

1 **Thermal and non-thermal controls on benthic foraminiferal Mg/Ca,**
2 **Sr/Ca, B/Ca, and Mg/Li: global core-top compilation, revised calibrations,**
3 **and application to the geologic record**

6 **Andressa Nauter-Alves¹, David Evans², Miriam E. Katz^{3,4}, William R. Gray⁵, James W.**
7 **B. Rae⁶, Chiara Borrelli¹**

8 ¹Department of Earth and Environmental Sciences, University of Rochester, Rochester,
9 NY, USA

10 ²School of Ocean and Earth Science, University of Southampton, Southampton, UK

11 ³Department of Earth and Planetary Sciences, Rutgers, The State University of
12 New Jersey, Piscataway, NJ, USA

13 ⁴Geosciences Department, Union College, Schenectady, NY, USA

14 ⁵Laboratoire des Sciences du Climat et de l'Environnement (LSCE/IPSL), Gif-sur-Yvette,
15 France.

16 ⁶School of Earth and Environmental Sciences, University of St Andrews, St Andrews,
17 UK

18 **Abstract**

19 Benthic foraminiferal Mg/Ca, Sr/Ca, B/Ca, and Mg/Li data are used to
20 reconstruct deep-sea temperature and seawater carbonate chemistry. However, the
21 concurrent influence of various environmental parameters on these proxies is not fully
22 understood. Here, we compile published and unpublished element/Ca core-top data
23 from *Cibicidoides mundulus*, *C. pachyderma*, *Lobatula wuellerstorfi*, *Oridorsalis*
24 *umbonatus*, *Nuttallides umbonifera*, and *Uvigerina* spp. to determine the sensitivity of
25 these species' Mg/Ca, Sr/Ca, B/Ca, and Mg/Li to temperature, calcite saturation state
26 (Ω_{calcite}), dissolved inorganic carbon (DIC), and salinity. By applying multivariate linear
27 regression analysis, we disentangle the effects of these environmental parameters on
28 the element/Ca incorporation into these species. As a result, we provide multivariate
29 element/Ca calibrations with temperature and Ω_{calcite} sensitivities for all species-
30 elemental system combinations. Overall, our analysis reveals that: 1) the Mg/Ca-
31 sensitivity to temperature is substantially lower compared to most previous
32 approaches when accounting for the effect of Ω_{calcite} ; 2) Sr/Ca is driven dominantly by
33 Ω_{calcite} ; 3) B/Ca can be linearly related to Ω_{calcite} ; and 4) Mg/Li-temperature calibrations
34 have substantially lower degrees of unexplained variance compared to Mg/Ca. Our
35 calibrations offer a statistically robust approach that also allows us to quantify the

36 uncertainties in the reconstruction of temperature and carbonate chemistry when
37 these element/Ca proxies are used. Finally, we develop a user-friendly data processing
38 software ('ElCaRBenthic') that can simultaneously solve up to two benthic
39 foraminiferal element/Ca datasets for temperature and Ω_{calcite} (e.g., Mg/Ca and Sr/Ca),
40 while providing the ability to correct for long-term changes in seawater elemental
41 chemistry and propagating all sources of uncertainty.

42 **Plain Language Summary**

43 The ratios of magnesium (Mg) to calcium (Mg/Ca) and Mg to lithium (Mg/Li)
44 measured in benthic foraminifera shells are commonly used to reconstruct deep-sea
45 temperature, whereas the ratios of strontium (Sr) and boron (B) to Ca (Sr/Ca and B/Ca)
46 are used to reconstruct ocean chemistry. However, the application of these elemental
47 ratios is not straightforward. In this study, we compiled and analyzed a dataset
48 including Mg/Ca, Sr/Ca, B/Ca, and Li/Ca collected from five benthic foraminiferal
49 species and one genus to understand how each elemental ratio responds to different
50 environmental parameters (i.e., temperature, carbonate chemistry, salinity). The
51 results of our statistical analyses show that the benthic foraminiferal Mg/Ca is less
52 sensitive to temperature when Ω_{calcite} (a measure of how saturated seawater is with
53 respect to calcite) is taken into account, and that Mg/Li provides more consistent
54 temperature reconstructions compared to Mg/Ca. Benthic foraminiferal Sr/Ca and
55 B/Ca are mainly influenced by changes in Ω_{calcite} . To support paleoceanographic
56 reconstructions, we develop a software called ElCaRBenthic. This software allows users
57 to solve for temperature and Ω_{calcite} using paired elemental ratios, while also
58 propagating uncertainties.

59
60 **Key Points**

61 • We compile ~1500 core-top benthic foraminiferal Mg-Sr-B-Li/Ca measurements
62 from five species and one genus to produce new calibrations
63 • Many benthic foraminiferal elemental systems require multivariate calibrations to
64 account for temperature and seawater carbonate chemistry
65 • We present data processing software that can simultaneously solve for
66 temperature and Ω_{calcite} and fully propagate uncertainty

67 **1. Introduction**

68 The ratio of magnesium to calcium (Mg/Ca) in the tests of foraminifera
69 provides one of the most widely applied proxies to reconstruct ocean
70 paleotemperatures. Rooted in thermodynamics, this proxy is based on the fact that
71 higher temperatures will be recorded as higher Mg/Ca ratios in foraminifera (Chave,
72 1954; Izuka, 1988; Morse and Bender, 1990; Nürnberg et al., 1996; Rosenthal et al.,
73 1997; Lea et al., 1999; Branson et al., 2013), a feature also observed in inorganic calcite
74 (e.g., Katz, 1973; de Choudens-Sanchez & Gonzalez, 2009).

75 Benthic foraminiferal Mg/Ca (Mg/Ca_{bf}) has become a useful tool in the
76 investigation of past deep-sea temperatures (e.g., Rosenthal et al., 1997; Martin et al.,
77 2002; Elderfield et al., 2006; Bryan and Marchitto, 2008; Tisserand et al., 2013; Lear et
78 al., 2015; Lo Giudice Cappelli et al., 2015; Kubota et al., 2015), in particular, because
79 few other methodologies are available for quantitative environmental reconstruction
80 in this realm (e.g., organic geochemical proxies are limited to the upper part of the
81 ocean). In addition, Mg/Ca_{bf} can be used in combination with benthic foraminiferal
82 oxygen isotopes ($\delta^{18}O_{bf}$) to determine the $\delta^{18}O$ of seawater and thus reconstruct sea
83 level and ice volume changes throughout the Cenozoic (e.g., Cramer et al., 2011; Lear
84 et al., 2000; 2015; Miller et al., 2020).

85 Notwithstanding the impact that benthic Mg/Ca data had on our understanding
86 of past climates, the application of Mg/Ca_{bf} is complicated by additional nonthermal
87 controls. For example, Martin et al. (2002) analyzed *Lobatula* (*Cibicidoides*)
88 *wuellerstorfi* Mg/Ca and observed that specimens collected at the deepest Atlantic and
89 Pacific sites were characterized by Mg/Ca values that were not in agreement with the
90 overall Mg/Ca-temperature relationship for this species. Based on this finding,
91 different relationships between Mg/Ca_{bf} and seawater carbonate chemistry were
92 proposed, usually parameterized as a function of carbonate ion saturation $\Delta[CO_3^{2-}]$
93 ($\Delta[CO_3^{2-}] = [CO_3^{2-}]_{in situ} - [CO_3^{2-}]_{saturation}$, where $[CO_3^{2-}]_{saturation}$ is that which would yield
94 $\Omega_{calcite} = 1$). Subsequent studies established a Mg/Ca_{bf} species-specific sensitivity to
95 $\Delta[CO_3^{2-}]$, suggesting that species-specific calibrations ideally should be used where
96 possible (Elderfield et al., 2006; Healey et al., 2008; Raitzsch et al., 2008; Yu and

97 Elderfield, 2008). In practical terms, the impact of $[\text{CO}_3^{2-}]$ in down core records has
98 been addressed by applying region-specific calibrations (e.g., Sosdian & Rosenthal,
99 2009) with the assumption that temperature and $[\text{CO}_3^{2-}]$ covaried through time to the
100 same degree as the calibration dataset. In situations where this is unlikely to have
101 been the case, Mg/Ca data were corrected for changes in seawater carbonate
102 chemistry by independently estimating the degree to which this factor has changed.
103 For example, Cramer et al. (2011) parameterized deep ocean $\Delta[\text{CO}_3^{2-}]$ as a function of
104 the calcite compensation depth (CCD). However, this approach suffers from the likely
105 inter- and intra-basinal variability in $\Delta[\text{CO}_3^{2-}]$ given that the dataset drew on globally
106 distributed sites, as well as the uncertainty in deriving $\Delta[\text{CO}_3^{2-}]$ from a CCD record
107 (Greene et al., 2019).

108 Other approaches to correcting $\text{Mg}/\text{Ca}_{\text{bf}}$ for changes in deep ocean carbonate
109 chemistry focused on other element/Ca ratios in foraminiferal CaCO_3 . Bryan &
110 Marchitto (2008) showed that normalization of benthic foraminiferal Mg to lithium
111 (i.e., $\text{Mg}/\text{Li}_{\text{bf}}$) resulted in a proxy that combined the temperature sensitivity of Mg
112 incorporation while removing the (presumably) biologically-driven impact of seawater
113 carbonate chemistry. The basis of the approach lies in the fact that, while Li
114 incorporation into calcite is also sensitive to temperature (e.g., Marriot et al., 2004b;
115 Hall and Chan, 2004, Bryan and Marchitto, 2008), it can also be influenced by changes
116 in deep water saturation, which is supported by the overall decrease in foraminiferal
117 Li/Ca with water depth (e.g., Lear and Rosenthal, 2006; Doss et al., 2018). Therefore,
118 the use of $\text{Mg}/\text{Li}_{\text{bf}}$ can account for the impact that $\Delta[\text{CO}_3^{2-}]$ changes have on biological
119 calcification processes that ultimately influence $\text{Mg}/\text{Ca}_{\text{bf}}$, such as Ca^{2+} utilization at the
120 calcification site (Rayleigh distillation) and growth rate (Marchitto et al., 2018). A
121 number of studies have since built on this. For example, Lear et al. (2010) combined *O.*
122 *umbonatus* Mg/Ca and Li/Ca data to subtract the seawater carbonate chemistry effect
123 from the $\text{Mg}/\text{Ca}_{\text{bf}}$ temperature signal (see also Dawber and Tripati, 2011; Lear and
124 Rosenthal, 2006). The development of boron to Ca ($\text{B}/\text{Ca}_{\text{bf}}$) as a proxy for seawater
125 carbonate chemistry offers an alternative tool to directly reconstruct changes in
126 $\Delta[\text{CO}_3^{2-}]$ (e.g., Yu & Elderfield, 2007; Yu et al., 2010; Rae et al., 2011; Brown et al.,
127 2011; Raitzsch et al., 2011).

128 Benthic foraminiferal strontium to Ca ($\text{Sr}/\text{Ca}_{\text{bf}}$) has also been suggested as an
129 alternate proxy to reconstruct $\Delta[\text{CO}_3^{2-}]$ in geological records (Dawber & Tripati, 2012
130 Yu et al., 2014; Lo Giudice Capelli et al., 2015; Keul et al., 2017; Lawson et al., 2014).
131 However, several studies have shown that $\text{Sr}/\text{Ca}_{\text{bf}}$ can also be influenced by
132 temperature, pH, salinity, and ontogenetic effects (Rathburn and Deckker, 1997;
133 Reichart et al., 2003; Rosenthal et al., 2006; Dissard et al., 2010; Lo Giudice Capelli et
134 al., 2015). In addition, the influence of temperature on the $\text{Sr}/\text{Ca}_{\text{bf}}$ of species
135 commonly used in paleoenvironmental investigations remains unclear (Rosenthal et
136 al., 1997; Rathmann & Kuhnert 2008; Yu et al., 2014).

137 As a result, while much progress has been made over the last few decades in
138 the development and application of (trace) element systems in benthic foraminifera,
139 much remains to be understood in terms of the relative influence of temperature and
140 the seawater carbonate chemistry on these proxies. For example, previous work
141 suggested a wide range of $\text{Mg}/\text{Ca}_{\text{bf}}$ -temperature sensitivities for the same species
142 depending on how nonthermal influences were treated, and the degree to which
143 temperature and carbonate chemistry varied in the underlying dataset (e.g., Lear et al.,
144 2002; 2010; Yu & Elderfield, 2008; Raitzsch et al., 2008; Lo Giudice Capelli et al., 2015).

145 In this study, we compile published and unpublished globally distributed core-
146 top Mg/Ca , Sr/Ca , B/Ca , and Li/Ca . Our dataset includes *Cibicidoides mundulus*, *C.*
147 *pachyderma*, *Lobatula wuellerstorfi*, and *Nuttallides umboinifera*, described as
148 epifaunal species (e.g., Lutze and Thiel, 1989; McCorkle et al., 1990; Rathburn and
149 Corliss, 1994), as well as *Oridorsalis umbonatus* and *Uvigerina* spp., described as
150 shallow infaunal species (e.g., McCorkle et al., 1990; Rathburn and Corliss, 1994) (Table
151 1). We analyze this compilation using multivariate linear regression models for all
152 proxy system-species combinations with sufficient data and determine the potential
153 influence of temperature, carbonate chemistry, and salinity on these systems. Finally,
154 we present a Matlab script ('ElCaRBenthic') for the calculation of temperature, Ω_{calcite} ,
155 and $[\text{CO}_3^{2-}]$ from benthic foraminiferal Mg/Ca , Sr/Ca , B/Ca , or Mg/Li data. This
156 software: 1) simultaneously solves for temperature and seawater carbonate chemistry
157 when provided with sufficient data; (2) fully propagates all sources of uncertainty; and
158 (3) can be applied on timescales across which the elemental composition of seawater
159 has changed.

160 **2. Material and methods**

161 2.1 Core-top data compilation

162 We compiled core-top Mg/Ca, Sr/Ca, B/Ca, and Li/Ca data for five benthic
163 foraminiferal species (*Cibicidoides mundulus*, *C. pachyderma*, *Lobatula wuellerstorfi*,
164 *Nuttallides umbonifera*, *Oridorsalis umbonatus*) and one benthic foraminiferal genus
165 (*Uvigerina* spp.) (Fig. S1) from the Arctic, Atlantic, Indian, and Pacific Oceans (Figure 1;
166 Table 1) from 25 published studies (Rosenthal et al., 1997; Martin et al., 2002; Lear et
167 al., 2002; Rathmann et al., 2004; Lear and Rosenthal, 2006; Elderfield et al., 2006;
168 Marchitto et al., 2007a; Yu and Elderfield, 2007; Yu and Elderfield, 2008; Bryan and
169 Marchitto, 2008; Healey et al., 2008; Raitzsch et al., 2008, 2011; Lear et al., 2010;
170 Brown et al., 2011; Rae et al., 2011; Rae, 2011; Tisserand et al., 2013; Yu et al., 2014;
171 Gray, 2015; Kubota et al., 2015; Lo Giudice Cappelli et al., 2015; Gussone et al., 2016;
172 Stirpe et al., 2021; Lawson et al., 2024) and two PhD theses (Gray, 2015; Rae, 2011; see
173 Table 1). The compiled dataset can be found in Pangaea (Nauter-Alves et al., XXXX;
174 doi:XXXXXX).

175 The compiled data span a temperature range from -1.15 to 18.6°C, a $\Delta[\text{CO}_3^{2-}]$
176 range from -35.4 to 161 $\mu\text{mol/kg}$ (as reported in the original publications but
177 recalculated here, see Sec. 2.2), and a water depth range of 151 to 5159 m. Sediment
178 samples below 2 cm core depth and all outliers as flagged in the original publication
179 (e.g., those resulting from analytical error or diagenetic alteration) were excluded from
180 our analysis.

181 We note that in some publications, *L. wuellerstorfi* is reported as *Cibicidoides*
182 *wuellerstorfi*. According to Hayward et al. (2025) (WoRMS database;
183 <https://www.marinespecies.org/index.php>), this species is currently classified as *L.*
184 *wuellerstorfi*. Given this and considering the distinct element/Ca signature of this
185 species compared to the other *Cibicidoides* species considered in this study (see Sec.
186 3), we decided to follow the current nomenclature and therefore do not analyze *L.*
187 *wuellerstorfi* data together with *C. mundulus* and *C. pachyderma* when investigating
188 this genus-specific portion of the dataset.

189 There is also an ongoing taxonomic debate regarding the classification of
190 *Cibicidoides pachyderma*. While the studies utilized here classified this species as
191 *Cibicidoides*, genetic analyses disagree in the placement of this species within this
192 genus (*Cibicidoides* versus *Cibicides*; Schweizer et al., 2009; Holzmann and Pawlowski,
193 2017). Considering that the debate has not yet been resolved and in the light of the
194 broadly similar test chemistry of *C. pachyderma* and *C. mundulus* (discussed in detail
195 below), we considered *C. pachyderma* as a *Cibicidoides* species for the purposes of our
196 analyses.

197 Finally, we considered all *Uvigerina* data included here as *Uvigerina* spp. We did
198 so because the studies utilized here were based on *U. peregrina*, which includes
199 morphologically variable species as defined in Stirpe et al. (2021), and other species
200 that were classified only at the genus level (i.e., *Uvigerina*, Yu and Elderfield 2007,
201 2014; Gussone et al., 2016).

202 The compiled data were measured using different analytical methods, and
203 samples were prepared according to different cleaning procedures. Some datasets
204 were collected using a spatially-resolved analytical technique (laser ablation
205 inductively coupled plasma mass spectrometry - LA-ICPMS), which differs from bulk
206 ('solution') approaches in that it does not average over intra- and inter-specimen
207 geochemical heterogeneity. In order to convert these datasets to a form that is – as far
208 as possible – comparable to solution approaches, where data were reported as
209 individual laser spot measurements or as individual specimen values (Rathmann et al.,
210 2004; Stirpe et al., 2021), we calculated the average element/Ca per site and used that
211 value in our analysis.

212 2.2 Hydrographic data extraction

213 Seawater hydrographic parameters (temperature, seawater carbonate
214 chemistry, salinity) were recalculated from each original study to establish consistency
215 across datasets. We based all calculations on the 0.25° resolution World Ocean Atlas
216 (WOA) 2018 temperature and salinity datasets (Locarnini et al., 2018; Zweng et al.,
217 2019) and the 1° resolution GLODAP gridded total alkalinity (TALK) and pre-industrial
218 dissolved inorganic carbon (DIC) datasets (Lauvset et al., 2016). Data were extracted

219 from the nearest available grid cell by first extracting the closest matching depth in the
 220 gridded datasets to a given sample record, subsequently converting sample locations
 221 and the gridded datasets at that depth to cartesian coordinates, and finally deriving
 222 the closest match using the minimum distance between the sample site and all
 223 available grid cells. In the vast majority of cases, this yielded temperature and salinity
 224 datasets that agreed with the value given in the original publication within $\pm 1^\circ\text{C}$ and
 225 ± 0.2 units. However, in five cases, the WOA-derived temperature used here was offset
 226 by more than 2°C . In these instances, we reverted to the value given in the original
 227 publication, on the assumption that the originally reported value better represents the
 228 sample site (e.g., given the resolution of the gridded data products).

229 **Table 1:** Sources of the benthic foraminiferal element/Ca datasets used in this study.

Species	Mg/Ca	Sr/Ca	B/Ca	Li/Ca
<i>Cibicidoides mundulus</i>	(Elderfield et al., 2006; Healey et al., 2008; Raitzsch et al., 2008; Rae, 2011; Tisserand et al., 2013)	(Rae, 2011; Yu et al., 2014; Lawson et al., 2024)	(Yu and Elderfield, 2007; Rae et al., 2011)	(Rae, 2011)
<i>Lobatula wuellerstorfi</i>	(Russell et al., 1994; Rosenthal et al., 1997; Lear et al., 2002; Martin et al., 2002a; Elderfield et al., 2006; Healey et al., 2008; Raitzsch et al., 2008; Yu and Elderfield, 2008; Rae, 2011; Tisserand et al., 2013; Kubota et al., 2015; Lo Giudice Cappelli et al., 2015; Gussone et al., 2016)	(Rae, 2011; Yu et al., 2014; Lo Giudice Cappelli et al., 2015; Gussone et al., 2016; Lawson et al., 2024)	(Yu and Elderfield, 2007; Brown et al., 2011; Rae et al., 2011; Raitzsch et al., 2011)	(Rae, 2011)
<i>Oridorsalis umbonatus</i>	(Rathmann et al., 2004; Elderfield et al., 2006; Healey et al., 2008; Lear et al., 2010; Rae, 2011; Tisserand et al., 2013; Gussone et al., 2016)	(Rae, 2011; Gussone et al., 2016)	(Brown et al., 2011; Rae et al., 2011)	(Lear and Rosenthal, 2006; Rae, 2011)
<i>Nuttallides umbonifera</i>			(Brown et al., 2011)	

<i>Cibicidoides pachyderma</i>	(Rosenthal et al., 1997; Lear et al., 2002; Marchitto et al., 2007a; Bryan and Marchitto, 2008)	(Lawson et al., 2024)	(Bryan and Marchitto, 2008)
<i>Uvigerina peregrina</i>	(Elderfield et al., 2006; Bryan and Marchitto, 2008; Rae, 2011; Gray, 2015; Gussone et al., 2016; Stirpe et al., 2021)	(Rae, 2011; Gray, 2015; Gussone et al., 2016;)	(Rae et al., 2011; Gray, 2015)
<i>Uvigerina</i> spp.	(Gussone et al., 2016)		(Yu and Elderfield, 2007)

230

231 Seawater carbonate chemistry parameters were calculated with co2sys (Lewis
 232 et al., 1998), MATLAB version V3.1.2, using the closest available TALK, pre-industrial
 233 DIC, temperature, and salinity estimates as described above. To accomplish this, we
 234 applied: 1) the carbonate system dissociation constants of Mehrbach et al. (1973) refit
 235 by Dickson and Millero (1987), as recommended by Wanninkhof et al. (1999); 2) the
 236 HSO_4^- dissociation constant of Dickson (1990); 3) the HF dissociation constant of Perez
 237 and Fraga (1987); and 4) the total boron value of Lee et al. (2010). Our carbonate
 238 system calculation results in broadly but not unanimously good agreement with the
 239 values reported in the original publications. Specifically, ~5.5% of the reported $\Delta[\text{CO}_3^{2-}]$
 240 values ($n = 68$) are offset from our internally consistent calculation by more than 25
 241 $\mu\text{mol/kg}$, typically to lower $\Delta[\text{CO}_3^{2-}]$. While investigating the sources of this discrepancy
 242 is beyond the scope of this study, it likely stems from different choices of dissociation
 243 constants, calculation routines, and other parameters required to perform this
 244 calculation between studies, and/or between this study and the original publication.
 245 To ensure consistency in dissociation constants and parameters across the dataset, we
 246 used the recalculated $\Delta[\text{CO}_3^{2-}]$ in all instances.

247 **2.3 Multivariate linear regression analysis**

248 We determine the sensitivity of the proxy systems considered here to a
 249 number of environmental parameters using least-squares multivariate linear
 250 regression models. We used this approach to disentangle the effects of temperature
 251 and seawater carbonate chemistry on element incorporation into benthic foraminiferal

252 calcite. This carbonate chemistry effect is often ascribed to $[\text{CO}_3^{2-}]$ or $\Delta[\text{CO}_3^{2-}]$ (e.g.,
253 Elderfield et al., 2006; Yu et al., 2014), with the rationale that some of the apparent
254 sensitivity of (e.g.) Mg/Ca to temperature mechanistically reflects the impact of
255 seawater carbonate chemistry on test chemistry or vice versa, as is the case for
256 planktonic foraminifera (see Gray et al. (2018) and Gray and Evans (2019) for
257 discussion). Here, we alternatively use the saturation state of seawater with respect to
258 calcite, defined as:

259 $\Omega_{\text{calcite}} = [\text{Ca}^{2+}]_{\text{sw}} \bullet [\text{CO}_3^{2-}]_{\text{sw}} / K^*_{\text{sp}},$

260 Where K^*_{sp} is the stoichiometric solubility product of calcite. We decided to do
261 so for two reasons: 1) In the modern open ocean, $[\text{Ca}^{2+}]_{\text{sw}}$ is characterized by
262 extremely little spatial heterogeneity (Zeebe and Wolf-Gladrow, 2001). While pressure
263 and temperature impact carbon speciation and calcite solubility to different degrees,
264 Ω_{calcite} and $\Delta[\text{CO}_3^{2-}]$ are – in broad terms - a direct transformation of one another, such
265 that this has little practical impact on the application of the calibration model (see
266 Figure S2C-D). 2) $[\text{Ca}^{2+}]_{\text{sw}}$ has varied across multi-million-year timescales (e.g., Brennan
267 et al., 2013; Zhou et al., 2021), which resulted in a decoupling of $[\text{Ca}^{2+}]_{\text{sw}}$ and Ω_{calcite} (or
268 $\Delta[\text{CO}_3^{2-}]$) over the Cenozoic (Zeebe and Tyrrell, 2019), as Ω_{calcite} remained relatively
269 constant, while $[\text{Ca}^{2+}]_{\text{sw}}$ decreased and $[\text{CO}_3^{2-}]_{\text{sw}}$ underwent a large increase from the
270 Palaeogene to today. While it is not necessary to consider changes in $[\text{Ca}^{2+}]_{\text{sw}}$ when
271 interpreting Pleistocene data, formulating regression models with Ω_{calcite} avoids the
272 need to apply different calibrations to different time intervals (Dai et al., 2023). In
273 addition, while Ω_{calcite} and $\Delta[\text{CO}_3^{2-}]$ are not exactly linearly related, they imply a similar
274 mechanistic driver of element incorporation into foraminiferal calcite in the modern
275 ocean, given that they are (approximately speaking) different representations of the
276 same underlying parameters. We argue that the use of Ω_{calcite} is likely more
277 appropriate on longer timescales, assuming that, in terms of the mechanistic impact of
278 seawater carbonate chemistry on mineral growth kinetics or biomineralization, a lower
279 $[\text{CO}_3^{2-}]_{\text{sw}}$ is counteracted by a higher $[\text{Ca}^{2+}]_{\text{sw}}$ (Dai et al., 2023).

280 We explored parameterizing element/Ca data as a function of both Ω_{calcite} and
281 $1/\Omega^2_{\text{calcite}}$ (Tierney et al., 2019), which accounts for the possibility that element

incorporation into foraminiferal calcite becomes less sensitive to seawater carbonate chemistry under more highly oversaturated conditions. Our preferred regression models are those based on $\Omega^{-2}_{\text{calcite}}$ because these can explain a greater degree of the variance in the datasets and provide a more robust empirical fit. In addition, we note that it has been previously suggested that Mg/Ca and $\Delta[\text{CO}_3^{2-}]$ are best described by a linear relationship (e.g., Elderfield et al., 2006; Yu and Elderfield, 2008), in contrast to our preferred regression model which is linearized by log-transforming the Mg/Ca data (described below), thus implying an exponential relationship between Mg/Ca and Ω_{calcite} or $\Omega^{-2}_{\text{calcite}}$. We also tested regression models based on an exponential relationship between Mg/Ca and temperature, and a linear relationship between Mg/Ca and $\Omega^{-2}_{\text{calcite}}$, in order to assess this, and again find that linear models using $\log(\text{Mg/Ca})$ provide better fits to the data in almost all cases. However, coefficients for all of these alternate model forms are given in the supplementary materials (Table S2) for completeness.

In the case of the other element systems analyzed (Sr-B-Li/Ca_{bf}), in which reconstructing Ω_{calcite} or $[\text{CO}_3^{2-}]$ is the primary objective, we formulated the models in terms of Ω_{calcite} rather than $\Omega^{-2}_{\text{calcite}}$ because the latter option results in models that yield prediction uncertainties that vary widely across the range of Ω_{calcite} observed in both core-top and downcore applications, even though such variability is not evident in the model residuals for these elements.

Informed by the above constraints, all element/Ca datasets were related to seawater hydrographic and carbonate chemistry parameters via a regression of the form

$$\ln(\text{Mg/Ca}) = x_1 + x_2 T + x_3 \Omega^{-2}_{\text{calcite}} + x_4 [\text{DIC}] + x_5 S \quad (\text{Eq. 1})$$

in the case of Mg/Ca and

$$\text{El}_1/\text{El}_2 = x_1 + x_2 T + x_3 \Omega_{\text{calcite}} + x_4 [\text{DIC}] + x_5 S \quad (\text{Eq. 2})$$

in the case of Sr/Ca, B/Ca, Li/Ca, and Mg/Li). T denotes temperature, Ω_{calcite} the saturation state of seawater with respect to calcite, DIC the concentration of dissolved inorganic carbon, and S salinity. DIC and S were initially included in the model to explore whether the inclusion of these parameters could explain a greater proportion of the variance in the elemental datasets, as suggested for planktonic species

313 (Kısakürek et al., 2009; Haynes et al., 2023). This was broadly found not to be the case
314 here. For this reason, all preferred models discussed below include only the first three
315 terms of Eqs. 1 and 2 (see Eq. 3 and Eq. 4) with alternative models including a DIC and
316 salinity component given in the supplementary information (Table S3). Specifically in
317 the case of Mg/Ca:

318 $\ln(\text{Mg/Ca}) = x_1 + x_2 T + x_3 \Omega^2_{\text{calcite}}$ (Eq. 3)

319 And for all other elements:

320 $EI_1/EI_2 = x_1 + x_2 T + x_3 \Omega_{\text{calcite}}$ (Eq. 4)

321 Element/Ca datasets were retained as raw proxy values in all regressions except
322 Mg/Ca, which was first log-transformed to linearize the model, accounting for the
323 thermodynamic expectation that the partitioning of Mg into CaCO_3 is logarithmically
324 related to temperature (e.g., Lea et al., 1999). While this is also the case for other
325 (trace) elements in solid substitution with CaCO_3 , the more minor or absent resolvable
326 impact of temperature on element partitioning in most other cases means that this
327 transformation is unnecessary.

328 The above analysis is complicated to a degree by the broad covariance of
329 temperature, Ω_{calcite} , and DIC in the modern ocean (Fig. S2). Pearson's correlation
330 coefficients between temperature and $\Omega^2_{\text{calcite}}$ are -0.38, -0.70, -0.51, -0.68, and -0.79
331 for the *L. wuellerstorfi*, *Uvigerina* spp., *O. umbonatus*, *C. mundulus*, and *C. pachyderma*
332 datasets, respectively, and 0.63, 0.83, 0.51, 0.60, and 0.95 for $\Omega^2_{\text{calcite}}$ and DIC (in the
333 same order). As such, while the dataset overall is reasonably well suited to
334 disentangling the effects of temperature and Ω_{calcite} (although this is more challenging
335 in the case of *Uvigerina* spp. and *C. pachyderma*), it is not possible to distinguish
336 between a Ω_{calcite} and DIC control on *C. pachyderma* test chemistry based on the data
337 currently available. However, we consider this latter issue to be of minor importance
338 given that the inclusion of DIC in the regression analysis does not substantially improve
339 the predictive power of the model (Table S4).

340 2.4 Example down-core applications

341 To compare our regression models to previous approaches, we applied our
342 calibrations to two different down-core records for which multiple element/Ca

343 datasets were available. The first of these uses *L. wuellerstorfi* Sr/Ca (Yu et al., 2014)
344 and B/Ca (Yu et al., 2010a, 2010b, 2014) on samples from core VM28-122 (12°N, 79°W,
345 3620 m, with a sill depth of 1.8 km, Caribbean Basin). Samples for B/Ca were cleaned
346 following the 'Mg-cleaning' method (Yu and Elderfield, 2007), whereas samples for
347 Sr/Ca were cleaned according to the oxidative and reductive methods (Boyle and
348 Keigwin, 1985; Barker et al., 2003), prior to solution ICP-MS analysis. The dataset spans
349 the Holocene and late Pleistocene (the last ~160 kyr). The age model was based on
350 radiocarbon dating of planktonic foraminifera for samples younger than 30 ka. For
351 older samples, ages were calculated based on tuning the $\delta^{18}\text{O}_{\text{bf}}$ to the global benthic
352 foraminiferal stack of Lisiecki and Raymo (2005) (Yu et al., 2010a).

353 We also applied our calibrations to a second, longer record in order to
354 investigate the degree to which reconstructions based on benthic foraminiferal
355 element/Ca data are sensitive to nonthermal parameters that cannot be accounted for
356 in the modern ocean (i.e., changes in the elemental composition of seawater). To do
357 so, we used the benthic foraminiferal Mg/Ca, Sr/Ca, B/Ca, and Li/Ca records available
358 from the Ocean Drilling Program (ODP) Site 806 (0°19.1'N, 159°21.7'E, 2521 m, Ontong
359 Java Plateau; Lear and Rosenthal, 2006; Lear et al., 2003b, 2015a). Data were collected
360 from specimens of *L. wuellerstorfi*, *O. umbonatus*, and *C. mundulus*. Samples were
361 cleaned following Boyle and Keigwin (1985) 'Cd-cleaning' prior to solution ICP-MS
362 analysis. Sample ages were based on the biostratigraphic events reported in the Site
363 report (Shipboard Scientific Party, 1991), revised with the datum ages compiled by the
364 ODP Leg 199 Shipboard Scientific Party (Lear et al., 2003a). This age model was applied
365 to the whole Site 806 record, which ranged from 0.02 to 16.35 Ma (Pleistocene to early
366 Middle Miocene).

367 **3. Results**

368 Mg-Sr-B-Li/Ca data from each species is shown as a function of *in situ*
369 temperature, Ω_{calcite} , and dissolved inorganic carbon (DIC) in Figure 2. Multivariate
370 regression models (Eqs. 3 and 4) with two predictor terms (T and Ω_{calcite} or $\Omega^2_{\text{calcite}}$) are
371 given in Table 2. These models can explain a substantial portion of the variance in all
372 the elemental datasets considered here ($R^2 = 0.60$ to 0.84; Tab. 2), with the exception

373 of Sr incorporation into *L. wuellerstorfi* and *C. mundulus* ($R^2 = 0.37$ in both cases). In
374 these species, we find that Sr/Ca cannot be predicted with confidence from T and/or
375 Ω_{calcite} , although the Ω_{calcite} coefficient is nonetheless significant at the 95% confidence
376 level in both cases. The preferred models that show statistically significant coefficients
377 only for each species element/Ca are present in Table 3.

378 3.1. Mg/Ca

379 Benthic foraminiferal Mg/Ca spans 0.5 to 10.2 mmol/mol across the dataset
380 (Fig. 2A). All five species are characterized by a similar range of values. Instead,
381 *Uvigerina* spp. is characterized by a narrower (~1 mmol/mol) range and offset to lower
382 Mg/Ca for a given temperature or Ω_{calcite} .

383 Our dataset includes sample measurements conducted following different
384 cleaning procedures as well as samples that were not reductively and/or oxidatively
385 cleaned prior to analysis (i.e., samples analyzed by LA-ICPMS). It was suggested that
386 the choice of cleaning procedure (Barker et al., 2003) could impact (benthic)
387 foraminiferal Mg/Ca, with the reductive process potentially resulting in the
388 preferential loss of higher-Mg regions of the test (Rosenthal et al., 2004). As a result, in
389 some cases, a ~10% correction was applied to samples that were reductively cleaned
390 (e.g., Lawson et al., 2024). Assessing this complication is challenging because, even if
391 the reductive cleaning may result in lower Mg/Ca, this loss of Mg may represent the
392 desirable loss of Mg from a contaminant phase rather than the foraminiferal test. To
393 determine the impact of reductive versus oxidative cleaning, we examined the
394 difference in Mg/Ca resulting from these two cleaning processes using the residuals of
395 the regression models (see Supporting Information, Figure S10), in addition to
396 conducting statistical tests on subsets of the dataset based on cleaning procedure, and
397 providing comparative regression models based on these subsets of the database
398 (Table S6). As previously reported (e.g., Yu and Elderfield, 2008), we find that mean
399 Mg/Ca is higher in the oxidative cleaning-only group by 5.5%, 10.4%, and 9.0% for *L.*
400 *wuellerstorfi*, *Uvigerina* spp., and *O. umbonatus*, respectively. In the case of *C.*
401 *mundulus*, the opposite is the case (6.9% lower), while insufficient data exist to test
402 this for *C. pachyderma*. A two-tailed test was used to compare the means obtained
403 from the two cleaning procedures. The results indicate that *C. mundulus* is not

404 characterized by a significant difference between cleaning methods, while the
 405 opposite is true for *L. wuellerstorfi*, *Uvigerina* spp., and *O. umbonatus*. While this may
 406 support the application of a correction, we do not do so here because it is not
 407 currently clear whether it is desirable to include Mg lost during the reductive cleaning
 408 process in the average sample Mg/Ca value, i.e., reductive cleaning may remove
 409 contaminant phases as well as primary shell material characterized by higher Mg/Ca,
 410 and this is likely true to differing but poorly constrained extents between species and
 411 sample sites. Spatially resolved (e.g., laser ablation) measurements of benthic
 412 foraminifera with diagenetic coatings subjected to both procedures would be one way
 413 of addressing this issue, as has been conducted for planktonic foraminifera (Fritz-
 414 Endress & Fehrenbacher, 2020). We also note that while regression models based on
 415 only reductively and oxidatively cleaned samples do significantly differ from those
 416 based on the complete dataset in some respects (cf. Table 2 and Table S6, e.g., the *C.*
 417 *mundulus* Mg/Ca-T slope is nonsignificant in both cases), we cannot unambiguously
 418 attribute these differences to cleaning procedure versus differential sampling of the
 419 underlying parameter space between these data subsets. Given the minor absolute
 420 Mg/Ca offsets observed between reductively and oxidatively cleaned samples, and for
 421 the reasons outlined above, we base our discussion on the combined dataset with no
 422 reductive correction applied. However, we stress that this possible impact on benthic
 423 foraminifera shell chemistry needs to be better and mechanistically understood before
 424 this issue can be conclusively addressed.

425 **Table 2.** The relationship between element/Ca, temperature, and Ω_{calcite} ($\text{Mg/Ca} = X_1 + X_2 * T +$
 426 $X_3 * \Omega^2$ and $\text{El}_1/\text{El}_2 = X_1 + X_2 * T + X_3 * \Omega$) for each species considered in this study. Uncertainties
 427 are reported as $\pm 1\text{SE}$ (standard error), n denotes sample size. The p-value, R-squared (R^2), Root
 428 Mean Square Error (RMSE). The Ω -T column provides the coefficient of correlation between
 429 Ω_{calcite} and temperature for the subset of the overall dataset on which each individual
 430 regression model is based. *Cibicidoides* spp. refers to the regression model based on the
 431 combined *C. pachyderma* and *C. mundulus* datasets.

Ratio	X_1 (intercept)	p-value	X_2 (T)	p-value	X_3 (Ω^2 (Mg/Ca) or Ω)	p-value	Overall model				
							R^2	RMSE	p-value	Ω -T	n
<i>L. wuellerstorfi</i>											
Mg/Ca	0.55 \pm 0.03	<<0.05	0.031 \pm 0.004	<<0.05	-0.595 \pm 0.028	<<0.05	0.66	0.19	<<0.05	-0.47	448
Sr/Ca	1.09 \pm 0.01	<<0.05	-0.0020 \pm 0.0018	0.2734	0.146 \pm 0.001	<<0.05	0.37	0.06	<<0.05	0.39	305

B/Ca	99.03 ± 5.07	<<0.05	0.11 ± 1.11	0.9208	79.397 ± 4.290	<<0.05	0.69	14.81	<<0.05	0.34	181
<i>Uvigerina</i> spp.											
Mg/Ca	0.23 ± 0.06	<<0.05	0.02 ± 0.01	<<0.05	-0.30 ± 0.16	<<0.05	0.68	0.16	<<0.05	-0.70	108
Sr/Ca	0.69 ± 0.03	<<0.05	0.003 ± 0.003	0.30	0.16 ± 0.03	<<0.05	0.63	0.08	<<0.05	0.80	53
B/Ca	-9.02 ± 3.53	0.020	-1.87 ± 0.56	0.002	33.24 ± 4.91	<<0.05	0.71	7.33	<<0.05	0.90	38
Mg/Li	0.059 ± 0.002	<<0.05	0.004 ± 0.001	<<0.05	-0.002 ± 0.002	0.2960	0.64	0.01	<<0.05	0.84	85
<i>O. umbonatus</i>											
Mg/Ca	0.38 ± 0.06	<<0.05	0.076 ± 0.008	<<0.05	-0.056 ± 0.089	0.5350	0.60	0.19	<<0.05	-0.55	96
Sr/Ca	0.77 ± 0.06	<<0.05	-0.001 ± 0.010	0.9089	0.077 ± 0.059	0.2023	0.33	0.05	0.0156	0.90	24
B/Ca	18.07 ± 22.15	0.4321	-10.10 ± 5.84	0.1113	44.72 ± 27.36	0.1305	0.21	13.87	0.2647	0.92	14
<i>C. mundulus</i>											
Mg/Ca	0.72 ± 0.11	<<0.05	0.036 ± 0.014	0.0085	-0.61 ± 0.12	<<0.05	0.43	0.25	<<0.05	-0.68	125
Sr/Ca	1.01 ± 0.04	<<0.05	-0.002 ± 0.010	0.8608	0.13 ± 0.05	0.0028	0.37	0.06	<<0.05	0.88	67
B/Ca	88.25 ± 6.34	<<0.05	-2.64 ± 1.57	0.0978	43.38 ± 7.46	<<0.05	0.73	9.47	<<0.05	0.94	64
<i>C. pachyderma</i>											
Mg/Ca	0.25 ± 0.12	0.040	0.059 ± 0.007	<<0.05	0.0796 ± 0.4310	0.8539	0.63	0.17	<<0.05	-0.75	87
Sr/Ca	1.09 ± 0.02	<<0.05	-0.016 ± 0.007	0.0176	0.1678 ± 0.0254	<<0.05	0.84	0.05	<<0.05	0.97	55
Mg/Li	0.07 ± 0.01	<<0.05	0.014 ± 0.001	<<0.05	-0.0072 ± 0.046	0.1238	0.80	0.02	<<0.05	0.62	48
<i>N. umboifera</i>											
B/Ca	11.37 ± 21.55	0.60	-4.890 ± 15.479	0.7555	126.21 ± 42.13	0.0074	0.60	16.52	<<0.05	0.86	22
<i>Cibicidoides</i> spp.											
Mg/Ca	0.62 ± 0.06	<<0.05	0.040 ± 0.004	<<0.05	-0.77 ± 0.21	<<0.05	0.64	0.23	<<0.05	-0.79	212
Sr/Ca	1.04 ± 0.02	<<0.05	0.005 ± 0.004	0.2569	0.1032 ± 0.0193	<<0.05	0.81	0.06	<<0.05	0.96	122

432

433 **Table 3.** Preferred models with the significant coefficient for element/Ca, temperature, and
 434 Ω_{calcite} ($\text{Mg/Ca} = X_1 + X_2 * T + X_3 * \Omega^{-2}$ and $\text{El}_1/\text{El}_2 = X_1 + X_2 * T + X_3 * \Omega$) for each species considered
 435 in this study. Uncertainties are reported as $\pm 1\text{SE}$ (standard error), n denotes sample size. The
 436 p-value, R-squared (R^2), Root Mean Square Error (RMSE).

Ratio	X_1 (intercept) pt)	p- value	X_2 (T)	p-value	X_3 (Ω^{-2} (Mg/Ca) or Ω)	p-value	Overall model				
							R^2	RMSE	p- value	Ω -T	n
<i>L. wuellerstorfi</i>											
Mg/Ca	0.55 ± 0.03	<<0.05	0.031 ± 0.004	<<0.05	-0.596 ± 0.028	<<0.05	0.66	0.19	<<0.05	-0.47	448
Sr/Ca	1.095 ± 0.015	<<0.05	-	-	0.14 ± 0.01	<<0.05	0.37	0.06	<<0.05	0.39	305
B/Ca	99.09 ± 5.02	<<0.05	-	-	79.54 ± 4.01	<<0.05	0.69	14.77	<<0.05	0.34	181
<i>Uvigerina</i> spp.											

Mg/Ca	0.23 ±0.56	<<0.05	0.023 ±0.005	<<0.05	-0.30 ± 0.04	<<0.05	0.68	0.16	<<0.05	-0.70	108
Sr/Ca	0.68 ± 0.03	<<0.05	-	-	0.19 ± 0.02	<<0.05	0.62	0.08	<<0.05	0.80	53
B/Ca	-1.85 ± 3.19	<<0.05	-	-	18.34 ± 2.42	<<0.05	0.61	8.32	<<0.05	0.90	38
Mg/Li	0.057 ± 0.002	<<0.05	0.0038 ±0.0003	<<0.05	-	-	0.63	0.01	<<0.05	0.84	85
<i>O. umbonatus</i>											
Mg/Ca	0.346 ± 0.028	<<0.05	0.079 ±0.007	<<0.05	-	-	0.60	0.19	<<0.05	-0.55	96
Sr/Ca	0.76 ±0.04	<<0.05	-	-	0.08 ± 0.04	0.0035	0.33	0.06	0.01	0.90	24
B/Ca	47.32 ±15.47	<<0.05	-	-	-	-	0.00	14.98	0.93	0.92	14
<i>C. mundulus</i>											
Mg/Ca	0.72 ±0.11	<<0.05	0.036 ±0.014	0.0085	-0.61 ±0.12	<<0.05	0.43	0.25	<<0.05	-0.68	125
Sr/Ca	1.01 ±0.03	<<0.05	-	-	0.13 ±0.02	<<0.05	0.37	0.06	<<0.05	0.88	67
B/Ca	96.64 ±3.97	<<0.05	-	-	31.53 ±2.49	<<0.05	0.72	9.60	<<0.05	0.94	64
<i>C. pachyderma</i>											
Mg/Ca	0.27 ±0.06	<<0.05	0.058 ±0.005	<<0.05	-	-	0.63	0.16	<<0.05	-0.75	87
Sr/Ca	1.09 ±0.02	<<0.05	-0.016 ±0.007	0.0176	0.168 ±0.025	<<0.05	0.84	0.05	<<0.05	0.97	55
Mg/Li	0.057 ±0.012	<<0.05	0.013 ± 0.001	<<0.05	-	-	0.79	0.02	<<0.05	0.62	48
<i>N. umbonifera</i>											
B/Ca	13.25 ±20.25	<<0.05	-	-	114.74 ± 20.91	<<0.05	0.60	16.14	<<0.05	0.86	22

437

438 Overall model (in the form of $Mg/Ca = X_1 + X_2 * T + X_3 * \Omega^{-2}$) root mean square error
 439 (RMSE) ranges between 0.16-0.25 (Table 2) in units of $\ln(Mg/Ca)$, equivalent to an
 440 unexplained variance of 0.26-0.40 mmol/mol for a Mg/Ca value of 1.5 mmol/mol (very
 441 approximately equivalent to a calcification temperature of 5°C). Regression models for
 442 all species yield a significant temperature coefficient ($p < 0.05$), although there are
 443 substantial differences in the relationship between Mg/Ca and temperature.
 444 Specifically, the species investigated here broadly cluster into two groups, with *O.*
 445 *umbonatus* and *Cibicidoides pachyderma* characterized by sensitivities of 5.9-7.6 %/°C,
 446 while *L. wuellerstorfi*, *Uvigerina* spp., and *C. mundulus* are characterized by sensitivities
 447 of 2.3-3.6 %/°C (Fig. 3A, Tab. 2). The former group all lie within the uncertainty of the
 448 sensitivity of many species of planktonic foraminifera (Gray & Evans, 2019), whereas
 449 the latter are closer to, albeit slightly higher than, that of inorganic calcite (~2 %/°C;
 450 Burton and Walter, 1991). These sensitivities are in many cases significantly different

451 from previous calibration approaches (e.g., Lear et al., 2002; Healey et al., 2008),
452 broadly being offset to lower values.

453 In contrast to temperature, only two species and one genus – *C. mundulus*, *L.*
454 *wuellerstorfi*, and *Uvigerina* spp. – are characterized by a significant Mg/Ca- $\Omega^{-2}_{\text{calcite}}$
455 slope ($m = -0.61$, -0.60 , and -0.30 , respectively; Tab. 2), with *Uvigerina* spp.
456 characterized by a lower sensitivity. In these species/genus, a one-unit change in Ω_{calcite}
457 drives a Mg/Ca change equivalent to a $\sim 10^{\circ}\text{C}$ change in temperature, although we
458 stress that the nonlinear parameterization of the seawater carbonate system in the
459 regression model means that drawing a direct equivalence between the temperature
460 and Ω_{calcite} slopes is not possible. Conversely, *C. pachyderma* and *Oridorsalis* spp. are
461 not characterized by a significant sensitivity to the carbonate system. While this finding
462 agrees with previous studies that have reported a strong $\Delta[\text{CO}_3^{2-}]$ effect on Mg/Ca in *L.*
463 *wuellerstorfi* and *C. mundulus* (Elderfield et al., 2006; Yu and Elderfield, 2008; Yu et al.,
464 2014), and no resolvable impact on *Oridorsalis* spp. (Rathmann and Kuhnert, 2008;
465 Elderfield et al., 2010), we note that these datasets suffer from strong covariance
466 between T and $\Omega^{-2}_{\text{calcite}}$ (see above and the Ω -T column in Table 2), which may mask a
467 carbonate system control. More broadly, we acknowledge that a multiple linear
468 regression approach cannot remove collinearity in these (or any) regressions,
469 impacting the significance of the predictors and standard error of the overall model
470 (e.g., Allen, 1997). Therefore, the interpretation of the influence of each variable on
471 Mg/Ca should be made with caution where covariance in the underlying parameters is
472 an issue (Table 2). In addition, the *C. mundulus* dataset spans a narrow Ω_{calcite} range,
473 such that additional data are required to properly assess whether such a relationship
474 exists in this case. Removing $\Omega^{-2}_{\text{calcite}}$ from the regression model in the case of the two
475 species that lack a significant $\Omega^{-2}_{\text{calcite}}$ coefficient results in no significant change in the
476 estimated Mg/Ca-T sensitivity of either *O. umbonatus* (7.9 ± 1.3 versus $7.6 \pm 1.0\text{ \%}^{\circ}\text{C}$) or
477 *C. pachyderma* (5.8 ± 1.0 versus $5.9 \pm 1.0\text{ \%}^{\circ}\text{C}$).

478 To assess whether our linear regression models capture the key structure of
479 the datasets, we corrected Mg-Sr-B/Ca for the influence of all but one of the
480 regression parameters in turn, and plotted against the parameter for which no

481 correction was applied (see Text S1), as a means of determining whether there is any
482 residual structure in the data. The results of this analysis are shown as a function of
483 temperature, $\Omega^{-2}_{\text{calcite}}$, and DIC, with the alternate parameters subtracted out (Figure 4).
484 Specifically: 1) $\Omega^{-2}_{\text{calcite}}$ -corrected Mg/Ca displayed as a function of temperature shows
485 Mg/Ca values with the effects of $\Omega^{-2}_{\text{calcite}}$ and the intercept term removed ($\Omega^{-2} \bullet x_3 + x_1$);
486 2) T-corrected Mg/Ca displayed as a function of $\Omega^{-2}_{\text{calcite}}$ shows test Mg/Ca with the
487 effect of temperature and the intercept term subtracted out ($T \bullet x_2 + x_1$); and 3) T-and
488 $\Omega^{-2}_{\text{calcite}}$ -corrected Mg/Ca displayed as a function of DIC is a transformation in the
489 direction of the y axis of a plot of the full model residuals. This latter analysis shows
490 that while the inclusion of DIC in the regression model overall does suggest a
491 sensitivity of Mg/Ca to DIC in *L. wuellerstorfi* and *Cibicidoides* spp. (Supporting
492 information, Table S4), we find no significant residual relationship between corrected
493 Mg/Ca and DIC once the temperature and $\Omega^{-2}_{\text{calcite}}$ sensitivities have been removed
494 from these datasets, suggesting that the former finding is an artifact of (e.g.) DIC and
495 temperature covariance in the dataset.

496 3.2 Sr/Ca

497 The overall Sr/Ca dataset clusters into two groups, with *L. wuellerstorfi*, *C.*
498 *mundulus*, and *C. pachyderma* characterized by higher values than *O. umbonatus* and
499 *Uvigerina* spp. at similar temperatures (~1.0-1.6 versus ~0.8-1.0 mmol/mol
500 respectively; Fig. 2D). Otherwise, the most obvious difference between species is that
501 the degree of variance at a given temperature is greater in *L. wuellerstorfi* compared to
502 the other species, although we note that this dataset is also substantially larger (n =
503 305 versus 24-67).

504 Using the regression model described above ($\text{Sr/Ca} = X_1 + X_2 \bullet T + X_3 \bullet \Omega$), the
505 overall model R^2 ranges between 0.33-0.84 and RMSE between 0.051 - 0.077 (in units
506 of mmol/mol; Tab. 2). Only *C. pachyderma* is characterized by a significant Sr/Ca-T
507 coefficient of -0.016 mmol/mol per °C (Fig. S6M). Conversely, all species investigated
508 here are characterized by the same Sr/Ca- Ω_{calcite} relationship within uncertainty ($x_3 =$
509 0.13 to 0.17), with the exception of *O. umbonatus* (0.08). These findings are in broad
510 agreement with previous analyses of the underlying datasets (Lawson et al., 2024; Yu

511 et al., 2014), although we find no significant Sr-carbonate system relationship in *O.*
512 *umbonatus*, in contrast to Rathmann & Kuhnert (2008). As in the case of Mg/Ca, the
513 underlying covariance between temperature and Ω_{calcite} means that it is challenging to
514 separate the effect of these two factors on test chemistry in the case of *Uvigerina* spp.
515 and *C. pachyderma*. We also note that neither temperature nor Ω_{calcite} is a good
516 predictor of *C. mundulus* Sr/Ca (overall model $R^2 = 0.37$), a result of the narrow range
517 of sample site temperature (95% of the dataset falls between 0.5-4.5°C) and Ω_{calcite} (~ 1 -
518 2) in the context of the variance in the Sr/Ca data (Fig. 2; Fig. S6K).

519 Reformulating the regression models to exclude nonsignificant terms (i.e.,
520 removing the temperature coefficient (Fig. 3B) from the regression models for all
521 species except *C. pachyderma*) results in no substantial change in the overall model R^2
522 or RMSE, but does result in an overall marginally shallower Sr/Ca- Ω_{calcite} slope (0.08-
523 0.19 mmol/mol per unit; Fig. 3E).

524 We find a significant, albeit low slope between residual Sr/Ca and DIC (Fig. 4F)
525 in the case of *L. wuellerstorfi*, with a slope of 3.9×10^{-4} mmol/mol per $\mu\text{mol/kg}$ (Fig.
526 S6C). However, within the context of the model R^2 (0.05) and RMSE (0.062), the
527 uncertainty in reconstructed DIC using these slopes would be similar in magnitude to
528 the entire modern ocean range, such that this may be of mechanistic but not practical
529 interest.

530 3.3 B/Ca

531 B/Ca values are lower in *Uvigerina* spp. and *O. umbonatus* than the other
532 species (~ 0 -100 versus 100-250 $\mu\text{mol/mol}$ Fig. 2G-I). B/Ca ratios appear to be positively
533 correlated to temperature for *L. wuellerstorfi* and *C. mundulus* (Fig. S7A and J), which is
534 almost certainly an artifact of the covariance of temperature and the carbonate system
535 (discussed below). B/Ca ratios in *L. wuellerstorfi* collected from the Norwegian Sea
536 (Elderfield et al., 2006; Yu and Elderfield, 2007) at temperatures $< 0^\circ\text{C}$ appear to be
537 offset from the rest of the dataset (Fig. 2G); however, this can be reconciled by
538 considering the unusual carbonate chemistry of these sites. We note that the majority
539 of the *L. wuellerstorfi* and *N. umbonifera* B/Ca data come from sites characterized by a
540 temperature $< 5^\circ\text{C}$, such that further data are required to unambiguously resolve

541 whether or not temperature variability independent of seawater carbonate chemistry
542 may play a role in B incorporation into these foraminifera.

543 Multiple linear regression models that include T and Ω_{calcite} can explain much of
544 the variance in the B/Ca datasets of *L. wuellerstorfi* and *C. mundulus* ($R^2 = 0.69$ and
545 0.73 and RMSE = 14.8 and 9.5 $\mu\text{mol/mol}$, respectively; Tab. 2), as previously described
546 (e.g., Yu and Elderfield, 2007; Rae et al., 2011). Based largely on the *Uvigerina* spp.
547 data published in this paper, we also demonstrate that this is the case for this genus
548 ($R^2 = 0.71$, RMSE = 7.3 $\mu\text{mol/mol}$). However, this is not the case for *O. umbonatus* or *N.*
549 *umbonifera* ($R^2 = 0.21$ -0.60, RMSE = 11.4-16.5 $\mu\text{mol/mol}$). This division coincides with
550 the size of the underlying datasets, with the two species in the latter group
551 represented by 12-22 data points (compared to 38-181 in the former group). As such,
552 this finding may simply reflect that further data are required before boron
553 incorporation into *O. umbonatus*, and *N. umbonifera* can be properly assessed.

554 None of the species considered here are characterized by a significant
555 temperature coefficient in the B/Ca regression models except for *Uvigerina* spp., which
556 has a temperature slope of -1.9 $\mu\text{mol/mol}$ per $^{\circ}\text{C}$. In contrast, *L. wuellerstorfi*, *C.*
557 *mundulus*, and *N. umbonifera* are characterized by significant B/Ca- Ω_{calcite} sensitivities
558 (Fig. 2H). We note that we choose Ω_{calcite} as an independent variable for the sake of
559 ease of use and consistency in dataset treatment among elemental systems but
560 studies on planktonic foraminifera species demonstrated that B/Ca may be more
561 appropriately related to a more complex boron/carbon system parameter such as
562 $[\text{B}(\text{OH})_4^-]/\text{DIC}$ (e.g., Foster, 2008); on the other hand, studies focused on low-Mg
563 benthic foraminifera have shown that B/Ca better correlates with deep water
564 carbonate saturation state (e.g., Yu and Elderfield, 2007). While this is discussed in
565 more detail below, we note that there is no systematic trend in the residuals of the
566 B/Ca- Ω_{calcite} regressions, suggesting that a more complex approach may be unnecessary
567 for practical purposes (i.e. balancing the goal of building regressions that are
568 mechanistically meaningful with the practicalities of constraining multiple aspects of
569 seawater carbon and boron chemistry through time). Removing temperature from the
570 model for all species does not substantially change the goodness of fit of the
571 regressions, but it does result in a lower slope between B/Ca and Ω_{calcite} in the case of

572 *Uvigerina* spp. and *O. umbonatus*, this change in slope is justifiable given the
573 nonsignificant temperature coefficient in almost all cases. On the other hand, including
574 temperature in the model results in very similar *Uvigerina* spp., *C. mundulus*, and *O.*
575 *umbonatus* B/Ca- Ω_{calcite} sensitivities (33 to 45 $\mu\text{mol/mol}$ per Ω_{calcite} ; Fig. 3C), possibly
576 indicating mechanistic similarity in boron incorporation when temperature is
577 accounted for, while removing this component results in an even spread between 1
578 and 115 $\mu\text{mol/mol}$ (Fig. 3F). However, we focus on models without a temperature
579 component below, given the overall minor impact on the regression model statistics
580 and nonsignificant temperature coefficient, and given that omitting it simplifies the
581 practical application of B/Ca.

582 We find a significant albeit shallow residual correlation between B/Ca and DIC
583 in *L. wuellerstorfi* and *C. mundulus*, with slopes of -0.27 and $0.091 \mu\text{mol/mol}$ per
584 $\mu\text{mol/kg}$ (Figs. 4I, S8C and L). As for the Sr/Ca-DIC relationships discussed above, these
585 residual relationships imply possible B/Ca changes in response to DIC far smaller than
586 the magnitude of the residuals in the overall regression, given the magnitude of the
587 DIC variability across the modern ocean.

588 3.4 Li/Ca

589 Li/Ca data are available only for *O. umbonatus*, *Uvigerina* spp., and *C.*
590 *pachyderma*, and are, in isolation, poorly or noncorrelated to temperature in all cases
591 (Fig. 2J; note that six of the nine *C. mundulus* Li/Ca data points are from sites with a
592 temperature range of $\sim 1^\circ\text{C}$, and therefore we do not interpret the data). Likewise,
593 there is no significant relationship between *Uvigerina* spp. and *C. pachyderma* Li/Ca
594 and Ω_{calcite} or DIC (Fig. 2K, and L), although residual *O. umbonatus* Li/Ca is positively
595 correlated with Ω_{calcite} ($R^2 = 0.52$; Fig. 2K and L, and Fig. S9E). Given the absence of a
596 clear temperature or carbonate chemistry control on Li incorporation, as shown in
597 previous studies, we focus on combining these data with Mg/Ca. Linear Mg/Li
598 regression models are characterized by a far lower degree of relative residual variance
599 compared to Mg/Ca for the two species for which sufficient data exist (*Uvigerina* spp.
600 and *C. pachyderma*) (Fig. 5).

601 Specifically, *Uvigerina* spp. and *C. pachyderma* Mg/Li fit to temperature and
602 Ω_{calcite} results in overall model R^2 of 0.64 and 0.80, and RMSE of 0.013 and 0.023,
603 respectively (Tab. 2). We find no significant relationship between Ω_{calcite} and Mg/Li in
604 the case of *Uvigerina* spp., while the *C. pachyderma* data come from sites spanning a
605 range of Ω_{calcite} that is too small to test whether Mg/Li responds exclusively to
606 temperature in less oversaturated seawater. Removing Ω_{calcite} from the regression for
607 *Uvigerina* spp. and *C. pachyderma* does not substantially change the goodness of fit (R^2
608 = 0.63 and 0.79, RMSE = 1.3 and 2.4×10^{-2}), but does result in a moderately lower
609 Mg/Li-temperature sensitivity of 0.38 and 1.27×10^{-2} rather than 0.43 and 1.39×10^{-2}
610 mol/mmol per °C, implying a negative, albeit statistically nonsignificant Ω_{calcite} effect on
611 Mg/Li (Table 2). There is no significant residual Mg/Li sensitivity to DIC in either species
612 (see Fig. 5 and discussion in section 4.1).

613 **4. Discussion**

614 **4.1 Practical considerations in the application of elemental chemistry in**
615 **benthic foraminifera to paleoenvironmental reconstructions**

616 Our regression (re)analysis demonstrates that temperature can explain the
617 greatest portion of the variance in Mg/Ca for *O. umbonatus* and *C. pachyderma*, based
618 on the ratio of the calibration slopes to the range of these parameters in the
619 calibration dataset (20°C for temperature and 4 for Ω_{calcite}). In contrast, species with a
620 shallower temperature slope (*L. wuellerstorfi*, *Uvigerina* spp., and *C. mundulus*) are
621 characterized by a stronger Ω_{calcite} influence. In the case of Sr/Ca and B/Ca, Ω_{calcite} can
622 explain the greatest portion of the variance in these elements in all cases except B/Ca
623 and Sr/Ca in *O. umbonatus*, which cannot be predicted from any combination of
624 temperature and seawater carbonate based on the data currently available. This
625 finding also emerges when limiting the analysis to samples from locations with
626 temperature <5°C, which deviate from the general Sr-B/Ca-temperature but not the
627 Sr-B/Ca- Ω_{calcite} trend (Figures 2D and 2G), suggesting that Ω_{calcite} is indeed
628 mechanistically responsible for the majority of the observed variance in Sr and B; this
629 demonstrates that in these instances, our calibration approach correctly attributes
630 variability to Ω_{calcite} bearing in mind the Ω_{calcite} -T covariance in the dataset. This is in

631 good agreement with virtually all of the studies that reported the datasets analyzed
632 here (Rosenthal et al., 1997; Martin et al., 2002; Lear et al., 2002; Rathmann et al.,
633 2004; Lear and Rosenthal, 2006; Elderfield et al., 2006; Marchitto et al., 2007a; Yu and
634 Elderfield, 2007; Yu and Elderfield, 2008; Bryan and Marchitto, 2008; Healey et al.,
635 2008; Raitzsch et al., 2008, 2011; Lear et al., 2010; Brown et al., 2011; Rae et al., 2011;
636 Tisserand et al., 2013; Yu et al., 2014; Kubota et al., 2015; Lo Giudice Cappelli et al.,
637 2015; Gussone et al., 2016; Stirpe et al., 2021; Lawson et al., 2024), and is encouraging
638 in that, while we show that a multivariate approach is desirable and yields
639 substantially different sensitivities in many cases (described in more detail below), no
640 element/Ca response has been incorrectly apportioned to an underlying independent
641 variable as a result of covariance in the modern ocean (Fig. S2). In addition, our
642 analysis enables us to provide recommendations when designing paleoenvironmental
643 reconstructions based on benthic foraminiferal elemental test chemistry, discussed in
644 more detail below.

645 We find that the predictive power of Mg/Ca measured in the species
646 considered here principally depends on the sensitivity of the Mg/Ca-temperature
647 relationship (Fig. 3). Unexplained variance in the data is greatest in the two species
648 characterized by a significant Mg/Ca- $\Omega^{-2}_{\text{calcite}}$ sensitivity (*L. wuellerstorfi* and *C.
649 mundulus*; Fig. S3B and K) as well as *Uvigerina* spp., which has the lowest Mg/Ca-
650 temperature sensitivity (evident in the latter case in that the range of observed Mg/Ca
651 between 0-2°C is equivalent to that observed between 0-20°C; Fig. S3D). To quantify
652 the degree to which this is likely to impact temperature reconstructions, we calculated
653 the inverse prediction interval (IPI) for the key controlling seawater parameter
654 (temperature in the case of Mg/Ca). The IPI gives the 95% confidence interval of the
655 residual rotated onto the x-axis (McClelland et al., 2021) and thus provides an estimate
656 of the uncertainty associated with the approach based on the unexplained variance in
657 the dataset. We did this by first calculating temperature from Mg/Ca for all samples in
658 the database (Fig. 6) using the multiple linear regression models presented here (Tab.
659 2). Then, we calculated the IPI of the predicted versus observed temperature. For
660 example, the IPI of the *L. wuellerstorfi* Mg/Ca-derived temperatures is $\pm 12.1^{\circ}\text{C}$, that is,
661 95% of the predicted temperatures fall within this range of the *in situ* world ocean

662 atlas temperature (Fig. 6A). *C. mundulus* and *Uvigerina* spp. also have relatively high
663 IPI, of 13.4 and 13.7°C, respectively (Fig. 6D, J). Given that these temperature ranges
664 are similar in magnitude to the entirety of the Neogene change in deep ocean
665 temperature (Cramer et al., 2011; Meckler et al., 2022; Evans et al., 2024), it may be
666 preferable to focus attention on species with less unexplained variance in the
667 calibration.

668 Of the species examined here, *O. umbonatus* and *C. pachyderma* are
669 characterized by substantially lower IPI of 4.8 and 5.6°C, largely resulting from their
670 greater Mg/Ca-T sensitivities (Fig. 6G and M). Coupled with their insensitivity of Mg/Ca
671 to seawater carbonate chemistry, these species thus represent good targets for
672 temperature reconstruction.

673 We note that the IPI represents a best-case uncertainty estimate compared to
674 paleo applications because *in situ* $\Omega^2_{\text{calcite}}$ measurements are used to account for this
675 term in core-top samples, and it is unlikely that $\Omega^2_{\text{calcite}}$ can be determined with a
676 similar degree of accuracy in the geological past. On the other hand, it may be possible
677 (or likely) that the unexplained variance we find in the core-top dataset might not
678 characterize geological records featured downcore, because downcore samples may
679 average over longer time intervals than core-top samples. We refer the reader to a
680 more detailed discussion of this topic in McClelland et al. (2021), but note that
681 determining whether this is the case is necessary before choosing the calibration
682 uncertainties versus IPI values when propagating uncertainty. For example, Elderfield
683 et al. (2012) and Woodard et al. (2014) observed: 1) a far smaller degree of variance
684 downcore across multiple glacial-interglacial cycles (G-IG) than the IPI values suggest
685 we should expect; 2) and coherent G-IG Mg/Ca-derived temperature changes. This is
686 encouraging and suggests that core-top variance may overestimate uncertainty
687 downcore, particularly when normalizing paleo records to modern deep ocean
688 temperature.

689 Mg/Li has been suggested as an alternative paleothermometer in foraminifera
690 and other marine calcifying organisms (e.g., Bryan and Marchitto, 2008; Lear et al.,
691 2010; Marchitto et al., 2018; Raddatz et al., 2013), with the rationale that normalizing

692 to Li potentially removes some of the vital effect-driven variance in Mg/Ca that results
693 from (e.g.) carbonate chemistry, $[Ca^{2+}]$, or Ca^{2+} utilization at the calcification site. Our
694 analysis confirms the conclusions reached in the original publications that regressions
695 based on Mg/Li rather than Mg/Ca under similar temperature ranges are characterized
696 by substantially improved IPIs for both *Uvigerina* spp. and *C. pachyderma* (Fig. 5G, H),
697 an improvement of a factor of 0.5 and 0.7, respectively. This suggests that Mg/Li
698 temperature reconstructions based on these two species may be less error-prone
699 compared to Mg/Ca despite the fact that Li is present at a much lower concentration in
700 foraminiferal calcite and is therefore analytically more challenging to measure.
701 However, in deeper time this approach may suffer from the limited information
702 available regarding past changes in seawater $[Li^+]$ (Weldegehebriel and Lowenstein,
703 2023) compared to Mg/Ca (e.g., Coggon et al., 2010; Gothmann et al., 2015; Evans et
704 al., 2018).

705 In addition to the above analysis, we present Mg/Ca multivariate regression
706 models that combine: 1) data from all the species considered here (Fig. 7A); and 2)
707 data from *C. mundulus* and *C. pachyderma* combined to generate a generic *Cibicidoides*
708 spp. calibration (Fig. 7B). The former may be useful when working with extinct species,
709 whereas the latter may be preferable where insufficient specimens exist to
710 differentiate between *Cibicidoides* species. The multispecies regression is
711 characterized by an IPI of 12.9 °C, generally worse than the species-specific approaches
712 discussed above (Fig. 6). Combining both *Cibicidoides* species results in a regression in
713 which both the temperature and $\Omega^2_{\text{calcite}}$ coefficients are significant (Tab. 2), and in
714 common with the individual species for which this is the case, an IPI around double
715 (10.8 °C; Fig. 7B) that of species that lack a carbonate chemistry sensitivity (*C.*
716 *pachyderma* and *O. umbonatus*). Whenever possible, more precise and accurate
717 temperature reconstructions can be achieved by utilizing *C. pachyderma* alone (cf. Fig.
718 7B and 5M).

719 We investigated the IPI of Sr/Ca-derived Ω_{calcite} using the same approach
720 described above to explore the magnitude of the unexplained variance in the
721 regression models. The IPI ranges from 0.60 to 1.3 (Fig. 6B, E, H, K, N), approximately
722 equivalent to $[CO_3^{2-}]$ of 55-120 $\mu\text{mol/kg}$, if $[Ca^{2+}]_{\text{sw}}$ is equal to modern. Combining both

723 *Cibicidoides* species together yields an overall model with similar statistics to those
724 described above ($R^2 = 0.81$, RMSE = 0.058 mmol/mol per unit, Ω_{calcite} slope = 0.10),
725 although a marginally worse Ω_{calcite} IPI (1.1; Fig. 7C), driven by the fact that *C.*
726 *pachyderma* has a significant Sr/Ca-temperature sensitivity, while *C. mundulus* does
727 not (Figs. 3E, S6J, K, M, N). As per the application of Mg/Ca data from these
728 foraminifera, it is clearly desirable to work with single species where possible.

729 Seawater carbonate chemistry reconstructions based on B/Ca are characterized
730 by substantially lower degrees of unexplained variance compared to Sr/Ca, with an
731 overall trade-off between ease of making accurate measurements and the utility of the
732 resulting data. The B/Ca- Ω_{calcite} IPI for *L. wuellerstorfi* and *C. pachyderma* are 0.37 and
733 0.60, respectively (Fig. 6C and L), which translates into an ~20-60% improvement in the
734 uncertainty with which Ω_{calcite} can be determined compared to using Sr/Ca in the same
735 species. We find a slightly worse *Uvigerina* spp. B/Ca- Ω_{calcite} IPI (0.91) and no predictive
736 power of *O. umbonatus* B/Ca in reconstructing Ω_{calcite} (Fig. 6F and I), but stress that this
737 may reflect the limited data available for these species, or variability in Ω_{calcite} in their
738 infaunal pore-water habitats compared to overlying bottom waters (see Rae et al.,
739 2011).

740 To produce the most accurate reconstructions, the above analysis indicates
741 that *O. umbonatus* is a good choice for Mg/Ca palaeothermometry, because it is
742 characterized by the lowest IPI and no resolvable sensitivity to $\Omega_{\text{calcite}}^2$. This species also
743 has one of the lowest Sr/Ca-derived Ω_{calcite} IPI, although we find no significant
744 relationship between B/Ca and Ω_{calcite} based on the available data. Given that B/Ca-
745 derived Ω_{calcite} reconstructions are generally more precise than those based on Sr/Ca
746 (Fig. 6), then the best way to derive information about temperature and seawater
747 carbonate chemistry would be to combine *O. umbonatus* Mg/Ca with *L. wuellerstorfi*
748 B/Ca (if there are sufficient specimens of these species). This strategy will produce
749 temperature and Ω_{calcite} reconstructions with the lowest uncertainty propagated based
750 on the unexplained variance in the regression models, although this should again be
751 viewed with the caveat that core-top variance may not necessarily be present in
752 downcore reconstructions, as discussed above.

753 While we focus on unexplained variance in the regression models in the above
754 discussion, we also report uncertainty based only on the multivariate least squares
755 linear models, often termed ‘calibration uncertainty’ when applied to fossil datasets.
756 Following Gray & Evans (2019), we account for covariance between the model
757 intercept and temperature coefficient by bootstrapping the regression. We then derive
758 67% and 95% confidence intervals for our regression models using a Monte Carlo
759 approach, sampling all coefficients within uncertainty 10,000 times and calculating the
760 temperature difference from the central estimate for a range of test Mg/Ca covering
761 the majority of the data analyzed here (Fig. S11). At the extreme ends of the
762 investigated Mg/Ca range, the calibration uncertainty exceeds the IPIs described
763 above, but is generally <3 °C (95% CI) over the typical range of Mg/Ca values in the
764 case of all species without Mg/Ca sensitivity to seawater carbonate chemistry. In
765 contrast, those that do have this sensitivity (*L. wuellerstorfi*, *Uvigerina* spp., and *C.*
766 *mundulus*) are characterized by 95% confidence intervals derived from calibration
767 uncertainty alone of >10°C at the upper end of the range of test Mg/Ca for these
768 species, driven by a combination of the addition of an $\Omega^{-2}_{\text{calcite}}$ coefficient term to the
769 calibration and the fact that these species are characterized by the shallowest Mg/Ca-
770 temperature slopes (Fig. 3). Coupled with the fact that all the aforementioned
771 confidence intervals assume that Ω_{calcite} is perfectly known, this suggests that
772 temperature reconstructions based on Mg/Ca of these two species are likely to be
773 associated with relatively high uncertainty even if seawater carbonate chemistry can
774 be accounted for.

775 Here, we formulate regression models to a different carbonate system
776 parameter compared to most previous studies (Ω and Ω^{-2} versus $\Delta[\text{CO}_3^{2-}]$), to facilitate
777 the applicability of these models to deep time (pre-Pleistocene) intervals when $[\text{Ca}^{2+}]_{\text{sw}}$
778 was different than today. As such, comparison between the slopes of our multivariate
779 models and those that have been published for Sr/Ca and B/Ca is not possible.
780 However, we can highlight the difference in Mg/Ca-temperature slope between this
781 and previous analyses. Compared to regression analyses that only consider
782 temperature as the independent variable (e.g., Rosenthal et al., 1997; Martin et al.,
783 2002; Rathmann et al., 2004; Rosenthal et al., 2006), the temperature coefficients of

784 the models reported here (Tab. 2) are substantially lower (2.3-7.6 %/°C versus ~10
785 %/°C) for all species, as a result of the broad positive relationship between
786 temperature and $[CO_3^{2-}]$ in the ocean, and the positive relationship between Mg/Ca
787 and both of these factors. More recent studies that recognize this covariance have
788 broadly reported reduced temperature sensitivities (Lear et al., 2010), although those
789 reported here are nonetheless lower than some previous multivariate calibrations (Yu
790 and Elderfield, 2008) that do not consider the effect of both temperature and
791 carbonate chemistry (Raitzsch et al., 2008), or samples from a narrow range of
792 temperature (Healey et al., 2008). Overall, the multivariate models presented here
793 would result in a substantially larger magnitude of reconstructed temperature change
794 for a given measured Mg/Ca change compared to most previous calibrations, all else
795 being equal (i.e., no covarying change in $[CO_3^{2-}]$).

796 4.2. Element/Ca ratios in benthic foraminifera: a biomineralization and
797 ecological perspective

798 All species included in this study belong to the order Rotaliida (Lankester,
799 1885). Even so, they exhibit different Mg/Ca sensitivities to temperature, with *L.*
800 *wuellerstorfi* and *Uvigerina* spp. characterized by lower sensitivities than *O.*
801 *umbonatus*, *C. mundulus*, and *C. pachyderma*. In contrast, the sensitivity of Sr/Ca to
802 $\Omega_{calcite}$ is similar for every species investigated except for *O. umbonatus*, while the
803 response of B/Ca to $\Omega_{calcite}$ is more heterogeneous (Fig. 3). Although we might expect
804 biomineralization mechanisms to be broadly similar among species within the same
805 order (cf. de Nooijer et al., 2023), our data indicate that the relative importance of
806 different ion transport processes and/or ecological preferences might play a role in the
807 species element/Ca sensitivity to the environmental parameters considered here.

808 The exact nature of the biological influence on the chemical composition of
809 foraminifera is an active area of research (e.g., Branson et al., 2025). However, several
810 mechanisms have been hypothesized to influence the incorporation of elements other
811 than Ca within the foraminiferal test. For example, the presence of internal reservoirs
812 used during calcification in perforate foraminifera (ter Kuile and Erez, 1988) is likely to
813 be an important feature because it implies a fractionation step between the extraction

814 of elements from seawater and delivery to the calcification site (cf. Elderfield et al.,
815 1996), which may be influenced by biological processes. In particular, the presence of
816 an inorganic carbon pool might influence the incorporation of B, because the
817 incorporation of this element has been suggested to be driven by the growth-rate
818 dependent attachment/detachment rates of borate (B(OH)_4^-) and boric acid (B(OH)_3^-)
819 to the growing crystal surface, the former of which is de-hydroxylated to BO_3^{3-} and
820 substituted for HCO_3^- or CO_3^{2-} in the calcite lattice (e.g., Hemming and Hanson, 1992;
821 Branson et al., 2015). Similarly, the presence of an internal Ca pool might influence the
822 concentration of cations, like Mg, Sr, and Li at the calcification site if this store is a solid
823 phase (e.g., Erez, 2003). Additional biomineralization-related processes that have been
824 proposed as potential 'vital effects' include variable precipitation rates (e.g., Erez,
825 2003), Rayleigh fractionation (Elderfield et al., 1996; Evans et al., 2018), the effect that
826 symbionts have on the foraminifer's microenvironment (e.g., Sadekov et al., 2005; van
827 Dijk et al., 2019), transmembrane transport in combination with the passive transport
828 of other cations (Nehrke et al., 2013), the involvement of organic linings during
829 chamber formation (e.g., Bentov and Erez, 2006, Erez, 2003), diffusion of boric acid
830 along with CO_2 to the calcifying space (Gagnon et al., 2021), the possible involvement
831 of an amorphous calcium carbonate phase during calcification (Bentov and Erez, 2006;
832 Evans et al., 2020), and the ability of foraminifera to control the fluid composition in
833 the extracellular calcification space (e.g., Bentov and Erez, 2006) and/or to modify pH
834 during calcite precipitation (Evans et al., 2015). Banding could also arise during calcite
835 precipitation due to chemical oscillatory zoning (Branson et al., 2015).

836 Here, we show that the Mg/Ca of different species is affected differently by
837 oxidative versus reductive cleaning (*L. wuellerstorfi*, *Uvigerina* spp., and *O.*
838 *umbonatus*), while for *C. mundulus*, there is no significant difference in test Mg/Ca
839 when an oxidative versus reductive cleaning protocol is applied. While this may simply
840 reflect differential cleaning between studies focused on different regions, it may
841 alternatively be because different species are characterized by different degrees of
842 intratest elemental heterogeneity. For example, Glock et al. (2019) reported a
843 different distribution of nitrogen, sulfur, and iodine between cleaned (following Glock
844 et al., 2016) and uncleaned specimens. Importantly, in this study the inside of the test

845 wall was affected by the cleaning procedure even if the specimens were cleaned
846 without breaking the test. This has important implications because if the test is
847 characterized by a microporous texture, the reagents used during the oxidative
848 cleaning can penetrate deep within the test and remove parts of the organic phase
849 (Glock et al., 2019), potentially affecting the elements associated with it, to different
850 degrees in different species. Whether or not this applies to the proxy systems in the
851 species considered in this study requires intratest chemical heterogeneity to be
852 comprehensively documented in each species.

853 Ecological preferences, like microhabitats, can also impact a species' chemical
854 signature. For example, epifaunal species like *L. wuellerstorfi* and *C. mundulus* (e.g.,
855 Lutze and Thiel, 1989; Elderfield et al., 2006; Raitzsch et al., 2008; Rae et al., 2011) are
856 exposed to bottom water conditions. In contrast, shallow to deep infaunal species,
857 such as *O. umbonatus* and *Uvigerina* spp. (e.g., Brow et al., 2011; Rathmann and
858 Kuhnert, 2008; Stirpe et al., 2021), are affected by porewater chemistry (e.g., Jorissen,
859 1999). Infaunal benthic foraminiferal species are characterized by lower apparent
860 partition coefficients for Cd, Sr, and B compared to epifaunal species (e.g., Tachikawa
861 and Elderfield, 2002; Yu et al., 2014). In addition, pore waters often are characterized
862 by altered or somewhat buffered saturation state compared to bottom water due to
863 organic matter remineralization and CaCO_3 dissolution in the upper ~10 cm of
864 sediment (e.g., Zeebe 2007; Cetiner et al., 2025). Species that calcify from a pore
865 water-derived fluid have been suggested to have a weak or absent carbonate
866 saturation effect for this reason (e.g., Elderfield et al., 2006, Raitzsch et al., 2008;
867 Brown et al., 2011). This is in overall agreement with our observations, namely, that *L.*
868 *wuellerstorfi* and *C. mundulus* (epifaunal species) are characterized by a similar Mg/Ca
869 sensitivity, within uncertainties, to $\Omega^{-2}_{\text{calcite}}$ (Fig. 3A), whereas no resolvable sensitivity
870 exists for the infaunal species *O. umbonatus*. However, this is not the same for
871 *Uvigerina* spp., which exhibits a negative sensitivity to $\Omega^{-2}_{\text{calcite}}$. This may be a result of
872 variability in porewater chemistry and related differences in biomineralization
873 strategy, e.g., as an adaptation to undersaturated conditions. This is additionally
874 supported by the much lower B/Ca- Ω_{calcite} sensitivities of the infaunal species (Fig. 3C),
875 although not all species fit this pattern, with *C. pachyderma*, which is also an epifaunal

876 species (e.g., Rathburn et al., 1996; McCorkle et al., 1997) characterized by a $Mg/Ca - \Omega^{2}_{\text{calcite}}$
877 sensitivity similar to the infaunal species (Fig. 3A). In addition, while microhabitat
878 similarities can partially explain the observed Mg/Ca and B/Ca sensitivities to Ω_{calcite} ,
879 we observe similar $Sr/Ca - \Omega_{\text{calcite}}$ slopes among all species considered here. Why Mg/Ca
880 and B/Ca should be buffered in infaunal species while Sr/Ca is not is mechanistically
881 problematic and potentially suggests that the grouping of sensitivities between
882 infaunal/epifaunal species discussed above may be coincidental, and perhaps better
883 ascribed to (e.g.) biologically-driven differences in calcification site carbonate
884 chemistry among species that inhabit different niches, rather than the direct effect of
885 bottom/pore water chemistry.

886 Interestingly, *C. pachyderma* juveniles were observed up to 4.6 cm deep in the
887 sediment, which suggests that this species changes microhabitat during different
888 ontogenetic stages (Rathburn and Corliss, 1994). This could explain the lack of $\Omega^{2}_{\text{calcite}}$
889 influence on *C. pachyderma* Mg/Ca (Table 2) as observed for *O. umboatus* and
890 *Uvigerina* spp.. Microhabitat changes during a species life cycle have also been invoked
891 to explain the Mg/Ca offset between *C. mundulus* and *L. wuellerstorfi*, with the former
892 migrating within the sediment during ontogenesis (Rathburn and Corliss, 1994;
893 Raitzsch et al., 2008). However, it should also be borne in mind that benthic
894 foraminiferal distributions in sediment can vary geographically and temporally (e.g.,
895 Kitazato, 1994; Jorissen, 1999). For example, *L. wuellerstorfi* is commonly described as
896 epifaunal, although the presence of this species below 1 cm was also reported (e.g.,
897 Rathburn and Corliss, 1994), possibly as a consequence of bioturbation in deep-sea
898 sediments (e.g., Jorissen, 1999).

899 Overall, we conclude that while microhabitat preferences broadly align with
900 test geochemical relationships, this factor alone cannot explain differential inter-
901 species element/Ca sensitivity to environmental parameters. Further studies are
902 necessary to better understand different species' ecological preferences and their
903 relationship with element/Ca incorporation in foraminiferal tests.

904 **4.3. ElCaRBenthic: Calibration implementation, uncertainty**
905 **propagation, and downcore application examples**

906 To apply the calibrations described above (Sec. 3.2 and 4.1), we provide
907 *EICaRBenthic* (<https://www.github.com/dbjevans/EICaRBenthic>), a Matlab script
908 for the calculation of temperature, Ω_{calcite} , and $[\text{CO}_3^{2-}]$ from benthic foraminiferal
909 Mg/Ca, Sr/Ca, B/Ca, or Mg/Li data. Following *MgCaRB* (for planktonic foraminifera;
910 Gray & Evans, 2019), the program provides a convenient implementation of the above
911 multivariate regression models. In addition, *EICaRBenthic* can account for nonthermal
912 controls on Mg/Ca and non-carbonate chemistry controls on Sr/Ca (where relevant),
913 and it fully propagates uncertainty associated with data, calibration, and all relevant
914 input datasets (seawater chemistry, Ω_{calcite} in the case of Mg/Ca and temperature in the
915 case of Sr/Ca, the relationship between seawater and test chemistry). The calibration
916 inverse prediction intervals (IPI) are also returned. While we focus on Ω_{calcite} and $\Omega^{2-}_{\text{calcite}}$
917 in the regression models, given that this is easier to implement over a range of
918 timescales, *EICaRBenthic* also converts Ω_{calcite} to $[\text{CO}_3^{2-}]$ using optional temperature and
919 depth inputs (or using Mg/Ca-derived temperature). Where a given system is sensitive
920 to more than one factor (e.g., Mg/Ca in the case of *L. wuellerstorfi*, Sr/Ca in the case of
921 *C. pachyderma*; Fig. 3), more than one trace element dataset can be input at the same
922 time to simultaneously solve for both dependent variables. The script derives the
923 multivariate calibrations discussed above from the database file provided here each
924 time it is run, such that it can easily be applied to new datasets or to an updated
925 version of our compilation database in future.

926 To achieve this, several reference datasets are required to correct foraminiferal
927 test chemistry datasets for changes in the minor and major ion composition of
928 seawater for application in deep time (>1 Ma). Specifically, we use a compilation of
929 Mg/Ca_{sw} datasets (Dickson, 2004; Coggon et al., 2010; Brennan et al., 2013; Rausch et
930 al., 2013; Gothmann et al., 2015; Evans et al., 2016, 2018; Zhou et al., 2020) with
931 uncertainty derived from the 2.5th and 97.5th percentiles of 10³ bootstrapped LOWESS
932 fits to the data, resampled within uncertainty. The seawater [Li⁺] and [Ca²⁺] datasets
933 required to correct Mg/Li (via Li/Ca) are, to our knowledge, only available in the former
934 case from modeling the composition of fluid inclusions measured in marine evaporitic
935 sequences (Weldeghebriel and Lowenstein, 2023). As such, these [Li⁺]_{sw} data were
936 combined with fluid inclusion and foraminiferal [Ca²⁺]_{sw} reconstructions (Brennan et

937 al., 2013; Zhou et al., 2021), resampled within uncertainty, and fit using a LOWESS
938 function 10³ times, with the propagated uncertainty taken from the 2.5th and 97.5th
939 percentiles of the resulting dataset.

940 In addition to knowledge of how the composition of seawater has changed
941 through geological time, correcting foraminiferal test chemistry data requires that the
942 shape of the relationship between seawater and test elemental ratios is known
943 (exponent H in Eq. 5). This has been argued to be nonlinear in the case of Mg/Ca
944 (Segev and Erez, 2006; Hasiuk and Lohmann, 2010; Evans and Müller, 2012; Evans et
945 al., 2015a), although linear (i.e., a constant apparent distribution coefficient)
946 relationships have been applied (e.g., Lear et al., 2000) and even the absence of such a
947 relationship has been argued for (Lear et al. 2015). Given that no consensus and/or
948 insufficient data exist to constrain this aspect of deep ocean benthic foraminifera, we
949 leave H and associated uncertainty as a required *EiCaRBenthic* input field when
950 processing samples older than 800 ka, where:

951
$$\text{Mg/Ca}_{\text{corrected}} = \text{Mg/Ca}_{\text{measured}} \times 5.2^H / \text{Mg/Ca}_{\text{sw}}^H \quad (\text{Eq. 5})$$

952 Mg/Ca_{sw} is the value at a time interval of interest, and 5.2 is the modern ratio. In the
953 examples below, we follow the recommendation of Evans et al. (2016) regarding the
954 value of H for the species utilized here, but note that this analysis likely requires
955 revision in light of the multivariate calibration models presented here.

956 Foraminiferal Li/Ca are corrected for changes in Li/Ca_{sw} using a constant
957 distribution coefficient, because the limited amount of data from foraminifera cultured
958 in seawater with varying Li/Ca indicates that foraminiferal test and seawater Li/Ca are
959 linearly related (Hauzer et al., 2024):

960
$$\text{Li/Ca}_{\text{corrected}} = \text{Li/Ca}_{\text{measured}} \times 2.6 / \text{Li/Ca}_{\text{sw}} \quad (\text{Eq. 6})$$

961 where Li/Ca_{sw} is that at a time interval of interest and 2.6 is the modern ratio (in
962 mmol/mol). We do not include a correction for past changes in Sr/Ca_{sw} because the
963 (high quality) available data suggest that [Sr²⁺]_{sw} and [Ca²⁺]_{sw} have covaried over the
964 course of the Cenozoic (Gothmann et al., 2015), but note that this could be accounted
965 for by correcting the data along similar lines to our processing of Li/Ca (Eq. 6) if
966 desired. Likewise, no attempt has been made to correct B/Ca data for past changes in

967 seawater chemistry given that the secular evolution of the total boron seawater
968 concentration is very poorly constrained (Lemarchand et al., 2002).

969 **Example application 1: late Pleistocene deep ocean carbonate chemistry.** To
970 demonstrate the utility of B/Ca and Sr/Ca for carbonate chemistry reconstruction, we
971 applied our *L. wuellerstorfi* B/Ca and Sr/Ca calibrations to site VM28-122 (3620 m,
972 Caribbean Basin, e.g., Yu et al., 2010; see Sec. 2.3 for further details and data sources).
973 We derive Ω_{calcite} directly from the above calibrations, as well as $[\text{CO}_3^{2-}]$ calculated from
974 Ω_{calcite} using the depth of the core site and assuming $T = 3^\circ\text{C}$ in the absence, to our
975 knowledge, of published Mg/Ca data for this site (Figure 8). We reconstructed glacial-
976 interglacial Ω_{calcite} changes of $\sim 0.5\text{--}0.75$ units and B/Ca-derived $[\text{CO}_3^{2-}] \sim 40 \mu\text{mol/kg}$
977 higher during the last two glacials compared to the Holocene. This latter
978 reconstruction agrees with Yu et al. (2013) to better than $10 \mu\text{mol/kg}$ throughout the
979 duration of the record. While this reflects that our *L. wuellerstorfi* B/Ca calibration
980 does not differ substantially from previous approaches, we fully propagate all sources
981 of uncertainty, demonstrating significant G-IG $[\text{CO}_3^{2-}]$ change when considering
982 calibration uncertainty alone, although with an IPI of the same order of magnitude as
983 the late Pleistocene variability. Moreover, our new Sr/Ca-derived Ω_{calcite} and $[\text{CO}_3^{2-}]$
984 reconstructions are in good agreement with those based on B/Ca (Fig. 8). While several
985 previous studies have noted the downcore correlation between B/Ca and Sr/Ca (Yu et
986 al., 2013; Lo Giudice Cappelli et al., 2015), and Sr/Ca-derived $\Delta[\text{CO}_3^{2-}]$ have been
987 reported based on a regression of down core Sr/Ca against B/Ca-derived $\Delta[\text{CO}_3^{2-}]$
988 (Lawson et al., 2024), our analysis may represent the first truly independent Sr/Ca-
989 based reconstruction; that is, the Sr/Ca-derived reconstructions (Fig. 8) are based on
990 independent core-top data rather than an ad hoc downcore correlation. The B/Ca and
991 Sr/Ca datasets show a good degree of coherence over the last two G-IG cycles,
992 further demonstrating the utility of benthic foraminiferal Sr/Ca data for the purposes
993 of carbonate chemistry reconstruction, especially given that it is more routinely and
994 more easily measured, as

995 as noted by Lawson et al. (2024). However, as discussed in Sec. 4.1, Sr/Ca based
996 carbonate chemistry reconstructions are a factor of ~ 2 times more uncertain than
997 those based on B/Ca. In addition, we observe substantial offsets in certain intervals;

998 for example, Ω_{calcite} during the last interglacial period was ~ 0.3 units lower based on
999 Sr/Ca, outside of the combined calibration uncertainty of the two approaches. This
1000 highlights that calibration uncertainty alone cannot account for all sources of
1001 uncertainty when applying these systems downcore (in contrast to the IPI intervals in
1002 this case) and warrants caution when interpreting the details of reconstructions such
1003 as these at this scale.

1004 **Example application 2: Neogene temperature and carbonate chemistry changes.**
1005 Using ODP Site 806 as an example, we used data (see Sec. 2.3) reported by Lear et al.
1006 (2003; 2015) and Lear & Rosenthal (2006) to assess Mg/Ca and Sr/Ca-derived
1007 temperature and Ω_{calcite} reconstructions across the last 15 Myr from *L. wuellerstorfi*, *O.*
1008 *umbonatus*, and *C. mundulus* using the multivariate calibration models presented
1009 above. This enabled us to compare results from species that are and are not
1010 characterized by a Mg/Ca sensitivity to Ω_{calcite} , including the use of Sr/Ca to reconstruct
1011 and correct for the Ω_{calcite} influence on Mg/Ca-based temperature reconstructions
1012 across multi-million-year timescales.

1013 Calculating temperature from Mg/Ca without accounting for the possible
1014 effects of changing Ω_{calcite} (i.e., assuming $\Omega_{\text{calcite}} = 1 \pm 0.5$ throughout the duration of
1015 these records) results in reconstructions that deviate from each other by up to 8°C in
1016 the late Neogene (<4 Ma) and by $>15^{\circ}\text{C}$ before then (Fig. 9A). Although all
1017 reconstructions agreed within uncertainty, our results demonstrate the substantial
1018 influence of nonthermal factors on Mg/Ca. In particular, *L. wuellerstorfi* yielded an
1019 unreasonable degree of cooling in the last 5 Myr. At the same time, temperatures
1020 based on *C. mundulus* Mg/Ca were 5-10 $^{\circ}\text{C}$ higher between 5-10 Ma compared to the
1021 temperatures obtained using *O. umbonatus*, the only species in the comparison that is
1022 not characterized by a Mg/Ca- Ω_{calcite} sensitivity. Reconstructing Ω_{calcite} using the species-
1023 specific Sr/Ca calibrations presented here (Fig. 9C) results in somewhat different Ω_{calcite}
1024 reconstructions, albeit within uncertainty, with the *O. umbonatus* and *C. mundulus*
1025 records in broadly good agreement, while the reconstruction based on *L. wuellerstorfi*
1026 Sr/Ca would imply that Site 806 was undersaturated with respect to calcite prior to 4
1027 Ma. This disagreement may reflect habitat differences among the species analyzed, or
1028 could highlight the overall limitations of the approach given that a portion of the

1029 inferred change in Ω_{calcite} based on *L. wuellerstorfi* is partially driven by the
1030 unreasonably large reconstructed temperature change. We also stress that these
1031 reconstructions are within uncertainty of each other at the level of the calibration IPI.

1032 Our results emphasize that using species-specific results to correct for the
1033 influence of Ω_{calcite} on Mg/Ca temperature reconstructions results in much better
1034 agreement among the three species (Fig. 9B), even though the magnitude of cooling
1035 and Pleistocene temperatures remain unrealistic in the case of *L. wuellerstorfi*. Overall,
1036 this highlights: 1) the power of interpreting Mg/Ca and Sr/Ca (or B/Ca) data together,
1037 without which temperature would have been greatly overestimated using species with
1038 an Ω_{calcite} sensitivity; and 2) the large degree of inherent uncertainty in using *L.*
1039 *wuellerstorfi* Mg/Ca for temperature reconstruction.

1040 **5. Conclusions**

1041 We compiled published Mg/Ca, Sr/Ca, B/Ca, and Li/Ca core-top data for *C.*
1042 *mundulus*, *C. pachyderma*, *L. wuellerstorfi*, *N. umbonifera*, *O. umbonatus*, and
1043 *Uvigerina* spp. Multivariate regression models that relate these geochemical datasets
1044 to *in situ* environmental parameters demonstrate that temperature can explain much
1045 of the variance in the Mg/Ca datasets with species clustering in two main groups: (1)
1046 *O. umbonatus* and *C. pachyderma*, with sensitivities of 7.6 and 5.9%/°C respectively
1047 (similar to planktonic foraminifera); and (2) *L. wuellerstorfi*, *Uvigerina* spp., and *C.*
1048 *mundulus*, with sensitivities of 2.3-3.6%/°C (similar to inorganic calcite). As previously
1049 suggested, we find that Ω_{calcite} is a significant predictor of *C. mundulus* and *L.*
1050 *wuellerstorfi* Mg/Ca, with the implication that temperature reconstructions using these
1051 species are inherently more uncertain irrespective of whether Ω_{calcite} can be
1052 independently constrained. The combination of Mg/Ca with Li/Ca data results in a
1053 proxy (Mg/Li; e.g., Bryan & Marchitto, 2008) that is characterized by lower calibration
1054 uncertainties despite the possible analytical challenges compared to Mg/Ca (e.g.
1055 determination via ICP-OES may not be possible).

1056 Our results also demonstrate that all species, with the exception of *O.*
1057 *umbonatus*, are characterized by the same Sr/Ca- Ω_{calcite} relationship within uncertainty,
1058 suggesting that Sr/Ca data are likely to be a useful means for reconstructing Ω_{calcite} , as

1059 recently suggested by Lawson et al. (2024). Only *C. pachyderma* exhibits a significant
1060 Sr/Ca-T sensitivity, thus requiring correction before interpreting Sr/Ca data in terms of
1061 seawater carbonate chemistry.

1062 B/Ca is tightly correlated with Ω_{calcite} in *C. mundulus*, *L. wuellerstorfi*, and *N.*
1063 *umbonifera* without a resolvable temperature influence, whereas the infaunal species
1064 are characterized by substantially lower test B concentrations and no significant
1065 sensitivity to Ω_{calcite} in the case of *O. umbonatus* (although limited data are available).

1066 Overall, the most precise and accurate temperature reconstructions can be
1067 obtained from *O. umbonatus* Mg/Ca (or *C. pachyderma* Mg/Li), whereas the most
1068 precise and accurate Ω_{calcite} reconstructions can be generated using *L. wuellerstorfi*
1069 B/Ca. However, if data from only one species can be generated, and a choice of species
1070 is available, combined *C. pachyderma* Mg/Ca and Sr/Ca measurements provide the
1071 best compromise of temperature and Ω_{calcite} data quality, although there is no
1072 substitute for *L. wuellerstorfi* B/Ca-derived Ω_{calcite} reconstructions that is not associated
1073 with a substantial (factor ~2) increase in uncertainty. Combining the dataset for all the
1074 species analyzed here into genus or multispecies calibrations results in substantially
1075 larger uncertainties, emphasizing the importance of single-species analysis whenever
1076 possible.

1077 Finally, based on the species-specific multivariate calibrations generated here,
1078 we develop a new processing software (ElCaRBenthic) that provides a convenient way
1079 of applying our calibrations to fossil samples. The software can account for the thermal
1080 and non-thermal controls on Mg/Ca, Sr/Ca, B/Ca, and Mg/Li by accepting up to two
1081 datasets simultaneously (e.g., Mg/Ca and Sr/Ca), as well as fully propagating
1082 uncertainty and correcting for changes in seawater elemental chemistry throughout
1083 the Cenozoic. Applying this tool to two published example datasets demonstrates for
1084 the first time the accuracy (i.e. match to B/Ca derived estimates) of Sr/Ca-derived
1085 Ω_{calcite} reconstructions based on core-top calibrations vs. ad hoc down core Sr/Ca
1086 calibration. It also highlights the importance of accounting for Ω_{calcite} in Mg/Ca-based
1087 temperature reconstructions, which increases the agreement among records
1088 generated using different species.

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1102

1103 **Declaration of Competing Interest**

1104 The authors declare that they have no competing financial or personal interests that
1105 could have affected the work presented in this paper.

1106

1107 **Open Research**

1108 The dataset used in this research was deposited to the Pangaea data repository and
1109 can be found online (Nauter-Alves et al., XXXX; doi:XXXXXXXX). The “El/CaRBenthic”
1110 Matlab code is available on Github <https://github.com/dbjevans/ElCaRBenthic> (Evans,
1111 2025), and the code written to perform the data analysis and produce all figures is
1112 available at <https://zenodo.org/records/XXXXXXXX>.

1113

1114 **Figure 1.** Site locations of the core-top samples included in this study. The map was generated
1115 using QGIS software version 3.38. See Table 1 for references. The compiled dataset can be
1116 found on Pangaea (Nauter-Alves et al., XXXX; doi:XXXXXXXX).

1117

1118 **Figure 2.** Mg/Ca, Sr/Ca, B/Ca, and Li/Ca data are shown as a function of *in situ* temperature (A,
1119 D, G, J), Ω_{calcite} (C, F, I, L), and dissolved inorganic carbon concentration (DIC) (B, E, H, K) for all

1120 species considered in this study. See Figs. S3, S5, and S7 for equivalent plots with each species
1121 displayed in individual panels.

1122
1123 **Figure 3.** Temperature and $\Omega^{-2}_{\text{calcite}} / \Omega_{\text{calcite}}$ regression coefficients for the preferred models
1124 presented here for (A) Mg/Ca, (B) Sr/Ca, and (C) B/Ca; y axis units are the same as those given
1125 on the x axes, per $\Omega^{-2}_{\text{calcite}}$ or unit Ω_{calcite} . The vertical gray shaded region in panels A and D
1126 shows the range of Mg/Ca-T sensitivities for planktonic foraminifera for which sufficient
1127 laboratory culture data exist to characterize this relationship (Gray & Evans, 2019). Other gray
1128 bars delineate the position of zero. Error bars show 1SE coefficient uncertainties. (D-F) show
1129 equivalent plots based on alternative regression models in which nonsignificant parameters
1130 have been removed (e.g., the $\Omega^{-2}_{\text{calcite}}$ coefficient in the case of *C. pachyderma* and *O.*
1131 *umbonatus* Mg/Ca, see text for details).

1132
1133 **Figure 4.** Corrected test Mg/Ca (A, B, C), Sr/Ca (D, E, F), and B/Ca (G, H, I) data shown as a
1134 function of *in situ* temperature, Ω_{calcite} , and dissolved inorganic carbon concentration (DIC).
1135 Corrected data plotted as a function of temperature are those with the Ω_{calcite} component
1136 subtracted out (e.g., $\Omega^{-2}_{\text{calcite}}$ -corrected Mg/Ca). Corrected data plotted as a function of Ω_{calcite}
1137 are those with the temperature component subtracted out (e.g., T-corrected Mg/Ca).
1138 Corrected data plotted against DIC show the residual variance in the data after the
1139 temperature and Ω_{calcite} components have been removed (e.g., T and $\Omega^{-2}_{\text{calcite}}$ -corrected Mg/Ca)
1140 (see text for details). See Figs. S4,6,8 for equivalent plots with each species displayed in
1141 individual panels and significant linear regressions overlain. Note that we plot corrected Mg/Ca
1142 as a function of Ω_{calcite} for consistency, although the regression and correction were applied
1143 using $\Omega^{-2}_{\text{calcite}}$.

1144
1145 **Figure 5.** Controls on Mg/Li for the genus and species for which sufficient data are available to
1146 assess this (*Uvigerina* spp. and *C. pachyderma*). (A-C) Mg/Li shown as a function of *in situ*
1147 temperature, Ω_{calcite} , and DIC. (D-F) As panels A-C, except with the Ω_{calcite} (D) or temperature (E)
1148 component subtracted out. Panel (F) shows the residual variance in the data as a function of
1149 DIC. (G-H) *In situ* versus predicted temperature using our preferred regression model, with the
1150 dashed confidence intervals showing the inverse prediction interval (IPI, see Fig. 6 caption). (I)
1151 The temperature and Ω_{calcite} coefficients of the least-squares linear regression model.

1152
1153 **Figure 6.** *In situ* temperature versus predicted temperature based on Mg/Ca (left panels). *In*
1154 *situ* Ω_{calcite} versus predicted Ω_{calcite} based on Sr/Ca (central panels) and B/Ca (right panels)
1155 using the preferred regression model form (see text for additional details). The inverse
1156 prediction interval (the 2SD of residual variance rotated onto the x-axis; McClelland et al.,
1157 2021) is shown by dashed lines.

1158
1159 **Fig. 7.** *In situ* temperature versus predicted based on the (A) multi-species Mg/Ca
1160 and (B) *Cibicidoides* spp. Mg/Ca regressions presented here. (C) *In situ* versus predicted $\Omega^{-2}_{\text{calcite}}$
1161 based on the *Cibicoides* spp. Sr/Ca model presented here. The inverse prediction interval (the
1162 2SD of residual variance rotated onto the x axis; McClelland et al., 2021) is shown by dashed
1163 lines.

1164
1165 **Figure 8.** Example application of ElCaRBenthic to reconstruct seawater carbonate chemistry
1166 using *L. wuellerstorfi* B/Ca and Sr/Ca from VM28-122 (Caribbean, 3.6 km; with a sill depth of
1167 1.8 km, data from Yu et al. 2010a, 2010b, and 2014). The three shaded regions show the 67 %
1168 and 95 % calibration uncertainties, as well as the prediction intervals derived from the
1169 unexplained variance in the calibration dataset. The regression model is formulated in terms of
1170 Ω_{calcite} (panel A) and is converted into $[\text{CO}_3^{2-}]$ (panel B) using the core depth assuming a

1170 constant $S = 35$ and $T = 3$ °C, in order to facilitate comparison to a previous calibration
1171 approach.
1172
1173 **Figure 9.** Application example of ElCaRBenthic to reconstruct temperature and seawater
1174 carbonate chemistry at ODP Site 806 over the last 15 Myr (western Equatorial Pacific, 2.5 km,
1175 data from Lear et al., 2015). (A) Mg/Ca-derived temperature assuming constant $\Omega_{\text{calcite}} = 1$
1176 where relevant (*L. wuellerstorfi* and *C. mundulus*). The three shaded regions show the 67% and
1177 95% calibration uncertainties, as well as the prediction intervals derived from the unexplained
1178 variance in the calibration dataset. (B) Mg/Ca-derived temperature corrected for changes in
1179 Ω_{calcite} where relevant. (C) Sr/Ca-derived Ω_{calcite} . No correction has been applied for potential
1180 changes in Sr/Ca_{sw}, although Sr/Ca_{sw} uncertainty is fully propagated into the reconstructions.

1181

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