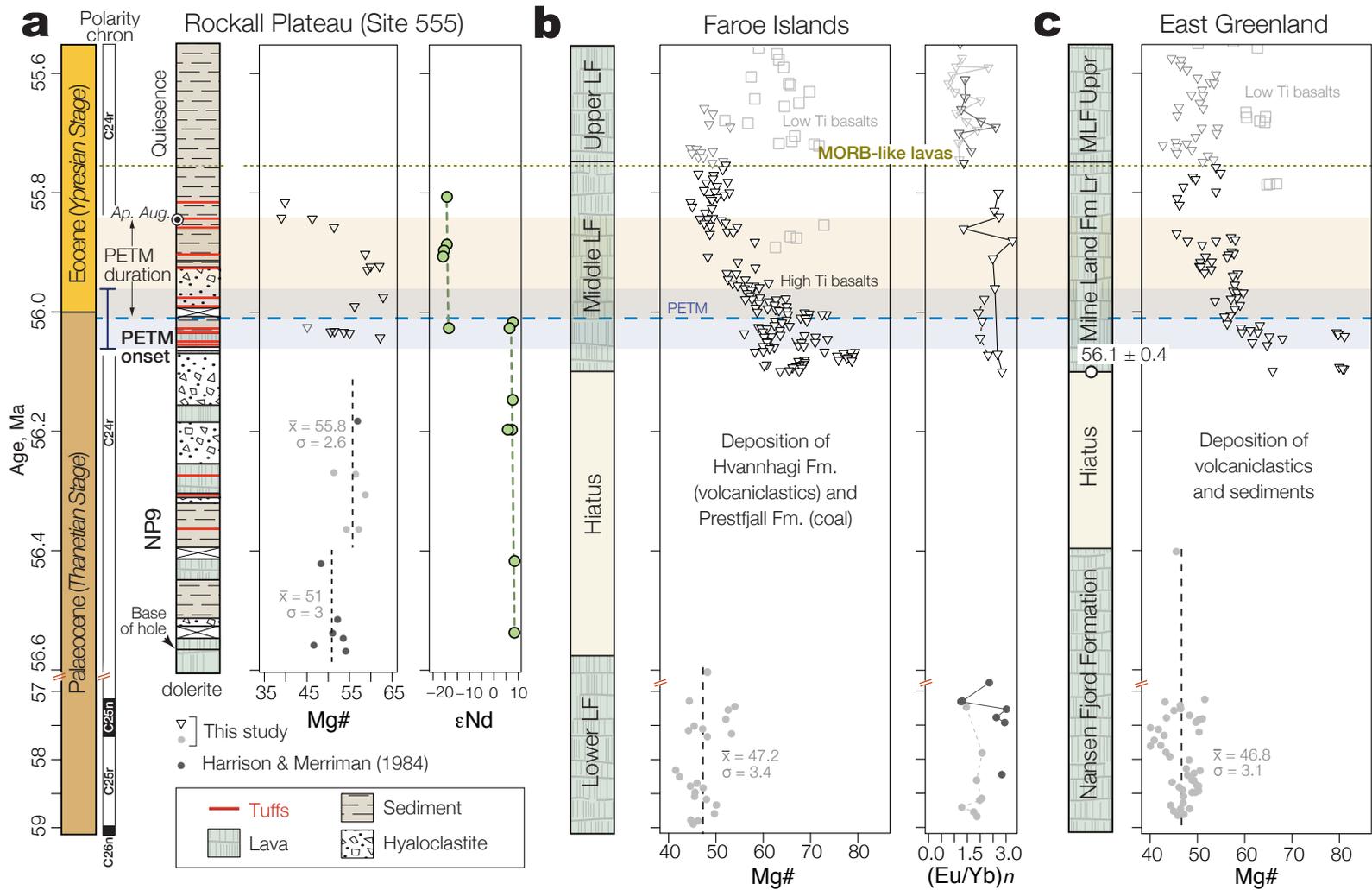
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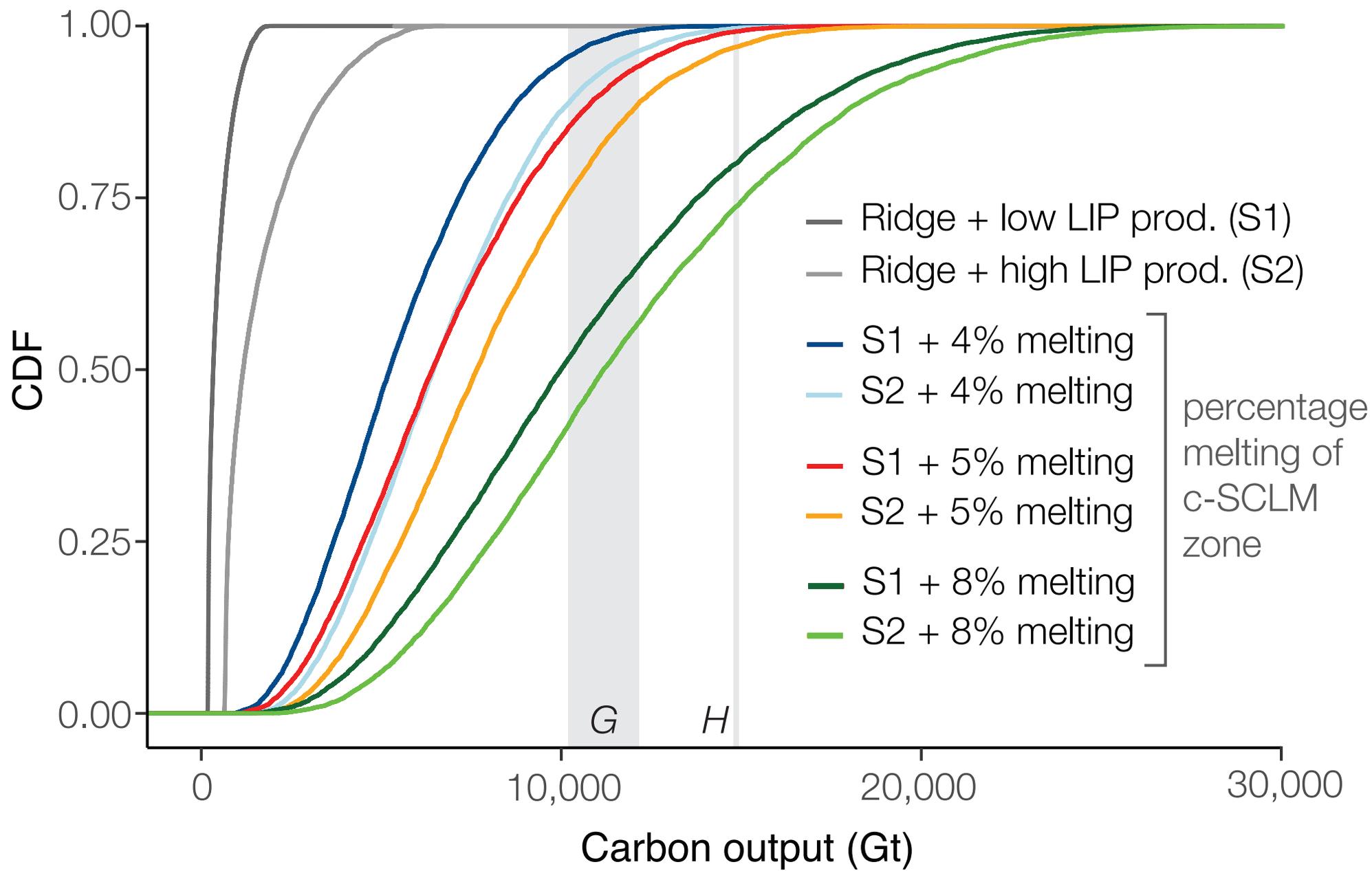
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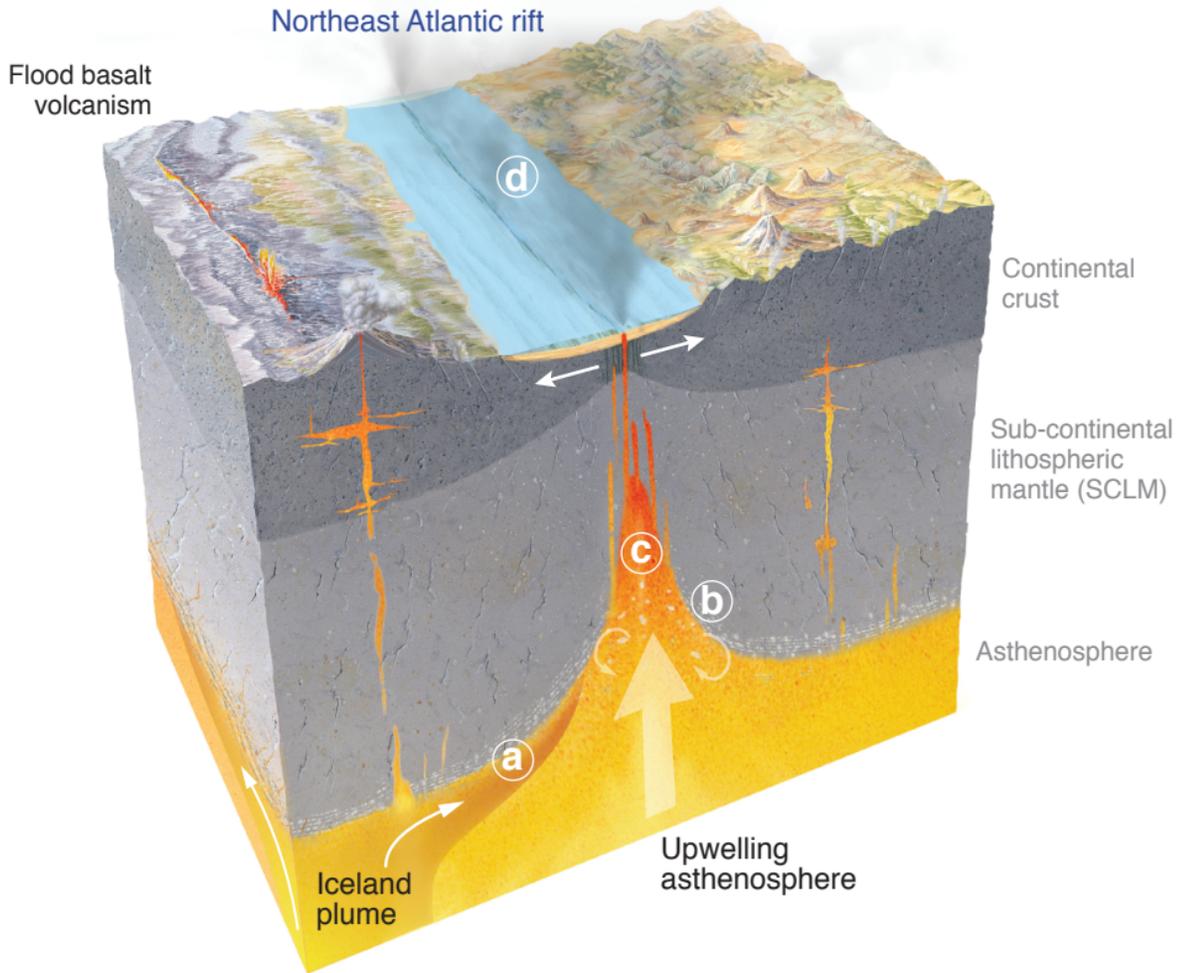
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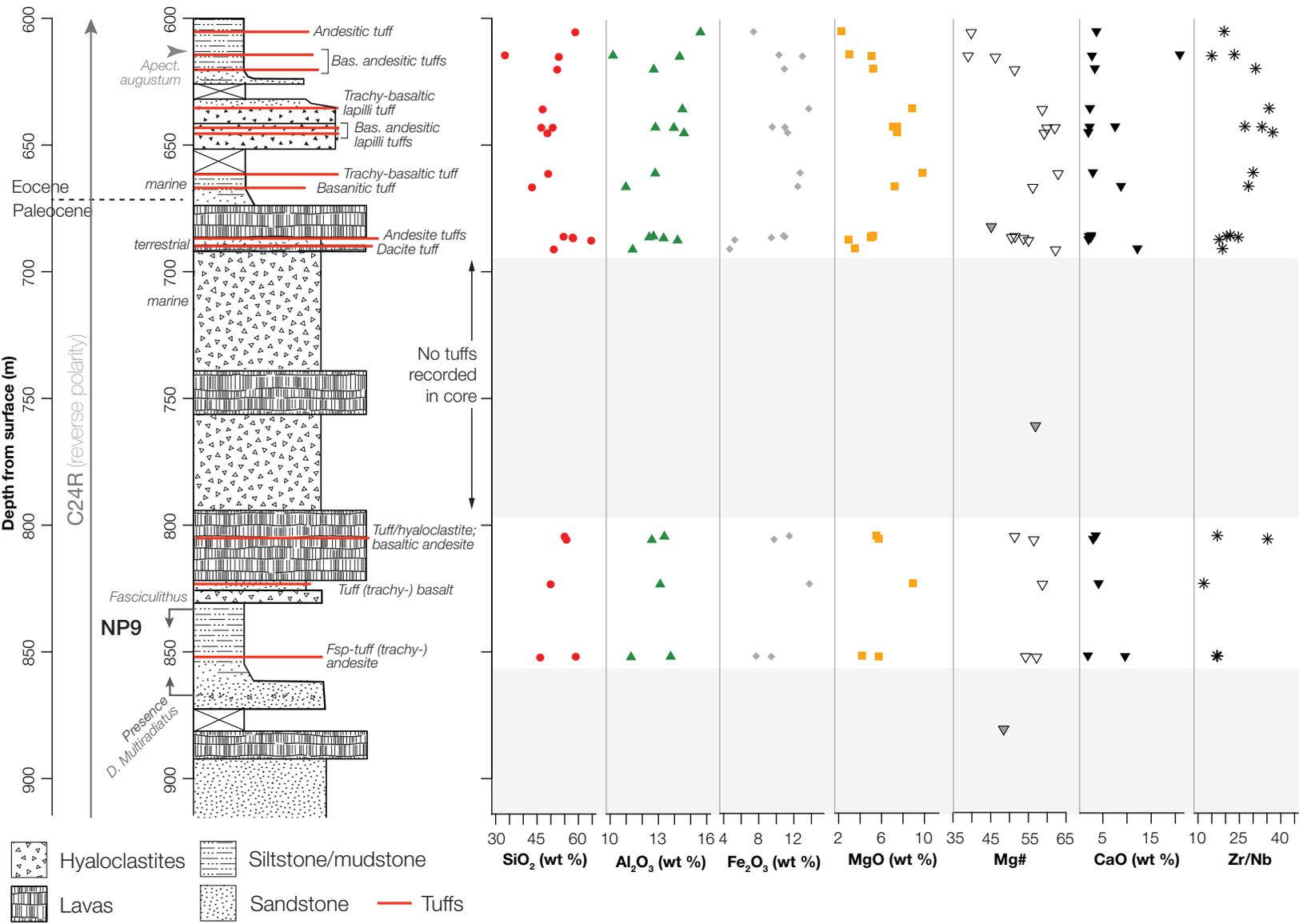
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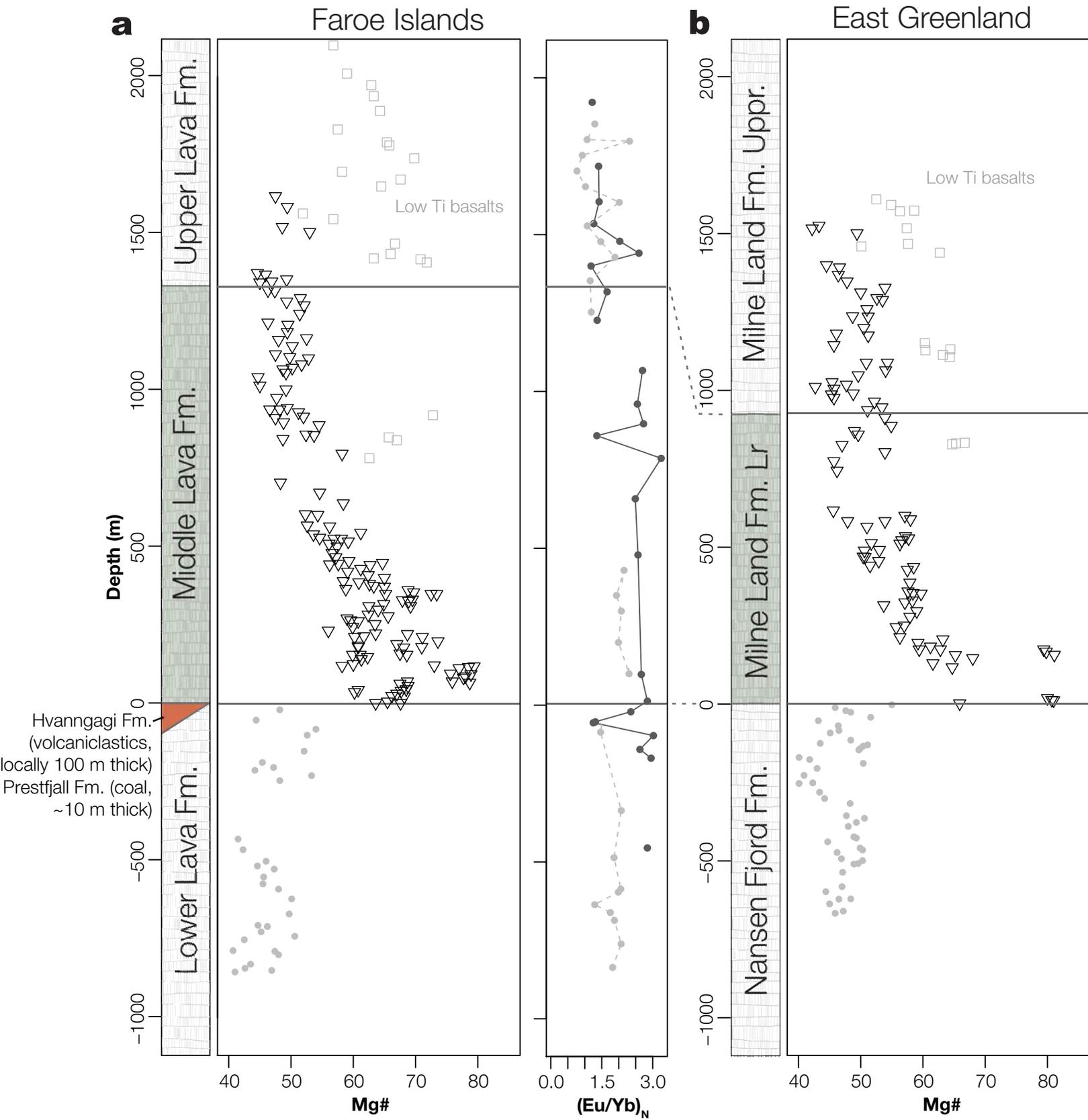


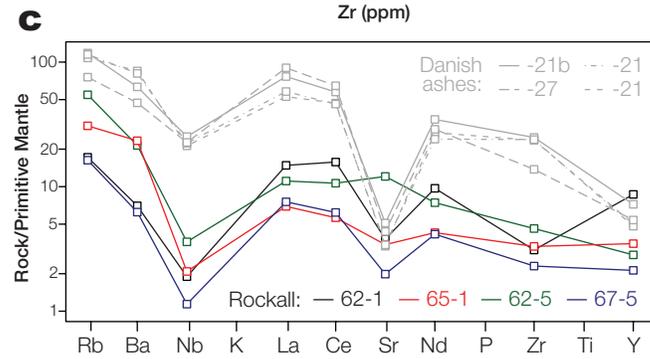
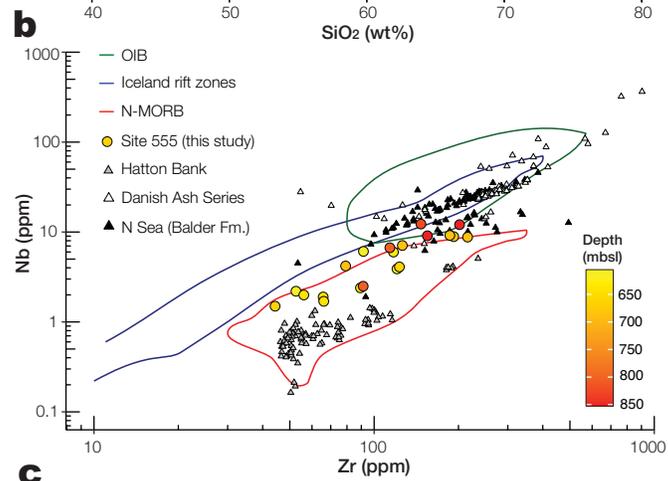
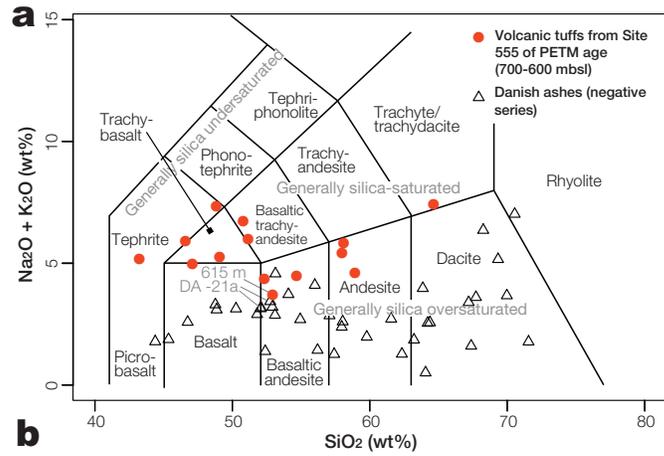


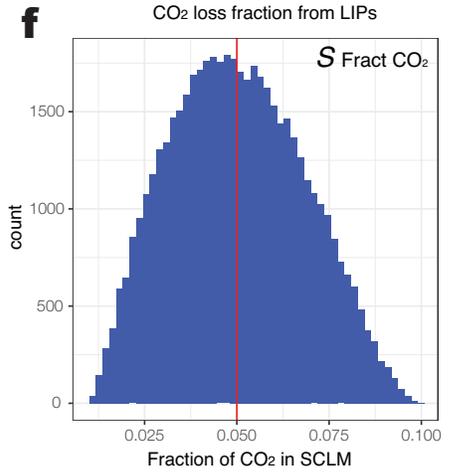
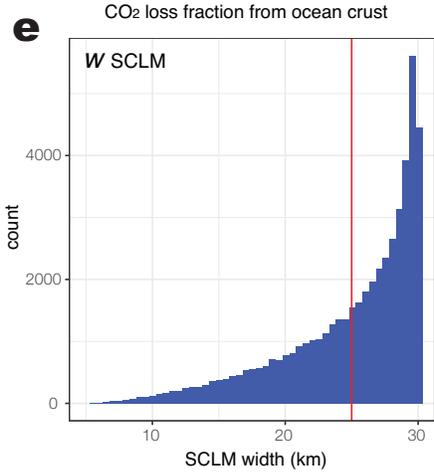
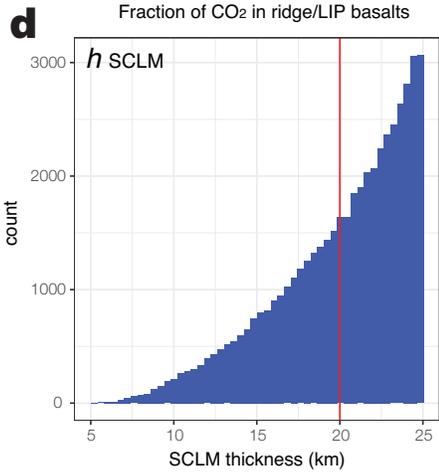
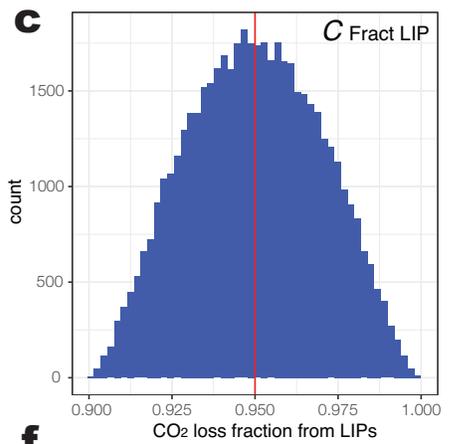
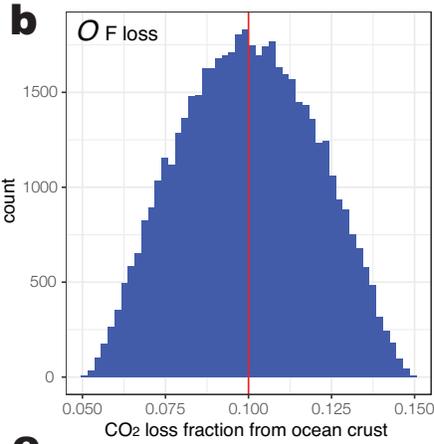
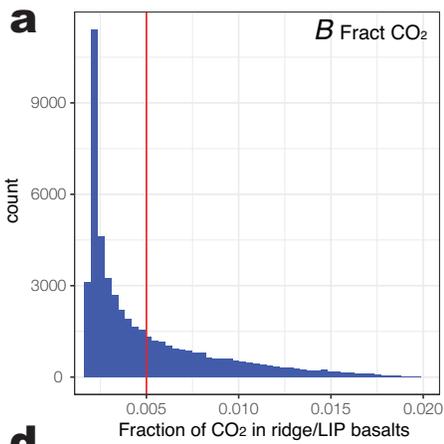












Sample name	TG1	TG2	TG3	TG4	TG5	TG6	TGT-01	TGT-02	TGT-03	TGT-04	TGT-05	TGT-06	TGT-07	TGT-08	TGT-09	TGT-10	TGT-11	TGT-12	TGT-13	TGT-14
DSDP core site	555	555	555	555	555	555	555	555	555	555	555	555	555	555	555	555	555	555	555	555
Sample depth (mbsl)	614.69	620.17	643.07	666.58	804.5	852.1	605.55	615.15	635.88	643.22	645.34	661.2	686.01	686.39	686.77	687.63	691.25	805.84	823.28	851.88
Major elements (wt.%)																				
SiO2	33.40	52.31	46.57	43.23	55.00	46.18	58.89	52.92	47.08	50.76	48.82	49.07	54.65	57.95	58.08	64.62	51.12	55.68	49.90	59.07
Al2O3	10.18	12.71	12.81	10.98	13.37	11.30	15.61	14.32	14.49	13.96	14.59	12.80	12.69	12.43	13.32	14.19	11.40	12.59	13.11	13.76
Fe2O3	10.30	10.92	9.54	12.44	11.48	9.43	7.41	12.97	13.69	10.97	11.28	12.71	10.81	10.98	9.44	5.26	4.69	9.75	13.76	7.70
MgO	2.99	5.23	7.08	7.25	5.51	5.72	2.23	5.07	8.83	7.46	7.43	9.79	5.20	5.10	4.97	2.92	3.50	5.74	8.90	4.15
CaO	20.93	3.34	7.55	8.76	3.52	9.60	3.67	2.75	2.32	2.26	2.00	2.93	2.58	2.09	2.55	2.03	12.15	2.99	4.12	1.90
Na2O	1.99	2.72	4.86	4.75	2.84	2.06	2.75	2.54	4.26	5.84	6.69	3.93	2.29	2.42	2.90	2.97	2.41	2.21	3.37	2.66
K2O	0.49	1.63	1.06	0.42	1.71	2.14	1.86	1.16	0.70	0.88	0.65	1.32	2.19	3.02	2.92	4.45	3.59	3.05	1.93	4.88
TiO2	0.83	0.84	0.80	0.67	0.89	0.86	0.87	1.06	1.11	0.97	1.03	1.05	0.99	0.83	0.86	0.50	0.32	0.97	1.29	0.82
MnO	0.99	0.04	0.13	0.36	0.04	0.18	0.04	0.06	0.15	0.12	0.12	0.19	0.06	0.05	0.05	0.03	0.20	0.13	0.11	0.07
P2O5	0.12	0.11	0.09	0.07	0.11	0.15	0.09	0.11	0.10	0.09	0.12	0.12	0.14	0.15	0.14	0.11	0.09	0.18	0.09	0.15
LOI	17.16	10.00	9.27	10.71	4.89	11.60	6.48	6.57	6.67	6.27	6.95	6.03	8.29	5.05	4.34	2.50	10.10	6.72	3.65	4.31
Total	99.38	99.84	99.76	99.65	99.37	99.23	99.91	99.53	99.41	99.58	99.67	99.94	99.88	100.06	99.58	99.59	99.57	100.01	100.23	99.45
Mg#	39.01	51.35	62.02	56.18	51.35	57.17	39.82	46.25	58.69	59.94	59.19	62.90	51.44	50.55	53.71	55.05	62.19	56.44	58.72	54.28
Trace elements by XRF (ppm)																				
Cr	188.80	201.90	192.50	222.30	103.90	170.10	267.60	160.10	212.00	207.00	203.40	206.90	192.70	194.10	243.80	242.60	138.00	110.60	207.10	188.40
Ni	86.90	56.00	69.20	66.70	43.90	62.90	54.30	65.80	88.70	74.80	78.50	81.40	56.60	48.70	48.70	26.70	28.20	62.10	43.80	64.20
Cu	155.70	102.20	133.50	125.60	54.80	77.00	106.80	163.40	175.40	158.00	170.70	158.70	80.50	55.00	72.10	29.10	22.50	169.20	56.00	80.50
Zn	112.10	109.20	79.30	76.60	71.80	112.10	104.30	130.20	90.80	86.30	90.20	118.40	120.80	117.40	105.40	62.40	46.40	82.90	113.70	127.80
Rb	13.39	40.83	20.52	11.28	61.64	57.02	46.26	38.22	15.90	18.31	15.00	26.56	56.51	94.01	75.31	101.05	78.53	25.35	110.00	74.81
Sr	120.91	326.40	103.58	60.07	164.31	120.91	310.42	194.83	112.25	91.62	80.49	129.05	333.93	246.80	394.35	378.78	261.75	95.85	176.17	130.50
Y	52.10	16.08	19.71	12.65	17.19	29.70	13.35	35.35	23.24	29.30	32.73	27.98	26.67	25.77	25.77	12.55	21.02	40.29	24.05	31.52
Zr	52.56	120.69	56.11	44.35	113.80	154.86	117.24	91.59	89.36	65.84	66.15	123.12	192.37	186.29	215.59	126.06	79.12	91.39	146.95	201.80
Nb	2.21	3.85	2.00	1.49	6.73	9.09	6.01	6.11	2.41	1.90	1.70	4.06	8.88	9.19	8.78	7.14	4.16	2.52	12.27	12.07
Ba	64.20	148.70	126.90	39.00	318.60	187.00	276.30	148.70	111.30	140.40	134.30	179.30	304.80	449.00	505.70	916.70	829.80	90.70	595.50	269.70
V	316.50	281.40	301.80	284.50	164.10	261.20	266.70	310.20	348.80	319.90	349.50	333.60	214.30	179.50	192.20	98.30	79.10	380.50	157.80	172.60
Sc	50.40	48.60	50.80	52.20	27.30	40.30	46.90	48.10	56.50	52.00	55.30	54.90	34.60	30.40	31.90	16.30	15.30	60.30	25.80	34.40
La	15.50	8.40	4.90	7.50	11.50	30.60	14.60	18.70	6.30	5.00	10.40	9.10	20.80	21.30	17.90	19.90	28.50	28.00	25.40	4.90
Ce	42.50	23.30	11.90	15.30	24.00	72.40	33.00	54.80	12.00	12.70	26.70	20.90	51.90	53.50	42.50	34.80	41.90	77.20	49.10	14.60
Nd	21.60	13.50	9.30	10.60	14.90	34.10	17.60	27.30	11.50	10.30	16.20	13.90	25.40	24.30	22.20	17.40	18.10	34.50	26.20	12.00

Sample ID	Depth (mbsl)	Sample type	$^{143}\text{Nd}/^{144}\text{Nd}$ norm	2 SE	1 SE	ϵNd	Int. error ϵNd (2s)	Corrected 55 Ma $^{143}\text{Nd}/^{144}\text{Nd}$	Corrected 55 Ma ϵNd
<i>555-65-1 122 cm</i>	643.22	Tuff	0.511593	0.000010	0.000005	-20.38	0.19	0.511547	-21.28
<i>555-69-5 84 cm</i>	686.77	Tuff	0.511695	0.000008	0.000004	-18.40	0.16	0.511650	-19.27
<i>555-64-3 38 cm</i>	635.88	Tuff	0.511612	0.000009	0.000004	-20.01	0.17	0.511562	-21.00
<i>555-61-2 4.5 cm</i>	605.55	Tuff	0.511657	0.000009	0.000005	-19.13	0.18	0.511610	-20.05
<i>555-65-4 14.5 cm</i>	645.34	Tuff	0.511572	0.000008	0.000004	-20.79	0.15	0.511527	-21.67
<i>555-78-3 81 cm</i>	769.11	Hyaloclastite	0.512939	0.000005	0.000003	5.88	0.10	0.512891	4.94
<i>555-78-3 12.5 cm</i>	768.43	Hyaloclastite	0.513026	0.000006	0.000003	7.57	0.11	0.512978	6.63
<i>Published compositions from Macintyre and Hamilton (1984)⁷³</i>									
<i>555-69-2 21 cm</i>	681.70	Lava	0.512990	-	-	7.02	-	-	-
<i>555-69-4 85 cm</i>	685.30	Lava	0.512920	-	-	5.66	-	-	-
<i>555-76-1 69 cm</i>	747.20	Lava	0.512987	-	-	6.96	-	-	-
<i>555-90-1 100 cm</i>	880.50	Lava	0.513026	-	-	7.72	-	-	-
<i>555-96-3 100 cm</i>	940.60	Lava	0.513022	-	-	7.65	-	-	-

Sample ID	555-62-1	555-62-5	555-65-1	555-67-5
Depth in core	118.5 cm	67 cm	107 cm	2.5 cm
Rb	10.32	32.79	18.47	9.79
Ba	46.30	141.40	154.20	41.22
Nb	1.25	2.37	1.37	0.75
La	9.61	7.19	4.50	4.89
Ce	26.40	17.85	9.43	10.39
Sr	75.66	240.40	68.13	39.49
Nd	12.12	9.29	5.34	5.19
Zr	32.56	48.39	34.78	24.19
Y	37.24	12.18	14.99	9.15

	Olivine				Orthopyroxene				Clinopyroxene				Garnet				Spinel		
	Mean D	1σ	n	refs.	mean D	1σ	n	refs.	mean D	1σ	n	refs.	mean D	1σ	n	refs.	D	n	ref.
Ce	0.0003	0.0002	3	82,89,90	0.0057	0.0032	8	82,89,90	0.0741	0.0383	23	82,86–92	0.0205	0.0166	11	82,87,88,90,92	0.01	1	93
Sm	0.0009	0.0005	4	82,89,90	0.0246	0.0186	8	82,89,90	0.2423	0.155	23	82,86–92	0.2934	0.2714	11	82,87,88,90,92	0.01	1	93
Yb	0.0194	0.0165	4	82,89,90	0.1201	0.0673	8	82,89,90	0.3685	0.1541	22	82,86–92	3.7042	1.4169	11	82,87,88,90,92	0.01	1	93

Variable	Description
Sampled Variables:	
$B_{FractCO_2}$	CO ₂ content (wt%) of basalt, expressed as a fraction
O_{Floss}	Fraction of CO ₂ lost from the ocean crust
$C_{FractLIP}$	Fraction of LIP that is fully degassed (of CO ₂)
h_{SCLM}	Thickness of the SCLM (km)
w_{SCLM}	Width of the SCLM melting zone (km)
$S_{FractCO_2}$	Fraction of CO ₂ in the SCLM

Constant and Fixed values:

ρ_{basalt}	Density of basalt (3000 kg/m ³)
ρ_{SCLM}	Density of the SCLM (3200 kg/m ³)
$r_{basaltprod}$	Volumetric rate of basalt production from ridges (m ³ /yr)
$r_{LIPprod}$	Volumetric rate of basalt production from LIPs (m ³ /yr)
t_{rift}	Duration of rifting (yrs)
l_{SCLM}	Length of ridge-parallel SCLM melting zone (km)
F_{CSCLM}	Fraction of C released from the SCLM (calculated with 0.04, 0.05 and 0.08)

Outputs:

W_{COcean}	Total weight of carbon released from the ocean crust (Gt)
W_{CSCLM}	Total weight of carbon in the SCLM (Gt)
W_{TotalC}	Total weight of carbon produced from the SCLM and ocean crust (Gt)

Variable	Min	Max	Mean	Var	SD	alpha	beta	Variable name in code
Fraction of CO₂ in ridge/LIP basalts	0.002	0.02	0.005	1.30E-05	0.003602244	0.412	2.060	CO2_conc_beta_samples
CO₂ loss fraction from ocean crust	0.05	0.15	0.1	0.000398429	0.019960674	2.625	2.625	CO2_loss_beta
CO₂ loss fraction from LIP basalts	0.9	1	0.95	0.000399757	0.019993932	2.625	2.625	LIP_loss_beta
Thickness of the SCLM (km)	5	25	20	15.78193019	3.972647756	2.766	0.922	SCLM_thickness
Width of the SCLM melting zone (km)	5	30	25	25.09708683	5.009699275	2.400	0.600	SCLM_width
Fraction CO₂ in the SCLM	0.01	0.1	0.05	0.000323156	0.017976531	2.299	2.874	pc