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of germanosilicate glasses.

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**Abstract**

The one-quantum photo-reactions of oxygen deficient centers (GODC) in  
H<sub>2</sub> loaded germanosilicate preforms and fibers are studied using time-  
resolved photo-luminescence and ESR spectroscopy.

# Effects of molecular hydrogen on low-intensity UV photochemistry of germanosilicate glasses.

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## Introduction.

Germanosilicate glasses are known to possess photorefractive (photosensitive) properties [1], that allows to fabricate refractive index gratings in germanosilicate optical fibers [2]. Hydrogen loading silica and germanosilicate glasses enhances its photosensitivity significantly [3-6]. Since the germanium-oxygen deficient center (GODC) is the main photosensitive defect in germanosilicate glasses, it is of particular importance to reveal the role of molecular H<sub>2</sub> in the processes of GODC phototransformation. In this work, we study the influence of molecular hydrogen impregnated into the glass network on the triplet decay time  $\tau_T$ , luminescence intensity ( $T_1 - S_0$ ,  $\lambda_{\max}=397$  nm) and efficiency of GODC photodecay when excited with low-intensity UV light.

## Experimental.

The triplet luminescence kinetics was recorded as follows. The pulse radiation of nitrogen laser ( $\lambda=337$  nm,  $\tau_{\text{pulse}}=10^{-8}$  s,  $f=100$  Hz) was focused on preliminary cleaned optical fiber, the maximum irradiation density was 0.23 J/cm<sup>2</sup>. A piece of the fiber to be irradiated was fixed in a thermostat with variable temperature (from 90 to 600 K).

In measurements of temperature dependence of triplet photoluminescence intensity  $I_T$  for GODC excited into the singlet absorption band (transition  $S_0-S_1$ ), we used the same apparatus, but with a deuterium lamp serving as a source of irradiation. Besides, for the high- and low-intensity UV irradiation the KrF excimer laser has been used.

All the optical fibers studied were drawn from preforms produced by the MCVD method. For ESR measurements the MCVD germanosilicate films were used.

The fibers and films were treated in a hydrogen atmosphere at room temperature in a chamber under a gas pressure of 24 atm for 2 weeks. The hydrogen concentration was about  $5 \times 10^{19}$  cm<sup>-3</sup> in the core of single mode fiber and  $3 \times 10^{19}$  cm<sup>-3</sup> in the film.

The ESR measurements of irradiated samples were performed with RE-1306 radiospectrometer working in the X-range ( $f=9.4$  GHz).

## Results and Discussion.

The presence of molecular hydrogen in the network of germanosilicate glass strongly affects the temperature dependences of the decay time of the excited state  $T_1$ . Figure 1a displays the temperature dependences of  $\tau_T$  for the unloaded (curves 1-3) and H<sub>2</sub>-loaded (curves 4) optical fibers with various GeO<sub>2</sub> concentrations in fiber core. Measurements were carried out routinely for a week. As hydrogen escapes from the fiber core, i.e., with a decrease in the hydrogen concentration, the curves 4  $\tau_T(T)$  become similar in shape to curve 1-3 for the hydrogen-free sample.

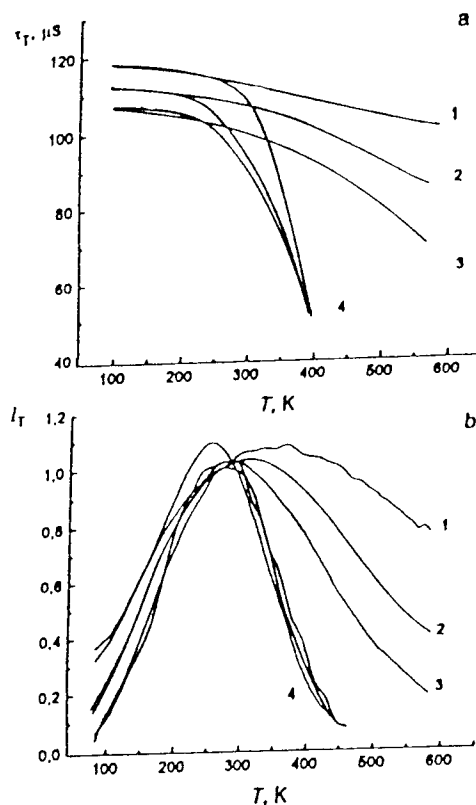


Fig.1. Temperature dependences of triplet lifetime (a) and intensity of triplet luminescence (b) in fibers with 2.3 (1), 11 (2) and 23 mol.%  $\text{GeO}_2$  (3);  $\text{H}_2$ -loaded fibers (4).

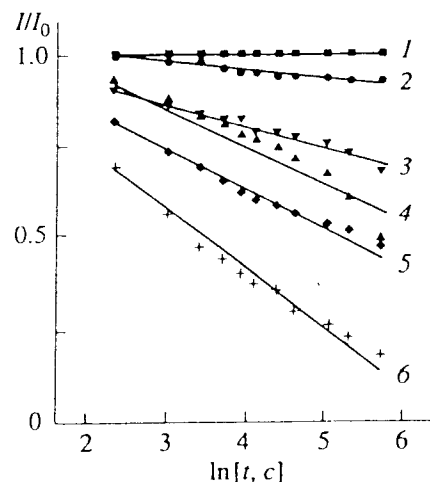


Fig.2. Dependence of the intensity of triplet photoluminescence in fiber with 23 mol.%  $\text{GeO}_2$  on the laser irradiation dose ( $\lambda=337 \text{ nm}$ ) at fixed irradiation temperatures  $T$ , K: (1) 290, (2) 390 and (3) 480 for the initial samples; and (4) 90, (5) 290, and (6) 390 for  $\text{H}_2$ -loaded samples.

Such a decrease in the decay time  $\tau_T$  in optical fibers saturated with hydrogen should manifest itself in the dependences  $I_T(T)$ , specifically at temperature higher than room temperature. Indeed, comparison of curves 1-3 and 4 in fig.1b shows that, at all germanium concentrations, the presence of molecular hydrogen in glasses leads to a strong decrease in  $I_T(T)$  with temperature. The maximum of hydrogen effect on  $I_T(T)$  appears in the fiber at the minimum  $\text{GeO}_2$  concentration, and eventually the high-temperature wings of curves 4 in Figs.1a and 1b turn out to be close to each other. This indicates that the presence of hydrogen molecules at high concentration almost completely determines the temperature behavior of the intensity of triplet photoluminescence of GODC, irrespective of the germanium concentration.

From our viewpoint, the results obtained from the hydrogen-containing fibers are explained by the participation of  $\text{H}_2$  in photochemical reactions of GODC in the excited  $T_1$  state. Note that the efficiency of these reactions increases with an increase in temperature of irradiation samples. We observed an increase in the efficiency of GODC decay in the  $\text{H}_2$ -loaded samples, which is a direct confirmation of the above reactions. The experimental data for hydrogen-containing samples with 23 mol.%  $\text{GeO}_2$  show photodecay efficiency of GODC considerably increases with an increase in the irradiation temperature. This manifests itself in a rise of the slope of the  $I_T(\ln t)$  dependences as the irradiation temperature increases (Fig.2).

It is seen from Fig.2 that no decay of GODC occurs in the hydrogen-free samples at room temperature and only at  $T > 300 \text{ K}$ , the efficiency of decay markedly increases. For  $\text{H}_2$ -loaded samples, the photodecay of GODC occurs even at temperature close to that of liquid nitrogen (Fig.2, curve 4); i.e.,  $\text{H}_2$  molecule interact with electron-excited GODC even at 90 K. No such effect is observed in similar processes in pure silica glasses [3]. This is caused by the fact, at liquid nitrogen temperatures, the diffusion of  $\text{H}_2$  molecules over the network of amorphous  $\text{SiO}_2$  appears to be "frozen", and at

rather small concentration of silicon ODC ( $10^{16} \text{ cm}^{-3}$ ), their nearest environment  $r < 10^{-8} \text{ cm}$  does not exhibit  $\text{H}_2$  molecules at the instant of photoexcitation. An estimated concentration of GODC in germanosilicate glasses is as high as  $10^{19}$ - $10^{20} \text{ cm}^{-3}$ . Because of excitation migration over GODC, there can appear some chance for frozen  $\text{H}_2$  molecules to be located in the nearest environment of GODC excited in  $T_1$  state. It is quite probable that the presence of hydrogen in the immediate vicinity of the excited center affects the efficiency of its decay.

The activation energy of the photodecay of the GODC was estimated from the experimental dependences  $\tau_T(T)$  and  $I_T(T)$  for the initial and  $\text{H}_2$ -loaded optical fibers. The activation energies  $E_a^\tau$  and  $E_a^I$  obtained from Arrhenius dependences  $\ln[1/\tau_T^H - 1/\tau_T]$  and  $\ln[I_T/I_T^H - 1]$  as functions  $1000/T$  are given in table:

C, mol.% GeO <sub>2</sub>	$E_a^\tau$ , eV	$E_a^I$ , eV	$E_a^R$ , eV
2.3	$0.37 \pm 0.05$	$0.36 \pm 0.05$	0.03
11	$0.22 \pm 0.05$	$0.36 \pm 0.05$	0.05
23	$0.25 \pm 0.05$	$0.37 \pm 0.05$	0.05

The activation energy of the reaction of GODC with hydrogen  $E_a^R$  can roughly be estimated from the experimental dependences of photodecay of GODC by analysis of their initial portions at the temperatures of 290 and 390 K (see Fig.2 and table).

One could expect the coincidence of activation energies  $E_a^\tau$ ,  $E_a^I$  and  $E_a^R$ , because in principle, they describe the same physical process: the interaction of molecular hydrogen with GODC in the metastable triplet  $T_1$  state. In our opinion, the activation energy  $E_a^\tau$  obtained from the temperature dependences of  $T_1$  state life time is closest to the realistic value. It is noteworthy that the activation energy  $E_a^R$  obtained from measurements of the photodecay yield is significantly less than  $E_a^\tau$  and  $E_a^I$ . This is because the  $E_a^R$  quantity is representative of some overall process in which the recombination (including both photo-stimulated processes and those after pulse), leading to the recovery of GODC, plays an important part. In addition, these processes are thermally activated.

It is known that at high-intensity laser irradiation of  $\text{H}_2$ -loaded silica glasses (two-quantum photochemistry of ODC) the passivation with hydrogen of laser induced radicals takes place [3]. The same situation is realized for germanosilicate glasses at  $F > 200 \text{ mJ/cm}^2$ . Some of laser induced Ge(n) radicals interacts with  $\text{H}_2$  molecules, creating a new stable optically unactive products.

The low-intensity UV irradiation (Hg-lamp; KrF laser,  $F = 2 \text{ mJ/cm}^2$ ) of GODC leads to formation of paramagnetic Ge(n) and H(II) centers as well. In contradistinction to the case of high-intensity laser irradiation the contribution of Ge(3) centers in total concentration of Ge(n) is increased, but the value  $[\text{Ge}(n)]/\Delta[\text{GeOGC}]$  is decreased. Thus, the one-quantum photoreactions of triplet excited GODC with  $\text{H}_2$  molecules gives quite different composition of paramagnetic products.

1. K. O. Hill, B. Malo, F. Bilodeau, D. C. Johnson, Ann. Rev. Mat. Sci. 23 (1993) 125.
2. R. Kashyap, Opt. Fiber Technology 1, 17 (1994).
3. V. A. Radzig, V. N. Bagratashvili, S. I. Tsypina, P. V. Chernov, A. O. Rybaltovskii, J. Phys. Chem. 99 (1995) 6640.
4. P.J.Lemaire, R.M. Atkins, V. Mizrahi, W.A. Reed, Electron. Lett. 29 (1993) 1191.
5. D.S. Starodubov, E.M.Dianov et al. SPIE Proc. 2998 (1997) 11.
6. Yu.S. Zavorotny, A.O. Rybaltovskii, P.V. Chernov, V.N. Bagratashvili, S.I. Tsypina, L. Dong, Glass Phys. and Chem. 23 (1997) 67.