

Ashia McManus<sup>a</sup>, Robert Raja<sup>a</sup>, Dominic Hudson<sup>b</sup>

[a] School of Chemistry, University of Southampton, SO17 1BJ (UK). [b] School of Engineering, University of Southampton, SO17 1BJ, (UK)

## Introduction

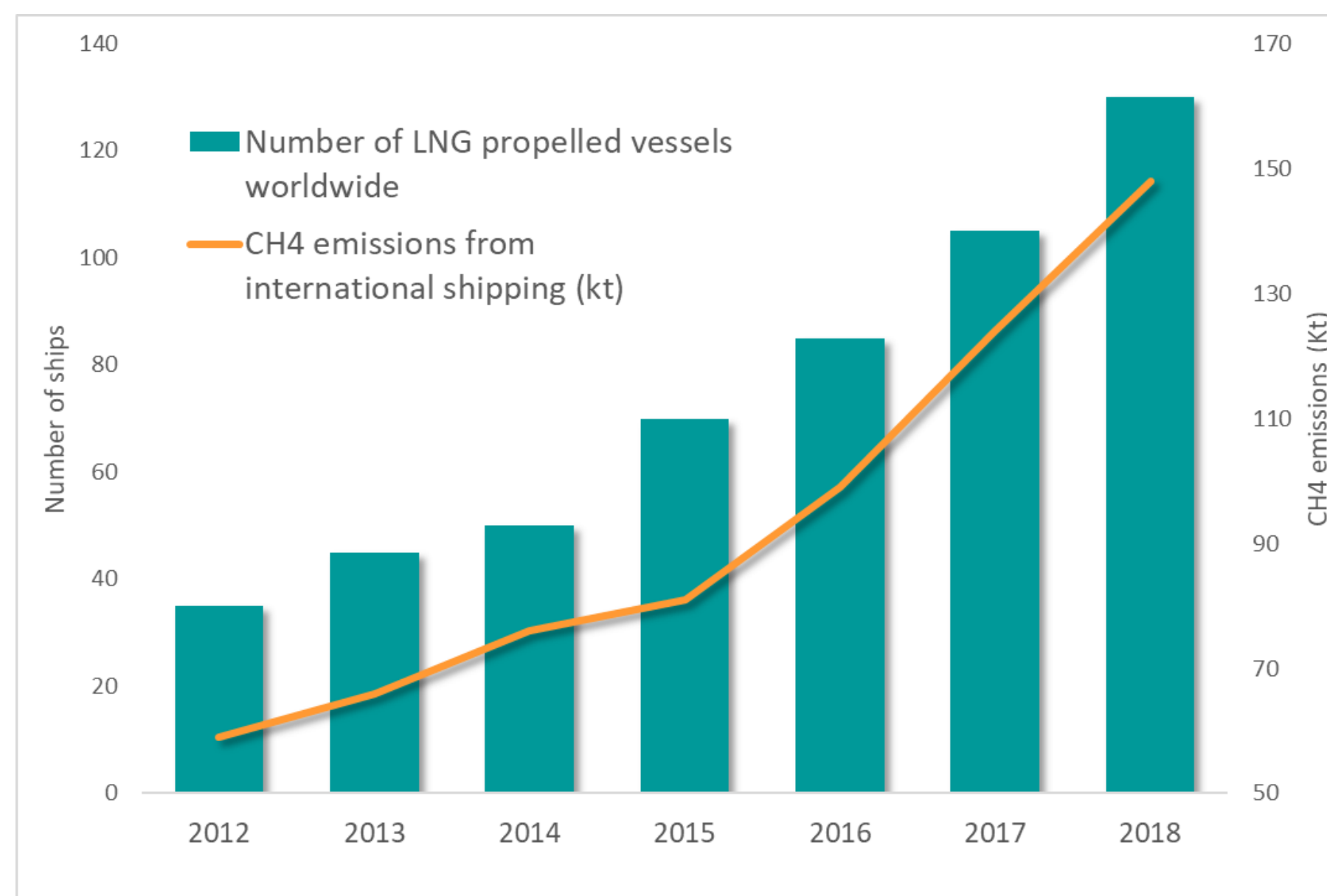
Shipping is a major source of greenhouse gas emissions

More and more ships are using LNG as an alternative fuel

LNG minimises NO<sub>x</sub>, SO<sub>x</sub> and CO<sub>2</sub> emissions

Releases unburnt methane into the atmosphere

As the number of LNG-powered ships has increased so have methane emissions produced from international shipping.<sup>1,2</sup>



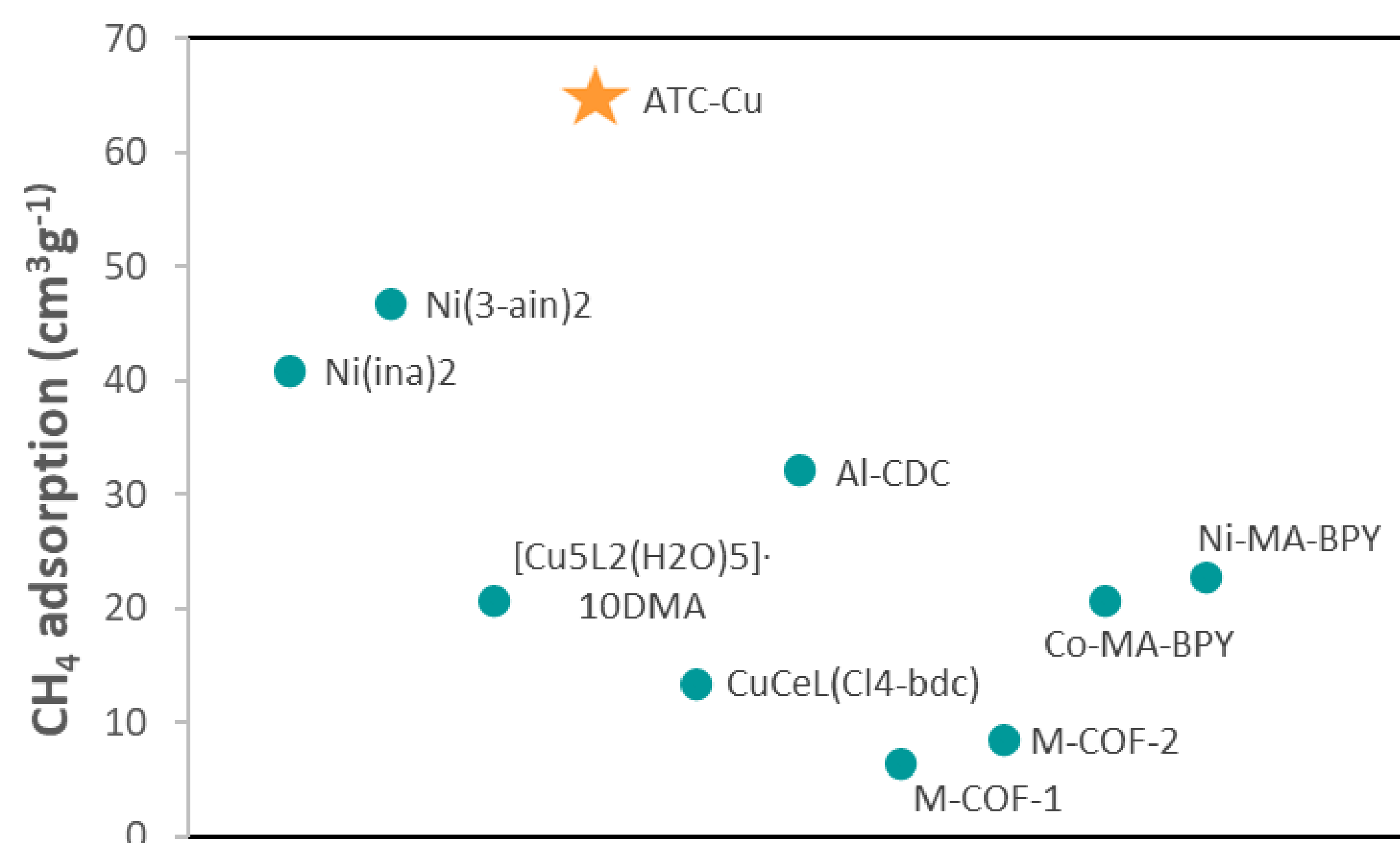
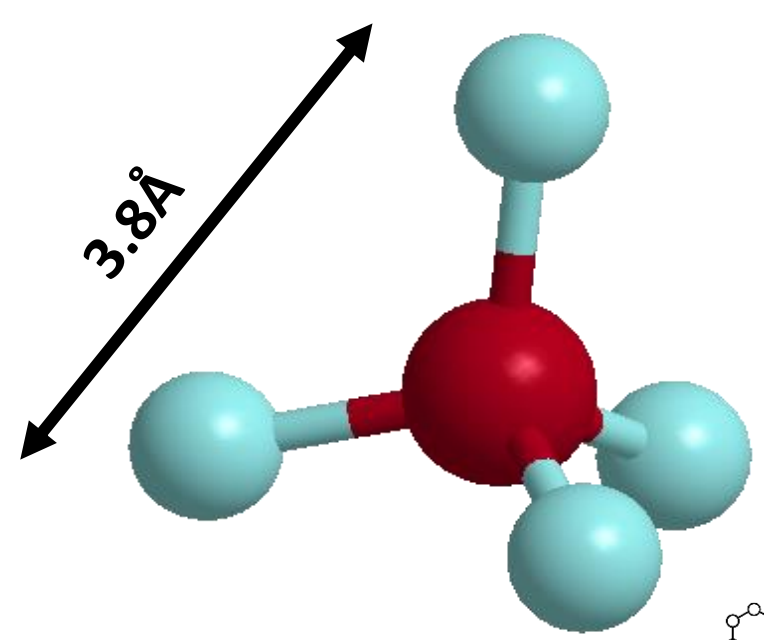
## Proposed methane slip abatement methods

	Maintains engine efficiency	Does not produce CO <sub>2</sub>	Reduce methane slip	Re-use Methane
In-cylinder methods	✗	✓	✓	✗
Total Oxidation Catalysts	✓	✗	✓	✗
Methane capture catalysts	✓	✓	✓	✓

## Methane Capture Catalysts

**Metal Organic Frameworks (MOFs)** are promising candidates for methane capture materials with advantageous features such as:

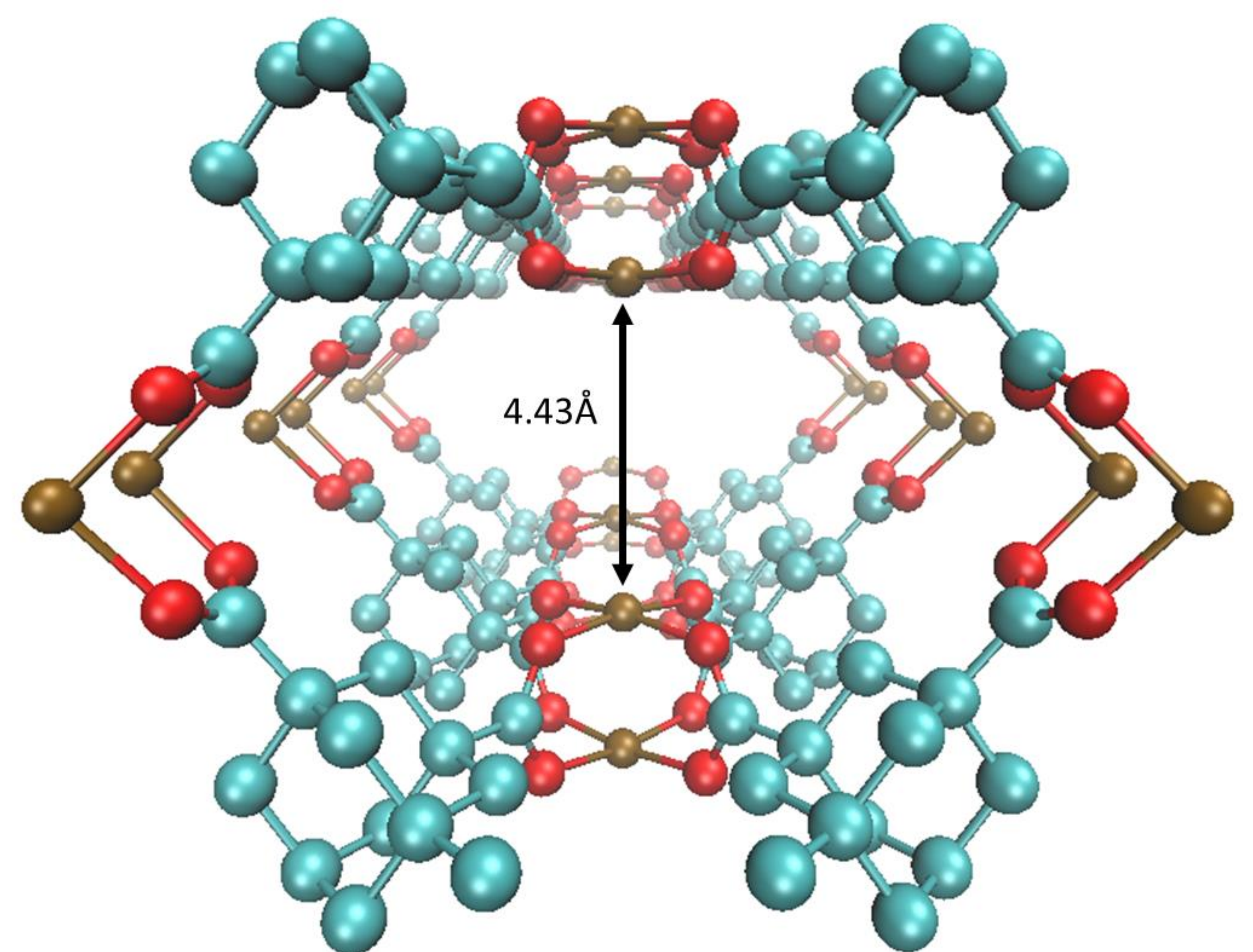
- ✧ Large surface area
- ✧ Tuneable pore sizes
- ✧ Introduce different metal sites
- ✧ Add functional groups to the framework



Various MOFs with differing frameworks and metal sites have been investigated for CH<sub>4</sub> capture.

Currently MOF ATC-Cu,<sup>3</sup> displays the highest CH<sub>4</sub> adsorption ability of 64.96cm<sup>3</sup>g<sup>-1</sup> at 298 K and 1 bar.

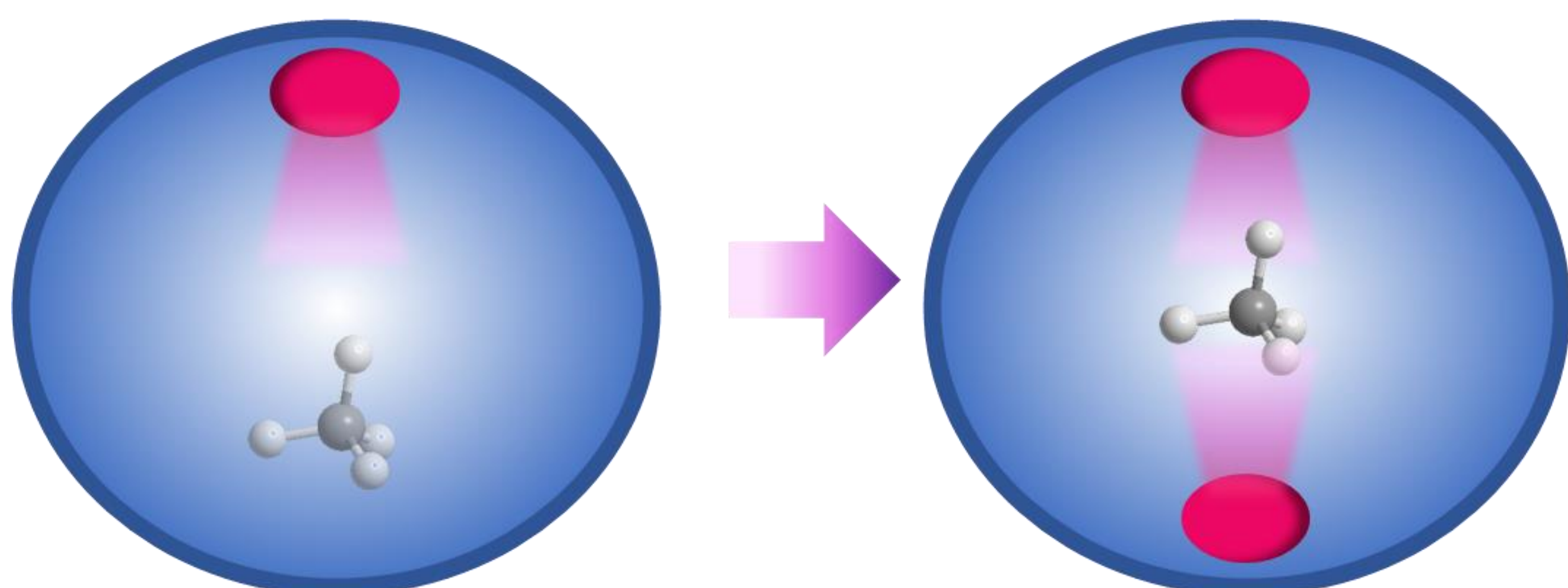
## ATC-Cu MOF



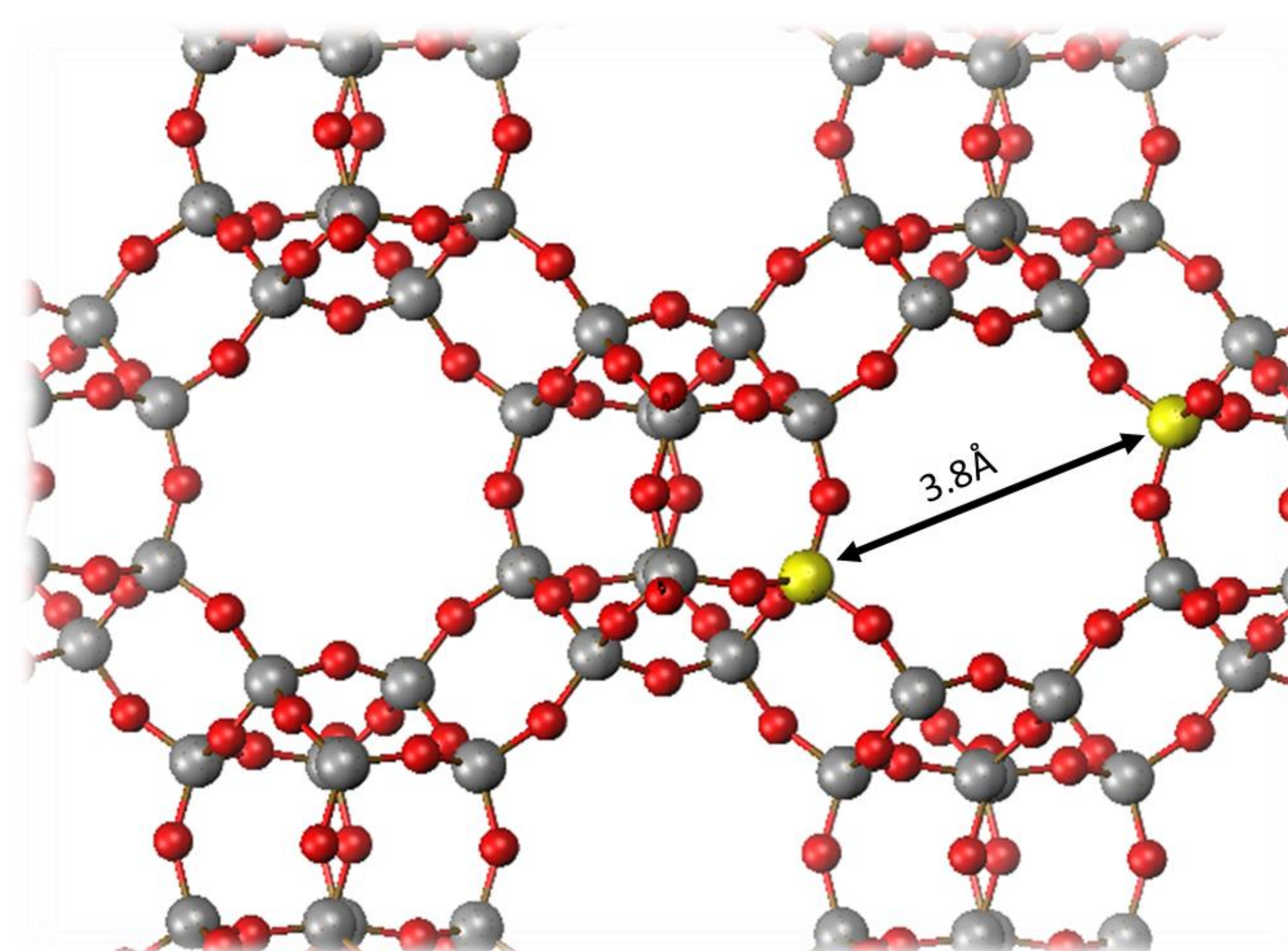
ATC-Cu MOF<sup>3</sup> is based on the H<sub>4</sub>ATC (1,3,5,7-adamantane tetracarboxylic acid) ligand acting as a 4 coordinated tetrahedral linker connecting Cu paddle-wheel secondary building units.

The MOF contains 3D rectangular channels of dimensions 4.43 x 5.39Å where the proximity of the 2 copper sites mediates a 'nano-trap' for methane.

## Key features of methane capture catalyst



- Unsaturated metal centre
- ★ Di-metal centre for enhanced binding
- Appropriate pore size
- Alkyl groups within the pore



Chabazite framework with isomorphous substitution could lead to di-copper sites.

Further possibilities of introducing other metals such as Co could create redox sites for oxidation to methanol.

## References

- [1] Fourth IMO GHG Study 2020 Full Report
- [2] Statista. (2021). Number of liquified natural gas-propelled (LNG) vessels worldwide from 2010 to 2020 with a forecast through 2027 <https://www.statista.com/statistics/1096072/trend-in-projected-global-supply-and-demand-for-ling-fueled-vessels/> (accessed: June 13, 2024)
- [3] Z. Niu *et al*, *Angew. Chem. Int. Ed.*, 2019, 58, 10138-10141.

I would like to thank the Southampton Marine and Maritime Institute for their funding.