



Borylative Coupling

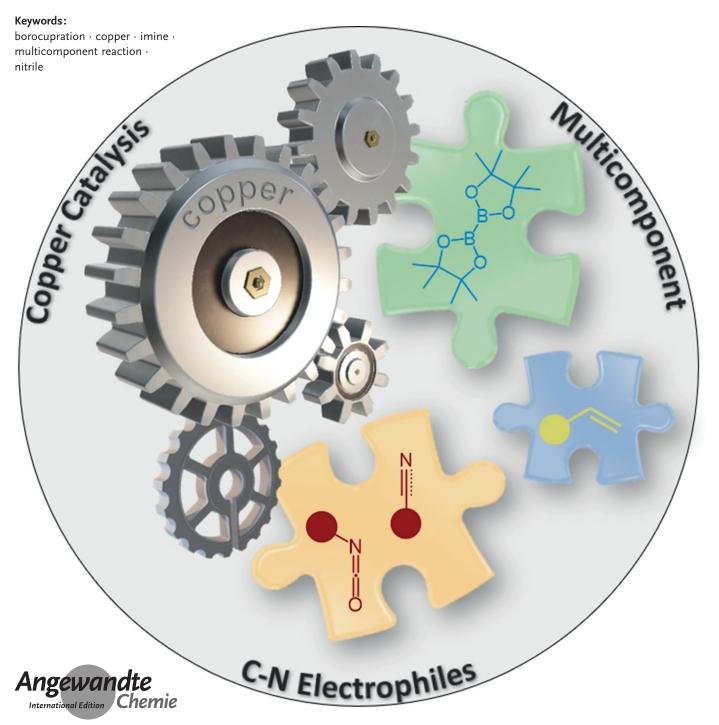
 How to cite: Angew. Chem. Int. Ed. 2020, 59, 20278 – 20289

 International Edition: doi.org/10.1002/anie.202007251

 German Edition: doi.org/10.1002/ange.202007251

Copper-Catalyzed Borylative Couplings with C-N Electrophiles

Fabien J. T. Talbot⁺, Quentin Dherbassy⁺, Srimanta Manna, Chunling Shi, Shibo Zhang, Gareth P. Howell, Gregory J. P. Perry, and David J. Procter*



Copper-catalyzed borylative multicomponent reactions (MCRs) involving olefins and C-N electrophiles are a powerful tool to rapidly build up molecular complexity. The products from these reactions contain multiple functionalities, such as amino, cyano and boronate groups, that are ubiquitous in medicinal and process chemistry programs. Copper-catalyzed MCRs are particularly attractive because they use a relatively abundant and non-toxic catalyst to selectively deliver high-value products from simple feedstocks such as olefins. In this Minireview, we explore this rapidly emerging field and survey the borylative union of allenes, dienes, styrenes and other olefins, with imines, nitriles and related C-N electrophiles.

1. Introduction

Building molecules that contain nitrogen is of great importance: amines constitute 80% of the bioactive targets used in drug discovery,[1] and the number of nitrile-containing drugs has been steadily increasing in recent decades.^[2] In addition, the versatile reactivity of nitrogen-containing functional groups make them highly useful building blocks in synthesis. Likewise, boron-containing compounds are involved in 11% of C-C bond forming reactions in process chemistry. [3] Thus, the union of nitrogen and boron-containing functional groups in defined molecular scaffolds is highly sought after. Indeed, to venture into underexplored regions of chemical space, novel disconnections of targets containing these important functionalities are needed.^[4] Multicomponent reactions (MCRs) figure amongst the most promising strategies for addressing this challenge as they transform readily available feedstocks into complex structures in a single step. [4b] In addition, due to their favorable atom and waste economies,^[5] MCRs are ideal for the synthesis of bioactive targets.[6]

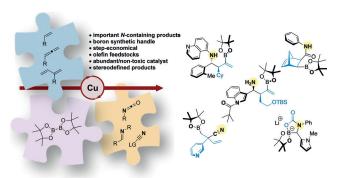
The need to replace precious metals with more abundant and less toxic elements, such as copper, is a pervading theme in contemporary synthesis.^[7] Since the first reports by Hosomi and Miyaura at the turn of the century, [8] the copper-catalyzed borylation of C-C multiple bonds, along with the powerful extension of this methodology in MCRs, has been extensively studied. [9] The aim of this Minireview is to highlight recent advances in copper-catalyzed borylative MCRs involving C-N electrophiles and olefins for the synthesis of highly functionalized amines, nitriles and other nitrogen-containing products (Scheme 1).

2. Mechanistic Aspects in Copper-Catalyzed **Borylative Couplings**

Organocopper reagents can be generated in situ through borocupration of olefins, and then intercepted by C-N electrophiles (Scheme 2A). A copper salt is first transformed into a copper(I) alkoxide complex int-1 by treatment with base.[10] Transmetallation then occurs between int-1 and a diboron reagent 1, typically via σ-bond metathesis (step

A).[11] NHCs and phosphines are popular ligands in these processes, in fact, the first isolated copper-boryl complexes int-2 featured NHCs as stabilizing ligands.[12] 1,2-Borocupration of an activated olefin 2 (internal, non-conjugated olefins are unreactive)^[13] with int-2 produces the borylated organocopper complex int-3 (step B and Scheme 2B).[14] The regioselectivity and stereoselectivity of the borocupration is not easy to predict as both kinetic and thermodynamic factors must be taken into account.[13] Furthermore, isomerization[15] and rearrange $ment^{[14]}$ can lead to epimerization $^{[16]}$ of

the organocopper. Finally, reaction with a suitable electrophile 3/4 delivers the product 5/6 (step C/D). The catalyst is regenerated using an equivalent of base and diboron, or by protonation of the product with an alcohol (step D). This simplified mechanism encompasses many of the transformations in this Minireview. As steps B and C determine the regio- and stereoselectivity of these reactions, they will be highlighted in the text when mechanistic evidence is available.



Scheme 1. Copper-catalyzed borylative multicomponent reactions involving olefins and C-N electrophiles. LG = leaving group. TBS = tbutyldimethylsilyl.

[*] F. J. T. Talbot, [*] Dr. Q. Dherbassy, [*] Dr. S. Manna, Dr. C. Shi, S. Zhang, Dr. G. J. P. Perry, Prof. Dr. D. J. Procter

Department of Chemistry, University of Manchester

Oxford Road, Manchester, M13 9PL (UK)

E-mail: David.J.Procter@manchester.ac.uk

Homepage: https://www.proctergroupresearch.com/

Dr. C. Shi

School of Material and Chemical Engineering, Xuzhou University of Technology, Xuzhou, 221018 (P. R. China)

Dr. G. P. Howell

Chemical Development Pharmaceutical Technology & Development, Operations, AstraZeneca, Macclesfield (UK)

[+] These authors contributed equally to this work

The ORCID identification number(s) for the author(s) of this article can be found under https://doi.org/10.1002/anie.202007251.

© 2020 The Authors. Published by Wiley-VCH Verlag GmbH & Co. KGaA. This is an open access article under the terms of the Creative Commons Attribution License, which permits use, distribution and reproduction in any medium, provided the original work is properly



3. Copper-Catalyzed Borylative Couplings with **Imines**

Classical syntheses of amines often involve the addition of organometallic reagents to imines.^[17] However, these reactions suffer from the inherent limitations associated with preformed organometallic reagents: cryogenic temperatures, air

and moisture sensitivity, safety risks and pre-functionalized starting materials. Copper species are well known to modulate the reactivity of organometallic reagents^[18] and can induce stereocontrol in additions to ketimines^[19] and aldimines.^[20] The use of copper catalysts to generate organocopper species in situ in MCRs is an attractive solution to the problems



Fabien Talbot received his BSc from the University of Angers (France) with one year spent at the University of Strathclyde (UK). He then undertook an MSc at the University of Strasbourg (France) and did his research project at the University of Manchester (UK) working on nickel catalysis in Prof. David Procter's research group. In 2017 he stayed in the Procter group for a PhD funded by AstraZeneca and the EPSRC. He is also the recipient of an SCI Scholarship. His research aims to develop new copper-catalyzed



Quentin Dherbassy received his MChem from the University of Strasbourg (France) in 2014 and his PhD in 2017, working in the group of Prof. Françoise Colobert (CNRS, Strasbourg, France). His doctoral studies focused on the control of axial chirality by sulfoxide-directed C-H activation. After a short post-doctoral post in the Colobert group, he joined the Procter Group as a PDRA in 2018, and is studying borylative copper-catalyzed asymmetric multicomponent reactions.



Srimanta Manna received his MSc from IIT Bombay (India) in 2012, and his PhD from the Max Planck Institute of Molecular Physiology, Dortmund (Germany) in 2017. His doctoral studies were carried out in the Antonchick group and focused on hypervalent iodine mediated C-H amination and coppercatalyzed oxidative cyclopropanation. In 2017, he joined the Procter group and was awarded a Marie Curie Fellowship in 2018. Currently, he is focusing on the development of copper-catalyzed asymmetric multicomponent coupling reactions.



Chunling Shi received her MChem from Jiangsu Normal University (P. R. China) in 2005. In 2009, she obtained her PhD from Southeast University (P. R. China). Her doctoral studies were carried out in the group of Prof. Min Ji on the organocatalytic synthesis of heterocyclic compounds. In 2009, she moved to a Lectureship at Xuzhou University of Technology and was promoted to Associate Prof. in 2013. In 2020, she joined the Procter Group as a visiting scholar.



Shibo Zhang received his undergraduate degree from Wuhan Institute of Technology (P. R. China) having carried out research in the area of green chemistry under the supervision of Prof. Zhibing Dong. In 2020, he joined the University of Manchester for his MPhil in the Procter group where he is working on borylative copper-catalyzed asymmetric multicomponent reactions.



Gareth P. Howell obtained an MChem in 2001 from the University of York including a one-year internship within Process Chemistry at GlaxoSmithKline. Focussing on asymmetric synthesis and catalysis, he obtained a PhD from the University of Nottingham with Professor James C. Anderson in 2004 before moving to Groningen (NL) to work with Professor Ben L. Feringa as a postdoctoral fellow sponsored by The Leverhulme Trust. He has worked within Process Chemistry at AstraZeneca in the UK since 2006 and is currently an Associate Principal Scientist, leading drug development projects.

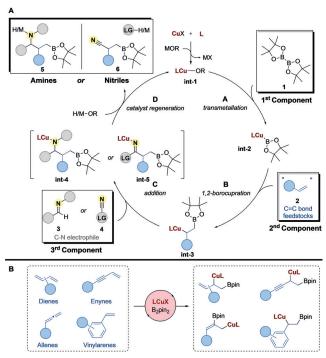
15213773, 2020, 46, Downloaded from https://onlinelibrary.wiley.com/doi/10.1002/anie.202007251 by University Of Southampton, Wiley Online Library on [21/06/2023]. See the Terms and Conditions (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles are governed by the applicable Creative Commons License



Gregory J. P. Perry obtained his PhD from the University of Manchester (UK) in 2016. His doctoral studies with Prof. Igor Larrosa focused on decarboxylative transformations. In 2017, he moved to Nagoya University (Japan) to work in the group of Prof. Kenichiro Itami, applying C-H activation in chemical biology. Since 2018, Greg has been working as a Lecturer in Organic Chemistry within the group of Prof. David Procter at the University of Manchester



David J. Procter obtained his PhD from the University of Leeds (UK) in 1995 working with Prof. Christopher Rayner on organosulfur and organoselenium chemistry. He then spent two years as a PDRA with Prof. Robert Holton at Florida State University (USA) working on the synthesis of Taxol. In 1997, he took up a Lectureship at the University of Glasgow (UK). In 2004, he moved to the University of Manchester (UK) for a Readership and was promoted to Professor in 2008.



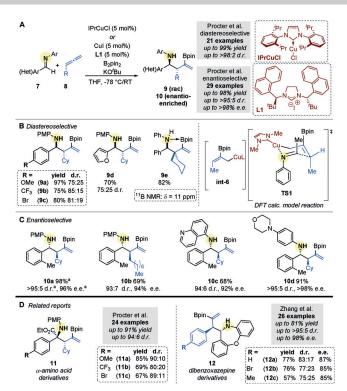
Scheme 2. General mechanism for multicomponent Cu-catalyzed borylative couplings.

associated with stoichiometric organometallic reagents in amine synthesis.

3.1. Allenes

In an effort to circumvent the need for pre-formed allylmetal reagents in additions to imines, [21] Procter and coworkers[22] reported the first multicomponent copper-catalyzed borylative coupling of imines and allenes in 2016. Homoallylic amines 9 were formed from the addition of allylcopper complexes (e.g. int-6), formed in situ by borocupration of 1-monosubstituted or 1.1-disubstituted allenes 8, to aldimines 7 (Scheme 3A). A range of substituents on the imine were well tolerated, including electron-rich and electron-deficient (hetero)aromatic groups (to give 9a-e, Scheme 3B). Interestingly, X-ray and ¹¹B NMR analysis of the products revealed a donor-acceptor interaction between the amine and the Bpin moiety (as illustrated in 9e). A density functional theory (DFT) study was performed to rationalize the observed anti-diastereoselectivity. After considering various possibilities, a lowest energy pathway featuring a 6membered, Zimmerman-Traxler-type transition-state TS1 from (Z)-allylcopper species int-6 and aldimine 7 was proposed.

Procter and co-workers^[23] also reported an enantioselective variant of the reaction. By using the enantiopure NHC precursor L1 with CuI (5 mol %), excellent levels of stereoinduction and good to excellent yields were obtained (Scheme 3C). Importantly, scalability was achieved using a low catalytic loading (1 mol %) on a 2 gram scale, affording



Scheme 3. A) Procter's Cu-catalyzed borylative coupling of aldimines and allenes. B) Diastereoselective variant. C) Enantioselective variant. [a] 2 gram scale reaction (1 mol% CuI/L1). D) Procter's and Zhang's coupling of allenes with imine derivatives.

10a in almost quantitative yield, and with excellent diastereoand enantioselectivity.

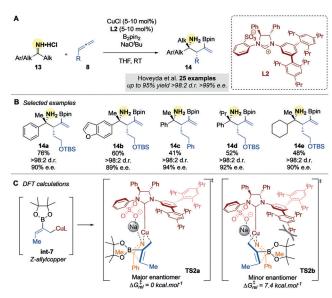
These reports triggered the development of similar MCRs involving imine derivatives. For example, the Procter group^[24a] used the approach to prepare quaternary α-amino acid derivatives 11 from ketiminoesters (Scheme 3D), for which an enantioselective variant has been reported by Chen et al. [24b] In addition, Zhang and co-workers [25] realized the enantioselective coupling of arylallenes with cyclic imines to access functionalized dibenzo-1,4-oxapines 12 (Scheme 3D).

In 2017, Hoveyda and co-workers^[26] developed an enantioselective, copper-catalyzed borylative coupling of allenes and N-H ketimines (Scheme 4A). The instability of ketimine electrophiles was cleverly managed by using the HCl salt of N-H ketimines 13, which were prepared through addition of an organolithium reagent to the corresponding nitrile and subsequent acidification. Using an enantiopure NHC ligand L2 with CuCl (5–10 mol %), the ketimine salts were combined with 1-substituted allenes 8 and B₂pin₂ to give the desired products in good to excellent yields and excellent diastereoand enantioselectivities (Scheme 4B). In agreement with the studies of Procter and co-workers (Scheme 3B), DFT calculations supported a mechanism involving a six-membered transition state TS2 (Scheme 4C). They postulated that a combination of N-Na coordination and steric repulsion between the ligand and the Bpin moiety accounts for the high enantioselectivity of the transformation.

. 46, Downloaded from https://onlinelibrary.wiley.com/doi/10.1002/anie.202007251 by University Of Southampton, Wiley Online Library on [21/06/2023]. See the Terms and Conditions (https://onlinelibrary.wiley.com/erms-and-conditions) on Wiley Online Library for rules of use; OA articles are governed by the applicable Creative Commons I



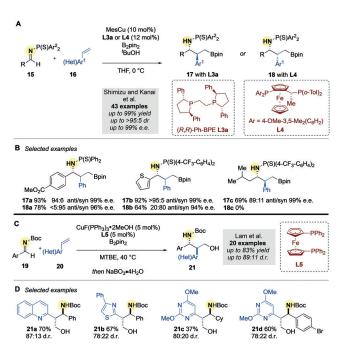
15213773, 2020, 46, Downloaded from https://onlinelibrary.wiley.com/doi/10.1002/anie.202007251 by University Of Southampton, Wiley Online Library on [2.1062023]. See the Terms and Conditions (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles as a governed by the applicable Creative Commons Information (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles as a governed by the applicable Creative Commons Information (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles as a governed by the applicable Creative Commons Information (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles as a governed by the applicable Creative Commons Information (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles as a governed by the applicable Creative Commons Information (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles as a governed by the applicable Creative Commons Information (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles as a governed by the applicable Creative Commons Information (https://onlinelibrary.wiley.com/terms-and-conditions) on the applicable Creative Commons Information (https://onlinelibrary.wiley.com/terms-and-conditions



Scheme 4. Hoveyda's Cu-catalyzed borylative couplings of N-H ketimines and allenes.

3.2. Vinylarenes

In 2018, Kanai, Shimizu and co-workers^[27] disclosed the first enantioselective copper-catalyzed borylative coupling of aldimines 15 and vinylarenes 16 using mesitylcopper (MesCu) as a pre-catalyst (Scheme 5 A). [28] Notably, they were able to selectively access either anti- or syn-diastereomeric products by varying the chiral ligand (L3a or L4). A wide range of products was obtained in high yields and high enantiomeric ratios (Scheme 5B). Provided that a large excess (10 equiv) of vinylarene was used, aliphatic imines were also suitable

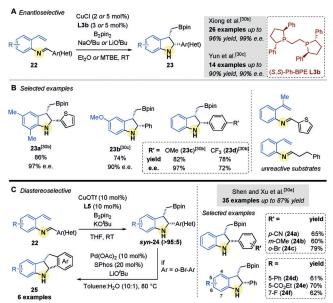


Scheme 5. A,B) Kanai's enantioselective Cu-catalyzed borylative couplings of aldimines and vinylarenes. C,D) Lam's Cu-catalyzed borylative coupling of vinylazaarenes with imines. MTBE = methyl-tert-butyl ether.

candidates in spite of their potential to tautomerize to enamines.

Prior to the report by the Kanai lab, Lam and coworkers^[29] had disclosed a related borylative coupling of arylsubstituted, N-Boc aldimines 19 with vinylazaarenes 20 (Scheme 5 C). A combination of CuF(PPh₃)₂·2MeOH and dppf L5 afforded, after in situ oxidation, 1,3-amino alcohols 21 with moderate to high diastereoselectivity (Scheme 5D).

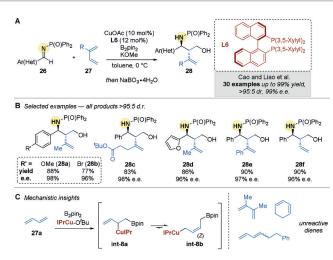
Following these reports, three groups simultaneously reported a related intramolecular coupling (Scheme 6);^[30] N-(2-Vinylphenyl) aldimines 22 were reacted with B₂pin₂ in the presence of a copper catalyst to give 2,3-disubstituted indolines 23. Both Xiong and co-workers [30b] and Yun and coworkers^[30c] reported an enantioselective coupling using (S,S)-Ph-BPE L3b (Scheme 6A). A range of (hetero)aryl substituents on the imines and aryl-substitutents on the alkene was well tolerated (Scheme 6B). Shen, Xu and co-workers[30a] reported a diastereoselective variant using dppf L5 (Scheme 6C). Interestingly, the products arising from the reaction of 2-bromo-aryl imines were further transformed into the corresponding tetrahydroindenoindoles 25 through a palladium-catalyzed Suzuki-Miyaura coupling.



Scheme 6. A,B) Xiong's and Yun's enantioselective Cu-catalyzed borylative couplings of aldimines and tethered aryl alkenes. C) Shan and Xu's diastereoselective variant.

3.3. Dienes

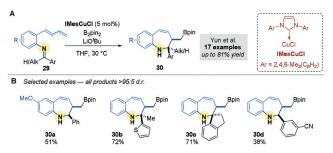
In 2016, Cao, Liao, and co-workers^[31] reported the coppercatalyzed diastereo- and enantioselective borylative coupling of imines 26 and dienes 27. Following in situ oxidation, 1,3amino alcohols 28 were obtained (Scheme 7A). Very high diastereoselectivities and excellent enantioselectivities were obtained with the bulky phosphine ligand (R)-DM-BINAP **L6** (Scheme 7). This three-component coupling reaction was effective with dienes ranging from the highly abundant 1,3butadiene (to give 28 f) and isoprene (to give 28 a, 28 b, 28 d),



Scheme 7. Cao and Liao's enantioselective Cu-catalyzed borylative couplings of aldimines and dienes.

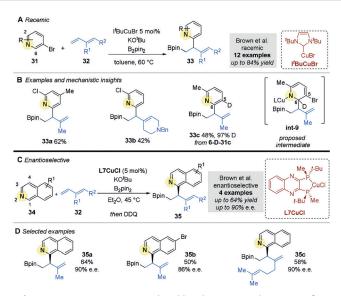
to more functionalized 2-substituted dienes (to give 28c). Unfortunately, the reactions with internal, cyclic, and 2,3substituted dienes were unsuccessful. The authors were able to isolate the (Z)-allyl copper species **int-8b**, which they suggested was formed from initial 1,2-borocupration of 1,3butadiene, followed by migration of copper to the least hindered position (Scheme 7C). This (Z) geometry is often thermodynamically favored in allylmetals. [32,33]

In 2018, Yun and co-workers^[34] reported an intramolecular copper-catalyzed borylative coupling of 1-dienylarenes **29** with tethered imines to give 7-membered benzo [b] azepines 30 (Scheme 8A). Aryl and alkyl aldimines, and aryl, alkyl ketimines, were suitable substrates and gave products with high diastereocontrol (Scheme 8B).



Scheme 8. Yun's Cu-catalyzed intramolecular borylative couplings of aldimines and dienes.

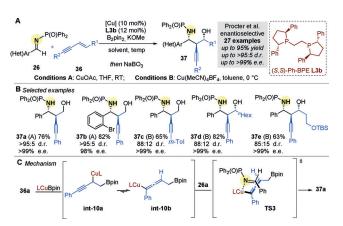
In 2018 Brown and co-workers^[35] serendipitously discovered the borylative *cine*-substitution of 3-bromopyridines 31 with dienes 32 (Scheme 9A). Labelling and crossover experiments suggested a concerted deuterium migration from the 6to the 5-position in intermediate int-9 (Scheme 9B). Quinolines and isoquinolines did not undergo cine-substitution, but direct addition after borocupration of 2-alkyl-1,3-dienes 32. For example, direct addition occurred at the 1-position of isoquinolines and afforded 1-hydroisoquinolines, which were prone to re-aromatization upon treatment with DDQ. This transformation was rendered enantioselective using L7CuCl (Scheme 9C,D).



Scheme 9. A,B) Brown's Cu-catalyzed borylative cine-substitution of 3bromopyridines. C,D) Brown's enantioselective Cu-catalyzed borylative couplings of isoquinolines and dienes.

3.4. Enynes

In 2020, Procter and co-workers^[36] used a copper-catalyzed borylative coupling of enynes 36 and aldimines 26 to prepare homopropargylic-1,3-aminoalcohols^[37] 37, containing up to three contiguous stereocenters, with high enantiocontrol (Scheme 10 A). Interestingly, (S,S)-Ph-BPE L3b



Scheme 10. Procter's enantioselective Cu-catalyzed borylative couplings of aldimines and enynes.

Schemes 5 and 6) was again used to impart enantioselectivity. A wide range of aromatic imines and electron-rich aromatic enynes were well tolerated (Scheme 10B). Based on a related DFT study involving carbonyl partners, [15] the authors suggested a mechanism involving rearrangement of the propargylic int-10a to the allenyl copper int-10b (Scheme 10C) followed by coupling via 6-membered **TS3** to give **37**.

, 46, Downloaded from https://onlinelibrary.wiley.com/doi/10.1002/anie.202007251 by University Of Southampton, Wiley Online Library on [21/06/2023]. See the Terms and Conditions (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles are governed by the applicable



5213773, 2020, 46, Downloaded from https://onlinelibrary.wiley.com/doi/10.1002/anie.202007251 by University Of Southampton, Wiley Online Library on [21:062023]. See the Terms and Conditions (https://onlinelibrary.wiley.com/rems-and-conditions) on Wiley Online Library for rules of use: OA articles are governed by the applicable Creative Commons I

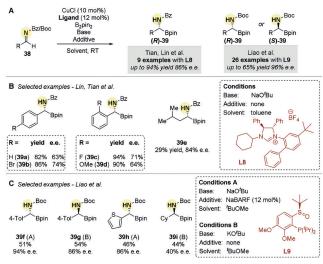
3.5. Direct Borylation of Imines

The hydroboration of imines affords important α -aminoboronic acids that are bio-isosteres of α -amino acids. [38]

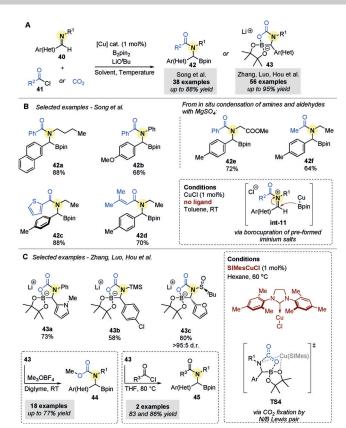
In 2013, Tian, Lin and co-workers^[39] used *N*-benzoyl arylaldimines **38** to obtain α -amido boronic esters **39** in good yields and moderate enantioselectivities (Scheme 11 A,B). In 2015, Liao and co-workers^[40] reported an improved coppercatalyzed enantioselective hydroboration of *N*-Boc aldimines **38** using a chiral sulfoxide-phosphine ligand **L9** (Scheme 11 A,C). Interestingly, the absolute configuration of the resulting α -amido boronic esters **39** could be controlled by the copper counter ion (Scheme 11 C).

The direct borylation of imines has been used in related multicomponent reactions. In 2019, Song and co-workers^[41] reported the copper-catalyzed boroacylation of aldimines **40** (Scheme 12 A). The two-step protocol involves formation of the iminium salt **int-11** then copper-catalyzed borylation to yield the desired *N*-acylated α-amino boronic esters **42**. The process was efficient across a wide range of aryl imines bearing alkyl and aryl *N*-substituents, and various aromatic and heteroaromatic acyl chlorides (Scheme 12 B). The direct conversion of aldehydes and amines through an in situ condensation/boroacylation sequence (to give **42e** and **42 f**) was also possible.

Shortly after, Zhang, Luo, Hou and co-workers^[42] reported a highly efficient copper-catalyzed borylative functionalization of aldimines involving CO₂ fixation (Scheme 12 A). Following borocupration of the aldimines, intramolecular N/B Lewis pair formation was proposed to efficiently activate CO₂ (TS4; supported by DFT calculations), yielding versatile borocarbamate salts 43. The methodology was applied to an extensive scope of aryl aldimines using only 1 mol% of a readily available NHC-ligated copper catalyst (Scheme 12 C). Finally, carbamate-containing α-amino boronic esters 44 and 45 were generated upon treatment with a methylating agent or an acyl chloride.



Scheme 11. A) Enantioselective Cu-catalyzed direct hydroboration of aldimines. B) Examples from Lin, Tian et al. C) Examples from Liao,



Scheme 12. Cu-catalyzed borylative couplings of aldimines and acyl chlorides, reported by Song and co-workers, and aldimines and CO₂, reported by Zhang, Luo, Hou and co-workers.

4. Copper-Catalyzed Borylative Couplings with Nitriles

The cyanation of olefins delivers versatile nitrile-containing products.^[2] Electrophilic cyanating agents, such as *N*-cyano-*N*-phenyl-*p*-methylbenzenesulfonamide (NCTS **46a**), have emerged as a safer and more practical alternative to traditional cyanating reagents, such as hydrogen cyanide and cyanide salts.^[43] They can be used to intercept nucleophilic organocopper species, and have recently featured as electrophilic partners in copper-catalyzed borylative MCRs.

4.1. Allenes

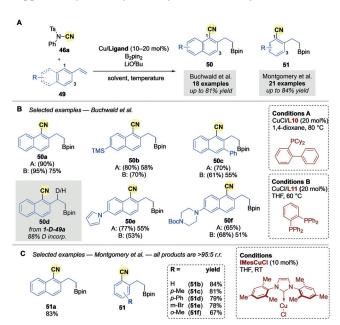
In 2016, Montgomery and co-workers^[44] reported the copper-catalyzed borylative trifunctionalization of terminal allenes **8** with the cyanating reagent NCTS **46a** (Scheme 13 A). The process demonstrated high diastereoselectivity, good functional group compatibility, and high yields (Scheme 13 B). As discussed previously (Scheme 2), initial borocupration forms an allyl copper intermediate **int-12** (Scheme 13 C). Subsequent cyanation with NCTS affords intermediate **48**. A second borocupration, followed by protodemetalation, delivers the trifunctionalized product **47** via **int-13**.

Scheme 13. Montgomery's Cu-catalyzed borylative cyanation of allenes. [a] MeOH was used as additive. [b] standard procedure used then NaOH/H2O2, overall yield shown for cyanodiboration and oxidation to the diol.

4.2. Vinylarenes

In 2014, Yang and Buchwald^[45] reported a copper-catalyzed, regioselective borylative cyanation of 2-vinylnaphthalene derivatives 49 using NCTS 46a (Scheme 14A). The reaction displayed a clear selectivity for ortho-cyanation over benzylic cyanation (observed in less than 5%), and an exclusive preference for the most hindered ortho position (C1 vs. C3). Substitution at the C6 position of the 2vinylnaphthalene derivatives was well tolerated (Scheme 14B). Interestingly, deuterium labelling and crossover experiments showed an intramolecular transfer of the C1 deuterium of labelled 1-D-49a to the benzylic position of 50d.

Shortly after, Montgomery and co-workers [46] reported the copper-catalyzed borylative cyanation of vinylarene deriva-



Scheme 14. Cu-catalyzed borylation/ortho-cyanation reported by Buchwald and by Montgomery. Yields in parentheses are crude yields.

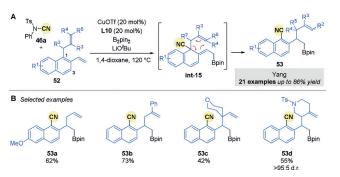
tives 49 with NCTS 46a (Scheme 14A). The same orthoselectivity was observed using the copper-NHC pre-catalyst IMesCuCl. For 2-vinylnaphthalene, cyanation at C1 was observed (51a). However, in the case of substituted styrenes, the nitrile group was directed to the least sterically hindered ortho-carbon (Scheme 14C).

The origin of regioselectivity was studied by Yang and Liu^[47] using DFT (Scheme 15). They proposed that benzylic cyanation is disfavored as a high energy 4-membered transition state **TS6** is required, whereas the *ortho*-cyanation features a more favorable six-membered TS (TS5a and TS5b). The selectivity for C1 over C3 cyanation in 2vinylnaphthalene derivatives comes from a lower disruption of the aromaticity of the naphthalene system in TS5a than in **TS5b**. Facile 1,2-elimination of the copper tosylamide from int-14 then leads to a dearomatized intermediate. Next, as previously demonstrated by Yang and Buchwald in a deuterium labeling experiment, rearomatization occurs by an intramolecular 1,3-migration of the C1 hydrogen via a sixmembered transition state TS7, followed immediately by benzylic protonation TS8.

Yang^[48] recognised the potential of the dearomatized intermediates to undergo rearomatization-driven sigmatropic rearrangement and disclosed a catalytic borylation/orthocyanation/Cope rearrangement sequence (Scheme 16A). After the initial borocupration and *ortho*-cyanation (vide supra), pre-installed allyl chains at the C1 position of 2-vinylnaphthalene derivatives 52 migrate to the benzylic position in a [3,3]-sigmatropic Cope rearrangement to re-establish aromaticity without the need for an alkoxide-assisted [1,3]-H



Scheme 15. Yang and Liu's computational study of the Cu-catalyzed regioselective borylation/ortho-cyanation of vinylarenes. M = Cu or Li.



Scheme 16. Yang's Cu-catalyzed 1,3-allyl group transfer borylation/ ortho-cyanation cascade.

, 46, Downloaded from https://onlinelibrary.wiley.com/doi/10.1002/anie.202007251 by University Of Southampton, Wiley Online Library on [21/06/2023]. See the Terms and Conditions (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles are governed by the applicable

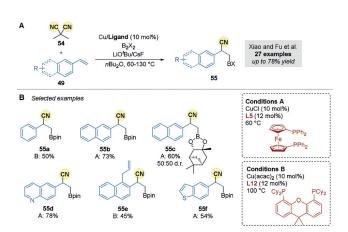


5213773, 2020,

, 46, Downloaded from https://onlinelibrary.wiley.com/doi/10.1002/anie.202007251 by University Of Southampton, Wiley Online Library on [21/06/2023]. See the Terms and Conditions (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles are governed by the applicable Creative Commons I

shift (see int-15, Scheme 16 A). Neither benzylic cyanation or functionalization at the allyl moiety were observed.

In 2018, Xiao, Fu and co-workers^[49] reported a complementary copper-catalyzed borylative benzylic cyanation of vinylarenes 49 (Scheme 17 A). The unprecedented regioselectivity was achieved by using dimethylmalononitrile (DMMN) 54, instead of NCTS derivatives 46, as the electrophilic cyanating agent. Benzylic nitriles were obtained in good yields from a wide range of vinylarenes, encompassing 2vinylnaphthalene and styrene derivatives (Scheme 17B). Notably, the presence of an allyl chain at the C1 position of 2-vinylnaphthalene did not lead to the Cope rearrangement seen by Yang.



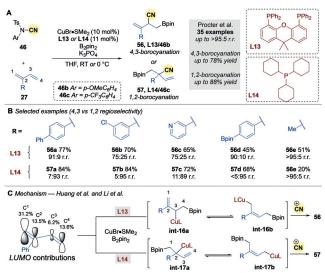
Scheme 17. Xiao and Fu's Cu-catalyzed borylation/benzylic cyanation of vinvlarenes.

4.3. Dienes

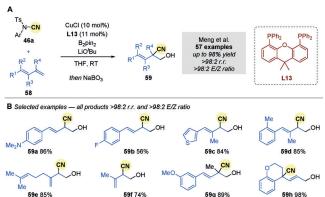
In 2018, Procter and co-workers^[50] reported the ligandcontrolled, regiodivergent borocyanation of 1,3-dienes 27 (Scheme 18A). The bidentate phosphine ligand XantPhos L13 and 4-methoxy substituted NCTS derivative 46b gave excellent regiocontrol and very good yields of the 4,3borocyanated product 56 (Scheme 18B). Whereas, switching to the monodentate phosphine ligand PCy₃ L14 and para-CF₃ NCTS derivative 46c enabled 1,2-borocyanation to give products 57 with good selectivity and excellent yields (Scheme 18B).

Shortly after, both Huang and co-workers^[51] and Li and co-workers^[52] investigated the origin of the regiodivergency through DFT calculations (Scheme 18C). They suggested that, due to the steric bulk of ligand L13, borocupration occurs across the less hindered C4-C3 double bond to give the allyl copper species int-16 a followed by rearrangement to int-16b. Subsequent cyanation gives the 4,3-borocyanated product 56. Conversely, borocupration occurs across C1-C2 with the less bulky ligand PCy3 L14 due to the greater contribution of C1 to the LUMO, ultimately leading to the 1,2-borocyanated isomer **57**.

Soon after, Meng and co-workers^[53] disclosed the coppercatalyzed cyanation of substituted 1,3-diene derivatives 58 (Scheme 19A). Excellent regioselectivity and E-stereoselec-



Scheme 18. A,B) Procter's Cu-catalyzed regiodivergent borocyanation of dienes. C) Huang's mechanistic DFT studies.



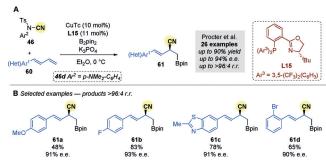
Scheme 19. Meng's Cu-catalyzed regioselective borocyanation of di-

tivity was achieved using CuCl and ligand L13 across a wide range of substrates (Scheme 19B). 4,3-Borocyanation dominated with 1-substituted, 1,2-disubstituted, and 1,3-disubstituted dienes yielding products 59 exclusively (see 59 a-g). On the other hand, 1,1-disubstituted dienes afforded products such as 59h, arising from 4,1-borocyanation.

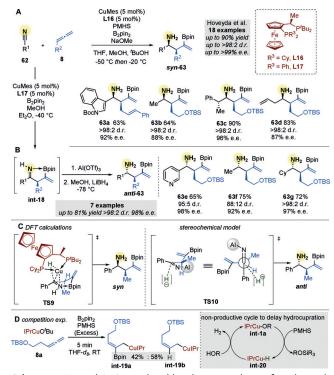
In the same year, Procter and co-workers^[54] disclosed an enantioselective 1,2-borocyanation of 1,3-dienes 60 that delivered enantiomerically enriched allylic nitriles 61 (Scheme 20 A). Substrates bearing heterocycles and substituted arenes were compatible, and gave the anticipated products with excellent regio- and enantiocontrol (Scheme 20B).

4.4. Addition Across Nitriles

In 2019, Hoveyda and co-workers^[55] presented a borylative coupling of allenes and nitriles to access primary alkylamines (Scheme 21 A). The process consists of two coppercatalyzed cycles: first, borocupration of 8 gives an allylcopper species that adds to nitriles 62, giving ketimine intermediates



Scheme 20. Procter's Cu-catalyzed regio- and enantioselective borocyanation of dienes. CuTc = copper(I) thiophene-2-carboxylate.

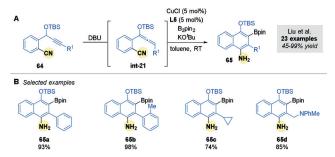


Scheme 21. Hoveyda's Cu-catalyzed borylative couplings of nitriles and allenes. PMHS = polymethylhydrosiloxane.

(c.f. int-18). In a second cycle, these intermediates are reduced to the desired amine 63 by a copper hydride species.^[56] Alkylallenes were coupled to a range of aromatic and aliphatic nitriles using L16 to afford syn-63 in good yields, with very good diastereo- and enantiocontrol (Scheme 21 A). Intramolecular N/B coordination was shown to be essential to activate the ketimine int-18 towards Cu-H reduction and to achieve excellent stereocontrol (TS9, Scheme 21 C). Through a three-step process, anti-63 products were obtained utilizing ligand L17 (Scheme 21 B). As spontaneous reduction does not occur with this system, Al(OTf)3 was added to promote decoordination of the N/B pair and allow LiBH4 reduction (**TS10**, Scheme 21 C).

Competition experiments (Scheme 21 D) showed that formation of the copper-hydride int-20, hydrocupration of the allene 8 and addition of the resulting organocopper species (e.g. int-19b) to the nitrile were all faster than the equivalent borylative process. To avoid this undesired process, Hoveyda and co-workers used an excess of the polymeric silane, polymethylhydrosiloxane (PMHS), and a finely tuned mixture of alcohols to engage the copper hydride catalyst in an unproductive cycle (Scheme 21 D).

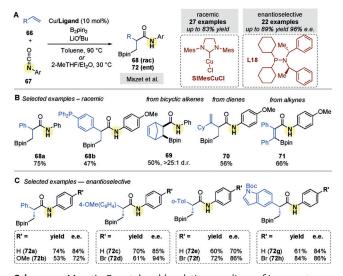
Prior to this work, Liu and co-workers^[57] had reported a related intramolecular borylative cyclization of o-(cyano)phenyl propargyl ethers 64 (Scheme 22 A). Allenes were first generated in situ by DBU isomerization (int-21). Borocupration by a copper/dppf complex followed by addition to nitriles formed ketimine intermediates that spontaneously isomerized to gain aromaticity and form naphthylamines 65 (Scheme 22B).



Scheme 22. Liu's Cu-catalyzed borylative cyclization of o-cyanophenyl propargyl ethers.

5. Copper-Catalyzed Borylative Couplings with Isocyanates

In 2020, Mazet and co-workers^[58] reported the synthesis of chiral β -borylated secondary amides 68/72 from the coupling of styrenes 66 with isocyanates 67 (Scheme 23 A). Good yields were obtained using an achiral NHC ligand (Scheme 23B). Other olefins were also evaluated as coupling partners: strained bicyclic alkenes, 1,3-dienes, and alkynes afforded 69, 70, and 71, respectively, in moderate to good yields, however,



Scheme 23. Mazet's Cu-catalyzed borylative couplings of isocyanates and olefins.

, 46, Downloaded from https://onlinelibrary.wiley.com/doi/10.1002/anie.202007251 by University Of Southampton, Wiley Online Library on [21/06/2023]. See the Terms and Conditions (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles are governed by the applicable



5213773, 2020, 46, Downloaded from https://onlinelibrary.wiley.com/doi/10.1002/anie.202007251 by University Of Southampton, Wiley Online Library on [21/06/2023]. See the Terms and Conditions (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles are governed by the applicable Cerative Commons License

substituted and heteroaromatic styrene derivatives were not effective. In addition, an enantioselective functionalization of styrenes was developed with the chiral phosphine ligand **L18** (Scheme 23 A,C).

6. Summary and Outlook

The past decade has seen significant progress made in the development of a suite of copper-catalyzed processes for the borofunctionalization of olefins using C-N electrophiles. Numerous olefins, ranging from complex polyenes to simple styrene feedstocks, have been validated as coupling partners. Similarly, various C-N inputs have been utilized. Early transformations using achiral catalysts have rapidly evolved into highly enantioselective processes and the approach now allows efficient, catalytic access to important synthetic building blocks.

A number of challenges face this nascent area of research. For example, copper-catalyzed MCRs are still limited to activated alkenes and extension of this methodology to unactivated, "simple" alkenes remains a goal for the future. Furthermore, a full understanding of the factors affecting regio- and stereoselectivity is needed and will aid our ability to predict reaction outcomes and to systematically develop new regio- and stereoselective borofunctionalizations.

Acknowledgements

We thank AstraZeneca (CASE Studentship to F.J.T.T.), EPSRC (Studentship to F.J.T.T. and Established Career Fellowship to D.J.P.; EP/M005062/1), SCI (Scholarship to F.J.T.T.), Jiangsu Oversea Visiting Scholar Program for University Prominent Young & Middle-aged Teachers and Presidents (Funding to C.S.), the Leverhulme Trust (PDRA to Q.D.; RPG-2016-360), the European Union Horizon 2020 (Fellowship to S.M.; Marie Sklodowska-Curie grant 798846-CuCAN) and the University of Manchester (Lectureship to G.J.P.P.).

Conflict of interest

The authors declare no conflict of interest.

- [1] S. D. Roughley, A. M. Jordan, J. Med. Chem. 2011, 54, 3451–3479.
- [2] a) F. F. Fleming, Nat. Prod. Rep. 1999, 16, 597-606; b) F. F. Fleming, L. Yao, P. C. Ravikumar, L. Funk, B. C. Shook, J. Med. Chem. 2010, 53, 7902-7917.
- [3] a) J. S. Carey, D. Laffan, C. Thomson, M. T. Williams, Org. Biomol. Chem. 2006, 4, 2337–2347; b) J. Magano, J. R. Dunetz, Chem. Rev. 2011, 111, 2177–2250.
- [4] a) D. G. Brown, J. Boström, J. Med. Chem. 2016, 59, 4443 4458;
 b) J. Boström, D. G. Brown, R. J. Young, G. M. Keserü, Nat. Rev. Drug Discov. 2018, 17, 709 727.
- [5] R. C. Cioc, E. Ruijter, R. V. A. Orru, Green Chem. 2014, 16, 2958–2975.

- [6] a) A. Dömling, W. Wang, K. Wang, Chem. Rev. 2012, 112, 3083–3135; b) T. Zarganes-Tzitzikas, A. Dömling, Org. Chem. Front. 2014, 1, 834–837; c) S. Javanbakht, A. Shaabani, ACS Appl. Bio Mater. 2020, 3, 156–174; d) D. Insuasty, J. Castillo, D. Becerra, H. Rojas, R. Abonia, Molecules 2020, 25, 505.
- [7] a) Cu is 4000 times more abundant in the Earth's crust than Pd, see: W. M. Haynes, in CRC Handbook of Chemistry and Physics. Section 14: Geophysics, Astronomy, and Acoustics, 97th ed. (Ed.: W. M. Haynes), CRC Press LLC, 2016, p. 17; b) Cu is considered less toxic in comparison to other transition metals, see pages 5/6 in: https://www.ema.europa.eu/en/documents/scientific-guideline/international-conference-harmonisation-technical-requirements-registration-pharmaceuticals-human-use en-32.
- [8] a) H. Ito, H. Yamanaka, J.-i. Tateiwa, A. Hosomi, *Tetrahedron Lett.* 2000, 41, 6821–6825; b) K. Takahashi, T. Ishiyama, N. Miyaura, *J. Organomet. Chem.* 2001, 625, 47–53.
- [9] a) Y. Shimizu, M. Kanai, Tetrahedron Lett. 2014, 55, 3727 3737;
 b) T. Fujihara, K. Semba, J. Terao, Y. Tsuji, Catal. Sci. Technol. 2014, 4, 1699 1709;
 c) K. Semba, T. Fujihara, J. Terao, Y. Tsuji, Tetrahedron 2015, 71, 2183 2197;
 d) Y. Tsuji, T. Fujihara, Chem. Rec. 2016, 16, 2294 2313;
 e) E. C. Neeve, S. J. Geier, I. A. I. Mkhalid, S. A. Westcott, T. B. Marder, Chem. Rev. 2016, 116, 9091 9161;
 f) A. P. Pulis, K. Yeung, D. J. Procter, Chem. Sci. 2017, 8, 5240 5247;
 g) D. Hemming, R. Fritzemeier, S. A. Westcott, W. L. Santos, P. G. Steel, Chem. Soc. Rev. 2018, 47, 7477 7494;
 h) G. J. P. Perry, T. Jia, D. J. Procter, ACS Catal. 2020, 10, 1485 1499.
- [10] T. Tsuda, T. Hashimoto, T. Saegusa, J. Am. Chem. Soc. 1972, 94, 658-659
- [11] C. Kleeberg, L. Dang, Z. Lin, T. B. Marder, Angew. Chem. Int. Ed. 2009, 48, 5350-5354; Angew. Chem. 2009, 121, 5454-5458.
- [12] D. S. Laitar, P. Müller, J. P. Sadighi, J. Am. Chem. Soc. 2005, 127, 17196–17197.
- [13] L. Dang, H. Zhao, Z. Lin, T. B. Marder, Organometallics 2007, 26, 2824–2832.
- [14] D. S. Laitar, E. Y. Tsui, J. P. Sadighi, Organometallics 2006, 25, 2405–2408.
- [15] F. Meng, F. Haeffner, A. H. Hoveyda, J. Am. Chem. Soc. 2014, 136, 11304–11307.
- [16] J. Lee, S. Radomkit, S. Torker, J. del Pozo, A. H. Hoveyda, *Nat. Chem.* 2018, 10, 99 108.
- [17] a) R. Bloch, Chem. Rev. 1998, 98, 1407-1438; b) S. Kobayashi,
 H. Ishitani, Chem. Rev. 1999, 99, 1069-1094.
- [18] S. Yamasaki, K. Fujii, R. Wada, M. Kanai, M. Shibasaki, J. Am. Chem. Soc. 2002, 124, 6536–6537.
- [19] R. Wada, T. Shibuguchi, S. Makino, K. Oisaki, M. Kanai, M. Shibasaki, J. Am. Chem. Soc. 2006, 128, 7687 7691.
- [20] E. M. Vieira, M. L. Snapper, A. H. Hoveyda, J. Am. Chem. Soc. 2011, 133, 3332–3335.
- [21] Y. Yamamoto, N. Asao, Chem. Rev. 1993, 93, 2207-2293.
- [22] J. Rae, K. Yeung, J. J. W. McDouall, D. J. Procter, Angew. Chem. Int. Ed. 2016, 55, 1102–1107; Angew. Chem. 2016, 128, 1114–1119.
- [23] K. Yeung, R. E. Ruscoe, J. Rae, A. P. Pulis, D. J. Procter, Angew. Chem. Int. Ed. 2016, 55, 11912–11916; Angew. Chem. 2016, 128, 12091–12095.
- [24] a) K. Yeung, F. J. T. Talbot, G. P. Howell, A. P. Pulis, D. J. Procter, ACS Catal. 2019, 9, 1655–1661; b) C.-Y. Zhao, H. Zheng, D.-W. Ji, X.-T. Min, Y.-C. Hu, Q.-A. Chen, Cell Rep. Phys. Sci. 2020, 1, 100067.
- [25] H. Deng, Z. Meng, S. Wang, Z. Zhang, Y. Zhang, Y. Shangguan, F. Yang, D. Yuan, H. Guo, C. Zhang, Adv. Synth. Catal. 2019, 361, 3582 – 3587.
- [26] H. Jang, F. Romiti, S. Torker, A. H. Hoveyda, Nat. Chem. 2017, 9, 1269 – 1275.

Minireviews





- [27] T. Itoh, Y. Kanzaki, Y. Shimizu, M. Kanai, Angew. Chem. Int. Ed. **2018**, 57, 8265 – 8269; Angew. Chem. **2018**, 130, 8397 – 8401.
- [28] M. Stollenz, F. Meyer, Organometallics 2012, 31, 7708-7727.
- [29] J. J. Smith, D. Best, H. W. Lam, Chem. Commun. 2016, 52, 3770-3772.
- [30] a) H.-M. Wang, H. Zhou, Q.-S. Xu, T.-S. Liu, C.-L. Zhuang, M.-H. Shen, H.-D. Xu, Org. Lett. 2018, 20, 1777 – 1780; b) G. Zhang, A. Cang, Y. Wang, Y. Li, G. Xu, Q. Zhang, T. Xiong, Q. Zhang, Org. Lett. 2018, 20, 1798-1801; c) D. Li, J. Kim, J. W. Yang, J. Yun, Chem. Asian J. 2018, 13, 2365-2368.
- [31] L. Jiang, P. Cao, M. Wang, B. Chen, B. Wang, J. Liao, Angew. Chem. Int. Ed. 2016, 55, 13854-13858; Angew. Chem. 2016, 128, 14058-14062.
- [32] a) R. B. Bates, W. A. Beavers, J. Am. Chem. Soc. 1974, 96, 5001 5002; b) P. v. R. Schleyer, J. Kaneti, W. Yun-Dong, J. Chandrasekhar, J. Organomet. Chem. 1992, 426, 143-157.
- [33] K. B. Smith, M. K. Brown, J. Am. Chem. Soc. 2017, 139, 7721-7724.
- [34] D. Li, Y. Park, J. Yun, Org. Lett. 2018, 20, 7526-7529.
- [35] K. B. Smith, Y. Huang, M. K. Brown, Angew. Chem. Int. Ed. **2018**, 57, 6146–6149; Angew. Chem. **2018**, 130, 6254–6257.
- [36] S. Manna, Q. Dherbassy, G. J. P. Perry, D. J. Procter, Angew. Chem. Int. Ed. 2020, 59, 4879-4882; Angew. Chem. 2020, 132, 4909 - 4912
- [37] M. Yus, J. C. González-Gómez, F. Foubelo, Chem. Rev. 2013, 113, 5595 - 5698.
- [38] P. Andrés, G. Ballano, M. I. Calaza, C. Cativiela, Chem. Soc. Rev. **2016**, 45, 2291 - 2307.
- [39] S.-S. Zhang, Y.-S. Zhao, P. Tian, G.-Q. Lin, Synlett 2013, 24, 437 –
- [40] D. Wang, P. Cao, B. Wang, T. Jia, Y. Lou, M. Wang, J. Liao, Org. Lett. 2015, 17, 2420-2423.
- [41] Q. Xia, H.-R. Chang, J. Li, J.-Y. Wang, Y.-Q. Peng, G.-H. Song, J. Org. Chem. 2020, 85, 2716-2724.
- [42] Z. Li, L. Zhang, M. Nishiura, G. Luo, Y. Luo, Z. Hou, J. Am. Chem. Soc. 2020, 142, 1966–1974.

- [43] a) J. Cui, J. Song, O. Liu, H. Liu, Y. Dong, Chem. Asian J. 2018, 13, 482-495; b) M. R. R. Prabhath, L. Williams, S. V. Bhat, P. Sharma, Molecules 2017, 22, 615.
- [44] W. Zhao, J. Montgomery, J. Am. Chem. Soc. 2016, 138, 9763-9766.
- Y. Yang, S. L. Buchwald, Angew. Chem. Int. Ed. 2014, 53, 8677 8681; Angew. Chem. 2014, 126, 8821-8825.
- [46] W. Zhao, J. Montgomery, Angew. Chem. Int. Ed. 2015, 54, 12683-12686; Angew. Chem. 2015, 127, 12874-12877.
- [47] Y. Yang, P. Liu, ACS Catal. 2015, 5, 2944 2951.
- [48] Y. Yang, Angew. Chem. Int. Ed. 2016, 55, 345-349; Angew. Chem. 2016, 128, 353-357.
- [49] S.-J. He, B. Wang, X. Lu, T.-J. Gong, Y.-N. Yang, X. X. Wang, Y. Wang, B. Xiao, Y. Fu, Org. Lett. 2018, 20, 5208-5212.
- [50] T. Jia, Q. He, R. E. Ruscoe, A. P. Pulis, D. J. Procter, Angew. Chem. Int. Ed. 2018, 57, 11305-11309; Angew. Chem. 2018, 130, 11475 - 11479.
- [51] X. Li, H. Wu, Z. Wu, G. Huang, J. Org. Chem. 2019, 84, 5514-5523.
- [52] X. Guo, T. Wang, Y. Zheng, L. Zhou, J. Li, J. Organomet. Chem. **2019**, *904*, 121014.
- [53] L. Wen, H. Zhang, J. Wang, F. Meng, Chem. Commun. 2018, 54, 12832 - 12835.
- [54] T. Jia, M. J. Smith, A. P. Pulis, G. J. P. Perry, D. J. Procter, ACS Catal. 2019, 9, 6744-6750.
- [55] S. Zhang, J. d. Pozo, F. Romiti, Y. Mu, S. Torker, A. H. Hoveyda, Science 2019, 364, 45-51.
- [56] B. H. Lipshutz, H. Shimizu, Angew. Chem. Int. Ed. 2004, 43, 2228-2230; Angew. Chem. 2004, 116, 2278-2280.
- [57] M. Xiong, X. Xie, Y. Liu, Org. Lett. 2017, 19, 3398–3401.
- [58] D. Fiorito, Y. Liu, C. Besnard, C. Mazet, J. Am. Chem. Soc. 2020, 142, 623 - 632.

Manuscript received: May 19, 2020 Accepted manuscript online: June 16, 2020 Version of record online: July 28, 2020

, 46, Downloaded from https://onlinelibrary.wiley.com/doi/10.1002/anie.202007251 by University Of Southampton, Wiley Online Library on [21/06/2023]. See the Terms and Conditions (https://onlinelibrary.wiley.com/terms-and-conditions) on Wiley Online Library for rules of use; OA articles are governed by the